

Contents



Gas Chromatography

Introduction

Basic principles of capillary GC	2
--	---

MN phases for GC · Summary	4
---	---

Phase selection by column parameters

Primary selection features

Phase polarity	8
Film thickness	11
Inner diameter	12
Column length	14

Secondary selection features

Temperature stability	15
Deactivation	16
Bleeding	18
Special selectivities	21

Other important GC parameters

The mobile phase in GC	23
Working conditions for GC	24
Conversion tables	25
Derivatisation in GC	26

Do's and don'ts in GC

Water on GC columns	27
Troubleshooting	28

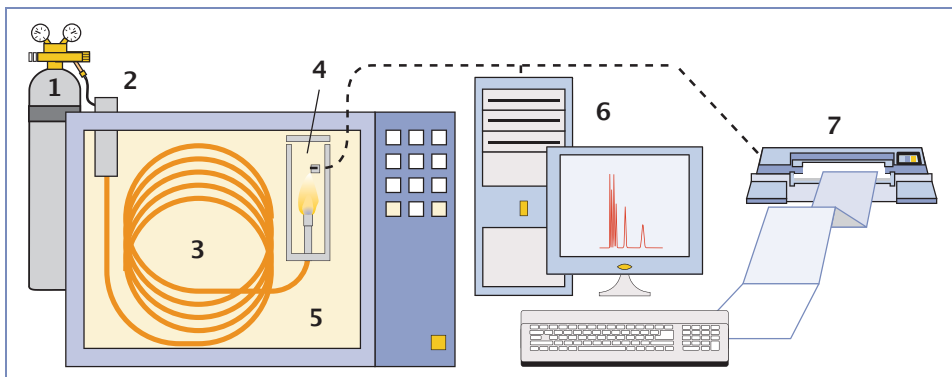
Application Gallery

Environmental pollutants	37
Solvents · Chemicals	118
Fragrances · Food and cosmetic components	152
Drugs · Pharmaceutical ingredients	180
Petrochemical products	195
Chiral separations	201

Appendices

Chromatogram index	276
Structure index (index of molecular formulas)	291

The GC system



1 – 5 compose the chromatograph:

1. **gas supplies:** carrier gas and e.g. burner gases for a flame ionisation detector (FID)
2. **sample injector:**
with **direct injection** the sample is introduced into the column without contact with other parts from glass or metal (on-column injection);
with **indirect injection** the sample is introduced into an evaporator, and the vapour is then transferred into the column either completely or partially (split technique); both techniques allow low temperatures, high temperatures or temperature programming
3. **capillary column:** the heart of the GC system
4. **detector:** indicates a substance by generation of an electrical signal (response). Some detectors are specific for certain classes of substances or for certain elements (P, N, etc.).
5. **oven** with temperature control

6 and/or 7 serve for evaluation of the separation

6. **data station** for digital evaluation of chromatograms
7. **recorder** for plotting or printing chromatograms

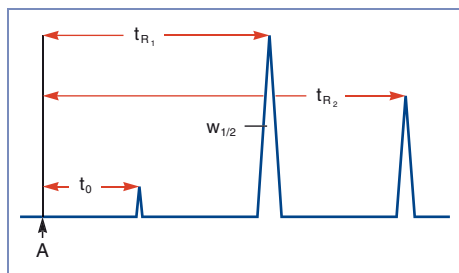
The separation process

Separation is achieved by repeated distribution of each sample component between two phases: in GC, the **mobile phase** is always a gas (mostly N_2 , H_2 , He). The **stationary phase** is a mostly viscous gum-like liquid coated to the inner wall of a capillary column (WCOT = Wall Coated Open Tubular). Transport of the analytes is achieved exclusively in the gas phase, separation is accomplished in the stationary phase. The quality of a separation (resolution) depends on how long the components to be separated stay in the stationary phase and on how often they interact with this phase. The type of interaction between component and phase (selectivity) is determined by the functional groups. The polarity of the phase is a function of stationary phase substituents.



The chromatogram

A chromatogram consists of a base line and a number of peaks. Peak areas allow quantitative determinations:



A: starting point of a chromatogram = time of injection of a dissolved solute
A component can be identified by its **retention time** (qualitative determination):

$$t_{Ri} = t_0 + t_{Ri}'$$

t_0 : dead time = void time = residence time of a solute in the mobile phase (time required by a component to migrate through the chromatographic system without any interaction with the stationary phase)

t_{Ri} : retention time = interval between peak i and the point of injection

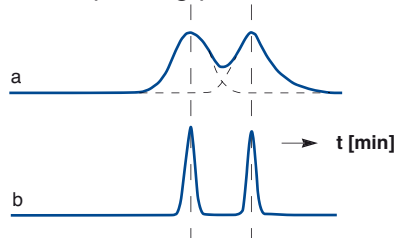
t_{Ri}' : net retention time = difference between total retention time and dead time t_0 . It indicates how long a substance stays in the stationary phase.

Other terms characterising a separation:

k' : capacity factor: a measure for the position of a sample peak in the chromatogram. The capacity factor is specific for a given compound and constant under constant conditions.

$$k'_i = \frac{t_{Ri} - t_0}{t_0}$$

The relative retention does not provide any information on the quality of a separation, since for equal values of α two very broad peaks may overlap, (as shown in trace a), or may be completely resolved (as in trace b), if they are correspondingly narrow.



R: resolution: a measure for the quality of a separation, taking the peak width at half height ($w_{1/2}$) into account according to

$$R = \frac{t_{R2} - t_{R1}}{(w_{1/2})_2 + (w_{1/2})_1}$$

N_{th} : number of theoretical plates: characterises the quality of a column (should be determined for $k' > 5$).

The height equivalent to a theoretical plate (h , HETP) is calculated by dividing the length L of the column by the number of theoretical plates N_{th} . The smaller this value the better works the column.

$$N_{th} = 5.54 \cdot \left(\frac{t_{Ri}}{w_{1/2}} \right)^2$$

$$h = \text{HETP} = \frac{L}{N_{th}}$$

MN phases for GC

Phase	Composition	Relative polarity ¹	max. Temperature ²	
OPTIMA® 1	100 % dimethylpolysiloxane		340/360 °C	
OPTIMA® 1 MS				
OPTIMA® 1 MS Accent				
OPTIMA® 5	5 % phenyl – 95 % dimethylpolysiloxane		340/360 °C	
OPTIMA® 5 MS	5 % diphenyl - 95 % dimethylpolysiloxane		340/360 °C	
OPTIMA® 5 MS Accent	silarylene phase equivalent to 5 % diphenyl – 95 % dimethylpolysiloxane		340/360 °C	
OPTIMA® δ-3	phase with autoselectivity		340/360 °C	
OPTIMA® XLB	phase with optimised silarylene content		340/360 °C	
OPTIMA® δ-6	phase with autoselectivity		340/360 °C	
OPTIMA® 1301	6 % cyanopropylphenyl – 94 % dimethylpolysiloxane		300/320 °C	
OPTIMA® 624	6 % cyanopropylphenyl – 94 % dimethylpolysiloxane		280/300 °C	
OPTIMA® 624 LB	as above, low bleed phase		280/300 °C	
¹ = nonpolar, = polar properties				
² first temperature for isothermal operation, second value for short isotherms in a temperature programme (please note, that for columns with 0.53 mm ID and for columns with thicker films temperature limits are generally lower)				

Summary



Structure	USP	Phases which provide a similar selectivity based on chemical and physical properties
$\left[\begin{array}{c} \text{CH}_3 \\ \\ \text{O} - \text{Si} - \\ \\ \text{CH}_3 \end{array} \right]_n$	G1 G2 G38	PERMABOND® SE-30, OV-1, DB-1, SE-30, HP-1, SPB-1, CP-Sil 5 CB, Rtx®-1, 007-1, BP1, MDN-1, AT™-1, ZB-1, OV-101 Ultra-1, DB-1MS, HP-1MS, Rtx®-1MS, Equity™-1, AT™-1MS, VF-1MS, CP-Sil 5 CB MS
$\left[\begin{array}{c} \text{CH}_3 \\ \\ \text{O} - \text{Si} - \\ \\ \text{C}_6\text{H}_5 \end{array} \right]_m \left[\begin{array}{c} \text{CH}_3 \\ \\ \text{O} - \text{Si} - \\ \\ \text{CH}_3 \end{array} \right]_n$	G27 G36	PERMABOND® SE-52, SE-54, SE-52, DB-5, HP-5, SPB-5, CP-Sil 8, Rtx®-5, 007-5, BP5, MDN-5, AT™-5, ZB-5
$\left[\begin{array}{c} \text{C}_6\text{H}_5 \\ \\ \text{O} - \text{Si} - \\ \\ \text{C}_6\text{H}_5 \end{array} \right]_m \left[\begin{array}{c} \text{CH}_3 \\ \\ \text{O} - \text{Si} - \\ \\ \text{CH}_3 \end{array} \right]_n$	G27 G36	DB-5MS, HP-5MS, Ultra-2, Equity™-5, CP-Sil 8CB low bleed/MS, Rtx®-5SIL-MS, Rtx®-5MS, 007-5MS, BPX5, MDN-5S, AT™-5MS, VF-5MS
$\left[\begin{array}{c} \text{C}_6\text{H}_5 \\ \\ \text{O} - \text{Si} - \\ \\ \text{C}_6\text{H}_5 \end{array} \right]_m \left[\begin{array}{c} \text{CH}_3 \\ \\ \text{O} - \text{Si} - \text{C}_6\text{H}_4 - \text{Si} - \\ \quad \quad \\ \text{CH}_3 \quad \text{H}_3\text{C} \quad \text{H}_3\text{C} \end{array} \right]_n \left[\begin{array}{c} \text{H}_3\text{C} \\ \\ \text{O} - \text{Si} - \\ \\ \text{H}_3\text{C} \end{array} \right]_o$	G27 G36	
see description on page 10	G49	no similar phases
$\left[\begin{array}{c} \text{C}_6\text{H}_5 \\ \\ \text{O} - \text{Si} - \\ \\ \text{C}_6\text{H}_5 \end{array} \right]_m \left[\begin{array}{c} \text{CH}_3 \\ \\ \text{O} - \text{Si} - \text{C}_6\text{H}_4 - \text{Si} - \\ \quad \quad \\ \text{CH}_3 \quad \text{H}_3\text{C} \quad \text{H}_3\text{C} \end{array} \right]_n \left[\begin{array}{c} \text{H}_3\text{C} \\ \\ \text{O} - \text{Si} - \\ \\ \text{H}_3\text{C} \end{array} \right]_o$	–	DB-XLB, Rtx®-XLB, MDN-12, VF-XMS
see description on page 10	–	no similar phases
$\left[\begin{array}{c} \text{C}_6\text{H}_5 \\ \\ \text{O} - \text{Si} - \\ \\ \text{NC} - (\text{CH}_2)_3 \end{array} \right]_m \left[\begin{array}{c} \text{CH}_3 \\ \\ \text{O} - \text{Si} - \\ \\ \text{CH}_3 \end{array} \right]_n$	G43	HP-1301, DB-1301, SPB-1301, Rtx®-1301, CP-1301, 007-1301 HP-624, HP-VOC, DB-624, DB-VRX, SPB-624, CP-624, Rtx®-624, Rtx®-Volatiles, 007-624, BP624, VOCOL
For special phases and phases tested for specific applications see table on page 21		

MN phases for GC

Phase	Composition	Relative polarity ¹	max. Temperature ²	
OPTIMA® 1701	14 % cyanopropylphenyl – 86 % dimethylpolysiloxane		300/320 °C	
OPTIMA® 35 MS	silarylene phase with selectivity similar to a 35 % diphenyl – 65 % dimethyl- polysiloxane phase		360/370 °C	
OPTIMA® 17	phenylmethylpolysiloxane, 50 % phenyl		320/340 °C	
OPTIMA® 210	trifluoropropylmethylpolysiloxane (50 % trifluoropropyl)		260/280 °C	
OPTIMA® 225	50 % cyanopropylmethyl – 50 % phenylmethylpolysiloxane		260/280 °C	
OPTIMA® 240	33 % cyanopropylmethyl – 67 % dimethylpolysiloxane		260/280 °C	
OPTIMA® WAX	polyethylene glycol 20 000 daltons		250/260 °C	
OPTIMA® FFAP	polyethylene glycol 2-nitro- terephthalate		250/260 °C	
¹  = nonpolar,  = polar properties ² first temperature for isothermal operation, second value for short isotherms in a temperature programme (please note, that for columns with 0.53 mm ID and for columns with thicker films temperature limits are generally lower)				

Summary



Structure	USP	Phases which provide a similar selectivity based on chemical and physical properties
$\left[\begin{array}{c} \text{Ph} \\ \\ \text{O} - \text{Si} \\ \\ \text{NC} - (\text{CH}_2)_3 \end{array} \right]_m \left[\begin{array}{c} \text{CH}_3 \\ \\ \text{O} - \text{Si} \\ \\ \text{CH}_3 \end{array} \right]_n$	G46	OV-1701, DB-1701, CP-Sil 19 CB, HP-1701, Rtx®-1701, SPB-1701, 007-1701, BP10, ZB-1701
$\left[\begin{array}{c} \text{Ph} \\ \\ \text{O} - \text{Si} \\ \\ \text{Ph} \end{array} \right]_m \left[\begin{array}{c} \text{CH}_3 \quad \text{H}_3\text{C} \\ \quad \\ \text{O} - \text{Si} - \text{C}_6\text{H}_4 - \text{Si} \\ \quad \\ \text{CH}_3 \quad \text{H}_3\text{C} \end{array} \right]_n \left[\begin{array}{c} \text{H}_3\text{C} \\ \\ \text{O} - \text{Si} \\ \\ \text{H}_3\text{C} \end{array} \right]_o$	G42	DB-35 MS, HP-35, SPB-35, Rtx-35, 007-35, BPX-35, MDN-35, AT-35 MS, ZB-35, OV-11, VF-35 MS
$\left[\begin{array}{c} \text{CH}_3 \\ \\ \text{O} - \text{Si} \\ \\ \text{Ph} \end{array} \right]_m$	G3	OV-17, DB-17, HP-50+, HP-17, SPB-50, SP-2250, Rtx®-50, CP-Sil 24 CB, 007-17, ZB-50
$\left[\begin{array}{c} \text{CH}_3 \\ \\ \text{O} - \text{Si} \\ \\ \text{F}_3\text{C} - (\text{CH}_2)_2 \end{array} \right]_n$	G6	OV-210, DB-210, Rtx®-200, 007-210
$\left[\begin{array}{c} \text{CH}_3 \quad \text{CH}_3 \\ \quad \\ \text{O} - \text{Si} - \text{O} - \text{Si} \\ \quad \\ \text{NC} - (\text{CH}_2)_3 \quad \text{Ph} \end{array} \right]_n$	G7 G19	DB-225, HP-225, OV-225, Rtx®-225, CP-Sil 43, 007-225, BP225
$\left[\begin{array}{c} \text{CH}_3 \\ \\ \text{O} - \text{Si} \\ \\ \text{NC} - (\text{CH}_2)_3 \end{array} \right]_m \left[\begin{array}{c} \text{CH}_3 \\ \\ \text{O} - \text{Si} \\ \\ \text{CH}_3 \end{array} \right]_n$	–	no similar phases
$\text{H} - \left[\begin{array}{c} \text{H} \quad \text{H} \\ \quad \\ \text{O} - \text{C} - \text{C} \\ \quad \\ \text{H} \quad \text{H} \end{array} \right]_n - \text{OH}$	G16	PERMABOND® CW 20 M, DB-Wax, Supelcowax™, HP-Wax, HP-INNOWax, Rtx®-Wax, CP-Wax 52 CB, Stabilwax, 007-CW, BP20, AT™-Wax, ZB-Wax
$\left[\begin{array}{c} \text{O} \quad \text{O} \\ \quad \\ \text{C} - \text{C}_6\text{H}_4 - \text{C} \\ \quad \\ \text{O}_3\text{N} \quad (\text{OCH}_2\text{CH}_2)_m - \text{O} \end{array} \right]_n$	G25 G35	PERMABOND® FFAP, DB-FFAP, HP-FFAP, CP-SIL 58 CB, 007-FFAP, CP-FFAP CB, Nukol
For special phases and phases tested for specific applications see table on page 21		

Phase selection in GC

Selecting the proper stationary phase is the most important decision when selecting a capillary column for a given separation task. In this chapter you will find some guidelines and concepts that simplify the process. There are four major column parameters (primary selection features) and four hardware- and problem adapted secondary selection features:

Primary selection features:

-  phase polarity
-  film thickness
-  inner diameter
-  column length

+ Secondary selection features:

-  temperature stability
-  deactivation
-  bleeding
-  special selectivities

Phase polarity

Phase selectivity is determined by the physicochemical interactions of the solute molecules with the stationary phase.

-  **General rule: only an adequate polarity of the GC phase leads to satisfying separation and peak shape**

Use similar polarities for phase and target compounds (e. g. nonpolar molecules require nonpolar polysiloxane phases in the column).

MN offers more than 40 different phases for gas chromatography, from very nonpolar to polar columns.

-  **Nonpolar** stationary phases, starting with 100% dimethylpolysiloxane, separate by volatility (i. e. boiling point) only.

Typical nonpolar phases:
OPTIMA® 1 or OPTIMA® 5

Typical analytes:
linear hydrocarbons (*n*-alkanes).

-  **Mid-polar** phases offer additional interactions, which may improve a separation. With increasing polarity, e. g. by introducing phenyl and/or cyanopropyl groups, the separation is increasingly influenced by differences in dipole moment and by charge transfer.

Typical mid-polar phase:

OPTIMA® 1701

Typical analytes:

molecules which can be polarised (e. g. aromatic compounds).

-  For **polar** components featuring medium to strong hydrogen bonding capacities, polyethylene glycol phases (WAX) are the best choice for a separation.

Typical polar phase:

OPTIMA® WAX

Typical analytes:

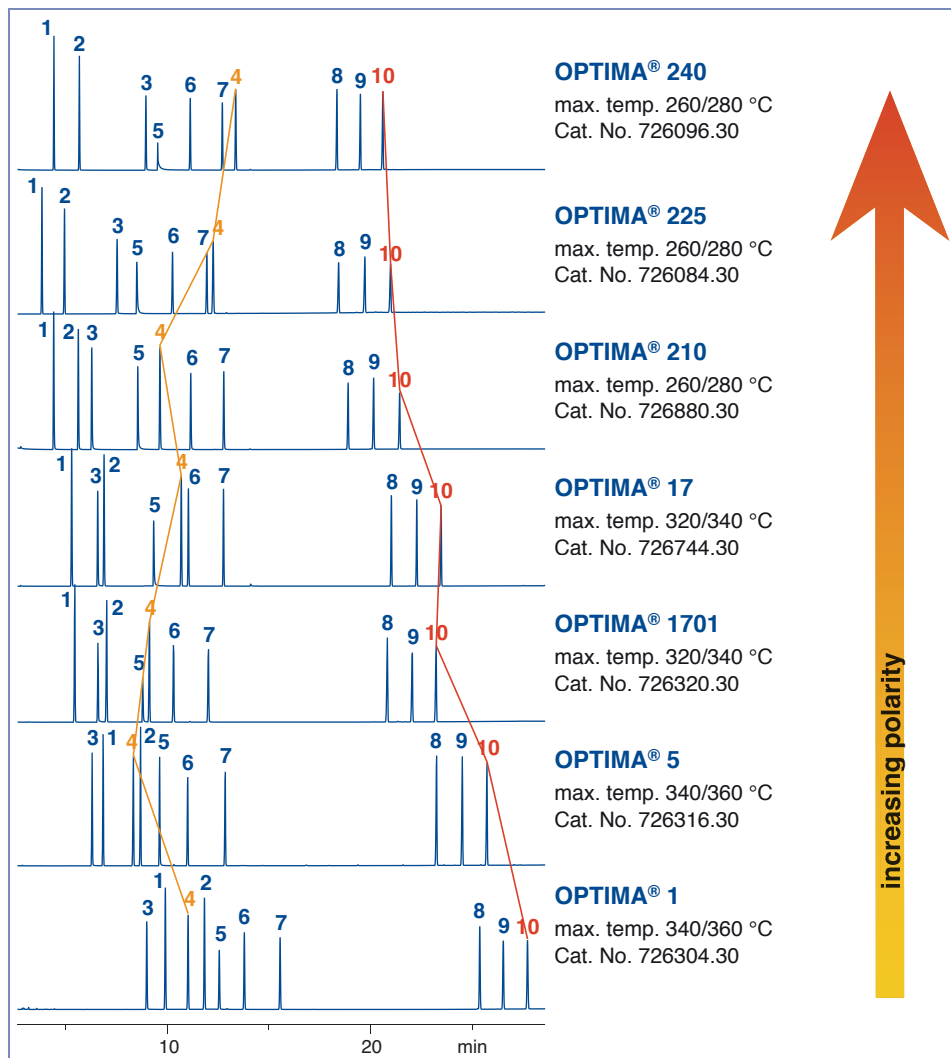
alcohols and carboxylic acids.

For structures and relative polarities of MN OPTIMA® phases please refer to the table on the preceding pages.

Phase polarity



Comparison of separation properties of selected OPTIMA® phases



All columns: 0.5 µm film, 30 m x 0.32 mm ID
Sample: MN-OPTIMA® test mixture (REF 722316)
Injection: 1.0 µl, split 1: 50
Carrier gas: 80 kPa N₂
Temperature: 80 °C → T_{max} (isothermal), 8 °C/min
Detector: FID, 260 – 300 °C

Peaks:
1. Undecane
2. Dodecane
3. Octanol
4. Dimethylaniline
5. Decylamine
6. Methyl decanoate
7. Methyl undecanoate
8. Henicosane
9. Docosane
10. Tricosane

Phase selection in GC

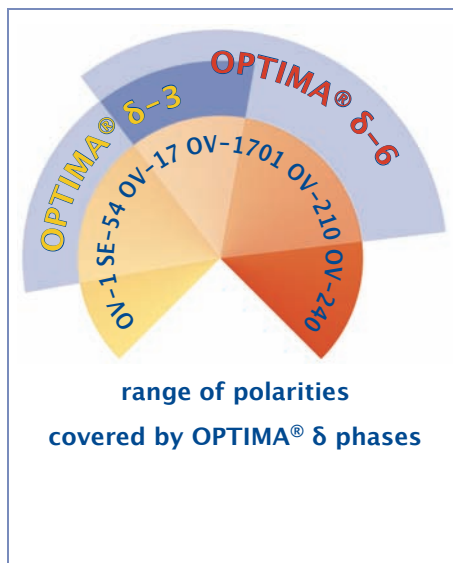
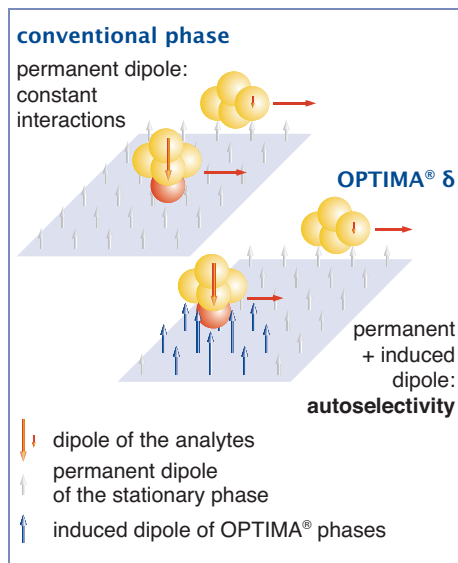
Autoselectivity · OPTIMA® δ phases

The polarity of all standard phases is defined by their composition.

In conventional mid-polar phases an increased polarity is achieved by an increase of the phenyl content in the dimethylpolysiloxane or by adding e.g. cyanopropyl or trifluoromethyl groups to the dimethylpolysiloxane often resulting in an increased tendency for bleeding.

OPTIMA® δ phases were developed to adjust themselves to the polarity of the target compounds: with non-polar compounds the phase reacts nonpolar, for polar compounds the phase increases its polarity.

The OPTIMA® δ phases consist of cross-linked polysiloxane block polymers with defined composition, exclusively produced for MN. Polar molecules are able to induce a dipole moment in the stationary phase, so that the analytes show stronger interactions. We call this phenomenon autoselectivity, because the stationary phase adjusts itself to the polarity of the analytes. Thus OPTIMA® δ phases cover broader ranges of polarity. Due to their structure, OPTIMA® δ phases show very high temperature limits of 340/360 °C and low bleed levels, making them well suited for MSD or PND.



Isomeric phenols are difficult to analyse on standard phases due to coelutions. The autoselective OPTIMA® δ -3 readily separates all of 22 isomeric phenols as shown in application 250060 on page 61.

Primary selection features



Polarity · Most frequently used phases

Selectivity has to be optimized for the critical pair of components or the main component.

You should always select the least polar column which solves your separation task.

About 70 % of all GC separations can be performed on non- to mid-polar columns.

In fact, 5-type phases (5 % phenyl – 95 % methylpolysiloxane) are the most commonly used GC phases in the world, because they generally feature high temperature stability, low column bleeding and good deactivation.

The second most common phases are WAX and/or FFAP. They represent the other end of the polarity range and are well suited for all compounds with strong hydrogen bonding capacity (e. g. compounds with OH, COOH or NH_x functionalities).

Mid-polar phases like 1301, 35, 17, 1701, 210 and 225 feature alternative polarities for special separations or for confirming analytical results.

Film thickness

Film thickness of MN capillary columns reaches from 0.1 to 5.0 μm .

Standard film thickness is 0.25 μm .

Thin films (0.1 – 0.2 μm) are very well suited for high-boiling or temperature labile compounds, fast separations, or very closely eluting substances.

Increasing film thickness will increase the capacity, the retention time for low-boiling compounds, and improve inertness. This is especially useful for samples with widely differing concentrations, or for the separation of volatile polar substances.

Better coverage of the column wall by a thicker film and a reduction of the column surface by reducing column length is favourable for extremely active substrates, which in many cases cause noticeable tailing, if they come in contact with uncoated spots of the column wall.

Thick films also mean more phase in the column, and consequently higher bleeding. This results in lower maximum operating temperatures for thick film columns. In addition, thick film columns may have a lower efficiency. Variation of the film thickness is often better than increasing the column length as may be seen in applications 200030 and 200041 page 196.

Phase selection in GC

Inner diameter

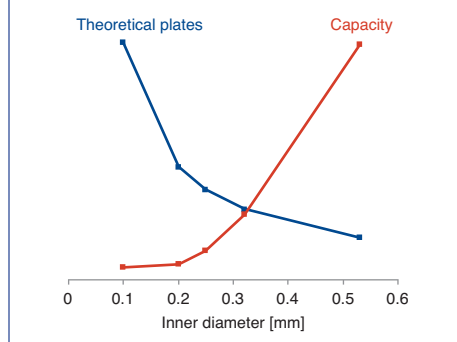
The lower the ID of a GC column, the higher is the theoretically possible number of plates per meter.

-  **0.1 – 0.2 mm ID**
for high resolution and short retention times with low carrier gas flows (Fast GC)
-  **0.25 mm ID**
all-purpose columns for analyses of complex mixtures
-  **0.32 mm ID**
for routine analyses with short retention times, but increased capacity
-  **0.53 mm ID**
for rapid separations with inert surface and highest capacity

Plate number as a function of ID

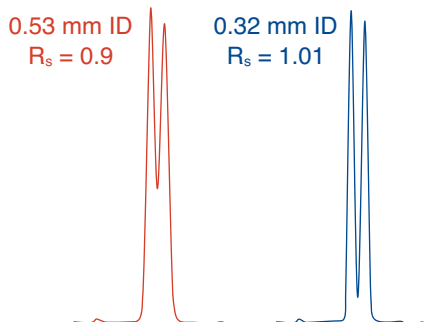
Column ID [mm]	Theoretical plates/m
0.10	12500
0.20	5940
0.25	4750
0.32	3710
0.53	2240

Plate number and capacity as function of inner diameter



Influence of inner diameter on resolution

Column: 30 m, identical film thickness



Decreasing the inner diameter and thus increasing resolution is useful for high speed/Fast GC procedures, but some side effects have to be considered:

-  retention time increases as well (at identical head pressure)
-  the back pressure of a small ID column increases due to restriction
-  the capacity of columns decreases (the loadability decreases, see table on previous page)
-  the hardware requirements for injection systems (high pressure, high speed of injection) and detectors (highest speed of detection) increase

Because of this, GC procedures can not always take profit from the increase in plate numbers!

Primary column parameters



Sample capacity (loadability) of a 30 m column in ng

Film thickness [μm]	Inner diameter [mm]			
	0.2	0.25	0.32	0.53
0.10	20 – 35	25 – 50	35 – 75	50 – 100
0.25	35 – 75	50 – 100	75 – 125	100 – 250
0.50	75 – 150	100 – 200	125 – 250	250 – 500
1.00	150 – 250	200 – 300	250 – 500	500 – 1000
3.00		400 – 600	500 – 800	1000 – 2000
5.00		1000 – 1500	1200 – 2000	2000 – 3000

Phase ratio of capillary columns

Column length (L) and retention times (for identical flow and pressure parameters) show a linear relation; resolution is proportional to the square root of L. If other column parameters (ID, film thickness) have to be adapted – for example in case of a new GC with different dimension requirements – a comparable separation can only be achieved with columns with similar phase ratio!

In general a higher phase ratio implies less retention power and shorter retention times and vice versa.

Calculation of the phase ratio:

$$\beta = \frac{r}{2 d_f}$$

with r = column radius and d_f = film thickness

Phase ratio as a function of inner diameter and film thickness

↓ ID \ d _f →	0.1	0.2	0.25	0.35	0.5	1	1.5	2	3	5
0.1	250	125.0	100	71.43	50	25.0	16.67	12.50	8.3	5.0
0.2	500	250.0	200	142.86	100	50.0	33.33	25.00	16.7	10.0
0.25	625	312.5	250	178.57	125	62.5	41.67	31.25	20.8	12.5
0.32	800	400.0	320	228.57	160	80.0	53.33	40.00	26.7	16.0
0.53	1325	662.5	530	378.57	265	132.5	88.33	66.25	44.2	26.5

Phase selection in GC

Column length

Column length is directly proportional to the separation efficiency (number of plates N).

Longer columns show increased resolution, however, R is proportional to the square root of N , i.e. a doubled column length only increases the resolution by a factor of 1.4, while cost and analysis time are proportional to the length.

Routine separations are most frequently performed on 25 or 30 m columns, while complex mixtures may require 50 or 60 m columns. 10 m columns with 0.10 mm ID are used for Fast GC.

As a rule of thumb:
use about 1 m column length per component in your mixture.

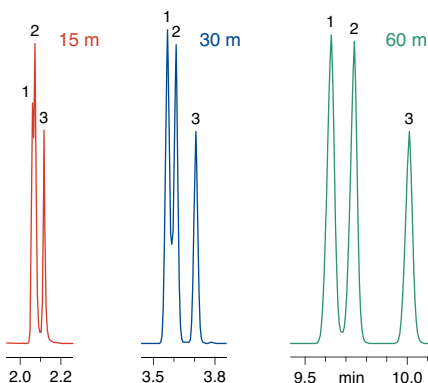
Influence of column length on resolution and retention time

Columns: OPTIMA® 1, 0.25 mm ID, 0.25 μ m film

Carrier gas: 1 ml/min He, isothermal

Peaks: FAMES

1. C18:2, 2. C18:1, 3. C18



Fast GC

Characteristics of Fast GC are decreased column diameters, high heating rates and decreased column length for faster GC separations with high resolution efficiency.

- ⊞ small inner diameters combined with fast temperature programs can reduce the analysis time by up to 80%
- ⊞ high heating rates place special demands on stationary phases: OPTIMA® columns meet exactly this requirement, as they show very low bleeding and provide long lifetimes, even when continuously subjected to high heating rates
- ⊞ small inner diameters result in high column inlet pressures and a lower volume flow of the mobile phase, which as a consequence require very fast injection of very small samples against a high pressure
- ⊞ the amount of sample, which can be injected, is limited by the inner diameter and the thin film
- ⊞ high sensitivity detectors with small volume and extremely short response time, as well as a very rapid data acquisition and processing are required

For comparison of a separation on a 50 m standard capillary with a separation on a 10 m fast GC column see application 211260 on page 151.

Secondary column parameters

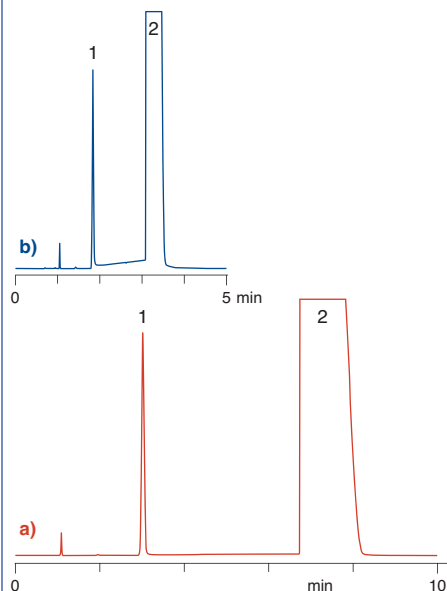


Temperature stability

High temperature stability of a column is beneficial for high-boiling solutes with very low vapour pressures, which normally have very long retention times and rather broad peak shapes. Since OPTIMA® columns can be operated at increased temperatures, they elute high-boiling compounds faster and with better peak shapes.

Improved peak shape at higher temperatures

Injection: 0.6 µl, split 1:100
Carrier gas: 0.4 bar He
a) Temperature: 100 °C isothermal
width of peak 2: 1.4 min
b) Temperature: 130 °C isothermal
width of peak 2: 0.5 min



Maximum operating temperatures for OPTIMA® phases

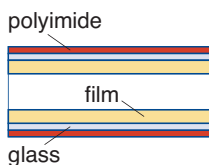
OPTIMA® phase	max. Temp. [°C]	
1	340/360	
1 MS	340/360	
1 MS Accent	340/360	
5	340/360	
5 MS	340/360	
5 MS Accent	340/360	
δ-3	340/360	
XLB	340/360	
δ-6	340/360	
1301	300/320	
624 / 624 LB	280/300	
1701	300/320	
35 MS	360/370	
17	320/340	
210	260/280	
225	260/280	
240	260/280	* first temperature for isothermal operation, second value for short isotherms in a temperature programme (please note, that for columns with 0.53 mm ID and for columns with thicker films temperature limits are generally lower)
WAX	250/260	
FFAP	250/260	

Phase selection in GC

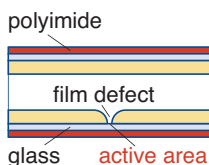
Deactivation

The quality of deactivation in GC columns is influenced by

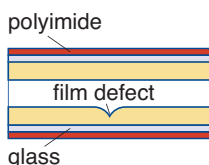
- 🌀 the deactivation of the glass tube itself
- 🌀 the binding process of the poly-siloxane or polyethylene glycol to the glass wall
- 🌀 the composition and thickness of the film
- 🌀 the constancy / quality of the coverage of the column wall



optimal column with even film and perfectly covered walls



column with film defect causing an active area, where analytes can interact with the glass wall, resulting in tailing



thicker films help to avoid active areas, even when the film is not perfectly smooth resulting in less or no tailing

The possibility of film defects increases with increasing column length and decreasing film thickness. Thus especially for polar compounds thin film columns may cause problems like tailing or deformed peaks due to uncontrolled interactions with active spots on the column wall.

Quality control of OPTIMA® columns

The deactivation process of OPTIMA® columns is optimized for excellent chromatographic performance, improved peak shape, efficiency and sensitivity.

Performance of each GC column from MN is tested by stringent quality control procedures including

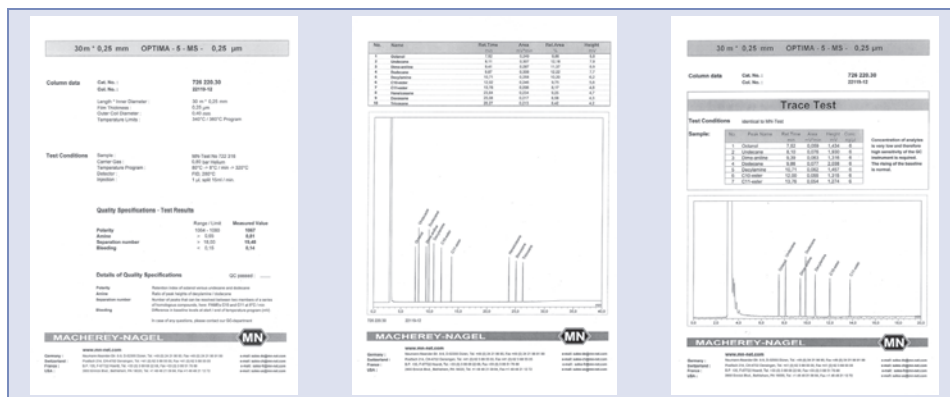
- 🌀 **column efficiency** by measuring the separation number (number of resolved peaks between two members of a series of homologue compounds, in this case FAMES C11 and C12) in a temperature programme (8 °C/min)
- 🌀 **polarity** by measuring the retention index of octanol versus decane and undecane
- 🌀 **bleeding** as difference in baseline values at start and end of the temperature programme
- 🌀 **inertness** by measuring the peak height ratio for decylamine/undecane (for non- to mid-polar phases)

All MN columns are supplied with test chromatograms of standard test mixtures stating the actual performance of the column (MS columns additionally include the chromatogram of a trace test diluted 1:100). Additionally, we include the corresponding test mixture.

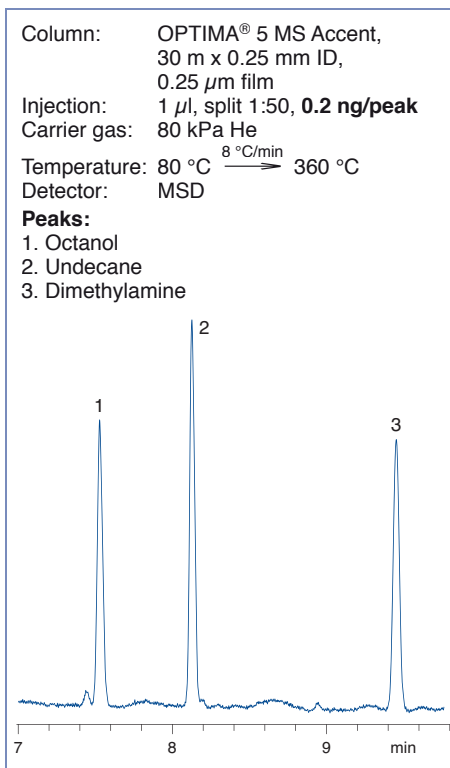
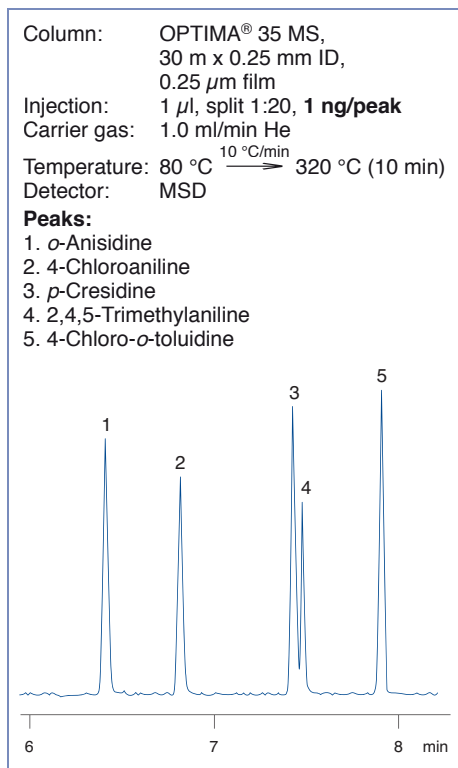
Secondary column parameters



Typical test chromatograms of an OPTIMA® column



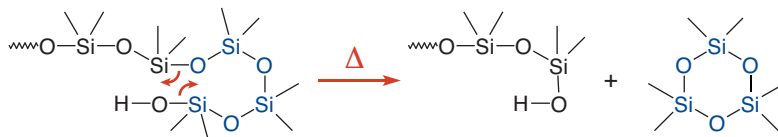
Examples of OPTIMA® column performance and deactivation



Phase selection in GC

Column bleeding

Degradation of a methylpolysiloxane chain forming hexamethylcyclotrisiloxane



Column bleeding is a thermal degradation of the siloxane polymer caused by splitting of siloxane bonds with lower bond energies (see formula above).

Bleeding is accelerated near the upper temperature limit of the stationary phase (normally above 180 °C). In the presence of oxygen or water bleeding even starts at much lower temperatures. This is why the purity of the carrier gas is very important, especially when working at high temperatures.

Column bleed increases as column length increases, because longer columns contain more of the stationary phase. However, don't let this fact discourage you to use a longer column if necessary.

Column bleed also increases as film thickness increases. Since thicker films are more retentive, later eluting peaks may shift into a region of higher column bleed when increasing film thickness as is shown in the chromatogram below.

Column bleeding at elevated temperature

Column: OPTIMA® 5 Amine, 30 x 0.32 mm ID, 1.5 µm film, REF 726356.30, max. temperature 300/320 °C

Injection: 1 µl, split 1:35

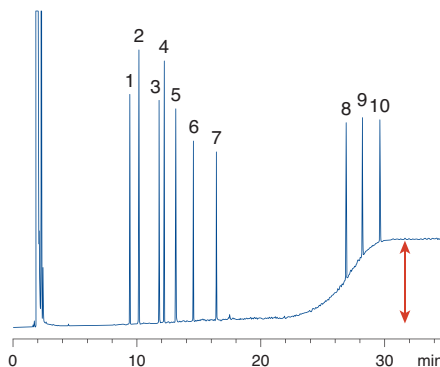
Carrier gas: 1.5 bar He

Temperature: 80 °C $\xrightarrow{8\text{ °C/min}}$ 320 °C (10 min)

Detector: FID

Peaks:

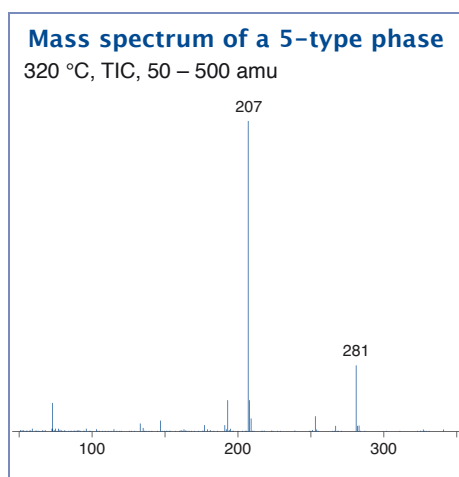
1. Octanol
2. Undecane
3. Dimethylaniline
4. Dodecane
5. Decylamine
6. Methyl decanoate
7. Methyl undecanoate
8. Henicosane
9. Docosane
10. Tricosane



Secondary column parameters

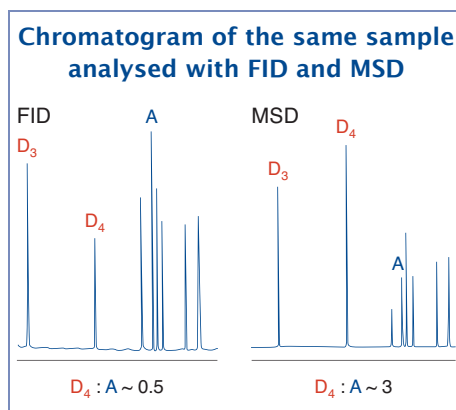


As is shown in the mass spectrum below, the major “bleed” ion from a 5-type phase is $m/z = 207$ resulting from hexamethylcyclotrisiloxane (D_3), the second important ion $m/z = 281$ results from the cyclic degradation product with 4 units (D_4).



Influence of the type of detector

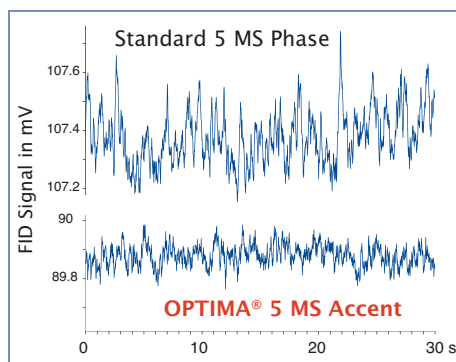
The impact of bleeding on the signal to noise ratio of a chromatogram is much more pronounced for ion trap MS detectors than for FID detectors as is shown in the graphs below. With FID detection, the ratio of D_4 to the analyte peak A is about 0.5; with MS detection it is about 3! This is why columns with low bleed characteristics are especially important for use with MS detection.



Benefits of low bleed columns in GC

Low bleed columns allow

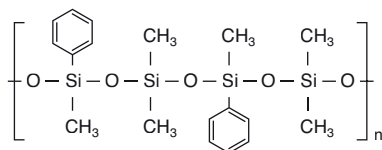
- higher operating temperatures resulting in shorter run times
- less pollution of the detection system
- improved detectability of solutes in GC/MS analyses
- increased detectability due to an improved signal to noise ratio as is shown in the traces at right



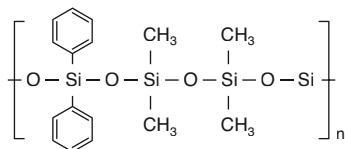
Phase selection in GC

Low-bleed GC/MS columns

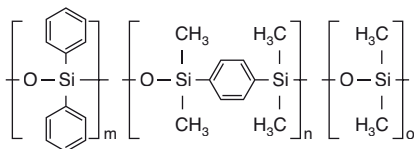
Standard columns with 5-type phases contain phenyl and methyl substituted siloxanes in the ratio 5:95 as shown in the structure below



The first generation of low-bleed GC/MS columns contains diphenyl siloxane groups which improve bleed characteristics due to steric hindrance of the formation of D₃ and D₄.



Silarylene-type columns (see figure below) are stabilised by insertion of arylenes groups in the siloxane chain, offering lowest bleeding and high inertness even at higher temperatures.

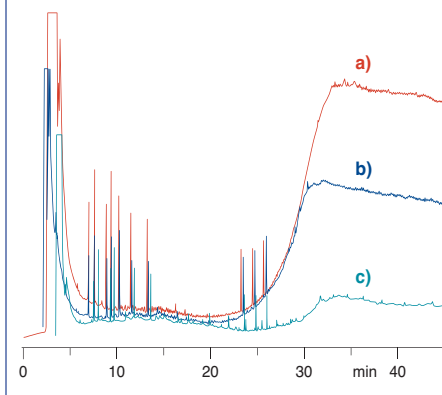


Comparison of bleeding characteristics of 5-type columns

Columns: 30 m x 0.25 mm ID,
0.25 μm film
Injection: 1 μl , split 1:50
Carrier gas: 80 kPa He
Temperature: 80 $^{\circ}\text{C}$ $\xrightarrow{8^{\circ}\text{C}/\text{min}}$ 360 $^{\circ}\text{C}$
Detector: FID

Chromatograms:

- a) conventional 5-type phase
- b) conventional 5-MS phase
- c) OPTIMA® 5 MS Accent





Achiral columns for special separations

Certain analytical separations can be performed more easily with chromatographic columns, which have been especially developed or tested for the respective task. The following table summarises our programme of GC speciality capillaries. Page numbers refer to the cited applications.

Phase composition	Recommended application	Typical application	Page
OPTIMA® 5 Amine			
5 % phenyl – 95 % methyl– polysiloxane especially deactivated	polyfunctional amines such as ethanolamines, amino– functionalised diols	250050 210280	138 140
FS–CW 20 M–AM			
polyethylene glycol, basic, non–immobilised	amines	201520 201530	139 139
PERMABOND® P–100			
dimethylpolysiloxane	petrochemical products	200071	199
PERMABOND® SE–54 HKW			
1 % vinyl – 5 % phenyl – 94 % methylpolysiloxane	volatile halogenated hydrocarbons	212480 200150	40 46
OPTIMA® 1–TG			
dimethylpolysiloxane	triglycerides according to carbon number	200050	197
OPTIMA® 17–TG			
phenyl–methyl–polysiloxane (50 % phenyl)	triglycerides according to degree of unsaturation	201800	176
PERMABOND® Silane			
	monomeric silanes and chlorosilanes	200090	148
PERMABOND® CW 20 M–DEG			
polyethylene glycol, tested for diethylene glycol	diethylene glycol in wine	201500	160

Phase selection in GC

Chiral columns for enantiomer separation

For chiral columns it is not possible to make a general prediction, which phase could solve a given separation task. Even for compounds with small structural differences the enantio-differentiation can be quite different. For numerous chiral separations – arranged by increasing molecular size – refer to pages 201 – 275. Page numbers in the table below refer to the cited typical applications.

Phase	Composition	max. Temp. [° C]	Typical application	Page
LIPODEX® cyclodextrin phases				
LIPODEX® is patented under EP 0407 412 and US Re. 36.092				
LIPODEX® A	hexakis-(2,3,6-tri-O-pentyl)-α-cyclodextrin	200/220	202851	212
LIPODEX® B	hexakis-(2,6-di-O-pentyl-3-O-acetyl)-α-cyclodextrin	200/220	202861	205
LIPODEX® C	heptakis-(2,3,6-tri-O-pentyl)-β-cyclodextrin	200/220	201880	210
LIPODEX® D	heptakis-(2,6-di-O-pentyl-3-O-acetyl)-β-cyclodextrin	200/220	202871	227
LIPODEX® E	octakis-(2,6-di-O-pentyl-3-O-buteryl)-γ-cyclodextrin	200/220	212761	202
LIPODEX® G	octakis-(2,3-di-O-pentyl-6-O-methyl)-γ-cyclodextrin	220/240	250410	259
HYDRODEX cyclodextrin phases, diluted with optimized polysiloxanes				
HYDRODEX β-PM	heptakis-(2,3,6-tri-O-methyl)-β-cyclodextrin	230/250	202030	272
HYDRODEX β-3P	heptakis-(2,6-di-O-methyl-3-O-pentyl)-β-cyclodextrin	230/250	201920	266
HYDRODEX β-6TBDM	heptakis-(2,3-di-O-methyl-6-O- <i>t</i> -butyldimethyl-silyl)-β-cyclodextrin	230/250	250170	274
HYDRODEX β-TBDAC	heptakis-(2,3-di-O-acetyl-6-O- <i>t</i> -butyldimethyl-silyl)-β-cyclodextrin	220/240	212430	265
HYDRODEX γ-TBDAC	octakis-(2,3-di-O-acetyl-6-O- <i>t</i> -butyldimethyl-silyl)-γ-cyclodextrin	220/240	212980	262

The mobile phase in GC



Sometimes it may be difficult to apply the data of a given GC application in reality, because every manufacturer of GC equipment has its own standards and specifications (depending on the country). The physical parameters are fixed, but an adaption has to be made.

Sometimes an older application uses different or out-dated parameters which are no longer in use. The tables and information in this chapter were compiled to separate the “jungle” in order to give practicable professional hints.

Selection of the optimum carrier gas

For optimum column performance, hydrogen carrier gas offers some strong advantages over helium or nitrogen. Hydrogen yields higher plate numbers (= better resolution) at rapid linear velocities and achieves higher velocities at lower pressures. At the same time, the presence of extra hydrogen is advantageous for flame ionization and other detectors that use hydrogen fuel gas. Nevertheless, in practical work H_2 is not used so often, because it is the gas with the highest price and the highest danger potential!

Helium is not so dangerous, cheaper, with a similar separation efficiency and for this reason the most recommended gas for modern routine analysis. For sure nitrogen has the lowest price, but many separation problems can be avoided or solved directly by switching from N_2 to He or H_2 . The following tables show, that H_2 has the highest gas velocity and the lowest viscosity of all commonly used gases for GC (H_2 , He, N_2).

We recommend: if a high resolution (e.g. for chiral separations) or a short analysis time is required, use hydrogen!

Physical parameters of GC gases

Gas	Viscosity *		Linear velocity at 25 °C					
	300 K	500 K	g/mol	m/s	km/h	Factor N_2	Faktor He	Faktor H_2
H_2	9.0	12.7	2.016	1920	6920	3.8	1.4	1
He	20.0	28.4	4.003	1370	4930	2.7	1	0.71
N_2	17.9		28.014	510	1840	1	0.37	0.26
O_2			32	480	1720			

* from Handbook of Chemistry and Physics, 72nd Edition

Working conditions in GC

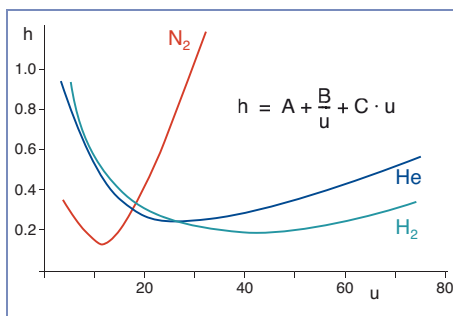
Optimized working conditions for GC

The table below shows the optimum working conditions for standard columns with common lengths and inner diameters. If you adjust your GC parameters to these values, you can be sure to work in the optimum chromatographic region according to the van Deemter curves. If your parameters are outside these values, you will lose separation efficiency and the results may no longer be ideal (poor separation, analysis time too long, dead time too short).

The short dead times mirror the high speed of H₂ compared to the other gases. If the same inlet pressures are applied for hydrogen as for helium, peaks are eluted in less time, because the linear velocity of H₂ is faster than that of helium. In general, when switching from helium to hydrogen retention times will be roughly cut in half if the inlet pressure is unchanged.

Plate height and gas velocity

The Van Deemter equation shows how the plate height h depends on the flow velocity u for 3 different GC gases:



A Eddy diffusion; for WCOT capillary columns $A = 0$

B molecular axial diffusion; B is a function of the diffusion coefficient of the component in the respective carrier gas

C resistance to mass transfer

In practice often higher velocities than u_{opt} are chosen, if separation efficiency is sufficient, since higher carrier velocities mean shorter retention times.

Gas	Column length [m]	Column ID [mm]	Average velocity [cm/s]	Outlet flow [ml/min]	Head pressure [kPa]	Head pressure [psi]	Head pressure [bar]	Dead time [min]
Nitrogen	10	0.10	8 – 16	0.04 – 0.1	55 – 112	8 – 16	0.5 – 1.1	2.1 – 1.0
	25	0.20	8 – 16	0.14 – 0.33	34 – 69	5 – 10	0.3 – 0.69	5.2 – 2.6
	30	0.25	8 – 16	0.21 – 0.48	26 – 53	3.8 – 7.6	0.26 – 0.53	6.3 – 3.1
N ₂	30	0.32	8 – 16	0.33 – 0.72	16 – 32	2.3 – 4.6	0.16 – 0.32	6.3 – 3.1
	30	0.53	8 – 16	0.88 – 1.79	5.8 – 11.7	0.8 – 1.7	0.06 – 0.11	6.3 – 3.1
Helium	10	0.10	20 – 40	0.14 – 0.45	156 – 330	22 – 48	1.6 – 3.3	0.83 – 0.42
	25	0.20	20 – 40	0.46 – 1.29	95 – 200	14 – 29	0.95 – 1.99	2.1 – 1.0
	30	0.25	20 – 40	0.65 – 1.74	72 – 150	10 – 22	0.7 – 1.5	2.5 – 1.25
He	30	0.32	20 – 40	0.95 – 2.29	43 – 89	6.3 – 12.9	0.43 – 0.89	2.5 – 1.25
	30	0.53	20 – 40	2.3 – 4.92	16 – 32	2.3 – 4.6	0.16 – 0.32	2.5 – 1.25
Hydrogen	10	0.10	30 – 55	0.18 – 0.47	113 – 215	16 – 31	1.1 – 2.1	0.55 – 0.3
	25	0.20	30 – 55	0.62 – 1.43	69 – 130	10 – 19	0.69 – 1.3	1.4 – 0.7
	30	0.25	30 – 55	0.9 – 2.0	53 – 99	7.6 – 14	0.5 – 0.9	1.7 – 0.9
H ₂	30	0.32	30 – 55	1.35 – 2.8	32 – 59	4.6 – 8.5	0.32 – 0.59	1.7 – 0.9
	30	0.53	30 – 55	3.35 – 6.5	11.5 – 21.4	1.7 – 3.1	0.12 – 0.21	1.7 – 0.9

Oven temperature 100 °C, split 10:1, injector temperature 175 °C

Conversion tables



Pressure conversion chart

Multiply value for unit in the left column by the factor given in the table below the desired unit

↓ Unit →	kPa	bar	atm	psi	kg cm ⁻²	Torr
kPa	1	0.0100	0.00987	0.1450	0.0102	7.52
bar	100	1	0.9869	14.5038	1.0197	751.88
atm	101.32	1.0133	1	14.696	1.0332	760
psi	6.8948	0.06895	0.068	1	0.0703	51.713
kg cm ⁻²	98.06	0.9806	0.9678	14.223	1	735.56
Torr	0.1330	0.00133	0.00132	0.0193	0.00136	1

Examples: to convert 250 bar to psi, multiply 250 by 14.5038 = 3626 psi
to convert 3000 psi to kPa, multiply 3000 by 6.8948 = 20684.4 kPa

Conversion of concentrations and amounts of sample per injection

The table shows common concentrations and injection volumes for GC. The upper half is the standard region of detectable concentrations (TCD and FID detectors) and the lower half is the region of ultra trace analysis (MS, Ion Trap, Quadrupol MS, ECD)

[%]	Concentration				Injection 1 µl	
	[g/g] [g/ml]	[g/kg] [g/l]	[ppm]	[ppb]	splitless [ng]	split 1: 10 [ng]
100	1	1 000	1 000 000	1 000 000 000	1 000 000	100 000
10	0.1	100	100 000	100 000 000	100 000	10 000
1	0.01	10	10 000	10 000 000	10 000	1 000
0.1	0.001	1	1 000	1 000 000	1 000	100
0.01	0.000 1	0.1	100	100 000	100	10
0.001	0.000 01	0.01	10	10 000	10	1
	[µg/g] [µg/ml]	[mg/kg] [mg/l]				
0.000 1	1	1	1	1 000	1	0.1
0.000 01	0.1	0.1	0.1	100	0.1	0.01
0.000 001	0.01	0.01	0.01	10	0.01	0.001
0.000 000 1	0.001	0.001	0.001	1	0.001	0.000 1
0.000 000 01	0.000 1	0.000 1	0.000 1	0.1	0.000 1	0.000 01
0.000 000 001	0.000 01	0.000 01	0.000 01	0.01	0.000 01	0.000 001
0.000 000 000 1	0.000 001	0.000 001	0.000 001	0.001	0.000 001	0.000 0001

Derivatisation in GC

Derivatisation

The purpose of derivatisation

In gas chromatography it is often advantageous to derivatise polar functional groups (mainly active hydrogen atoms) with suitable reagents. Prerequisite for successful derivatisation is quantitative, rapid and reproducible formation of only one derivative. Aim of this reaction is an improved volatility, better thermal stability or a lower limit of detection due to improved peak symmetry.

The halogen atoms introduced by derivatisation (e. g. trifluoroacetates) allow specific detection (ECD) with the advantage of high sensitivity. Elution orders and fragmentation patterns in mass spectroscopy can be influenced by a specific derivatisation. The derivatisation examples in the table are meant as a first orientation and have to be adjusted or optimized for special problems.

Selection guide for derivatisation of important functional groups in GC

Function	method	derivative	recommended reagents
Alcohols, Phenols	silylation	$R'O - TMS$	BSA, MSTFA, MSHFBA, TSIM, SILYL-2110, SILYL-21, SILYL-1139
$R'OH$	acylation	$R'O - CO - R$	TFAA, HFBA, MBTFA, MBHFBA
	alkylation	$R'O - R$	TMSH
ster. hindered	silylation	$R'O - TMS$	TSIM, BSTFA, SILYL-991
Amines	silylation	$R' - NR'' - TMS$	BSA, MSTFA, MSHFBA, SILYL-991
prim., sec.	acylation	$R' - NR'' - CO - R$	TFAA, HFBA, MBTFA, MBHFBA
hydrochlorides	silylation	$R' - NR'' - TMS$	MSTFA
Amides	silylation	not stable	
	acylation	$R' - CO - NH - CO - R$	TFAA, MBTFA, HFBA, MBHFBA
Amino acids	silylation	$R' - \begin{array}{c} \diagup CO - O - TMS \\ CH \\ \diagdown NH - TMS \end{array}$	BSA, BSTFA, MSTFA, MSHFBA
	alkyl. (a) + acyl. (b)	$R' - \begin{array}{c} \diagup CO - O - R \\ CH \\ \diagdown NH - CO - R \end{array}$	a) MeOH/TMCS, TMSH b) TFAA, HFBA, MBTFA, MBHFBA
Carboxylic acids	silylation	$R' - CO - O - TMS$	BSA, MSTFA, MSHFBA, TMCS, TSIM, SILYL-2110, SILYL-21, Silyl-1139
(fatty acids)	alkylation	$R' - CO - O - R$	DMF-DMA, MeOH/TMCS (1 M), TMSH
salts	silylation	$R' - CO - O - TMS$	TMCS
		susceptible to hydrolysis	
Carbo-hydrates	silylation		MSTFA, TSIM, HMDS, SILYL-1139
	acylation		TFAA, MBTFA
Steroids	silylation		BSA, TSIM
	acylation		TFAA, MBTFA, HFBA, MBHFBA



Water on GC columns

Water causes trouble right from the beginning because of its high expansion volume. This overload effect of the injector is called “backflash” and leads to poor separations and bad peak shapes. Reproducibility is not guaranteed for repeated injections. The table shows the different gas volumes of common GC solvents. Here you find the reason, why nonpolar solvents are always preferred in GC. Even very polar analytes should be injected in nonprotic solvents (if possible).

Volume expansion of solvents

Solvent	Gas volume [μ l]
n-Hexane	140
Ethyl acetate	185
Acetone	245
Dichloromethane	285
Acetonitrile	350
Methanol	450
Water	1010

Injection 1 μ l, injector temperature 250 °C, pressure 20 psi

Attention: a water sample can contain a (maybe unknown) concentration of salts. This salt load can damage each column immediately.

If water cannot be avoided, some points have to be considered:

- ☞ In order to prevent condensation on the column, the start temperature in the oven should be at least 100 °C or higher for a better peak shape.

- ☞ Water can extinguish the FID flame or decrease the sensitivity of an ECD.
- ☞ Because nonpolar GC phases cannot be wetted well by water, a compound with good solubility in water can show poor peak shapes or double peaks.

GC phases which can be used with water samples – without risk of a direct damage – are phases without oxidizable groups, e.g. OPTIMA® 1, 5, 35, 17, Wax, FFAP types and OPTIMA® δ -3 and δ -6.

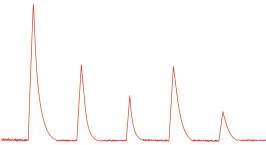
More critical are 1301, 1701, 225 and 240 types, because the cyano groups can be oxidised or destroyed catalytically (by acids, bases or metals). The phase has still its correct film thickness, but it is chromatographically “dead”!

All Cyclodextrin columns are directly destroyed by water, so the samples must be dried before injection. We recommend the use of drying cartridges like CHROMAFIX® Dry (REF 731852).

Rule: the higher film thickness is, the more stable the column is towards water.

Water itself is not so critical, but the (possible) ingredients can be (salts, catalysts, acids, bases). When columns have been run with aqueous samples, they should be heated at least at 150 °C in order to remove the water film completely from the column. Lower temperatures have no effect resulting in poor reproducibility.

Do's and don'ts in GC

Observation / Possible causes	Suggested remedy
No peaks, no gas flow	
detector has no power	check detector power supply and cables
no FID flame	check FID; reignite it
syringe defective / clogged	use a different syringe or clean it
temperature too low for the analytes	check temperature programme, oven temperature (external thermometer)
detector / software / computer hardware failure	check integrator, cables; restart computer
no gas flow	check gas tubes, valves, seals; test gas flow; shorten front of GC column, change injection septum
column connection leaks	use new ferrules
broken GC column	if breakage is at the beginning or at the end, remove the short piece; breakage in the middle can be mended with a glass connector; for multiple breakages replace column
Tailing	
	
sample vaporises too slowly, not evenly or condenses	increase injector temperature (consider max. temperature limits of the column)
high-boiling analytes	derivatise polar, basic or high boiling compounds
system leaks	check column installation; search for leaks; replace ferrules
analytes coelute	change temperature programme or use column with different selectivity
sample decomposes	check temperature programme, oven temperature (external thermometer); if analytes are not temperature-stable, reduce injector temperature; replace liner by a deactivated one
column absorbs or decomposes analytes	check capillary ends; check intact deactivation using the test mixture; for poor results shorten both column ends by about 10 cm; or replace column; if column test does not show any defects: - use a column with thicker film - use phase with better deactivation - use column with special selectivity
split rate too low	increase split rate
analytes always tending to tail	no chance for symmetric peaks

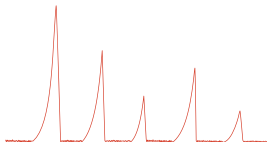
GC troubleshooting



Observation / Possible causes

Suggested remedy

Fronting



column overload

decrease injection volume; dilute sample

sample vaporises too slowly, not evenly or condenses

increase injector temperature (consider max. temperature limits of the column)

analytes coelute

change temperature programme or use column with different selectivity

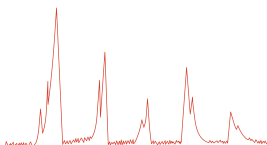
sample decomposes

check temperature programme, oven temperature (external thermometer); if analytes are not temperature-stable, reduce injector temperature; replace liner

column absorbs or decomposes analytes

check capillary ends; check intact deactivation using the test mixture; for poor results shorten both column ends by about 10 cm; or replace column; if column test does not show any defects:
a) use a column with thicker film
b) use phase with better deactivation
c) use column with special selectivity

Small peaks on fronting or tailing of bigger peaks



column not properly installed

check capillary ends; check tight and correct fit in injector and detector

temperature of injection too low

check injector temperature; if analytes are stable, increase temperature

solvent not compatible with GC phase

change solvent

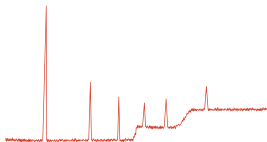
splitter defect

measure flow and adjust splitter

poorly deactivated column, film thickness too low

check capillary ends; check intact deactivation using the test mixture; for poor results shorten both column ends by about 10 cm; or replace column

Plateaus at certain temperatures



steps in temperature programme too drastic

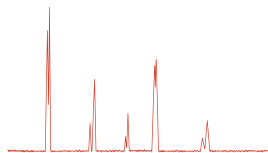
avoid very short and strong heating periods

Do's and don'ts in GC

Observation / Possible causes

Suggested remedy

Double peaks, doubled peak tops



solvent and column not compatible

change solvent or use guard column

solvent mixtures with large differences in boiling point and polarity

use just one solvent

sample decomposes

check temperature programme, oven temperature (external thermometer); if analytes are not temperature-stable, reduce injector temperature; replace liner by a deactivated one

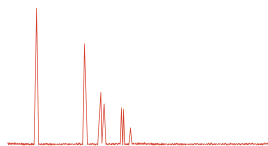
analytes coelute

modify temperature programme or use longer column; possibly change column polarity

detector overload

inject less; control make-up flow

Missing or overlapping peaks, poor separation efficiency



syringe defective / clogged

use a different syringe or clean it

sample too diluted

increase injection volume; concentrate sample

sample concentration too high

decrease injection volume; dilute sample

column connection leaks, column not properly installed

check column installation; search for leaks; replace ferrules

perforated injection septum

replace septum

injector temperature too low

check temperature programme; increase injector temperature

sample decomposes in the injector

check temperature programme; reduce injector temperature; replace liner; check capillary ends

column oven too hot

check temperature programme, oven temperature (external thermometer); decrease temperature

wrong flow rate

measure flow and correct it if necessary

column absorbs or decomposes analytes

check capillary ends; check intact deactivation using the test mixture; for poor results shorten both column ends by about 10 cm; or replace column; if column test does not show any defects:

- use a column with thicker film
- use phase with better deactivation
- use column with special selectivity

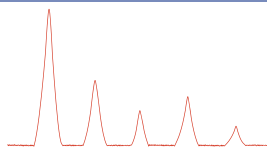
GC troubleshooting



Observation / Possible causes

Suggested remedy

Broad peaks



poor focussing

decrease start temperature of the programme

flow too high or too low

measure flow, control and adjust it if necessary

split rate too low

increase split rate

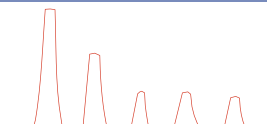
column overloaded

decrease injection volume, dilute sample or increase split flow

no solvent focussing effect

decrease oven temperature or use solvent with higher boiling point

Cut tops of peaks, broad peaks



detector overload

decrease injection volume; dilute sample; increase the split flow

column overload

decrease injection volume; increase split flow

zero point is outside the display

change scale

Strong noise, waves



leak at column entrance or injection septum

check column installation; search for leaks; replace ferrules

bleeding of septum / injector contaminated

make a run with lower injector temperature; if the baseline improves, replace liner, use low bleed or high temperature septa

septum particles in column entrance

cut 1 turn from column entrance; replace injection septum

column contaminated

cut two turns from column entrance; rinse column with solvent (only chemically bonded phases); otherwise replace column or use guard column

column not properly conditioned

condition column according to manufacturers' instructions (while column is not connected to the detector)

hardware defect

check temperature programme, oven temperature (external thermometer); contact your GC manufacturer

detector contaminated

clean detector

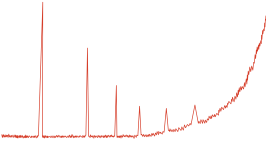
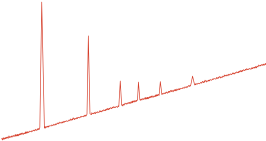
increase of temperature too fast

decrease temperature gradient and end temperature

poor gas quality

use gas grades recommended for GC; for longer supply lines from gas source to GC use gas purification cartridges directly connected to the GC

Do's and don'ts in GC

Observation / Possible causes	Suggested remedy
Increasing baseline, at high temperature bleeding or noise 	
septum particles in column entrance	cut one turn from column entrance; replace injection septum
column contaminated	cut two turns from column entrance; rinse column with solvent (only chemically bonded phases); otherwise replace column or use guard column
increase of temperature too fast / end temperature too high	decrease temperature gradient and end temperature
column not properly conditioned	condition column according to manufacturers' instructions (while column is not connected to the detector)
detector contaminated	clean detector
poor gas quality	use gas grades recommended for GC; for longer supply lines from gas source to GC use gas purification cartridges directly connected to the GC
film thickness in column too high	use column with thinner film
column stability not sufficient for detector sensitivity	use special low bleed column types
Constantly rising baseline 	
leak at column entrance or injection septum	check column installation; search for leaks; replace ferrules
bleeding of septum / injector contaminated	make a run at lower injector temperature; if the baseline improves, replace liner, use low bleed or high temperature septa
septum particles in column entrance	cut one turn from column entrance; replace injection septum
column contaminated	cut two turns from column entrance; rinse column with solvent (only chemically bonded phases); otherwise replace column or use guard column
detector contaminated	clean detector
increase of temperature too fast	decrease temperature gradient and end temperature
poor gas quality	use gas grades recommended for GC; for longer supply lines from gas source to GC use gas purification cartridges directly connected to the GC

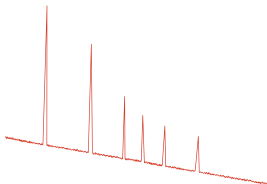
GC troubleshooting



Observation / Possible causes

Suggested remedy

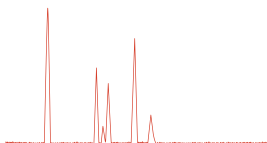
Constantly declining baseline



gas flow changes with temperature gradient

check gas content in gas cylinder; pressure must be a few bar above the required pressure at max. temperature; otherwise exchange gas cylinder

Short lifetime, poor resolution, lack of separation efficiency



impurities on the column

cut two turns from column entrance; rinse column with solvent (only chemically bonded phases); otherwise replace column or use guard column

contamination from vials / septa or sample preparation

check SPE and / or autosampler vials; use low bleed or high temperature septa

polymerisation on the column

use guard column (at least 10 m)

separation efficiency decreases for repeated injections, improves after reconditioning

use guard column; reduce injection volume; modify temperature programme; for repeated injections increase end temperature (if possible) and use longer temperature programme

temperature too high / temperature increase too fast

decrease oven temperature and / or temperature gradient (should not be higher than 25 °C/min)

cooling too fast

do not open oven door at high temperatures

temperature too low / condensation

increase injector temperature and/or start temperature

poor deactivation

check capillary ends; check intact deactivation using the test mixture; for poor results shorten both column ends by about 10 cm; or replace column; if column test does not show any defects:
a) use a column with thicker film
b) use phase with better deactivation
c) use column with special selectivity

air in the system

use oxygen absorber or gas grade with less oxygen

water content too high

reduce water content

head-space analysis: permanent air injections

displace oxygen from vials with an inert gas

Do's and don'ts in GC

Observation / Possible causes

Suggested remedy

Regular interfering peaks



bleeding of silicon septa

replace injection septum, use low bleed or high temperature septa

poor gas quality

use gas grades recommended for GC; for longer supply lines from gas source to GC use gas purification cartridges directly connected to the GC

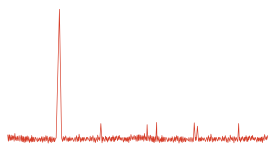
FID: dust or contaminants in the detector

clean detector; if particles are visible in the column or column ends are not cut precisely (frayed edges), cut two turns from the column entrance

electronic defect, damaged cable or detector

replace cable, contact your GC manufacturer

Irregular interfering peaks, spikes, ghost peaks



contamination from vials / septa or sample preparation

control SPE and / or autosampler vials; use low bleed or high temperature septa

derivatisation not quantitative

check derivatisation protocol; use more reactive derivatisation reagents

dirty syringe

use a different syringe or clean it

sample decomposes

check temperature programme, oven temperature (external thermometer); if analytes are not temperature-stable, reduce injector temperature; replace liner

column absorbs or decomposes analytes

check capillary ends; check intact deactivation using the test mixture; for poor results shorten both column ends by about 10 cm; or replace column; if column test does not show any defects:
a) use a column with thicker film
b) use phase with better deactivation
c) use column with special selectivity

sample volume too high, double injection

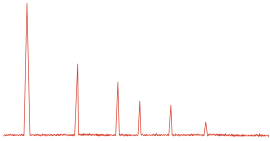
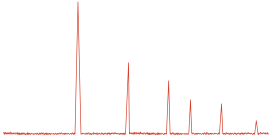
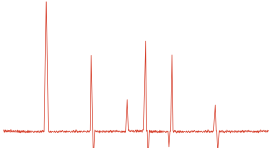
reduce sample volume or add a blank run after a high volume injection

poor gas quality

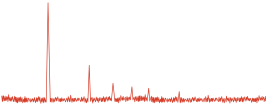
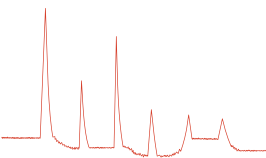
use gas grades recommended for GC; for longer supply lines from gas source to GC use gas purification cartridges directly connected to the GC

GC troubleshooting



Observation / Possible causes	Suggested remedy
Decreased or differing retention times 	
speed of gas too high	compare flow at column entrance and outlet with preset flow; check and / or clean gas tubes; in case of pressure build-up cut and remove one turn (20 cm) from the column or replace column
oven temperature too high	check temperature programme, oven temperature (external thermometer); decrease temperature
Increased or differing retention times/ low gas flow 	
speed of gas too low	increase flow
column connection leaks, column not properly installed	check column installation; search for leaks; replace ferrules.
oven temperature too low	check temperature programme, oven temperature (external thermometer); if the analytes are stable, increase temperature
strong decrease of gas pressure	replace septum; for an instrument with pressure / temperature control, flow must be higher than 15 psi above the demand at max. temperature of the programme
tubes / capillaries / column constricted or blocked	compare flow at column entrance and outlet with preset flow; check and / or clean gas tubes; in case of pressure build-up cut and remove one turn (20 cm) from the column or replace column
Negative peaks, negative signals 	
polarity of integrator is inverted	invert polarity at the instrument
sample injected into wrong column	inject sample in proper column
column overload	decrease injection volume; dilute sample
pressure fluctuations	check gas tubes, valves, seals; test gas flow; change injection septum; contact hardware manufacturer
detector contaminated	clean detector

Do's and don'ts in GC

Observation / Possible causes	Suggested remedy
Peaks too small, poor quantification, concentrations not reproducible 	
dirty syringe	use a different syringe or clean it
concentration of sample too low	increase injection volume; concentrate sample
split too high	reduce split
sensitivity of detector too low	inject standard in order to test detector sensitivity
column connection leaks, column not properly installed	check column installation; search for leaks; replace ferrules
injector temperature too low	check temperature programme, increase injector temperature
dirty ECD	clean ECD
FID, TCD gas flow too low	correct flow according to manufacturers' instructions
sample decomposes	check capillary ends; check intact deactivation using the test mixture; for poor results shorten both column ends by about 10 cm; or replace column; if column test does not show any defects: - use a column with thicker film - use phase with better deactivation - use column with special selectivity
Baseline increases or decreases before or after a peak 	
injection volume too high	decrease injection volume; dilute sample; clean injection system
column bleeding due to poor conditioning	condition column according to manufacturers' instructions (while column is not connected to the detector)
pressure fluctuations	check gas tubes, valves, seals; test gas flow; change injection septum; contact hardware manufacturer
baseline not properly adjusted	readjust baseline; calibrate integrator
injector temperature too low	check injector temperature; if the analytes are stable, increase temperature
injection septum perforated	replace septum
wrong TCD gas flow	adjust flow according to manufacturers' instructions

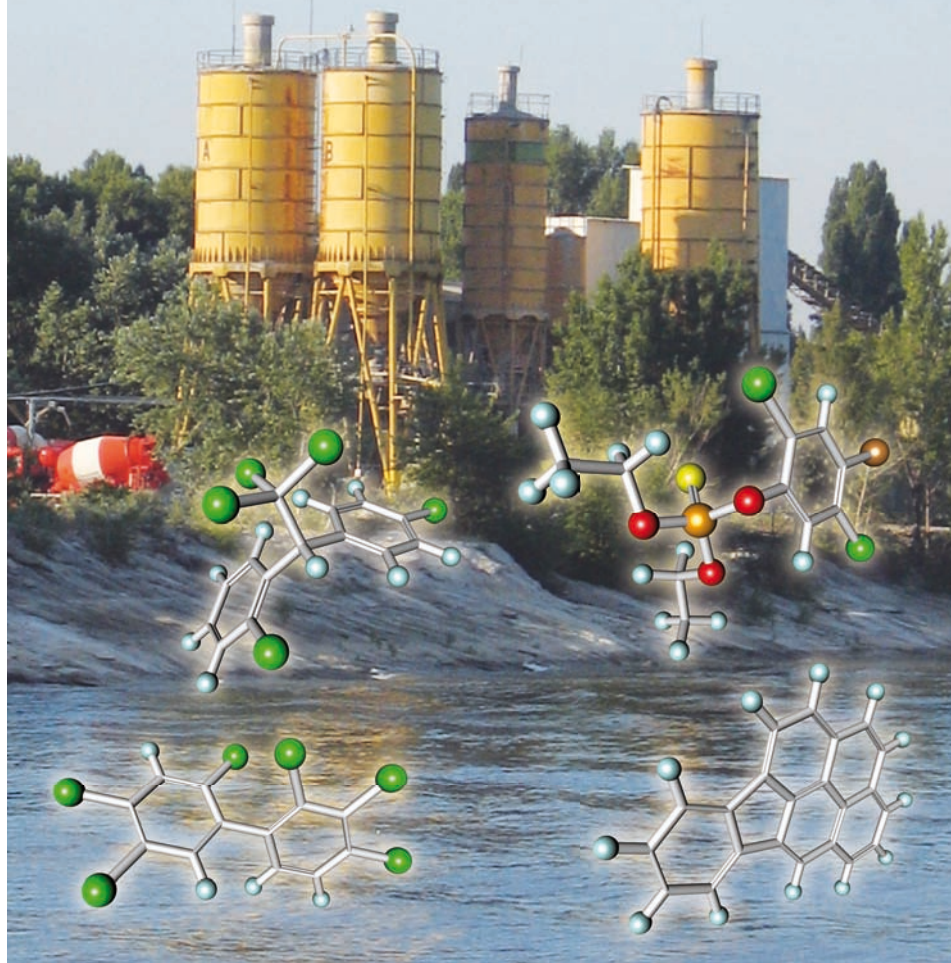


Environmental Pollutants

hydrocarbons · PAH

PCB · phenols

pesticides



Environmental pollutants

Separation of PAH, PCB, pesticides and phthalates (EPA 525)

MN Appl. No. 212810

Column: OPTIMA® XLB, 30 m x 0.25 mm ID, 0.25 µm film, REF 725850.30,
max. temperature 340/360 °C

Sample: standard according to EPA method 525, 5 ng per component

Injection: 1 µl, 300 °C, pressure pulsed (0.4 min 30 psi), splitless for 0.4 min

Carrier: 1.0 ml/min He

Temperature: 35 °C (2 min) $\xrightarrow{20\text{ °C/min}}$ 260 °C $\xrightarrow{6\text{ °C/min}}$ 330 °C (5 min)

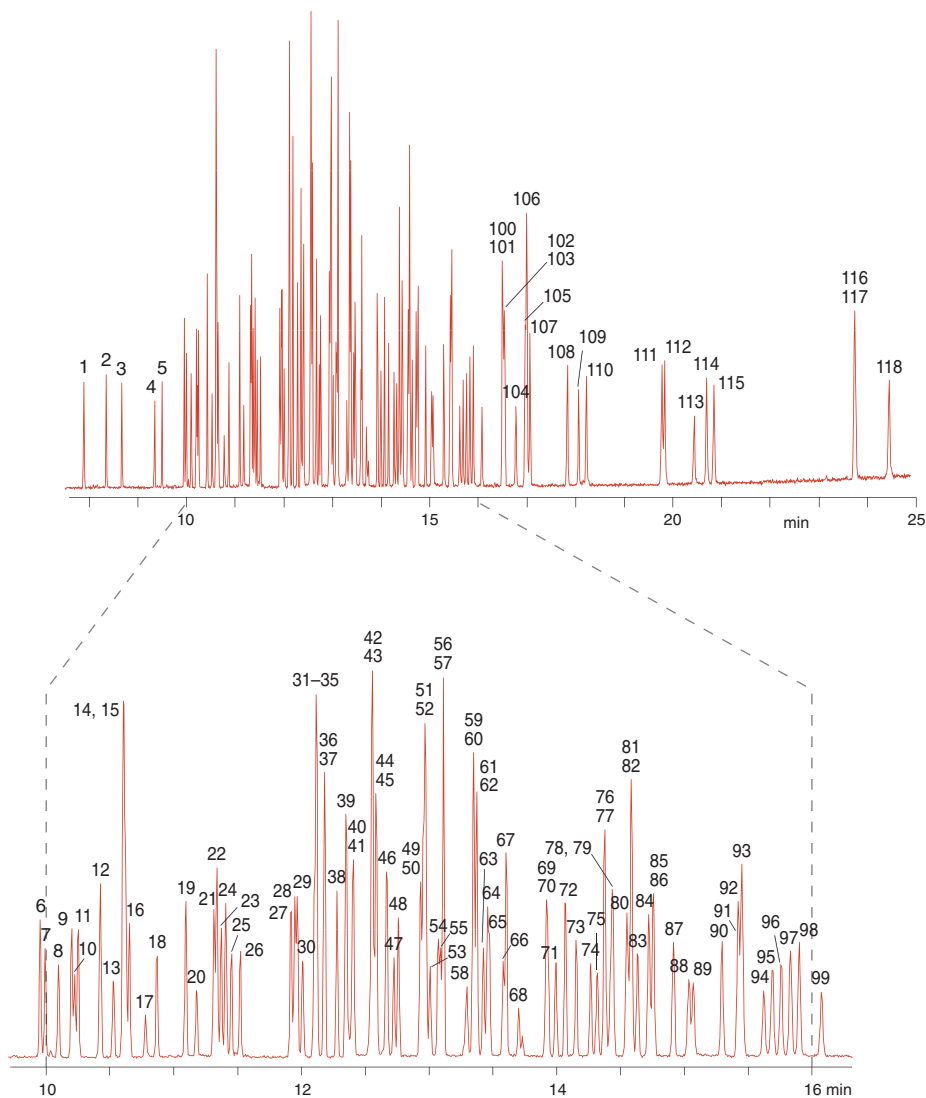
Detector: MSD, 280 °C, scan range 45 – 550 amu

Peaks:

- | | | |
|--|---|---|
| 1. Isophorone | 40. β-BHC | 79. <i>trans</i> -Nonachlor |
| 2. 2-Nitro- <i>m</i> -xylene | 41. Disulfoton | 80. Pyrene-d ₁₀ |
| 3. Dichlorvos | 42. Terbacil | 81. Pyrene |
| 4. Hexachlorocyclopentadiene | 43. Phenanthrene-d ₁₀ | 82. 4,4'-DDE |
| 5. EPTC = S-ethyl N,N-dipropyl-
thiocarbamate | 44. Parathion-methyl | 83. PCB-154 |
| 6. Butylate | 45. Phenanthrene | 84. <i>p</i> -Terphenyl-d ₁₄ |
| 7. Mevinphos | 46. Anthracene | 85. Dieldrin |
| 8. Vernolate | 47. γ-BHC (lindane) | 86. Carboxin |
| 9. Pebulate | 48. PCB-29 | 87. Chlorobenzilate |
| 10. Etridiazole (terrazole) | 49. Alachlor | 88. Tricyclazole |
| 11. Dimethyl phthalate | 50. Prometryn | 89. Endrin |
| 12. Acenaphthene | 51. Ametryn | 90. 4,4'-DDD |
| 13. 2,6-Dinitrotoluene | 52. Simetryn | 91. Bis(2-ethylhexyl) adipate |
| 14. Acenaphthene-d ₁₀ | 53. δ-BHC | 92. Butyl benzyl phthalate |
| 15. PCB-1 | 54. Heptachlor | 93. Endosulfan II |
| 16. Chloroneb | 55. Chlorothalonil | 94. Endrin aldehyde |
| 17. Tebuthiuron | 56. Di- <i>n</i> -butyl phthalate | 95. Norflurazon |
| 18. Molinate | 57. Terbutryn | 96. 4,4'-DDT |
| 19. Diethyl phthalate | 58. Bromacil | 97. Triphenyl phosphate |
| 20. 2,4-Dinitrotoluene | 59. Chlorpyrifos | 98. Hexazinone |
| 21. Propachlor | 60. Metolachlor | 99. Endosulfan sulphate |
| 22. Fluorene | 61. Chlorthal-methyl | 100. Bis(2-ethylhexyl) phthalate |
| 23. Ethoprop | 62. PCB-47 | 101. Methoxychlor |
| 24. Cycloate | 63. Aldrin | 102. PCB-201 |
| 25. Trifluralin | 64. Triadimefon | 103. PCB-171 |
| 26. Chlorpropham | 65. Cyanazine (Bladex) | 104. Endrin ketone |
| 27. PCB-5 | 66. MGK-264 (N-octyl bicyclo-
heptene dicarboximide) | 105. Benz[a]anthracene |
| 28. Atraton | 67. Diphenamid | 106. Chrysene-d ₁₂ |
| 29. Prometon | 68. Merphos | 107. Chrysene |
| 30. α-BHC | 69. PCB-98 | 108. Fenarimol |
| 31. Hexachlorobenzene | 70. Heptachlor epoxide B | 109. <i>cis</i> -Permethrin |
| 32. Propazine | 71. Heptachlor epoxide A | 110. <i>trans</i> -Permethrin |
| 33. Simazin | 72. Butachlor | 111. Benzo[b]fluoranthene |
| 34. Atrazine | 73. Stirofos (tetrachlorvinphos) | 112. Benzo[k]fluoranthene |
| 35. Metribuzin | 74. Fenamiphos | 113. Fluridone (Sonar®) |
| 36. Diazinon | 75. α-Chlordane | 114. Benzo[a]pyrene |
| 37. Terbufos | 76. Napropamide | 115. Perylene-d ₁₂ |
| 38. Pronamide (propyzamide) | 77. γ-Chlordane | 116. Dibenzo[ah]anthracene |
| 39. Pentachlorophenol | 78. Endosulfan I | 117. Indeno[1,2,3-cd]pyrene |
| | | 118. Benzo[ghi]perylene |

For formulas of selected compounds see structure index from page 291.

Complex mixture



Environmental pollutants

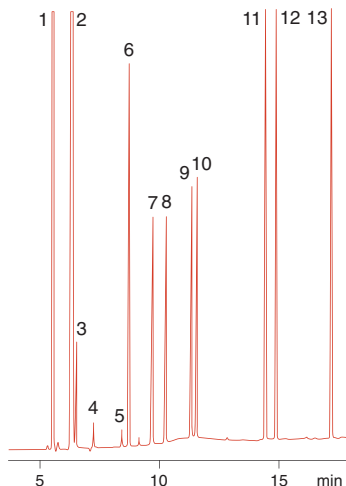
Analysis of volatile halogenated hydrocarbons

MN Appl. No. 210560

Column: OPTIMA® 5,
60 m x 0.32 mm ID,
1.0 µm film, REF 726325.60,
max. temperature 340/360 °C
Injection: 0.1 µl
Carrier gas: 56 ml/min N₂ $\xrightarrow{10\text{ °C/min}}$ 200 °C
Temperature: 50 °C (5 min) $\xrightarrow{10\text{ °C/min}}$ 200 °C
Detector: ECD 300 °C

Peaks:

1. Trichlorofluoromethane (F11)
2. 1,1,2-Trichlorotrifluoroethane (F113)
3. Dichloromethane
4. *trans*-1,2-Dichloroethene
5. *cis*-1,2-Dichloroethene
6. Trichloromethane
7. 1,1,1-Trichloroethane + 1,2-dichloroethane
8. Tetrachloromethane
9. Trichloroethene
10. Bromodichloromethane
11. Dibromochloromethane
12. Tetrachloroethene
13. Tribromomethane



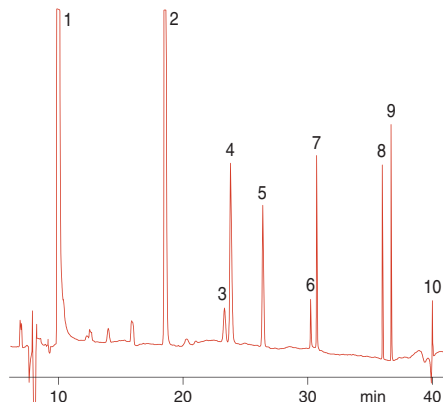
Analysis of a haloform test mixture

MN Appl. No. 212480

Column: PERMABOND® SE-54-HKW,
50 m x 0.32 mm ID,
REF 723945.50,
max. temperature 300/320 °C
Injection: 1 µl, split ~1:30
Carrier gas: 0.9 bar He $\xrightarrow{10\text{ °C/min}}$ 160 °C
Temperature: 35 °C (25 min) $\xrightarrow{10\text{ °C/min}}$ 160 °C
(5 min)
Detector: ECD 300 °C

Peaks [ng/ml]:

1. Dichloromethane [795]
2. Trichloromethane [75]
3. 1,1,1-Trichloroethane [67]
4. 1,2-Dichloroethane [100]
5. Tetrachloromethane [15.9]
6. Trichloroethene [14.6]
7. Bromodichloromethane [20]
8. Dibromochloromethane [122]
9. Tetrachloroethene [81]
10. Tribromomethane [28.9]





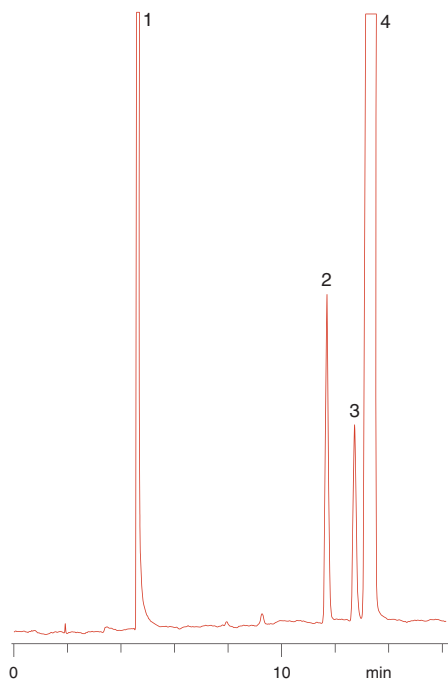
Separation of *cis*-1,2-dichloroethene and bromochloromethane from excess trichloromethane (1:1:1000)

MN Appl. No. 212570

Column: OPTIMA® 624 LB,
30 m x 0.32 mm ID,
1.8 µm film, REF 726786.30
max. temperature 280/300 °C
Injection: 1 µl, 280 °C, split 300 ml/min
Carrier gas: 55 kPa He (1.7 ml/min)
Temperature: isothermal, 40 °C
Detector: FID 280 °C

Peaks:

1. Ethanol (stabilizer)
2. *cis*-1,2-Dichloroethene
3. Bromochloromethane
4. Trichloromethane



Please note that for such a long isothermal separation little differences in film thickness may result in longer or shorter retention times. In this case it may be necessary to adjust pressure or flow rate.

Environmental pollutants

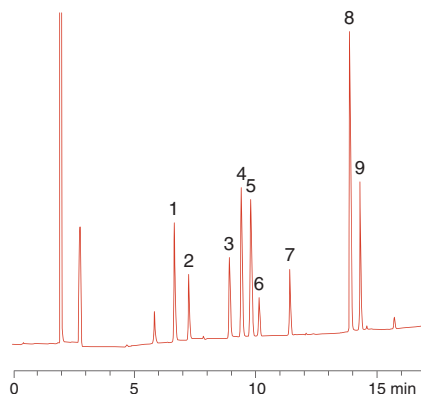
Analysis of volatile hydrocarbons with headspace GC

MN Appl. No. 210330

Column: OPTIMA® 624, 50 m x 0.32 mm ID, 1.8 µm film, REF 726787.50,
max. temperature 280/300 °C
Injection: HS Cryofocussing, splitless, 150 °C
Carrier gas: 2.11 ml/min N₂
Temperature: 36 °C (4 min) $\xrightarrow{10\text{ °C/min}}$ 200 °C (2 min)
Detector: ECD 260 °C

Peaks:

1. Dichloromethane
2. *trans*-1,2-Dichloroethene
3. *cis*-1,2-Dichloroethene
4. Trichloromethane
5. 1,1,1-Trichloroethane
6. Tetrachloromethane
7. Trichloroethene
8. 1,1,2-Trichloroethane
9. Tetrachloroethene



Courtesy of R. Mueller, G&P Torsten Plaar Umweltanalytik, Oldenburg, Germany

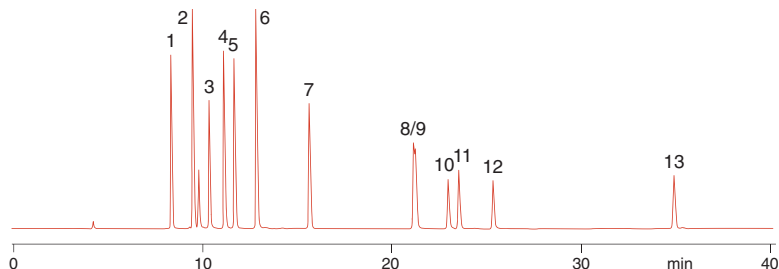
Analysis of chlorinated hydrocarbons

MN Appl. No. 210080

Column: OPTIMA® 624, 30 m x 0.25 mm ID, 1.4 µm film, REF 726785.30,
max. temperature 280/300 °C
Sample: 6.25 µl/l
Temperature: 35 °C (4 min) $\xrightarrow{4\text{ °C/min}}$ 180 °C (8 min)
Detector: FID

Peaks:

- | | | |
|-----------------------------|--------------------------------------|---|
| 1. Isoallyl chloride | 6. 1,5-Hexadiene | 11. 2-Chloromethyloxirane
(Epichlorohydrine) |
| 2. 1-Chloro-2-methylpropane | 7. 2,2-Dichloropropane | 12. (<i>E</i>)-1,3-Dichloropropene |
| 3. 1-Chloropropene | 8. 1,1-Dichloropropane | 13. 1,2,3-Trichloropropane |
| 4. 3-Chloropropene | 9. 2,3-Dichloropropene | |
| 5. 1-Chloropropane | 10. (<i>Z</i>)-1,3-Dichloropropene | |



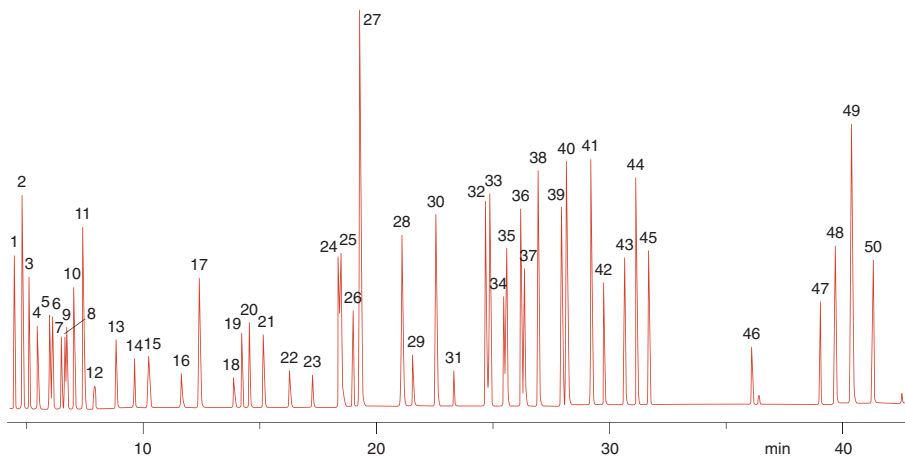
Courtesy of Mrs. Kehl, DOW, Stade, Germany



MN Appl. No. 250461

Column: OPTIMA® δ-6, 50 m x 0.20 mm ID, 0.2 μm film, REF 726465.50,
max. temperature 340/360 °C
Sample: EPA 502 Volatile Organics Calibration Mix, 20 μg/ml per component in methanol
Injection: 1 μl, split 20 ml/min
Carrier gas: 1.5 bar He
Temperature: 40 °C (5 min) $\xrightarrow{2.5\text{ °C/min}}$ 70 °C $\xrightarrow{3\text{ °C/min}}$ 100 °C $\xrightarrow{4\text{ °C/min}}$ 160 °C (10 min)
Detector: MSD

- | | | |
|--|---|---------------------------------|
| 1. 1,1-Dichloroethene | 17. Toluene | 34. 1,2,3-Trichloropropane |
| 2. Dichloromethane | 18. <i>trans</i> -1,3-Dichloropropene | 35. 2-Chlorotoluene |
| 3. <i>trans</i> -1,2-Dichloroethene | 19. 1,1,2-Trichloroethane | 36. 1,3,5-Trimethylbenzene |
| 4. 1,1-Dichloroethane | 20. Tetrachloroethene | 37. 4-Chlorotoluene |
| 5. 2,2-Dichloropropane | 21. 1,3-Dichloropropane | 38. <i>t</i> -Butylbenzene |
| 6. <i>cis</i> -1,2-Dichloroethene | 22. Dibromochloromethane | 39. 1,2,4-Trimethylbenzene |
| 7. Trichloromethane | 23. 1,2-Dibromoethane | 40. <i>sec</i> -Butylbenzene |
| 8. Bromochloromethane | 24. Chlorobenzene | 41. <i>p</i> -Isopropyltoluene |
| 9. 1,1,1-Trichloroethane | 25. Ethylbenzene | 42. 1,3-Dichlorobenzene |
| 10. 1,1-Dichloroprop-1-ene
+ tetrachloromethane | 26. 1,1,1,2-Tetrachloroethane | 43. 1,4-Dichlorobenzene |
| 11. Benzene | 27. <i>m</i> -Xylene + <i>p</i> -xylene | 44. <i>n</i> -Butylbenzene |
| 12. 1,2-Dichloroethane | 28. <i>o</i> -Xylene | 45. 1,2-Dichlorobenzene |
| 13. Trichloroethene | 29. Styrene | 46. 1,2-Dibromo-3-chloropropane |
| 14. 1,2-Dichloropropane | 30. Isopropylbenzene | 47. Hexachlorobutadiene |
| 15. Dibromomethane
+ bromodichloromethane | 31. Tribromomethane | 48. 1,2,4-Trichlorobenzene |
| 16. <i>cis</i> -1,3-Dichloropropene | 32. <i>n</i> -Propylbenzene | 49. Naphthalene |
| | 33. Bromobenzene
+ 1,1,2,2-tetrachloroethane | 50. 1,2,3-Trichlorobenzene |



Environmental pollutants

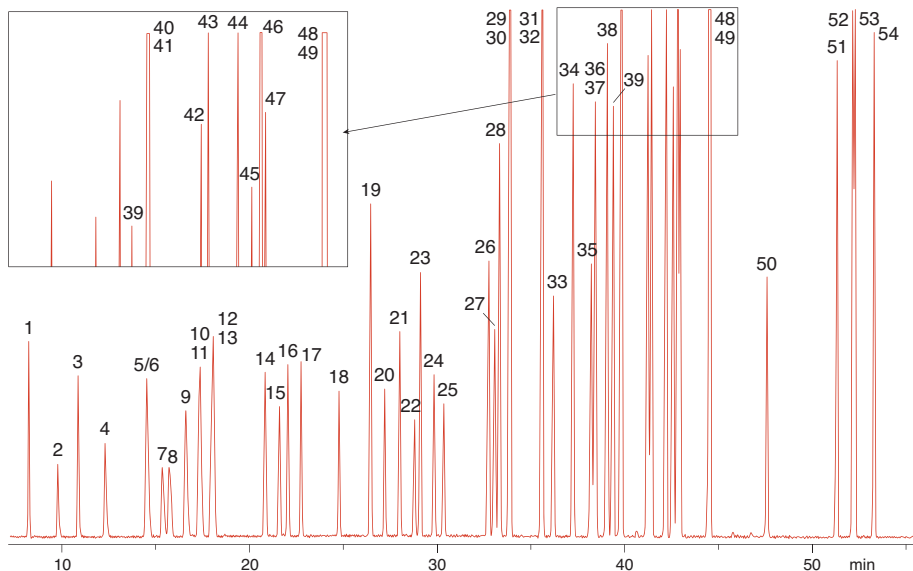
Analysis of volatile organic compounds (EPA 502/524)

MN Appl. No. 211280

Column: OPTIMA® 624, 50 m x 0.25 mm ID, 1.4 µm film, REF 726785.50,
max. temperature 280/300 °C
Sample: EPA 502 Volatile Organics Calibration Mix, Supelco 502 111,
2000 µg/ml per component in methanol
Injection: 1 µl, split 1:50
Carrier gas: 1.5 bar He
Temperature: 40 °C (5 min) $\xrightarrow{2.5\text{ °C/min}}$ 70 °C $\xrightarrow{3\text{ °C/min}}$ 100 °C $\xrightarrow{4\text{ °C/min}}$ 220 °C (5 min)
Detector: MSD

Peaks:

- | | | |
|-------------------------------------|---------------------------------------|---------------------------------|
| 1. 1,1-Dichloroethene | 19. Toluene | 37. Bromobenzene |
| 2. Dichloromethane | 20. <i>trans</i> -1,3-Dichloropropene | 38. <i>n</i> -Propylbenzene |
| 3. <i>trans</i> -1,2-Dichloroethene | 21. 1,1,2-Trichloroethane | 39. 2-Chlorotoluene |
| 4. 1,1-Dichloroethane | 22. 1,3-Dichloropropane | 40. 1,3,5-Trimethylbenzene |
| 5. 2,2-Dichloropropane | 23. Tetrachloroethene | 41. 4-Chlorotoluene |
| 6. <i>cis</i> -1,2-Dichloroethene | 24. Dibromochloromethane | 42. <i>t</i> -Butylbenzene |
| 7. Bromochloromethane | 25. 1,2-Dibromoethane | 43. 1,2,4-Trimethylbenzene |
| 8. Trichloromethane | 26. Chlorobenzene | 44. <i>sec</i> -Butylbenzene |
| 9. 1,1,1-Trichloroethane | 27. 1,1,1,2-Tetrachloroethane | 45. 1,3-Dichlorobenzene |
| 10. 1,1-Dichloropropene | 28. Ethylbenzene | 46. <i>p</i> -Isopropyltoluene |
| 11. Tetrachloromethane | 29. <i>m</i> -Xylene | 47. 1,4-Dichlorobenzene |
| 12. 1,2-Dichloroethane | 30. <i>p</i> -Xylene | 48. <i>n</i> -Butylbenzene |
| 13. Benzene | 31. <i>o</i> -Xylene | 49. 1,2-Dichlorobenzene |
| 14. Trichloroethene | 32. Styrene | 50. 1,2-Dibromo-3-chloropropane |
| 15. 1,2-Dichloropropane | 33. Tribromomethane | 51. 1,2,4-Trichlorobenzene |
| 16. Dibromomethane | 34. Isopropylbenzene | 52. Hexachlorobutadiene |
| 17. Bromodichloromethane | 35. 1,1,2,2-Tetrachloroethane | 53. Naphthalene |
| 18. <i>cis</i> -1,3-Dichloropropene | 36. 1,2,3-Trichloropropane | 54. 1,2,3-Trichlorobenzene |



Volatile hydrocarbons



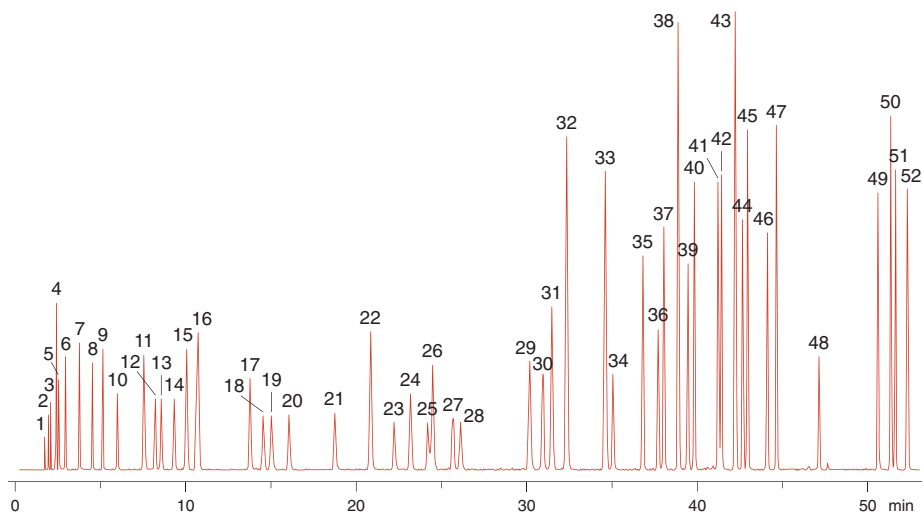
Analysis of volatile organic compounds (EPA 624)

MN Appl. No. 250200

Column: OPTIMA® 624, 25 m x 0.20 mm ID, 1.1 µm film, REF 726784.25, max. temperature 280/300 °C
 Sample: EPA 624 Volatile Organics, 20 µg/ml per component in methanol
 Injection: 1 µl, split 1:50
 Carrier gas: 20 cm/s He
 Temperature: 40 °C (8 min) $\xrightarrow{1.5\text{ °C/min}}$ 70 °C $\xrightarrow{4\text{ °C/min}}$ 180 °C
 Detector: MSD, EI

Peaks:

- | | | |
|-------------------------------------|---|---------------------------------|
| 1. Dichlorofluoromethane | 18. 1,2-Dichloropropane | 37. 1,2,3-Trichloropropane + |
| 2. Chloromethane | 19. Dibromomethane | 1,1,2,2-tetrachloroethane |
| 3. Vinyl chloride | 20. Bromodichloromethane | 38. Benzyl chloride |
| 4. Bromomethane | 21. <i>cis</i> -1,3-Dichloropropene | 39. 2-Chlorotoluene |
| 5. Chloroethane | 22. Toluene | + 3-chlorotoluene |
| 6. Trichlorofluoromethane | 23. <i>trans</i> -1,3-Dichloropropene | 40. 1,3,5-Trimethylbenzene |
| 7. 1,1-Dichloroethene | 24. 1,1,2-Trichloroethane | 41. <i>t</i> -Butylbenzene |
| 8. Dichloromethane | 25. 2-Chloro-1-propene | 42. 1,2,4-Trimethylbenzene |
| 9. <i>trans</i> -1,2-Dichloroethene | 26. Tetrachloroethene | 43. <i>sec</i> -Butylbenzene |
| 10. 1,1-Dichloroethane | 27. Dibromochloromethane | + 1,3-dichlorobenzene |
| 11. <i>cis</i> -1,2-Dichloroethane | 28. 1,2-Dibromoethane | 44. 1,4-Dichlorobenzene |
| 12. Bromochloromethane | 29. Chlorobenzene | 45. <i>o</i> -Isopropyltoluene |
| + 2,2-dichloropropane | 30. 1,1,1,2-Tetrachloroethane | 46. 1,2-Dichlorobenzene |
| 13. Trichloromethane | 31. Ethylbenzene | 47. <i>n</i> -Butylbenzene |
| 14. 1,1,1-Trichloroethane | 32. <i>m</i> -Xylene + <i>p</i> -xylene | 48. 1,2-Dibromo-3-chloropropane |
| 15. Tetrachloromethane | 33. Styrene + <i>o</i> -xylene | 49. 1,2,4-Trichlorobenzene |
| + 1,1-dichloropropane | 34. Tribromomethane | 50. Naphthalene |
| 16. Benzene | 35. Isopropylbenzene | 51. Hexachloro-1,3-butadiene |
| + 1,2-dichloroethane | 46. Bromobenzene | 52. 1,2,3-Trichlorobenzene |
| 17. Trichloroethene | | |



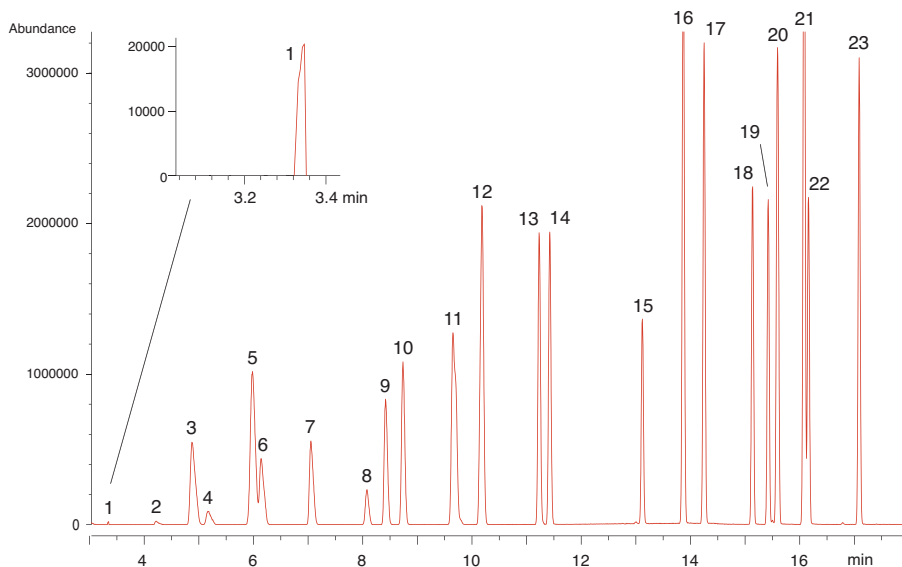
Environmental pollutants

Analysis of volatile halogenated hydrocarbons and BTX MN Appl. No. 200150

Column: PERMABOND® SE-54-HKW, 50 m x 0.32 mm ID, REF 723945.50,
max. temperature 300/320 °C
Injection: 1 µl, 0.2% in methanol, split 15 ml/min
Carrier gas: 1.5 ml/min He (constant flow)
Temperature: 40 °C (5 min) $\xrightarrow{10\text{ °C/min}}$ 160 °C
Detector: MSD

Peaks:

- | | |
|--|-----------------------------------|
| 1. Vinyl chloride | 13. Trichloroethene |
| 2. Ethanol | 14. Dichlorobromomethane |
| 3. Trichlorofluoromethane (F 11) | 15. Toluene |
| 4. Pentane | 16. Chlorodibromomethane |
| 5. 1,1,2-Trichlorotrifluoroethane (F 113) | 17. Tetrachloroethene |
| 6. Dichloromethane | 18. Chlorobenzene |
| 7. <i>trans</i> -1,2-Dichloroethene | 19. Ethylbenzene |
| 8. Hexane | 20. <i>m</i> - + <i>p</i> -Xylene |
| 9. <i>cis</i> -1,2-Dichloroethene | 21. Tribromomethane |
| 10. Trichloromethane | 22. <i>o</i> -Xylene |
| 11. 1,1,1-Trichloroethane + 1,2-dichloroethane | 23. Bromobenzene |
| 12. Tetrachloromethane + benzene | |



Chlorinated hydrocarbons



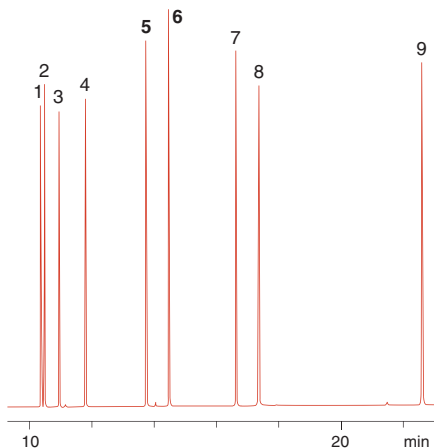
Analysis of chlorinated hydrocarbons (EPA 612)

Peaks:

- | | | |
|------------------------|---------------------------|------------------------------|
| 1. 1,3-Dichlorobenzene | 4. Hexachloroethane | 7. Hexachlorocyclopentadiene |
| 2. 1,4-Dichlorobenzene | 5. 1,2,4-Trichlorobenzene | 8. 2-Chloronaphthalene |
| 3. 1,2-Dichlorobenzene | 6. Hexachlorobutadiene | 9. Hexachlorobenzene |

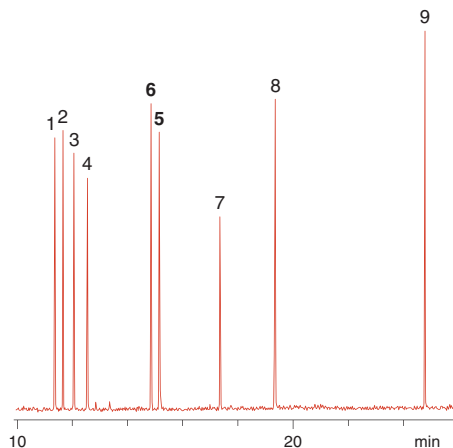
MN Appl. No. 211290

Column: OPTIMA® 5,
50 m x 0.20 mm ID,
0.20 μ m film, REF 726857.50,
max. temperature 340/360 °C
Sample: chlorinated hydrocarbon
standard EPA 612,
2000 μ g/ml per component in
isooctane
Injection: 1 μ l, split 1:20
Carrier gas: 2.5 bar He
Temperature: 50 °C $\xrightarrow{8\text{ °C/min}}$ 260 °C (10 min)
Detector: MSD



MN Appl. No. 250440

Column: OPTIMA® δ -6,
50 m x 0.20 mm ID,
0.20 μ m film, REF 726465.50,
max. temperature 340/360 °C
Sample: chlorinated hydrocarbon
standard EPA 612,
200 μ g/ml per component in
isooctane
Injection: 1 μ l, split 1:20
Carrier gas: 2.0 bar He
Temperature: 50 °C $\xrightarrow{8\text{ °C/min}}$ 260 °C (10 min)
Detector: MSD



Environmental pollutants

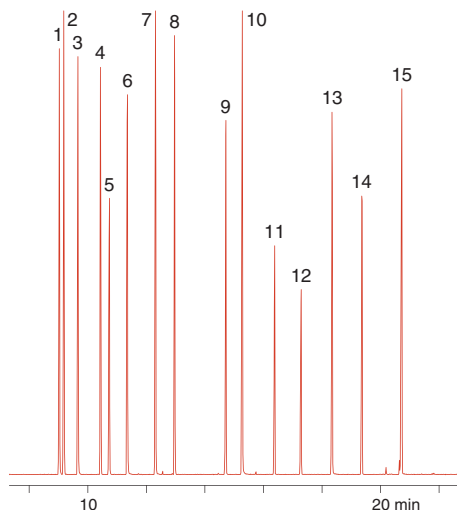
Analysis of a mixture of neutral and basic organic compounds

MN Appl. No. 200320

Column: OPTIMA® 5,
25 m x 0.20 mm ID,
0.2 µm film, REF 726857.25,
max. temperature 340/360 °C
Sample: 200 µg/ml each in 2-propanol
Injection: 1.0 µl; split 1:150
Carrier gas: 25 cm/s He
Temperature: 50 °C (5 min) $\xrightarrow{10\text{ °C/min}}$ 220 °C
 $\xrightarrow{20\text{ °C/min}}$ 330 °C
Detector: MSD

Peaks:

- 1,3-Dichlorobenzene
- 1,4-Dichlorobenzene
- 1,2-Dichlorobenzene
- Hexachloroethane
- Nitrobenzene
- Isophorone
- 1,2,4-Trichlorobenzene
- Hexachlorobutadiene
- Hexachlorocyclopentadiene
- 1-Chloronaphthalene
- 2-Methyl-1,3-dinitrobenzene
- 1-Methyl-2,4-dinitrobenzene
- Azobenzene
- Hexachlorobenzene
- Carbazole

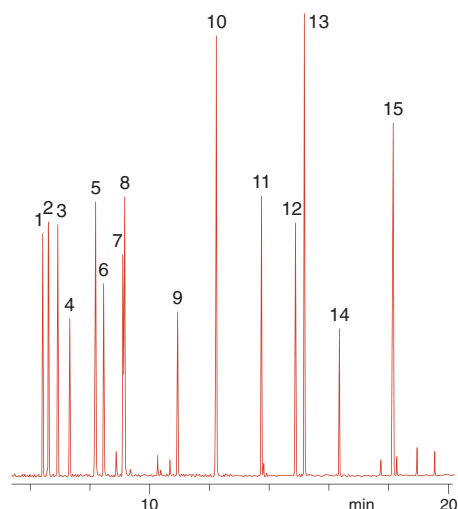


MN Appl. No. 250070

Column: OPTIMA® δ-3,
25 m x 0.20 mm ID,
0.20 µm film, REF 726400.25,
max. temperature 340/360 °C
Injection: 1 µl, split 1:150
Carrier gas: 25 cm/s He
Temperature: 50 °C (5 min) $\xrightarrow{10\text{ °C/min}}$ 220 °C
 $\xrightarrow{20\text{ °C/min}}$ 330 °C
Detector: MSD

Peaks:

- 1,4-Dichlorobenzene
- 1,3-Dichlorobenzene
- 1,2-Dichlorobenzene
- Hexachloroethane
- Nitrobenzene
- Isophorone
- 1,2,4-Trichlorobenzene
- Hexachlorobutadiene
- Hexachlorocyclopentadiene
- 1-Chloronaphthalene
- 2-Methyl-1,3-dinitrobenzene
- 1-Methyl-2,4-dinitrobenzene
- Azobenzene
- Hexachlorobenzene
- Carbazole



Halogenated hydrocarbons



Analysis of chlorotoluene isomers

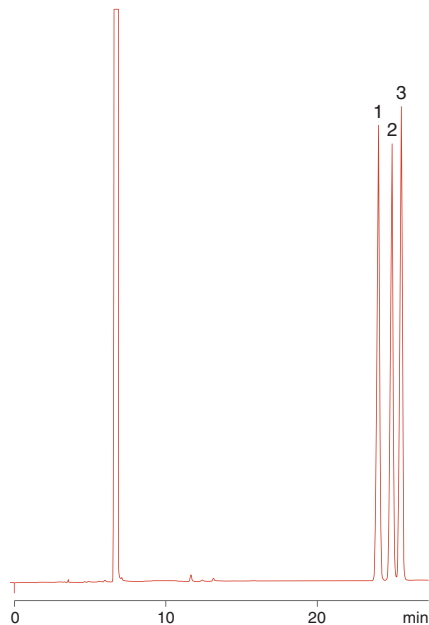
Column: OPTIMA® 624, 50 m x 0.32 mm ID, 1.8 µm film, REF 726787.50,
max. temperature 280/300 °C
Carrier gas: 150 kPa H₂ (7.7 ml/min)
Temperature: 70 °C
Detector: FID 250 °C

MN Appl. No. 200340

Injection: 1 µl (1 % each in toluene)
split 220 ml/min

Peaks:

1. *o*-Chlorotoluene
2. *m*-Chlorotoluene
3. *p*-Chlorotoluene

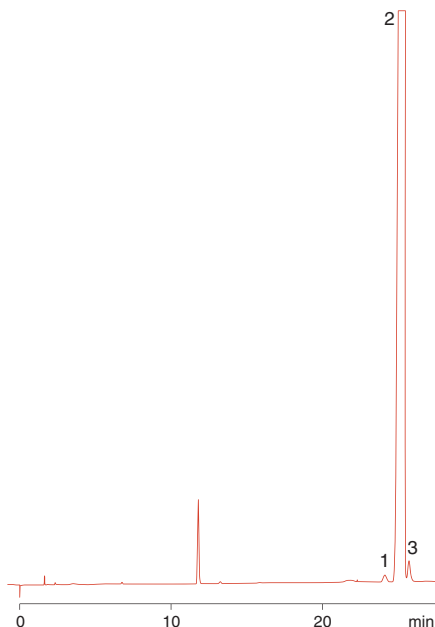


MN Appl. No. 200350

Injection: 1 µl (undiluted sample)
split 220 ml/min

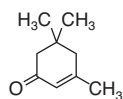
Peaks:

1. *o*-Chlorotoluene [0.16 %]
2. *m*-Chlorotoluene [98.31 %]
3. *p*-Chlorotoluene [0.39 %]

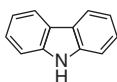


Structures of miscellaneous organics

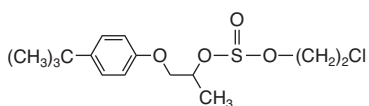
Isophorone



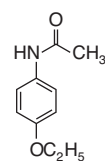
Carbazole



Aramite



Phenacetin



Environmental pollutants

Analysis of semivolatile organics (EPA 8270)

MN Appl. No. 212790

Column: OPTIMA® 5 MS Accent, 30 m x 0.25 mm ID, 0.25 µm film, REF 725820.30, max. temperature 340/360 °C

Sample: 16 µg/ml in CH₂Cl₂

Injection: 1 µl splitless (hold 0.4 min), 300 °C

Carrier gas: 1.0 ml/min He

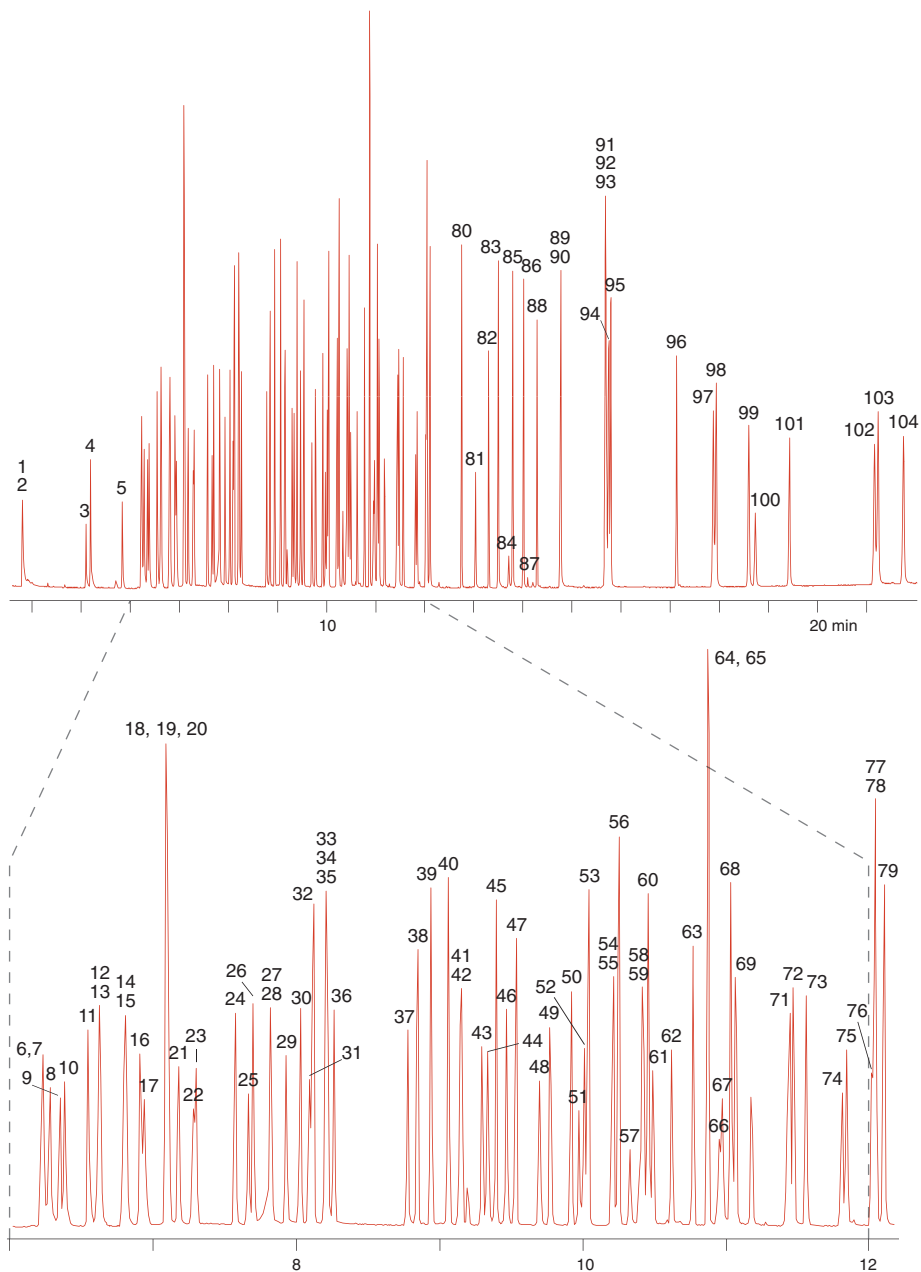
Temperature: 35 °C (2 min) $\xrightarrow{20\text{ °C/min}}$ 260 °C $\xrightarrow{6\text{ °C/min}}$ 330 °C (1 min)

Detection: MSD 280 °C

Peaks:

- | | | |
|---|----------------------------------|---|
| 1. <i>N</i> -Nitrosodimethylamine | 35. Hexachloropropene | 70. 2,4,6-Tribromophenol |
| 2. Pyridine | 36. Hexachlorobutadiene | 71. Phenacetin |
| 3. Methyl methanesulfonate | 37. 4-Chloro-3-methylphenol | 72. 4-Bromophenyl phenyl ether |
| 4. 2-Fluorophenol | 38. Isosafrole | 73. Hexachlorobenzene |
| 5. Ethyl methanesulfonate | 39. 2-Methylnaphthalene | 74. Pentachlorophenol |
| 6. Phenol-d ₆ | 40. 1-Methylnaphthalene | 75. Pentachloronitrobenzene |
| 7. Phenol | 41. Hexachlorocyclopentadiene | 76. Phenanthrene-d ₁₀ |
| 8. Aniline | 42. 1,2,4,5-Tetrachlorobenzene | 77. Dinoseb |
| 9. Bis(2-chloroethyl) ether | 43. 2,4,6-Trichlorophenol | 78. Phenanthrene |
| 10. 2-Chlorophenol | 44. 2,4,5-Trichlorophenol | 79. Anthracene |
| 11. 1,3-Dichlorobenzene | 45. 2-Fluorobiphenyl | 80. Di- <i>n</i> -butyl phthalate |
| 12. 1,4-Dichlorobenzene-d ₄ | 46. Safrole | 81. 4-Nitro-quinoline 1-oxide |
| 13. 1,4-Dichlorobenzene | 47. 2-Chloronaphthalene | 82. Isodrin |
| 14. 1,2-Dichlorobenzene | 48. 2-Nitroaniline | 83. Fluoranthene |
| 15. Benzyl alcohol | 49. 1,4-Naphthoquinone | 84. Benzinide |
| 16. 2-Methylphenol | 50. Dimethyl phthalate | 85. Pyrene |
| 17. Bis(2-chloroisopropyl) ether | 51. 1,3-Dinitrobenzene | 86. <i>p</i> -Terphenyl-d ₁₄ |
| 18. Acetophenone | 52. 2,6-Dinitrotoluene | 87. Aramite |
| 19a. 4-Methylphenol | 53. Acenaphthylene | 88. Chlorobenzilate |
| 19b. 3-Methylphenol | 54. Acenaphthene-d ₁₀ | 89. Kepone |
| 20. <i>N</i> -Nitroso-di- <i>n</i> -propylamine | 55. 3-Nitroaniline | 90. Butyl benzyl phthalate |
| 21. Hexachloroethane | 56. Acenaphthene | 91. Benz[a]anthracene |
| 22. Nitrobenzene-d ₅ | 57. 2,4-Dinitrophenol | 92. 3,3'-Dichlorobenzidine |
| 23. Nitrobenzene | 58. Pentachlorobenzene | 93. Chrysene-d ₁₂ |
| 24. Isophorone | 59. 4-Nitrophenol | 94. Chrysene |
| 25. 2-Nitrophenol | 60. Dibenzofuran | 95. Bis(2-ethylhexyl) phthalate |
| 26. 2,4-Dimethylphenol | 61. 2,4-Dinitrotoluene | 96. Di- <i>n</i> -octyl phthalate |
| 27. Bis(2-chloroethoxy)methane | 62. 2,3,4,6-Tetrachlorophenol | 97. Benzo[b]fluoranthene |
| 28. Benzoic acid | 63. Diethyl phthalate | 98. Benzo[k]fluoranthene |
| 29. 2,4-Dichlorophenol | 64. Fluorene | 99. Benzo[a]pyrene |
| 30. 1,2,4-Trichlorobenzene | 65. 4-Chlorophenyl phenyl ether | 100. Perylene-d ₁₂ |
| 31. Naphthalene-d ₈ | 66. 4-Nitroaniline | 101. 3-Methylcholanthrene |
| 32. Naphthalene | 67. 2-Methyl-4,6-dinitrophenol | 102. Indeno[1,2,3-cd]pyrene |
| 33. 2,6-Dichlorophenol | 68. Diphenylamine | 103. Dibenz[ah]anthracene |
| 34. 4-Chloroaniline | 69. Azobenzene | 104. Benzo[ghi]perylene |

Semivolatile organics



Environmental pollutants

Analysis of semivolatile organics (EPA 625)

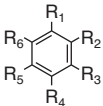
MN Appl. No. 212830

Column: OPTIMA® 5 MS Accent, 30 m x 0.25 mm ID, 0.25 µm film, REF 725820.30,
max. temperature 340/360 °C
Injection: 1.0 µl, 10 ppm (20 ppm int. std.), 20 psi 0.3 min, pulsed splitless (hold 0.3 min),
300 °C
Carrier gas: 1.0 ml/min He
Temperature: 35 °C (1 min) $\xrightarrow{18\text{ °C/min}}$ 270 °C $\xrightarrow{5\text{ °C/min}}$ 305 °C $\xrightarrow{30\text{ °C/min}}$ 330 °C (1 min)
Detector: MSD 280 °C

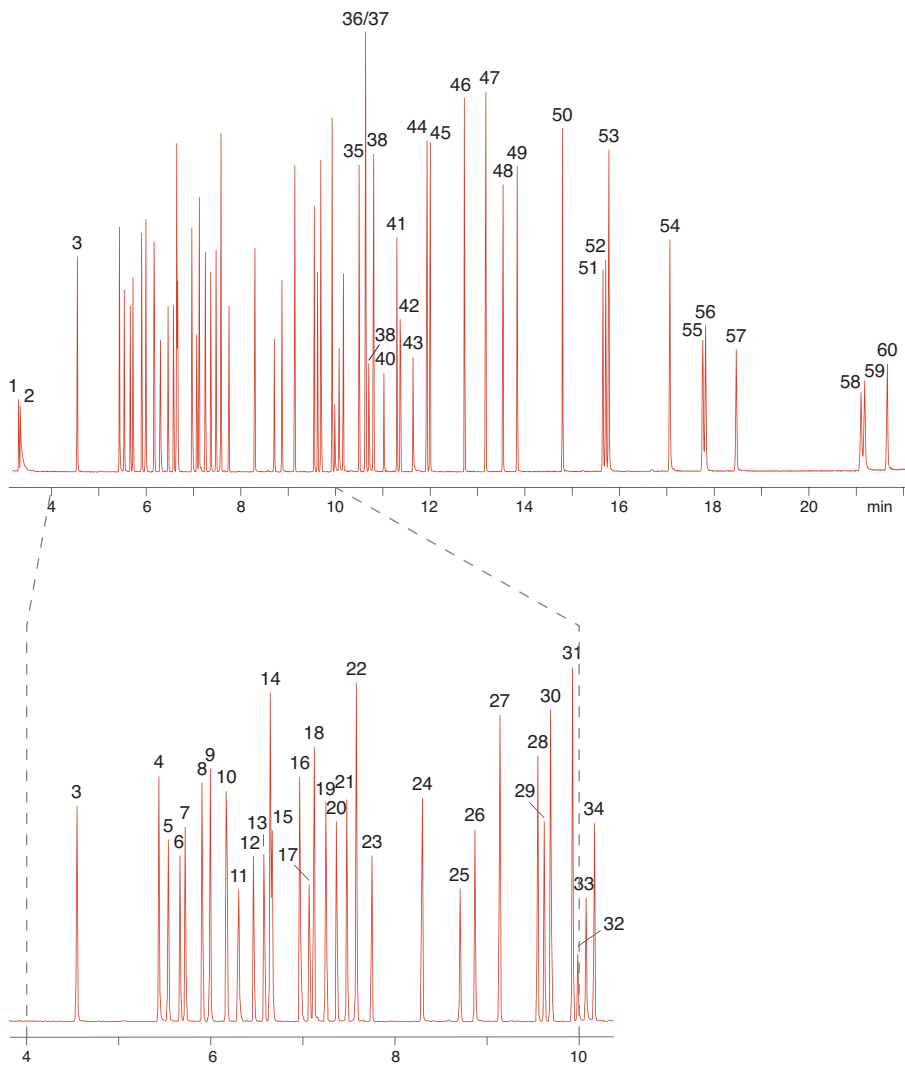
Peaks:

- | | | |
|---|-------------------------------------|--------------------------------------|
| 1. N-Nitrosodimethylamine | 21. 1,2,4-Trichlorobenzene | 41. 4-Bromophenyl phenyl ether |
| 2. Pyridine-d ₅ | 22. Naphthalene | 42. Hexachlorobenzene |
| 3. 2-Fluorophenol (surrogate std.) | 23. Hexachlorobutadiene | 43. Pentachlorophenol |
| 4. Pentafluorophenol (int. std.) | 24. 4-Chloro-3-methylphenol | 44. Phenanthrene |
| 5. Phenol | 25. Hexachlorocyclopentadiene | 45. Anthracene |
| 6. Bis(2-chloroethyl) ether | 26. 2,4,6-Trichlorophenol | 46. Di-n-butyl phthalate |
| 7. 2-Chlorophenol | 27. 2-Chloronaphthalene | 47. 4,4'-Dibromobiphenyl (int. std.) |
| 8. 1,3-Dichlorobenzene | 28. Dimethyl phthalate | 48. Fluoranthene |
| 9. 1,4-Dichlorobenzene | 29. 2,6-Dinitrotoluene | 49. Pyrene |
| 10. 1,2-Dichlorobenzene | 30. Acenaphthylene | 50. Butyl benzyl phthalate |
| 11. Bis(2-chloroisopropyl) ether | 31. Acenaphthene | 51. Benz[a]anthracene |
| 12. N-Nitroso-di-n-propylamine | 32. 2,4-Dinitrophenol | 52. Chrysene |
| 13. Hexachloroethane | 33. 4-Nitrophenol | 53. Bis(2-ethylhexyl) phthalate |
| 14. Nitrobenzene-d ₅ (int. std.) | 34. 2,4-Dinitrotoluene | 54. Di-n-octyl phthalate |
| 15. Nitrobenzene | 35. Diethyl phthalate | 55. Benzo[b]fluoranthene |
| 16. Isophorone | 36. Fluorene | 56. Benzo[k]fluoranthene |
| 17. 2-Nitrophenol | 37. 4-Chlorophenyl phenyl ether | 57. Benzo[a]pyrene |
| 18. 2,4-Dimethylphenol | 38. 4,6-Dinitro-2-methylphenol | 58. Indeno[1,2,3-cd]pyrene |
| 19. Bis(2-chloroethoxy)methane | 39. Diphenylamine | 59. Dibenz[ah]anthracene |
| 20. 2,4-Dichlorophenol | 40. 4,4'-Dibromooctafluoro-biphenyl | 60. Benzo[ghi]perylene |

Substituted benzene common names

Structure	Compound	R ₁	R ₂	R ₃	R ₄	R ₅	R ₆
	HCB = Hexachlorobenzene	Cl	Cl	Cl	Cl	Cl	Cl
	Chloroneb	Cl	OCH ₃	H	Cl	OCH ₃	H
	Chlorthal	Cl	COOH	Cl	Cl	COOH	Cl
	Dichlobenil	Cl	CN	Cl	H	H	H
	Chlorothalonil	Cl	CN	Cl	CN	Cl	Cl
	Tecnazene	Cl	Cl	NO ₂	Cl	Cl	H
	Quintozene	Cl	Cl	NO ₂	Cl	Cl	Cl
	Dicloran	Cl	NH ₂	Cl	H	NO ₂	H
	Dinoseb	OH	sec-C ₄ H ₉	H	NO ₂	H	NO ₂

Semivolatile organics



Environmental pollutants

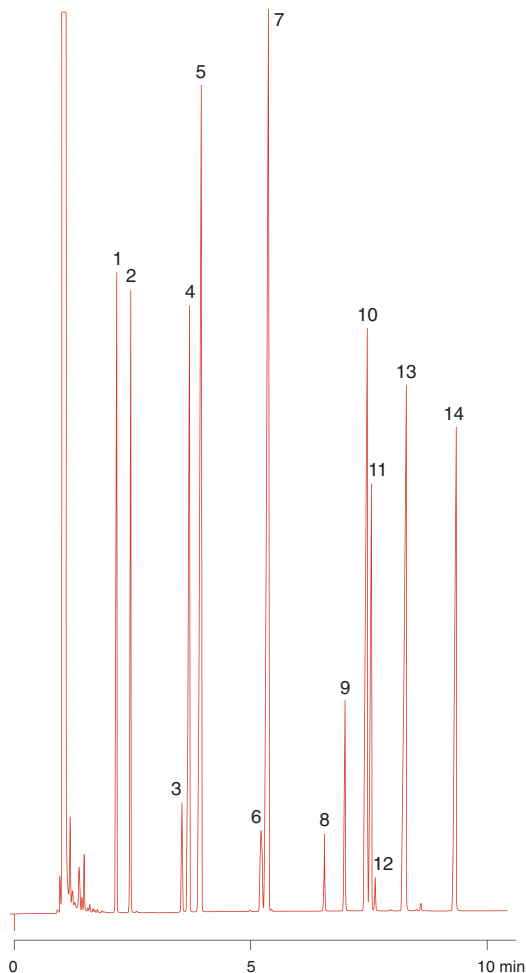
Analysis of nitro- and chloronitroaromatics

MN Appl. No. 200300

Column: OPTIMA® 1301, 25 m x 0.32 mm ID, 0.25 µm film, REF 726777.25,
max. temperature 300/320 °C
Injection: 0.5 µl, split 55 ml/min
Carrier gas: 60 kPa H₂ (3.3 ml/min)
Temperature: 120 °C (3 min) $\xrightarrow{15\text{ °C/min}}$ 230 °C
Detector: FID 250 °C

Peaks:

1. Nitrobenzene
2. Isophorone
3. 3-Nitrochlorobenzene
4. 2-Nitrochlorobenzene
5. 4-Nitrochlorobenzene
6. 2,5- and 2,6-Dichloronitrobenzene
7. 2,4-Dichloronitrobenzene
8. 2,6-Dinitrotoluene
9. 2,5-Dinitrotoluene
10. 2,4-Dinitrotoluene
11. 2,3-Dinitrotoluene
12. 3,5-Dinitrotoluene
13. 3,4-Dinitrotoluene
14. 2,4,6-Trinitrotoluene



Aromatic hydrocarbons



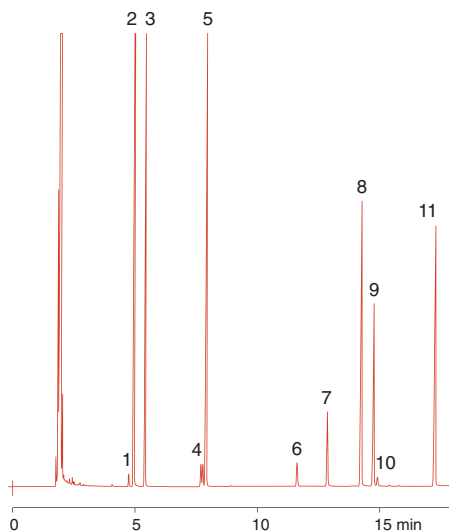
Analysis of nitro- and chloronitro-aromatics

MN Appl. No. 200290

Column: OPTIMA® 1701,
50 m x 0.32 mm ID,
0.25 µm film, REF 726318.50,
max. temperature 300/320 °C
Injection: 1 µl, split 125 ml/min
Carrier gas: 100 kPa H₂
Temperature: 140 °C (6 min) $\xrightarrow{4\text{ °C/min}}$ 190 °C
Detector: FID 250 °C

Peaks:

1. 3-Nitrochlorobenzene
2. 4-Nitrochlorobenzene
3. 2-Nitrochlorobenzene
4. 2,5- and 2,6-Dichloronitrobenzene
5. 2,4-Dichloronitrobenzene
6. 2,6-Dinitrotoluene
7. 2,5-Dinitrotoluene
8. 2,4-Dinitrotoluene
9. 2,3-Dinitrotoluene
10. 3,5-Dinitrotoluene
11. 3,4-Dinitrotoluene



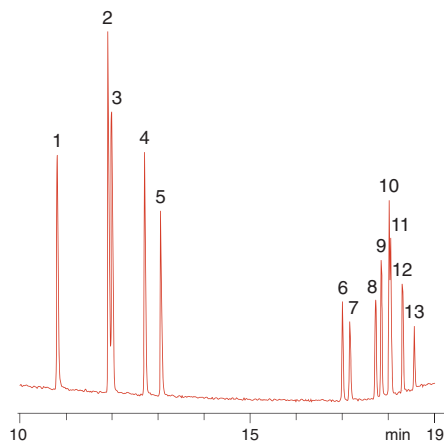
Analysis of nitroaromatic compounds (explosives)

MN Appl. No. 210350

Column: OPTIMA® δ-3,
30 m x 0.25 mm ID,
0.25 µm film, REF 726420.30,
max. temperature 340/360 °C
Injection: 1.0 µl, on column, 250 °C
Carrier gas: 15 psi He
Temperature: 40 °C (1 min) $\xrightarrow{10\text{ °C/min}}$ 250 °C
Detector: MSD

Peaks:

1. Nitrobenzene
2. 2-Nitrotoluene-D₇
3. 2-Nitrotoluene
4. 3-Nitrotoluene
5. 4-Nitrotoluene
6. 2,6-Dinitrotoluene
7. 1,3-Dinitrobenzene
8. 2,5-Dinitrotoluene
9. 2,3-Dinitrotoluene
10. 2,4-Dinitrotoluene-D₃
11. 2,4-Dinitrotoluene
12. 3,5-Dinitrotoluene
13. 3,4-Dinitrotoluene



Courtesy of Mr. Steinbach, Inst. of Organic
Chemistry, Univ. of Marburg, Germany

Environmental pollutants

Analysis of nitromusk compounds in domestic dust

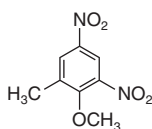
MN Appl. No. 211851

Column: OPTIMA® δ-3, 30 m x 0.25 mm ID, 0.25 μm film, REF 726420.30, max. temperature 340/360 °C
 Sample: 100 μl/l standard with 2-methyl-4,6-dinitroanisole in *n*-hexane
 Injection: 260 °C, split 10 ml/min
 Carrier gas: H₂
 Temperature: 150 °C (5 min) $\xrightarrow{7\text{ °C/min}}$ 290 °C (5 min)
 Detector: ECD 320 °C

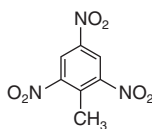
Peaks for applications 211851

– 211853:

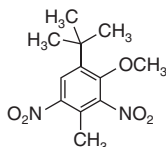
1. 2-Methyl-4,6-dinitroanisole



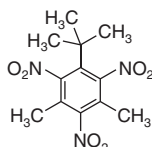
2. 2,4,6-Trinitrotoluene (TNT)



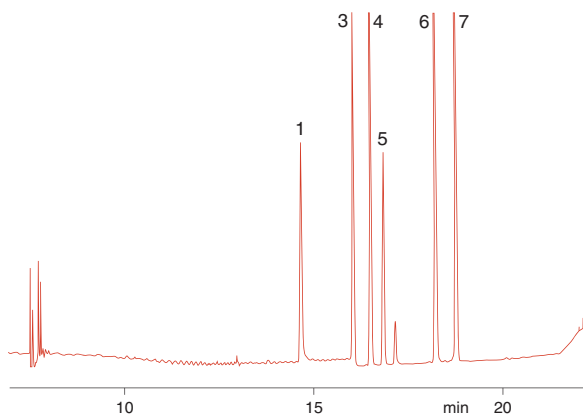
3. Musk ambrette (MA)



4. Musk xylene (MX)

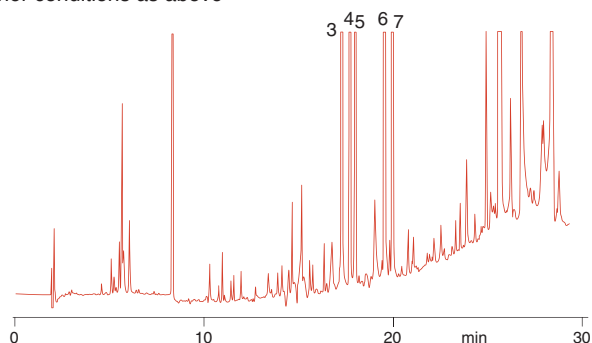


(continued next page)



MN Appl. No. 211852

Column: OPTIMA® δ-6, 30 m x 0.25 mm ID, 0.25 μm film, REF 726470.30, max. temperature 340/360 °C
 Sample: pooled dust sample extracted by steam distillation with *t*-butyl methyl ether, spiked with 50 μg/l standard and 2-methyl-4,6-dinitroanisole
 Other conditions as above





Analysis of nitromusk compounds in domestic dust (cont.)

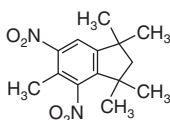
MN Appl. No. 211853

- Samples:
- a) hexane extract of a single dust sample, spiked with 100 $\mu\text{g/l}$ standard and 50 $\mu\text{g/l}$ 2-methyl-4,6-dinitroanisole
 - b) hexane extract of a pooled dust sample, spiked with 50 $\mu\text{g/l}$ standard, 2-methyl-4,6-dinitroanisole and 2,4,6-trinitrotoluene (TNT)
- Other conditions as in application 211851

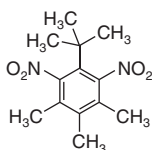
Peaks for applications 211851

– 211853 (cont.):

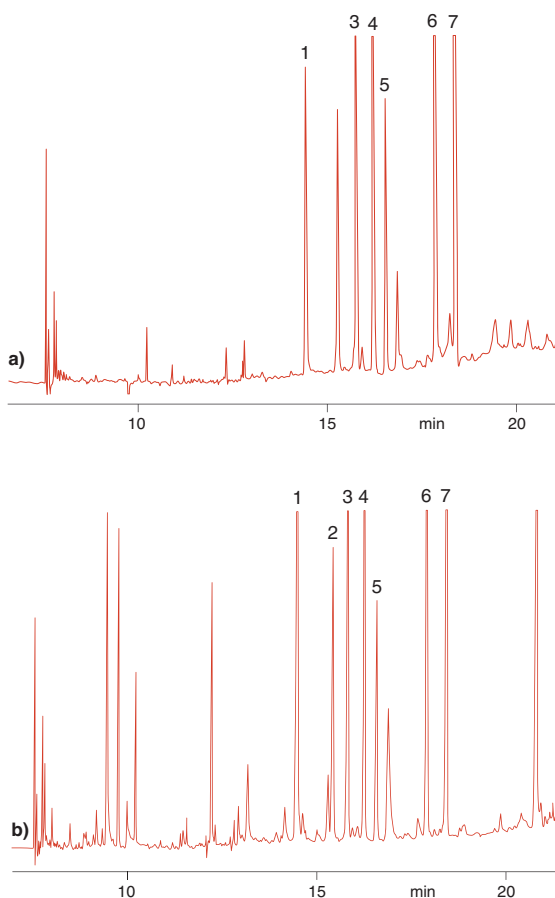
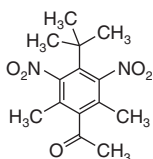
5. Musk moskene (MM)



6. Musk tibetene (MT)



7. Musk ketone (MK)



Courtesy of Sven Heegmann, AK Prof. Butte, Diplomarbeit, University of Oldenburg, Germany

Environmental pollutants

Analysis of nitroaromatics (EPA 609) MN Appl. No. 250450

Column: OPTIMA® δ -6,
50 m x 0.20 mm ID, 0.20 μ m
film, REF 726465.50,
max. temperature 340/360 °C

Sample: EPA 609 Nitroaromatics and
Isophorone Mix, 2000 μ g/ml
per substance in hexane

Injection: 1.0 μ l, split 1:50

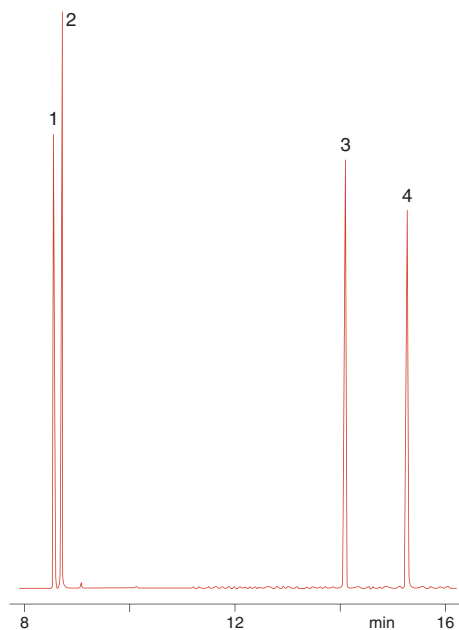
Carrier gas: 2.0 bar He

Temperature: 100 °C (1 min) $\xrightarrow{10\text{ °C/min}}$ 300 °C
(5 min)

Detector: MSD

Peaks:

1. Nitrobenzene
2. Isophorone
3. 2,6-Dinitrotoluene
4. 2,4-Dinitrotoluene



Analysis of nitroaromatics (EPA 609) MN Appl. No. 213200

Column: OPTIMA® 35 MS,
30 m x 0.25 mm ID,
0.25 μ m film, REF 726154.30,
max. temperature 360/370 °C

Injection: 1.0 μ l, split 1:50

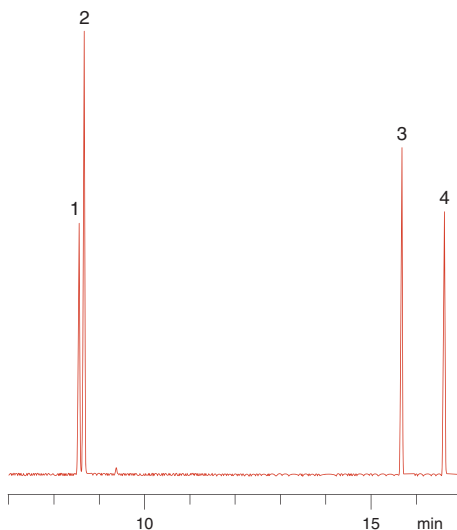
Carrier gas: 1 ml/min He

Temperature: 60 °C $\xrightarrow{8\text{ °C/min}}$ 250 °C

Detector: MSD

Peaks:

1. Nitrobenzene
2. Isophorone
3. 2,6-Dinitrotoluene
4. 2,4-Dinitrotoluene





Analysis of aromatic hydrocarbons

MN Appl. No. 200370

Column: OPTIMA® 210, 50 m x 0.25 mm ID, 0.5 µm film, REF 726874.50,
max. temperature 260/280 °C

Injection: 0.5 µl, split 64 ml/min

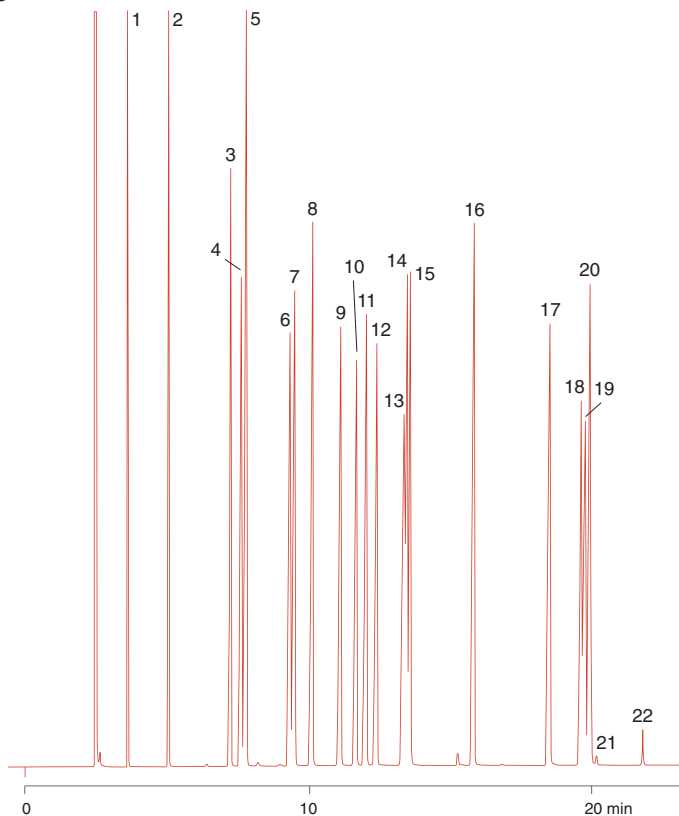
Carrier gas: 120 kPa H₂ (1.5 ml/min)

Temperature: 60 °C (10 min) $\xrightarrow{5\text{ °C/min}}$ 120 °C

Detector: FID 250 °C

Peaks:

1. Benzene
2. Toluene
3. Ethylbenzene
4. *p*-Xylene
5. Chlorobenzene
+ *m*-xylene
6. *o*-Xylene
7. *i*-Propylbenzene
8. Styrene
9. *n*-Propylbenzene
10. Bromobenzene
11. 1,3,5-Trimethylbenzene
12. *o*-Chlorotoluene
13. *m*-Chlorotoluene
14. *p*-Chlorotoluene
15. *p*-Isopropyltoluene
16. *n*-Butylbenzene
17. 2-Methylphenol
18. 2,6-Dichlorotoluene
19. 3-Methylphenol
20. 2,4-Dichlorotoluene
+ 3-methylphenol
21. 2,5-Dichlorotoluene
22. 2,3-Dichlorotoluene



Environmental pollutants

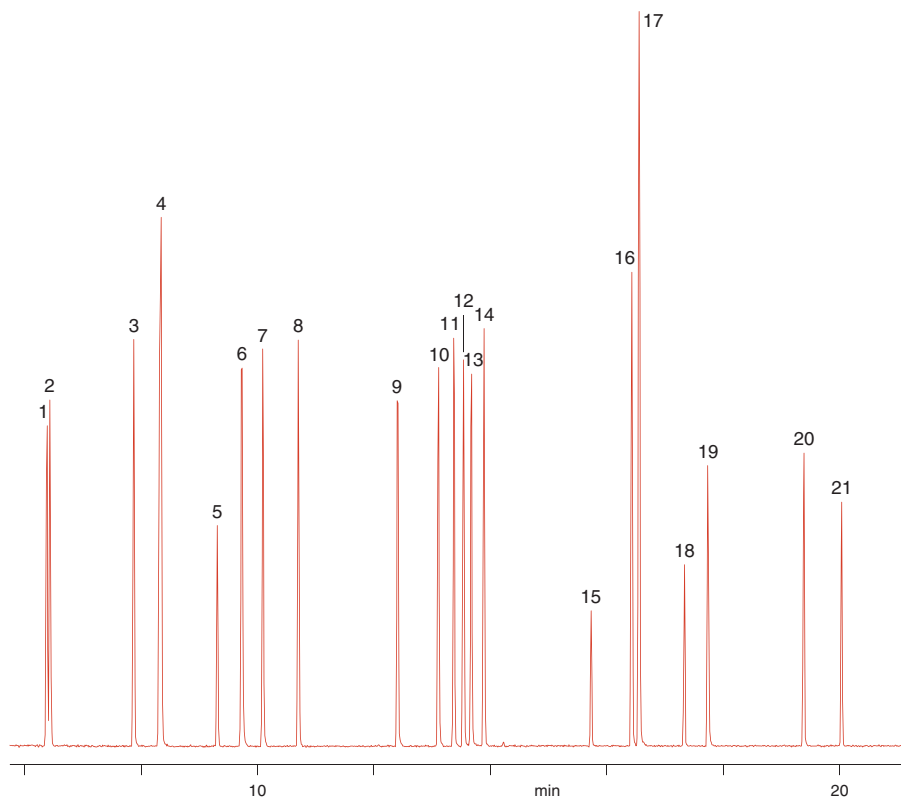
Analysis of phenols (EPA 8040 A)

MN Appl. No. 200410

Column: OPTIMA® 5, 25 m x 0.20 mm ID, 0.2 µm film, REF 726857.25,
max. temperature 340/360 °C
Sample: phenols (EPA 8040 A), 50 µg/ml in 2-propanol
Injection: 1.0 µl, split 1:150
Carrier gas: 25 cm/s He
Temperature: 60 °C (3 min) $\xrightarrow{8\text{ °C/min}}$ 330 °C
Detector: MSD

Peaks:

- | | | |
|-----------------------|----------------------------|-----------------------------------|
| 1. Phenol | 8. 2,6-Dichlorophenol | 15. 2,4-Dinitrophenol |
| 2. 2-Chlorophenol | 9. 4-Chloro-3-methylphenol | 16. 2,3,4,6-Tetrachlorophenol |
| 3. 2-Methylphenol | 10. 2,3,5-Trichlorophenol | 17. 2,3,5,6-Tetrachlorophenol |
| 4. 4-Methylphenol | 11. 2,4,6-Trichlorophenol | 18. 2-Methyl-4,6-dinitrophenol |
| 5. 2-Nitrophenol | 12. 2,4,5-Trichlorophenol | 19. 3,4,5-Trichlorophenol |
| 6. 2,4-Dimethylphenol | 13. 2,3,4-Trichlorophenol | 20. Pentachlorophenol |
| 7. 2,4-Dichlorophenol | 14. 2,3,6-Trichlorophenol | 21. 2-Isopropyl-4,6-dinitrophenol |



Phenols



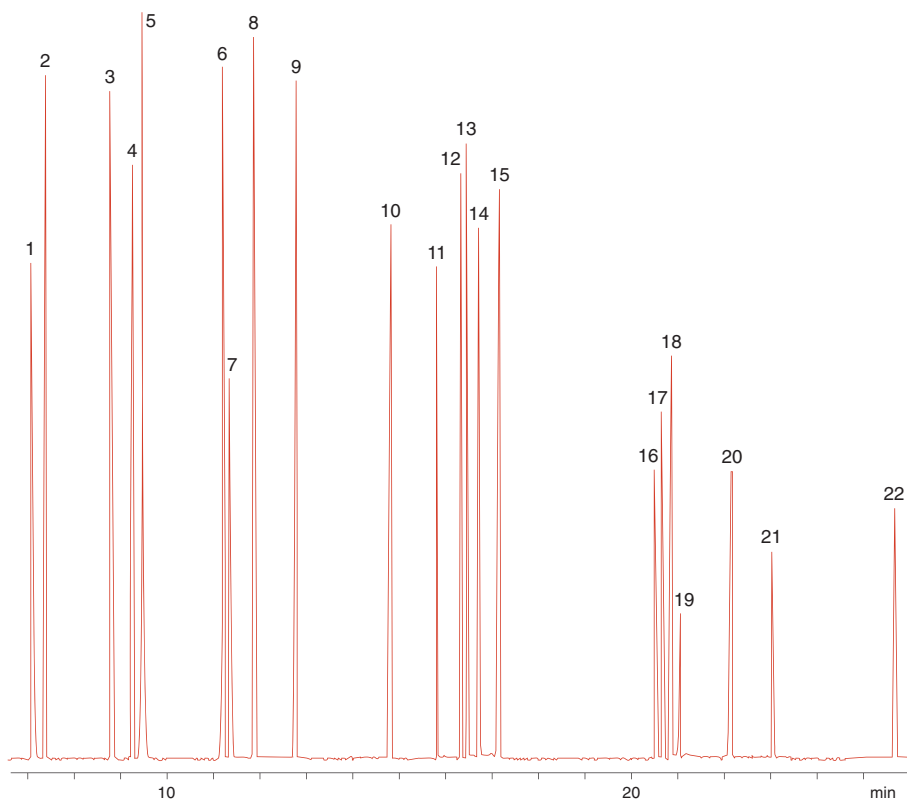
Analysis of isomeric phenols (EPA 8040 A)

MN Appl. No. 250060

Column: OPTIMA® δ -3, 60 m x 0.25 mm ID, 0.25 μ m film, REF 726420.60,
max. temperature 340/360 °C
Injection: 1.0 μ l, split 1:80
Carrier gas: 1.3 bar He
Temperature: 60 °C (3 min) $\xrightarrow{6\text{ }^{\circ}\text{C/min}}$ 320 °C
Detector: MSD

Peaks:

- | | | |
|-----------------------|-----------------------------|-----------------------------------|
| 1. Phenol | 9. 2,6-Dichlorophenol | 16. 2,3,5,6-Tetrachlorophenol |
| 2. 2-Chlorophenol | 10. 4-Chloro-3-methylphenol | 17. 2,3,4,5-Tetrachlorophenol |
| 3. 2-Methylphenol | 11. 2,3,5-Trichlorophenol | 18. 2,3,4,6-Tetrachlorophenol |
| 4. 4-Methylphenol | 12. 2,4,6-Trichlorophenol | 19. 2,4-Dinitrophenol |
| 5. 3-Methylphenol | 13. 2,4,5-Trichlorophenol | 20. 3,4,5-Trichlorophenol |
| 6. 2,4-Dimethylphenol | 14. 2,3,4-Trichlorophenol | 21. 2-Methyl-4,6-dinitrophenol |
| 7. 2-Nitrophenol | 15. 2,3,6-Trichlorophenol | 22. 2-Isopropyl-4,6-dinitrophenol |
| 8. 2,4-Dichlorophenol | | |



Environmental pollutants

Analysis of phenols (EPA 8040)

MN Appl. No. 213240

Column: OPTIMA® 35 MS, 30 m x 0.25 mm ID, 0.25 µm film, REF 726154.30,
max. temperature 360/370 °C

Sample: phenols (EPA 8040)

Injection: 1.0 µl, 280 °C, split 20 ml/min

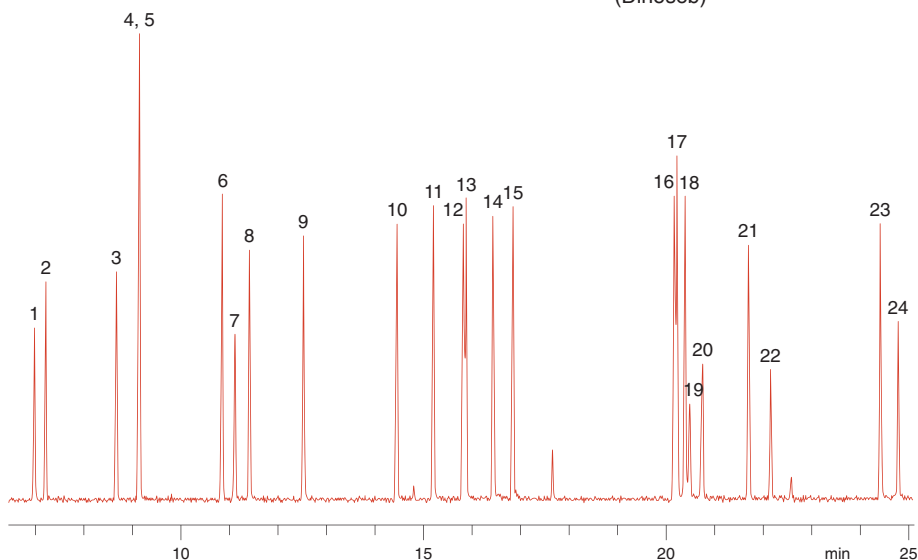
Carrier gas: 0.8 ml/min He

Temperature: 60 °C $\xrightarrow{6\text{ °C/min}}$ 280 °C

Detector: MSD

Peaks:

- | | | |
|-----------------------|-------------------------------|---|
| 1. Phenol | 9. 2,6-Dichlorophenol | 17. 2,3,4,5-Tetrachlorophenol |
| 2. 2-Chlorophenol | 10. 4-Chloro-3-methylphenol | 18. 2,3,4,6-Tetrachlorophenol |
| 3. 2-Methylphenol | 11. 2,3,5-Trichlorophenol | 19. 2,4-Dinitrophenol |
| 4. 3-Methylphenol | 12. 2,4,6-Trichlorophenol | 20. 4-Nitrophenol |
| 5. 4-Methylphenol | 13. 2,4,5-Trichlorophenol | 21. 3,4,5-Trichlorophenol |
| 6. 2,4-Dimethylphenol | 14. 2,3,4-Trichlorophenol | 22. 2-Methyl-4,6-dinitrophenol |
| 7. 2-Nitrophenol | 15. 2,3,6-Trichlorophenol | 23. Pentachlorophenol |
| 8. 2,4-Dichlorophenol | 16. 2,3,5,6-Tetrachlorophenol | 24. Methylpropyl-4,6-dinitrophenol
(Dinoseb) |



Phenols



Analysis of phenols (EPA 528)

MN Appl. No. 212840

Column: OPTIMA® 5 MS Accent, 30 m x 0.25 mm ID, 0.25 µm film, REF 725820.30, max. temperature 340/360 °C

Sample: 1 µl, 5 ppm, 5 ng/compound

Injection: pulsed splitless (hold 0.5 min), pulsed pressure 50 psi (hold 0.5 min)

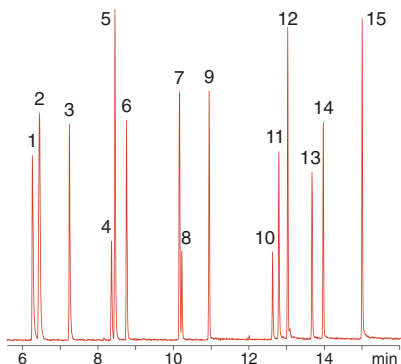
Carrier gas: 1.3 ml/min He

Temperature: 40 °C (1 min) $\xrightarrow{12\text{ °C/min}}$ 200 °C $\xrightarrow{30\text{ °C/min}}$ 300 °C (1 min)

Detector: MSD 280 °C

Peaks:

1. Phenol
2. 2-Chlorophenol-3,4,5,6-d₄ + 2-chlorophenol
3. 2-Methylphenol
4. 2-Nitrophenol
5. 2,4-Dimethylphenol-3,5,6-d₃ + 2,4-dimethylphenol
6. 2,4-Dichlorophenol
7. 4-Chloro-3-methylphenol
8. 1,2-Dimethyl-3-nitrobenzene (int. std. 1)
9. 2,4,6-Trichlorophenol
10. 2,4-Dinitrophenol
11. 4-Nitrophenol
12. 2,3,4,5-Tetrachlorophenol (int. std. 2)
13. 2-Methyl-4,6-dinitrophenol
14. 2,4,6-Tribromophenol (surrogate std.)
15. Pentachlorophenol



Analysis of phenols (EPA 604)

MN Appl. No. 213170

Column: OPTIMA® 35 MS, 30 m x 0.25 mm ID, 0.25 µm film, REF 726154.30, max. temperature 360/370 °C

Injection: 1.0 µl, split 1:20

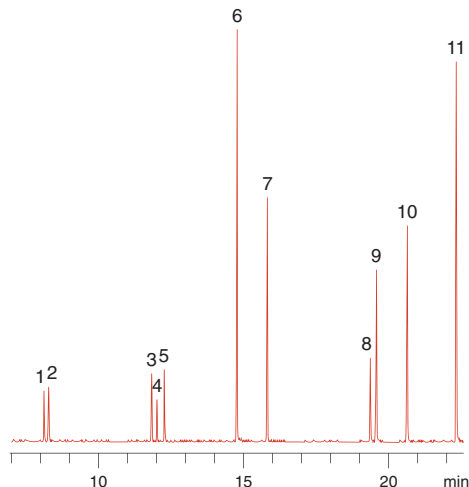
Carrier gas: 1.5 ml/min He

Temperature: 60 °C (4 min) $\xrightarrow{5\text{ °C/min}}$ 230 °C

Detector: MSD

Peaks:

1. Phenol
2. 2-Chlorophenol
3. 2,4-Dimethylphenol
4. 2-Nitrophenol
5. 2,4-Dichlorophenol
6. 4-Chloro-3-methylphenol
7. 2,4,6-Trichlorophenol
8. 2,4-Dinitrophenol
9. 4-Nitrophenol
10. 2-Methyl-4,6-dinitrophenol
11. Pentachlorophenol



Environmental pollutants

Analysis of various phenols

MN Appl. No. 210110

Column: OPTIMA® 5 MS,
30 m x 0.25 mm ID, 0.25 µm film,
REF 726220.30,
max. temperature 340/360 °C

Sample: 5 ppm of each compound except
N-*i*-propylaniline (9.4 ppm)

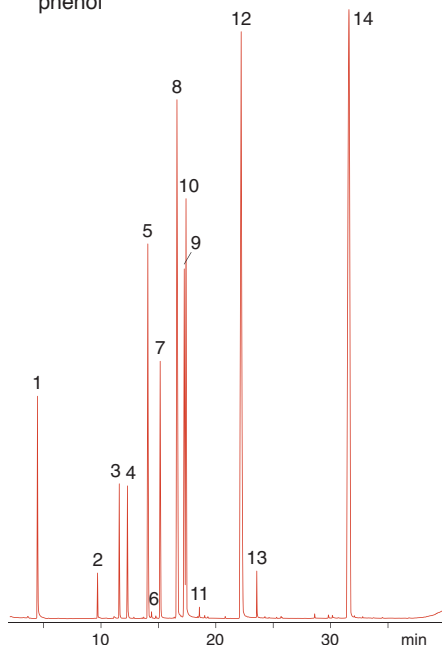
Method: SPME

Temperature: 40 °C (2 min) $\xrightarrow{20\text{ °C/min}}$ 320 °C $\xrightarrow{6\text{ °C/min}}$ 240 °C

Detector: MSD

Peaks:

- | | |
|---------------------------------------|------------------------------|
| 1. Toluene-D ₈ | 9. 3-Bromophenol |
| 2. Phenol | 10. 4-Chloro-3-methyl-phenol |
| 3. 2-Methylphenol (<i>o</i> -Cresol) | 11. 2,4-Dibromophenol |
| 4. Nitrobenzene-D ₅ | 12. 2-Hydroxybiphenyl |
| 5. <i>N</i> - <i>i</i> -Propylaniline | 13. 2-Cyclohexylphenol |
| 6. 2,4-Dichlorophenol | 14. Hexafluorobis-phenol A |
| 7. 4-Chlorophenol | |
| 8. 4-Bromo-2-chloro-phenol | |



Courtesy of Riedel-de-Haën, Seelze, Germany

PAH structures

Naphthalene



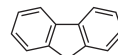
Acenaphthylene



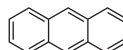
Acenaphthene



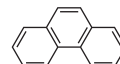
Fluorene



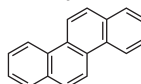
Anthracene



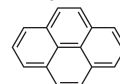
Phenanthrene



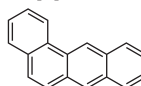
Chrysene



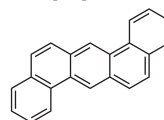
Pyrene



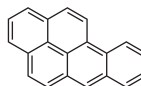
Benz[a]anthracene



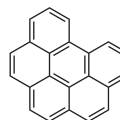
Dibenz[ah]anthracene



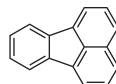
Benzo[a]pyrene



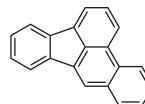
Benzo[ghi]perylene



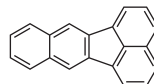
Fluoranthene



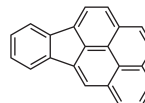
Benzo[b]fluoranthene



Benzo[k]fluoranthene



Indeno[1,2,3-cd]pyrene





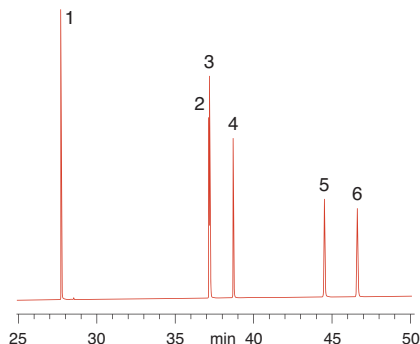
Analysis of 6 PAH according to German drinking water specifications

MN Appl. No. 212540

Column: OPTIMA® δ-6, 50 m x 0.2 mm, 0.2 µm film, REF 726465.50,
max. temperature 340/360 °C
Sample: test mixture acc. to German drinking water specifications (REF 722331),
20 µg/ml in toluene
Injection: 1 µl, 300 °C, splitless
Carrier gas: 0.7 ml/min He
Temperature: 100 °C (1 min) $\xrightarrow{7\text{ °C/min}}$ 340 °C
(15 min)
Detector: MSD 280 °C

Peaks:

1. Fluoranthene
2. Benzo[b]fluoranthene
3. Benzo[k]fluoranthene
4. Benzo[a]pyrene
5. Indeno[1,2,3-cd]pyrene
6. Benzo[ghi]perylene



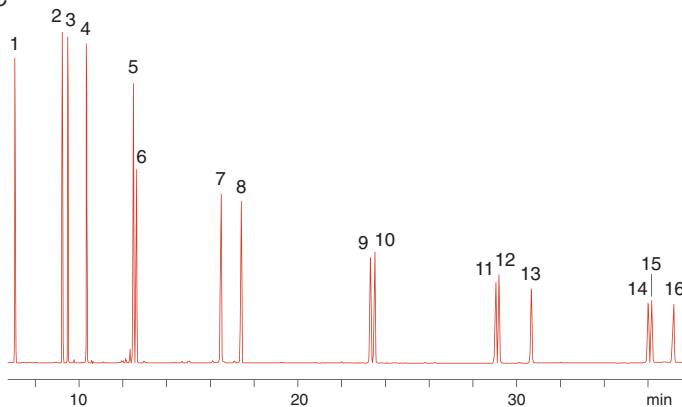
Analysis of 16 PAH (EPA 610)

MN Appl. No. 212800

Column: OPTIMA® 5 MS Accent, 30 m x 0.25 mm ID, 0.25 µm film, REF 725820.30,
max. temperature 340/360 °C
Injection: 0.1 µl (20 µg/ml in toluene), 300 °C, splitless (hold 1 min)
Carrier gas: 40 cm/s H₂
Temperature: 40 °C (1 min) $\xrightarrow{20\text{ °C/min}}$ 200 °C $\xrightarrow{4\text{ °C/min}}$ 310 °C (5 min)
Detector: FID 310 °C

Peaks:

1. Naphthalene
2. Acenaphthylene
3. Acenaphthene
4. Fluorene
5. Phenanthrene
6. Anthracene
7. Fluoranthene
8. Pyrene
9. Benz[a]anthracene
10. Chrysene
11. Benzo[b]fluoranthene
12. Benzo[k]fluoranthene
13. Benzo[a]pyrene
14. Indeno[1,2,3-cd]pyrene
15. Dibenzo[ah]anthracene
16. Benzo[ghi]perylene



Environmental pollutants

Analysis of 16 PAH (EPA 610)

MN Appl. No. 213190

Column: OPTIMA® 35 MS, 30 m x 0.25 mm ID, 0.25 µm film, REF 726154.30,
max temperature 360/370 °C

Injection: 1 µl, split 1:10

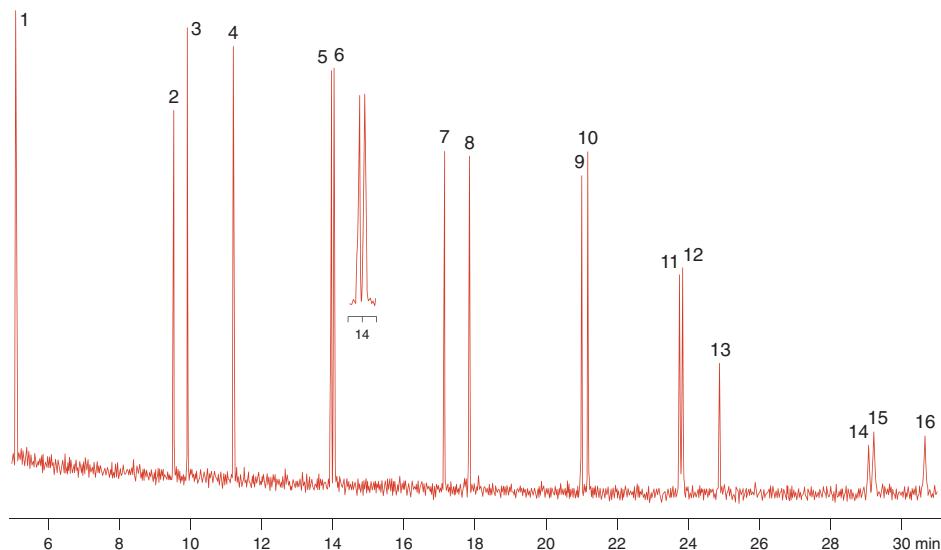
Carrier gas: 0.6 bar H₂

Temperature: 100 °C (3 min) $\xrightarrow{6\text{ °C/min}}$ 300 °C (10 min)

Detector: MSD

Peaks:

- | | | | |
|-------------------|-----------------|--------------------------|----------------------------|
| 1. Naphthalene | 5. Phenanthrene | 9. Benz[a]anthracene | 13. Benzo[a]pyrene |
| 2. Acenaphthylene | 6. Anthracene | 10. Chrysene | 14. Indeno[1,2,3-cd]pyrene |
| 3. Acenaphthene | 7. Fluoranthene | 11. Benzo[b]fluoranthene | 15. Dibenzo[ah]anthracene |
| 4. Fluorene | 8. Pyrene | 12. Benzo[k]fluoranthene | 16. Benzo[ghi]perylene |





Rapid separation of PCB and PAH

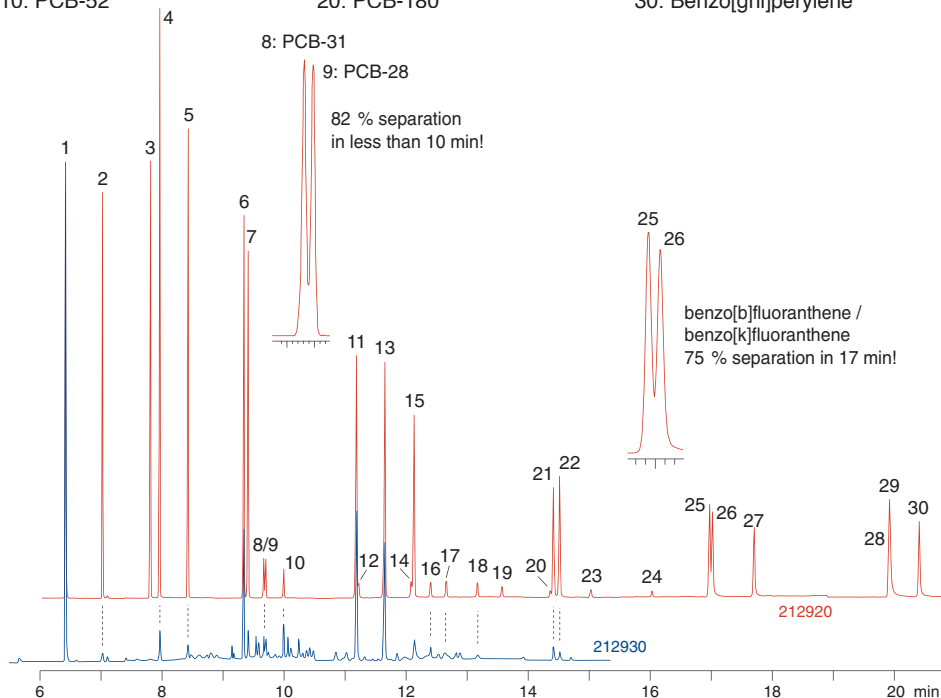
MN Appl. No. 212920/ 212930

Column: OPTIMA® XLB, 30 m x 0.25 mm ID, 0.25 µm film, REF 725850.30
 max. temperature 340/360 °C
 Injection: 1 µl, standard 0.005 ng/µl; 250 °C, pulsed, splitless, pulse 1.38 bar in 1 min
 Purge flow: He with purge flow 60 ml/min
 Temperature: 40 °C (2 min) $\xrightarrow{30\text{ °C/min}}$ 240 °C (2 min) $\xrightarrow{10\text{ °C/min}}$ 340 °C (5 min)
 Detection: MS source 230 °C, interface 280 °C, quadrupol 150 °C

a) application 212920: standard; b) application 212930 PCB and PAH from slag

Peaks:

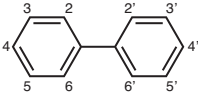
- | | | |
|------------------------|--------------------------|----------------------------|
| 1. Naphthalene | 11. Fluoranthene | 21. Benz[a]anthracene |
| 2. 2-Methylnaphthalene | 12. PCB-101 | 22. Chrysene |
| 3. Acenaphthylene | 13. Pyrene | 23. PCB-169 |
| 4. Acenaphthene | 14. PCB-77 | 24. PCB-194 |
| 5. Fluorene | 15. 2-Methylfluoranthene | 25. Benzo[b]fluoranthene |
| 6. Phenanthrene | 16. PCB-118 | 26. Benzo[k]fluoranthene |
| 7. Anthracene | 17. PCB-153 | 27. Benzo[a]pyrene |
| 8. PCB-31 | 18. PCB-138 | 28. Dibenz[ah]anthracene |
| 9. PCB-28 | 19. PCB-126 | 29. Indeno[1,2,3-cd]pyrene |
| 10. PCB-52 | 20. PCB-180 | 30. Benzo[ghi]perylene |



Courtesy of Centre d'Analyses de Recherche, Lab. d'Hydrologie, Illkirch, France

Environmental pollutants

PCB nomenclature

BZ	
	
1	2-Chlorobiphenyl
5	2,3-Dichlorobiphenyl
18	2,2',5-Trichlorobiphenyl
20	2,3,3'-Trichlorobiphenyl
28	2,4,4'-Trichlorobiphenyl
29	2,4,5-Trichlorobiphenyl
31	2,4',5-Trichlorobiphenyl
44	2,2',3,5'-Tetrachlorobiphenyl
47	2,2',4,4'-Tetrachlorobiphenyl
49	2,2',4,5'-Tetrachlorobiphenyl
52	2,2',5,5'-Tetrachlorobiphenyl
61	2,3,4,5-Tetrachlorobiphenyl
66	2,3',4,4'-Tetrachlorobiphenyl
70	2,3',4',5-Tetrachlorobiphenyl
98	2,2',3',4,6-Pentachlorobiphenyl
99	2,2',4,4',5-Pentachlorobiphenyl
101	2,2',4,5,5'-Pentachlorobiphenyl
103	2,2',4,5',6-Pentachlorobiphenyl
105	2,3,3',4,4'-Pentachlorobiphenyl
114	2,3,4,4',5-Pentachlorobiphenyl
118	2,3',4,4',5-Pentachlorobiphenyl
128	2,2',3,3',4,4'-Hexachlorobiphenyl
138	2,2',3,4,4',5'-Hexachlorobiphenyl
149	2,2',3,4',5',6-Hexachlorobiphenyl
153	2,2',4,4',5,5'-Hexachlorobiphenyl
154	2,2',4,4',5,6'-Hexachlorobiphenyl
156	2,3,3',4,4',5-Hexachlorobiphenyl
170	2,2',3,3',4,4',5-Heptachlorobiphenyl
171	2,2',3,3',4,4',6-Heptachlorobiphenyl
180	2,2',3,4,4',5,5'-Heptachlorobiphenyl
183	2,2',3,4,4',5',6-Heptachlorobiphenyl
189	2,3,3',4,4',5,5'-Heptachlorobiphenyl
194	2,2',3,3',4,4',5,5'-Octachlorobiphenyl
201	2,2',3,3',4,5',6,6'-Octachlorobiphenyl
209	Decachlorobiphenyl

Analysis of PCB (Aroclor 1248)

MN Appl. No. 210510

Column: OPTIMA® δ-3,
50 m x 0.20 mm ID, 0.20 μm
film, REF 726400.50,
max. temperature 340/360 °C

Injection: 1.0 μl, split 1:25

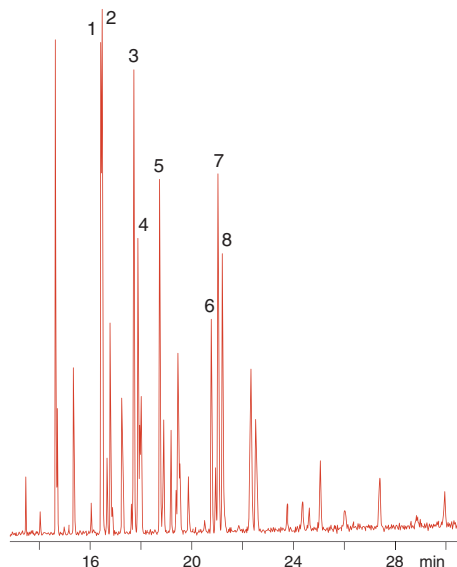
Carrier gas: 25 cm/s He

Temperature: 100 °C $\xrightarrow{12\text{ °C/min}}$ 220 °C
 $\xrightarrow{1\text{ °C/min}}$ 260 °C

Detector: MSD

Peaks:

1. PCB-31
2. PCB-28
3. PCB-52
4. PCB-49
5. PCB-44
6. PCB-61
7. PCB-70
8. PCB-66





Analysis of PCB (W22 congener mix)

MN Appl. No. 211180

Column: OPTIMA® 5 MS,
50 m x 0.20 mm ID,
0.2 µm film, REF 726210.50,
max. temperature: 340/360 °C

Injection: 1 µl PCB W22 congener mix,
10 µl/ml, split 80 ml/min

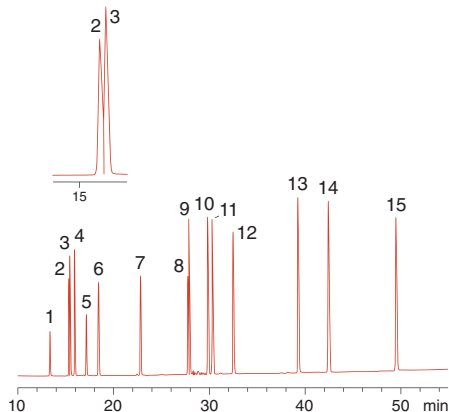
Carrier gas: 0.5 bar He

Temperature: 220 °C $\xrightarrow{1.5\text{ °C/min}}$ 300 °C (15 min)

Detector: ECD 300 °C

Peaks:

- | | |
|------------|-------------|
| 1. PCB-18 | 9. PCB-118 |
| 2. PCB-31 | 10. PCB-153 |
| 3. PCB-28 | 11. PCB-105 |
| 4. PCB-20 | 12. PCB-138 |
| 5. PCB-52 | 13. PCB-180 |
| 6. PCB-44 | 14. PCB-170 |
| 7. PCB-101 | 15. PCB-194 |
| 8. PCB-149 | |



Analysis of PCB

MN Appl. No. 210770

Column: OPTIMA® 8-3,
60 m x 0.25 mm ID, 0.25 µm
film, REF 726420.60,
max. temperature 340/360 °C

Injection: 1 µl, splitless, 36 pg/45 pg,
depending on component

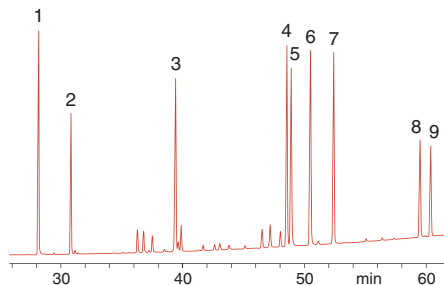
Carrier gas: 2.6 ml/min He

Temperature: 60 °C (1 min) $\xrightarrow{20\text{ °C/min}}$ 200 °C
 $\xrightarrow{1.5\text{ °C/min}}$ 290 °C (3 min)

Detector: ECD 300 °C,
25 ml/min Ar/CH₄ 5%

Peaks:

1. PCB-28
2. PCB-52
3. PCB-101
4. PCB-114
5. PCB-153
6. PCB-105
7. PCB-138
8. PCB-156
9. PCB-180

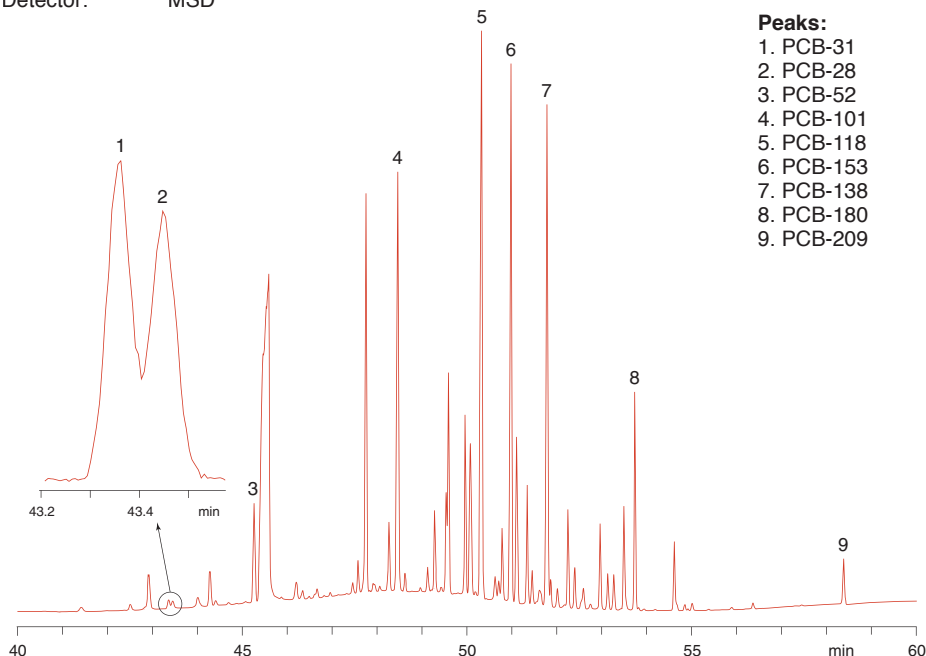


Courtesy of Mr. Bauer, Mr. Drichelt, Zentr. Inst.
d. San.dienstes BW, Kiel, Germany

Environmental pollutants

Analysis of PCB from window putty MN Appl. No. 212071

Column: OPTIMA® 5 MS, 50 m x 0.20 mm ID, 0.2 µm film, REF 726210.50,
max. temperature 340/360 °C
Injector temp.: 200 °C
Carrier gas: 32 psi He
Temperature: 120 °C (1 min) $\xrightarrow{10\text{ °C/min}}$ 200 °C $\xrightarrow{15\text{ °C/min}}$ 290 °C (5 min) $\xrightarrow{20\text{ °C/min}}$ 320 °C
Detector: MSD



Courtesy of N. Bertram, LTA Labor f. Toxikol. u. Analytik, Königswinter, Germany



Analysis of polychlorinated dibenzodioxins (PCDDs)

MN Appl. No. 210180

Column: OPTIMA® δ-3,
60 m x 0.32 mm ID,
0.25 µm film, REF 726440.60,
max. temperature 340/360 °C

Sample: ¹³C-labelled standards

Injection: 1.0 µl, splitless

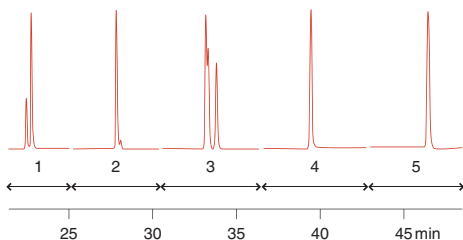
Carrier gas: He

Temperature: 200 °C $\xrightarrow{4\text{ °C/min}}$ 280 °C
 $\xrightarrow{2\text{ °C/min}}$ 320 °C (10 min)

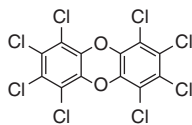
Detector: MSD

Peaks:

	m/z
1. Tetrachlorodibenzodioxins	333.9
2. Pentachlorodibenzodioxins	367.9
3. Hexachlorodibenzodioxins	401.9
4. Heptachlorodibenzodioxins	435.8
5. Octachlorodibenzodioxin	471.8



Courtesy of Dr. Klostermann, Mr. Ludwig, SGS Intercontrol, Wismar, Germany



Octachlorodibenzodioxin

Analysis of polychlorinated dibenzofurans (PCDFs) in dust

MN Appl. No. 250520

Column: OPTIMA® δ-3,
60 m x 0.25 mm ID,
0.25 µm film, REF 726420.60
max. temperature 340/360 °C

Injection: 2.0 µl, 6 – 160 pg/µl depending on component

Carrier gas: 1.4 ml/min He

Temperature: 120 °C (1 min) $\xrightarrow{25\text{ °C/min}}$ 220 °C
 $\xrightarrow{3\text{ °C/min}}$ 300 °C (25 min)

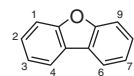
Detector: MSD

Peaks:

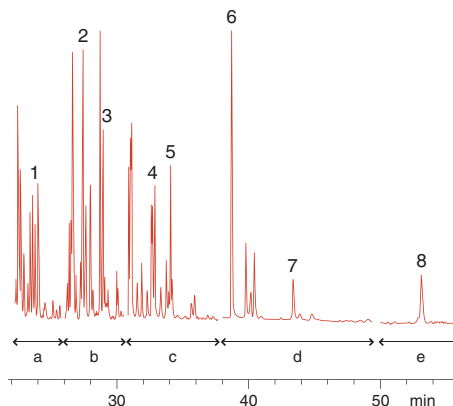
- 2,3,7,8-Tetrachlorodibenzofuran
- 1,2,3,7,8-Pentachlorodibenzofuran
- 2,3,4,7,8-Pentachlorodibenzofuran
- 1,2,3,4,7,8-Hexachlorodibenzofuran
+ 1,2,3,6,7,8-hexachlorodibenzofuran
- 2,3,4,6,7,8-Hexachlorodibenzofuran
- 1,2,3,4,6,7,8-Heptachlorodibenzofuran
- 1,2,3,4,7,8,9-Heptachlorodibenzofuran
- Octachlorodibenzofuran

m/z:

- 307
- 341
- 375
- 409
- 443



Dibenzofuran



Courtesy of A. Dockwiller, Tredi, Strassbourg, France

Environmental pollutants

Analysis of halogenated hydrocarbons, organochlorine pesticides and PCB

MN Appl. No. 200530

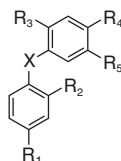
Column: OPTIMA® 1301, 25 m x 0.32 mm ID, 0.25 µm film, REF 726777.25,
max. temperature 300/320 °C
Injection: 0.8 µl on column
Carrier gas: 50 kPa H₂
Temperature: 80 °C $\xrightarrow{30\text{ °C/min}}$ 200 °C $\xrightarrow{4\text{ °C/min}}$ 270 °C
Detector: ECD 310 °C

Peaks: [pg]

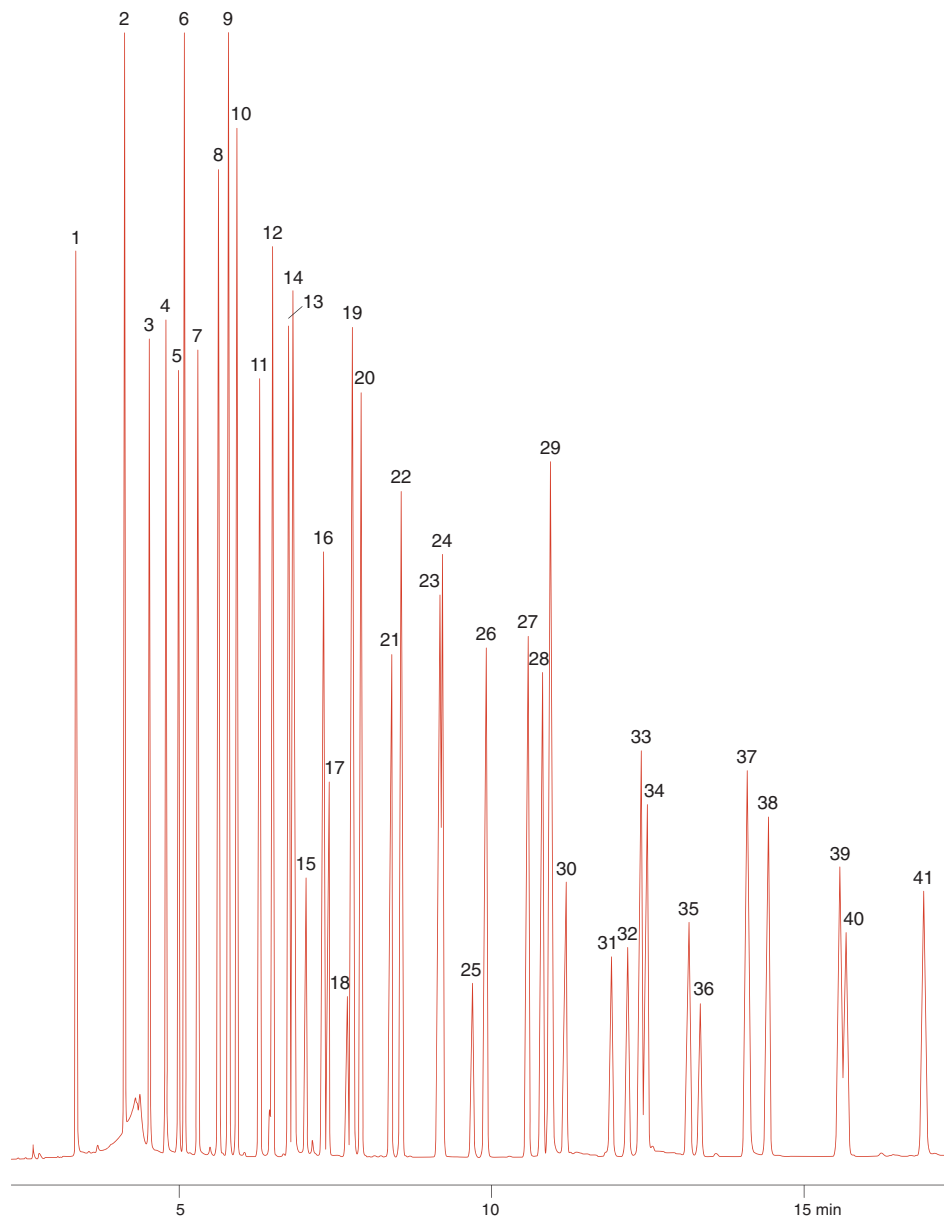
- | | | |
|---|--|--------------------------------|
| 1. Hexachlorocyclopentadiene [160] | 14. 2,3,4,5-Tetrachloroaniline [160] | 27. <i>p,p'</i> -DDE [80] |
| 2. Etridiazole [160] | 15. PCB-28 [80] | 28. Dieldrin [80] |
| 3. α -2,4-Trichloroacetophenone [80] | 16. Heptachlor epoxide [80] | 29. Chlorfenson [160] |
| 4. 2,4,5,6-Tetrachloroxylene [80] | 17. β -BHC [80] | 30. <i>o,p'</i> -DDD [80] |
| 5. Tecnazene [80] | 18. PCB-52 [80] | 31. <i>o,p'</i> -DDT [80] |
| 6. 2,3,5,6-Tetrachloroaniline [240] | 19. Methyl pentachlorophenyl sulphide [80] | 32. PCB-153 [80] |
| 7. Trifluralin [320] | 20. δ -BHC [80] | 33. Endosulfan β [80] |
| 8. HCB [80] | 21. Telodrin [80] | 34. <i>p,p'</i> -DDD [80] |
| 9. Pentachloroanisole [80] | 22. Octachlorostyrene [80] | 35. PCB-138 [80] |
| 10. α -BHC [80] | 23. <i>cis</i> -Heptachlor epoxide [80] | 36. <i>p,p'</i> -DDT [80] |
| 11. Quintozene [80] | 24. Alodan (int. std.) [80] | 37. Chlorbenside sulphone [80] |
| 12. Lindane [80] | 25. PCB-101 [80] | 38. Endosulfan sulphate [80] |
| 13. Bromocyclen [80] | 26. Endosulfan α [80] | 39. PCB-180 [80] |
| | | 40. Mirex [80] |
| | | 41. Tetradifon [80] |

Pesticide structures: bridged diphenyl derivatives

Structure	Compound	X	R ₁	R ₂	R ₃	R ₄	R ₅
	Bifenox	—O—	Cl	Cl	H	NO ₂	CO-O-CH ₃
	Fluorodifen	—O—	CF ₃	NO ₂	H	NO ₂	H
	Tetrasul	—S—	Cl	H	Cl	Cl	Cl
	Tetradifon	—SO ₂ —	Cl	H	Cl	Cl	Cl
	Chlorbenside sulphone	—CH ₂ -SO ₂ —	Cl	H	H	Cl	H
	Chlorfenson	—O-SO ₂ —	Cl	H	H	Cl	H
	<i>o,p'</i> -DDD	>CH-CHCl_2	H	Cl	H	Cl	H
	<i>p,p'</i> -DDD	>CH-CHCl_2	Cl	H	H	Cl	H
	<i>o,p'</i> -DDE	>C=CCl_2	H	Cl	H	Cl	H
	<i>p,p'</i> -DDE	>C=CCl_2	Cl	H	H	Cl	H
	<i>o,p'</i> -DDT	>CH-CCl_3	H	Cl	H	Cl	H
	<i>p,p'</i> -DDT	>CH-CCl_3	Cl	H	H	Cl	H
	<i>p,p'</i> -Methoxychlor	>CH-CCl_3	OCH ₃	H	H	OCH ₃	H
	Chlorobenzilate	$\text{>C(OH)-CO-O-C}_2\text{H}_5$	Cl	H	H	Cl	H
	Bromopropylate	$\text{>C(OH)-CO-O-}i\text{-C}_3\text{H}_7$	Br	H	H	Br	H
	Fenarimol		Cl	H	Cl	H	H
	Nuarimol		F	H	Cl	H	H



Halogenated hydrocarbons • Pesticides



Courtesy of Mr. Lembacher, Hipp N hrmittel, Pfaffenhofen, Germany

Environmental pollutants

Analysis of pesticides and PCB

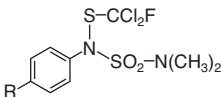
MN Appl. No. 210690

Column: OPTIMA® δ-6, 50 m x 0.20 mm ID, 0.20 µm film, REF 726465.50,
max. temperature 340/360 °C
Injection: 0.03 – 0.5 µg/ml depending on component, splitless, purge time 0.8 min, 270 °C
Carrier gas: 2.3 ml/min He
Temperature: 130 °C (1 min) $\xrightarrow{25\text{ °C/min}}$ 180 °C (5 min) $\xrightarrow{2\text{ °C/min}}$ 260 °C $\xrightarrow{6\text{ °C/min}}$ 310 °C (14 min)
Detector: ECD 300 °C

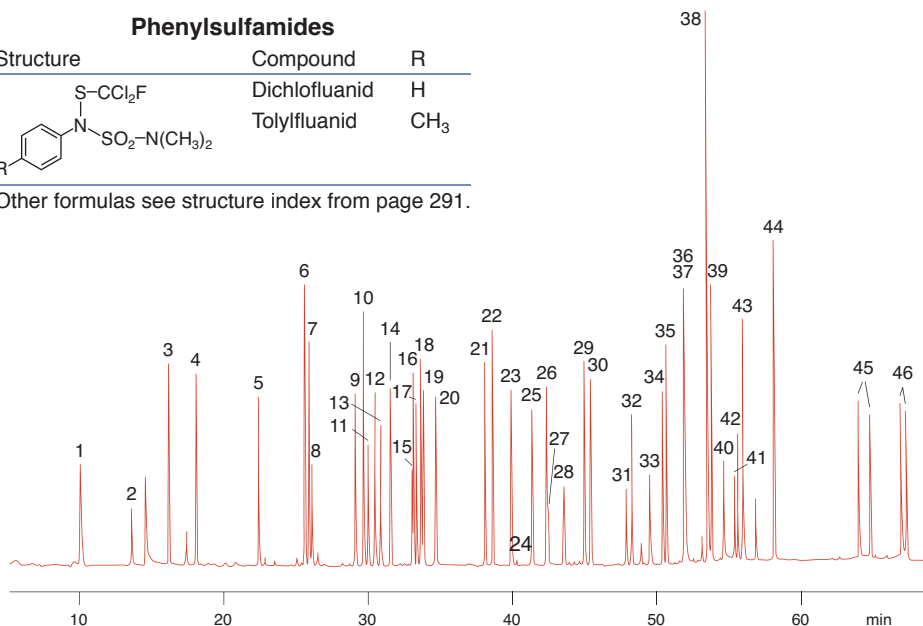
Peaks:

- | | | | |
|-------------------------|------------------------|------------------|-------------------------------|
| 1. Dichlobenil | 13. S 421 | 25. PCB-101 | 37. PCB-138 |
| 2. Chloroneb | 14. δ-BHC | 26. α-Endosulfan | 38. Nuarimol |
| 3. Trifluralin | 15. PCB-52 | 27. Captan | 39. Endosulfan sulphate |
| 4. Tecnazene | 16. Aldrin | 28. Folpet | 40. Bromopropylate |
| 5. α-BHC | 17. Chlorothalonil | 29. Dieldrin | 41. Captafol |
| 6. Dichloran | 18. Chlorpyrifos-ethyl | 30. o,p'-DDD | 42. Methoxychlor |
| 7. γ-BHC | 19. Dichlofluanid | 31. Endrin | 43. PCB-180 |
| 8. Quintozene | 20. Triadimefon | 32. o,p'-DDT | 44. Tetradifon |
| 9. β-BHC | 21. Fluorochloridone | 33. PCB-153 | 45. Flucythrinate (2 isomers) |
| 10. Vinclozolin | 22. Tolyfluanid | 34. Tetrasul | 46. Fluvalinate (2 isomers) |
| 11. PCB-28 + heptachlor | 23. Procymidone | 35. β-Endosulfan | |
| 12. p-Chloroaniline | 24. Anilazine | 36. p,p'-DDT | |

Phenylsulfamides

Structure	Compound	R
	Dichlofluanid	H
	Tolyfluanid	CH ₃

Other formulas see structure index from page 291.



Courtesy of Mr. Grabher, CLUA, Sigmaringen, Germany

Chlorinated hydrocarbons • Pesticides



Analysis of chlorinated hydrocarbons, organochlorine pesticides and PCB acc. to German Health Administration (BGA 1988)

MN Appl. No. 200700

Column: OPTIMA® 5, 60 m x 0.25 mm ID, 0.25 µm film, REF 726056.60,
max. temperature 340/360 °C

Split: 64 ml/min

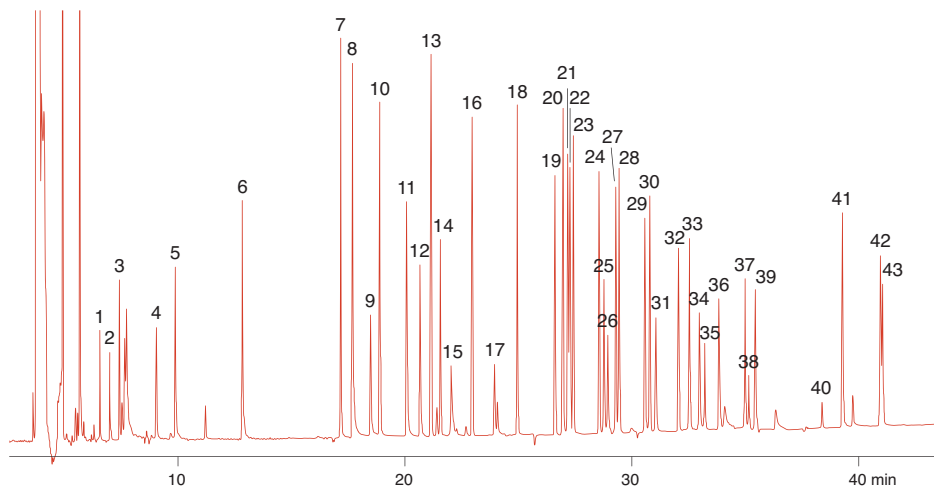
Carrier gas: 160 kPa H₂ (2.3 ml/min)

Temperature: 60 °C $\xrightarrow{40\text{ °C/min}}$ 150 °C (1 min) $\xrightarrow{3.5\text{ °C/min}}$ 300 °C (5 min)

Detector: ECD 320 °C

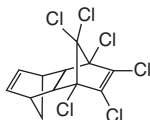
Peaks:

- | | | | |
|-------------------------------|------------------------------------|--------------------------------------|-------------------------|
| 1. 1,2,3-Trichlorobenzene | 13. Bromocyclen | 23. <i>trans</i> -Heptachlor epoxide | 34. <i>p,p'</i> -DDD |
| 2. 1,2,4-Trichlorobenzene | 14. Musk xylene | | 35. <i>o,p'</i> -DDT |
| 3. 1,3,5-Trichlorobenzene | 15. PCB-28 | 24. <i>trans</i> -Chlordane | 36. PCB-153 |
| 4. 1,2,4,5-Tetrachlorobenzene | 16. Heptachlor | 25. <i>o,p'</i> -DDE | 37. Endosulfan sulphate |
| 5. 1,2,3,4-Tetrachlorobenzene | 17. PCB-52 | 26. PCB-101 | 38. <i>p,p'</i> -DDT |
| 6. Pentachlorobenzene | 18. Aldrin | 27. Endosulfan α | 39. PCB-138 |
| 7. α-BHC | 19. Isodrin (int. std.) | 28. <i>cis</i> -Chlordane | 40. Methoxychlor |
| 8. HCB | 20. Octachlorostyrene | 29. <i>p,p'</i> -DDE | 41. PCB-180 |
| 9. β-BHC | 21. <i>cis</i> -Heptachlor epoxide | 30. Dieldrin | 42. PCB-170 |
| 10. γ-BHC (lindane) | 22. Oxychlordane | 31. <i>o,p'</i> -DDD | 43. Mirex |
| 11. δ-BHC | | 32. Endrin | |
| 12. ε-BHC | | 33. Endosulfan β | |

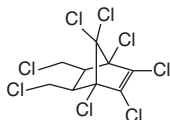


Organochlorine pesticide structures

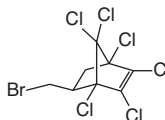
Aldrin



Alodan



Bromocyclen

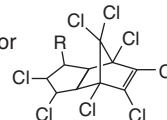


Chlordane

R = H

Nonachlor

R = Cl



Environmental pollutants

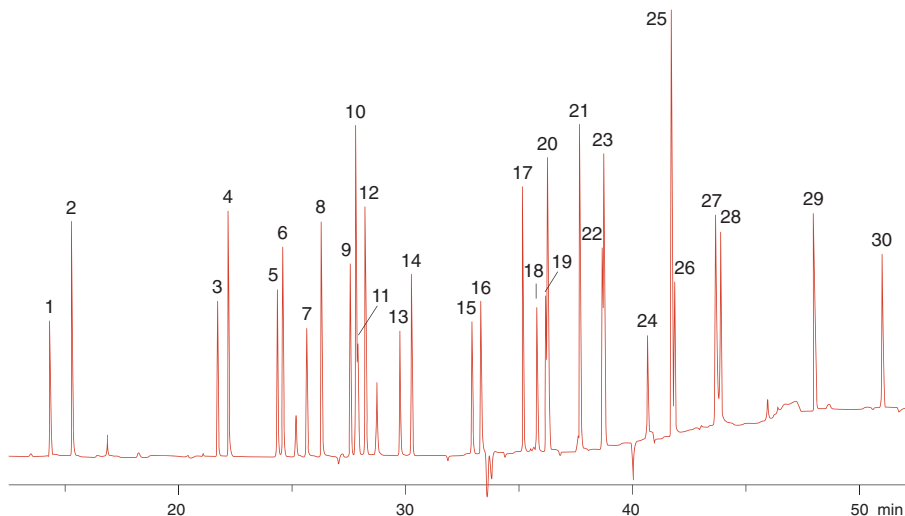
Analysis of organochlorine pesticides and PCB

MN Appl. No. 210230

Column: OPTIMA® δ -3, 50 m x 0.20 mm ID, 0.20 μ m film, REF 726400.50,
max. temperature 340/360 °C
Injection: 0.01 – 0.15 μ g/ml depending on component, splitless, purge time 0.5 min
Carrier gas: 1.37 ml/min He
Temperature: 130 °C (1 min) $\xrightarrow{25\text{ °C/min}}$ 180 °C (5 min) $\xrightarrow{3\text{ °C/min}}$ 260 °C (2 min) $\xrightarrow{4\text{ °C/min}}$ 300 °C
(20 min)
Detector: ECD

Peaks:

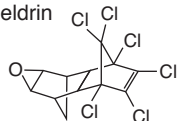
- | | | | |
|-----------------------|-----------------------------|-----------------------------|----------------------|
| 1. Chloroneb | 9. PCB-28 | 17. <i>o,p'</i> -DDE | 25. <i>p,p'</i> -DDD |
| 2. Pentachlorobenzene | 10. <i>p</i> -Chloroaniline | 18. PCB-101 | 26. PCB-153 |
| 3. α -BHC | 11. Heptachlor | 19. <i>trans</i> -Nonachlor | 27. <i>p,p'</i> -DDT |
| 4. HCB | 12. δ -BHC | 20. <i>cis</i> -Chlordane | 28. PCB-138 |
| 5. γ -BHC | 13. PCB-52 | 21. <i>p,p'</i> -DDE | 29. PCB-180 |
| 6. Quintozene | 14. Aldrin | 22. Dieldrin | 30. Mirex |
| 7. Bromocyclen | 15. Isodrin | 23. <i>o,p'</i> -DDD | |
| 8. β -BHC | 16. Oxychlordane | 24. Endrin | |



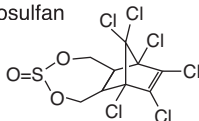
Courtesy of Mr. Grabher, CLUA, Sigmaringen, Germany

Organochlorine pesticide structures

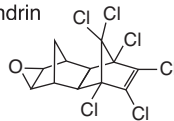
Dieldrin



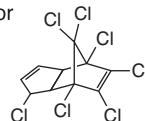
Endosulfan



Endrin



Heptachlor





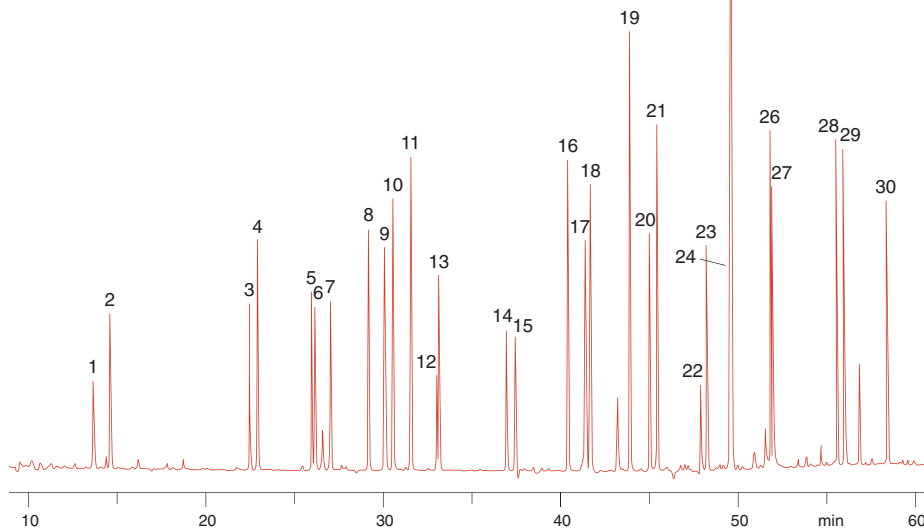
Analysis of organochlorine pesticides and PCB

MN Appl. No. 210700

Column: OPTIMA® δ -6, 50 m x 0.20 mm ID, 0.20 μ m film, REF 726465.50,
max. temperature 340/360 °C
Injection: 0.03 – 0.5 μ g/ml depending on component, splitless, purge time 0.8 min, 270 °C
Carrier gas: 2.3 ml/min He
Temperature: 130 °C (1 min) $\xrightarrow{25\text{ }^{\circ}\text{C/min}}$ 180 °C (5 min) $\xrightarrow{2\text{ }^{\circ}\text{C/min}}$ 260 °C $\xrightarrow{6\text{ }^{\circ}\text{C/min}}$ 310 °C (14 min)
Detector: ECD 300 °C

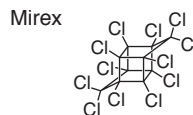
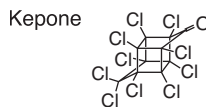
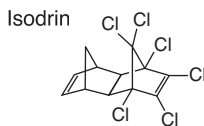
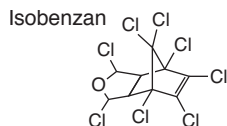
Peaks:

- | | | |
|-----------------------------|---------------------------|----------------------|
| 1. Chloroneb | 11. δ -BHC | 21. <i>o,p'</i> -DDD |
| 2. Pentachlorobenzene | 12. PCB-52 | 22. Endrin |
| 3. α -BHC | 13. Aldrin | 23. <i>o,p'</i> -DDT |
| 4. HCB | 14. Isodrin | 24. PCB-153 |
| 5. γ -BHC | 15. Oxychlorane | 25. <i>p,p'</i> -DDD |
| 6. Quintozene | 16. <i>o,p'</i> -DDE | 26. <i>p,p'</i> -DDT |
| 7. Bromocyclen | 17. PCB-101 | 27. PCB-138 |
| 8. β -BHC | 18. <i>cis</i> -Chlordane | 28. Methoxychlor |
| 9. PCB-28 + heptachlor | 19. <i>p,p'</i> -DDE | 29. PCB-180 |
| 10. <i>p</i> -Chloroaniline | 20. Dieldrin | 30. Mirex |



Courtesy of Mr. Grabher, CLUA, Sigmaringen, Germany

Organochlorine pesticide structures



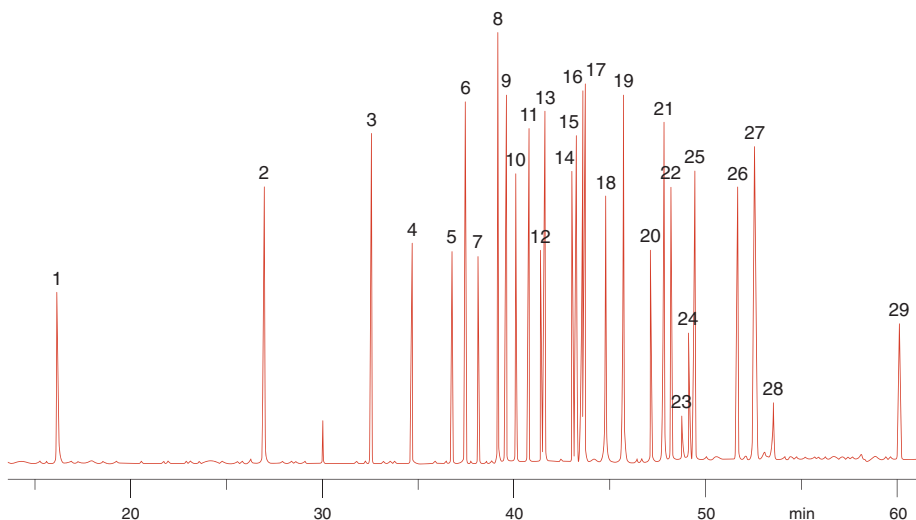
Environmental pollutants

Analysis of halogenated hydrocarbons, organochlorine pesticides and PCB MN Appl. No. 250220

Column: OPTIMA® 1701, 50 m x 0.20 mm ID, 0.20 µm film, REF 726841.50,
max. temperature 300/320 °C
Injection: accu-pesticide test mixture, 20 µg/ml, split 1:10
Carrier gas: 1.1 bar He
Temperature: 100 °C (3 min) $\xrightarrow{4\text{ °C/min}}$ 250 °C, then isothermal
Detector: MSD EI

Peaks:

1. Hexachlorobutadiene	11. Isobenzan	21. <i>o,p'</i> -DDD
2. Pentachlorobenzene	12. δ -BHC	22. PCB-118
3. Hexachlorobenzene	13. Isodrin	23. Endrin
4. α -BHC	14. Heptachlor epoxide A	24. <i>p,p'</i> -DDD
5. β -BHC	15. Heptachlor epoxide B	25. PCB-138
6. PCB-28	16. PCB-101	26. <i>o,p'</i> -DDT
7. Heptachlor	17. <i>o,p'</i> -DDE	27. PCB-153 + endosulfan 2
8. PCB-52	18. Endosulfan 1	28. <i>p,p'</i> -DDT
9. Aldrin	19. <i>p,p'</i> -DDE	29. PCB-180
10. γ -BHC	20. Dieldrin	





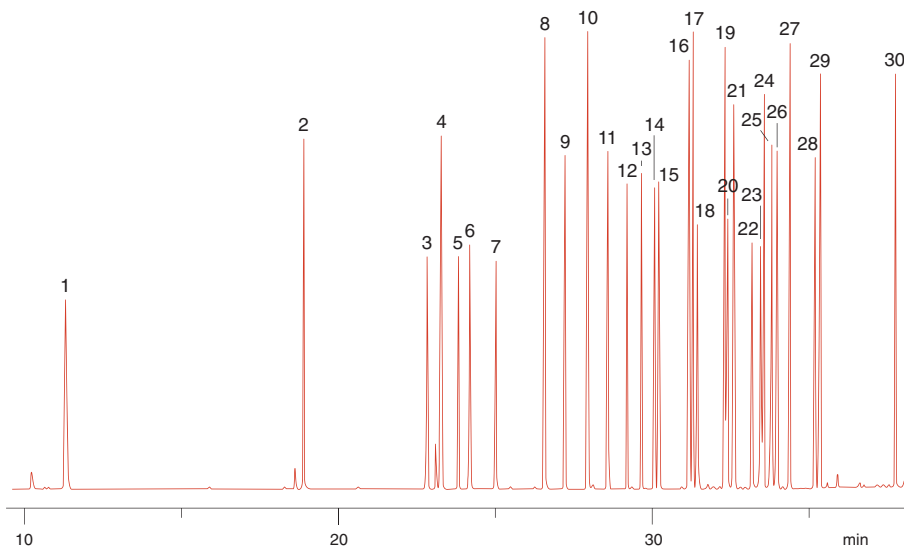
Analysis of halogenated hydrocarbons, organochlorine pesticides and PCB

MN Appl. No. 250350

Column: OPTIMA® 5 MS, 25 m x 0.20 mm ID, 0.35 µm film, REF 726215.25,
max. temperature 340/360 °C
Sample: accu-pesticide test mixture, 10 ng/peak
Injection: 1 µl, 1 min splitless
Carrier gas: 30 cm/s He
Temperature: 75 °C (1 min) $\xrightarrow{5\text{ °C/min}}$ 280 °C
Detector: MSD

Peaks:

- | | | |
|------------------------|--------------------------|----------------------|
| 1. Hexachlorobutadiene | 11. Aldrin | 21. <i>o,p'</i> -DDD |
| 2. Pentachlorobenzene | 12. Isobenzan | 22. Endrin |
| 3. α -BHC | 13. Isodrin | 23. Endosulfan 2 |
| 4. Hexachlorobenzene | 14. Heptachlor epoxide A | 24. PCB-118 |
| 5. β -BHC | 15. Heptachlor epoxide B | 25. <i>p,p'</i> -DDD |
| 6. γ -BHC | 16. <i>o,p'</i> -DDE | 26. <i>o,p'</i> -DDT |
| 7. δ -BHC | 17. PCB-101 | 27. PCB-153 |
| 8. PCB-28 | 18. Endosulfan 1 | 28. <i>p,p'</i> -DDT |
| 9. Heptachlor | 19. <i>p,p'</i> -DDE | 29. PCB-138 |
| 10. PCB-52 | 20. Dieldrin | 30. PCB-180 |



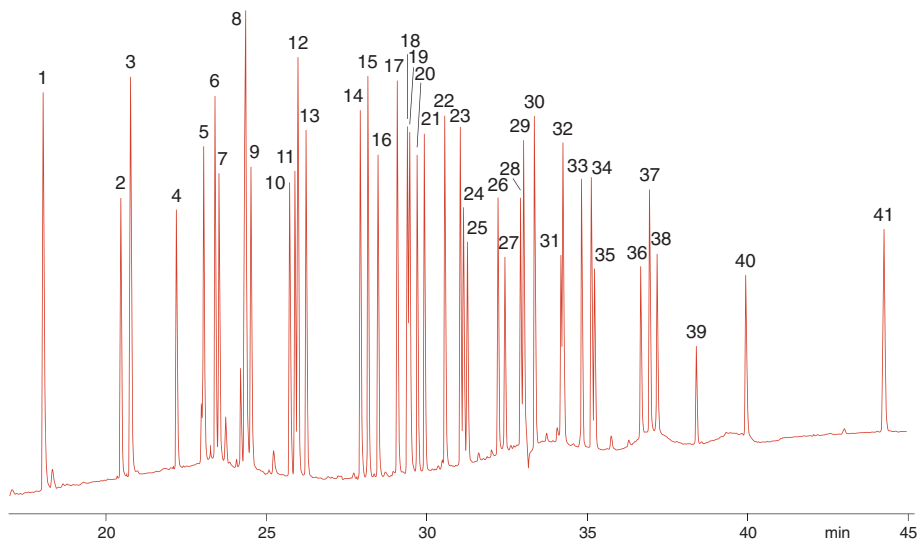
Environmental pollutants

Analysis of organochlorine pesticides and PCB MN Appl. No. 210500

Column: OPTIMA® δ-3, 60 m x 0.25 mm ID, 0.25 µm film, REF 726420.60,
max. temperature 340/360 °C
Injection: 1 µl, 3 – 18 pg/µl depending on component, 270 °C
Carrier gas: 1.37 ml/min H₂ (1.76 bar)
Temperature: 120 °C (1 min) $\xrightarrow{5\text{ °C/min}}$ 300 °C (45 min)
Detector: ECD 300 °C

Peaks:

- | | | |
|--|--------------------------------------|-------------------------|
| 1. TCNB = 1,2,4,5-Tetrachloro-3-nitrobenzene = Tecnazene | 14. Oxychlordane | 28. <i>p,p'</i> -DDD |
| 2. α-BHC | 15. <i>cis</i> -Heptachlor epoxide | 29. PCB-153 |
| 3. HCB | 16. <i>trans</i> -Heptachlor epoxide | 30. β-Endosulfan |
| 4. Lindane | 17. <i>o,p'</i> -DDE | 31. <i>p,p'</i> -DDT |
| 5. Bromocyclen | 18. <i>trans</i> -Chlordane | 32. PCB-138 |
| 6. Musk xylene | 19. PCB-101 | 33. PCB-183 |
| 7. β-BHC | 20. <i>cis</i> -Chlordane | 34. Endosulfan sulphate |
| 8. PCB-28/31 | 21. α-Endosulfan | 35. Parlar 50 |
| 9. Heptachlor | 22. <i>p,p'</i> -DDE | 36. PCB-156 |
| 10. PCB-52 | 23. Dieldrin | 37. PCB-180 |
| 11. PCB-49 | 24. <i>o,p'</i> -DDD | 38. Ketoendrin |
| 12. Aldrin | 25. Parlar 26 | 39. Parlar 62 |
| 13. Musk ketone | 26. Endrin | 40. PCB-189 |
| | 27. <i>o,p'</i> -DDT | 41. Deca-PCB = PCB-209 |



Courtesy of Mr. Münch, Staatl. Med. Lebensm. u. Vet. UA, Gießen, Germany



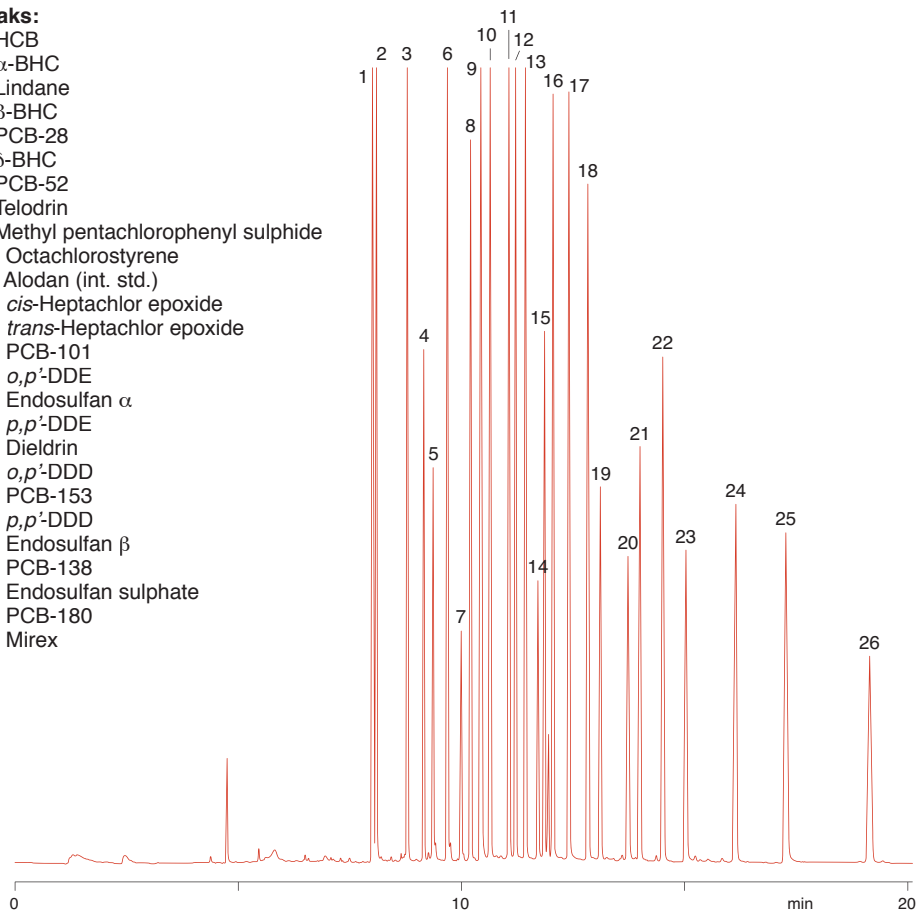
Analysis of organochlorine pesticides and PCB

MN Appl. No. 200610

Column: OPTIMA® 17, 50 m x 0.25 mm ID, 0.25 µm film, REF 726022.50,
max. temperature 320/340 °C
Injection: 0.5 µl (100 pg each), splitless, 280 °C
Carrier gas: 250 kPa H₂
Temperature: 80 °C $\xrightarrow{30\text{ °C/min}}$ 240 °C $\xrightarrow{6\text{ °C/min}}$ 280 °C
Detector: ECD 310 °C

Peaks:

1. HCB
2. α-BHC
3. Lindane
4. β-BHC
5. PCB-28
6. δ-BHC
7. PCB-52
8. Telodrin
9. Methyl pentachlorophenyl sulphide
10. Octachlorostyrene
11. Alodan (int. std.)
12. *cis*-Heptachlor epoxide
13. *trans*-Heptachlor epoxide
14. PCB-101
15. *o,p'*-DDE
16. Endosulfan α
17. *p,p'*-DDE
18. Dieldrin
19. *o,p'*-DDD
20. PCB-153
21. *p,p'*-DDD
22. Endosulfan β
23. PCB-138
24. Endosulfan sulphate
25. PCB-180
26. Mirex



Courtesy of Mr. Lembacher, Hipp Nahrungsmittel, Pfaffenhofen, Germany

Environmental pollutants

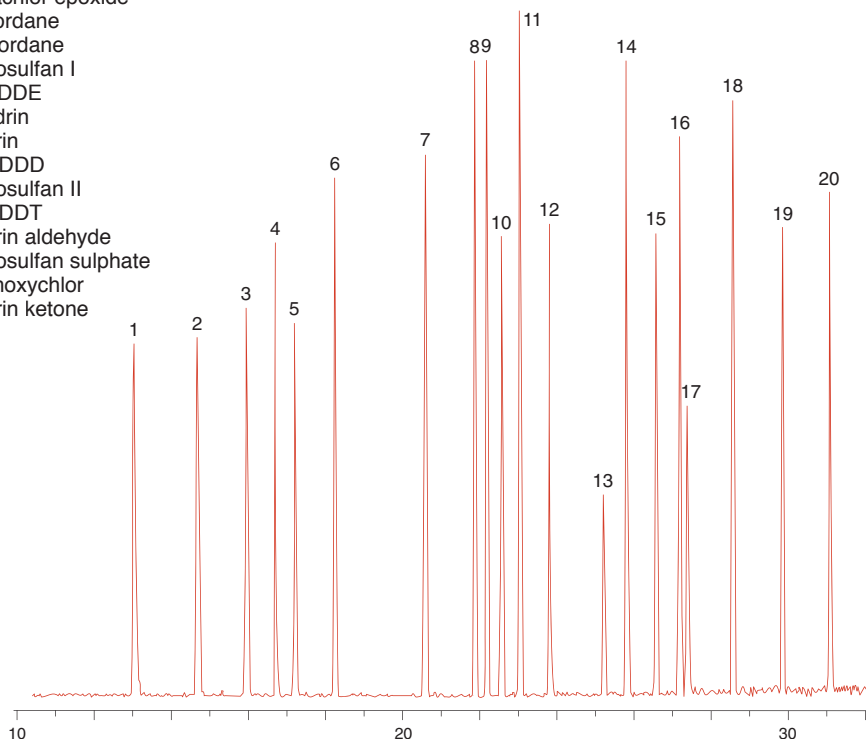
Separation of organochlorine pesticides (EPA 8081)

MN Appl. No. 250430

Column: OPTIMA® δ -6, 50 m x 0.20 mm ID, 0.2 μ m film, REF 726465.50,
max. temperature 340/360 °C
Sample: EPA 8081 organochlorine pesticide calibration mix (Restek), 200 μ g/ml each in
toluene – hexane (1:1, v/v)
Injection: 1 μ l, split 1: 30
Carrier gas: 2.0 bar He
Temperature: 180 °C $\xrightarrow{4\text{ }^{\circ}\text{C/min}}$ 300 °C (10 min)
Detector: MSD

Peaks:

1. α -BHC
2. γ -BHC (lindane)
3. β -BHC
4. Heptachlor
5. δ -BHC
6. Aldrin
7. Heptachlor epoxide
8. γ -Chlordane
9. α -Chlordane
10. Endosulfan I
11. 4,4'-DDE
12. Dieldrin
13. Endrin
14. 4,4'-DDD
15. Endosulfan II
16. 4,4'-DDT
17. Endrin aldehyde
18. Endosulfan sulphate
19. Methoxychlor
20. Endrin ketone



Organochlorine pesticides



Separation of organochlorine pesticides (EPA 608)

MN Appl. No. 213220

Column: OPTIMA® 35 MS, 30 m x 0.25 mm ID, 0.25 µm film, REF 726154.30,
max. temperature 360/370 °C

Sample: EPA 608 organochlorine pesticide mix

Injection: 1 µl, split 20 ml/min

Carrier gas: 0.8 ml/min He

Temperature: 160 °C $\xrightarrow{6\text{ °C/min}}$ 260 °C (10 min)

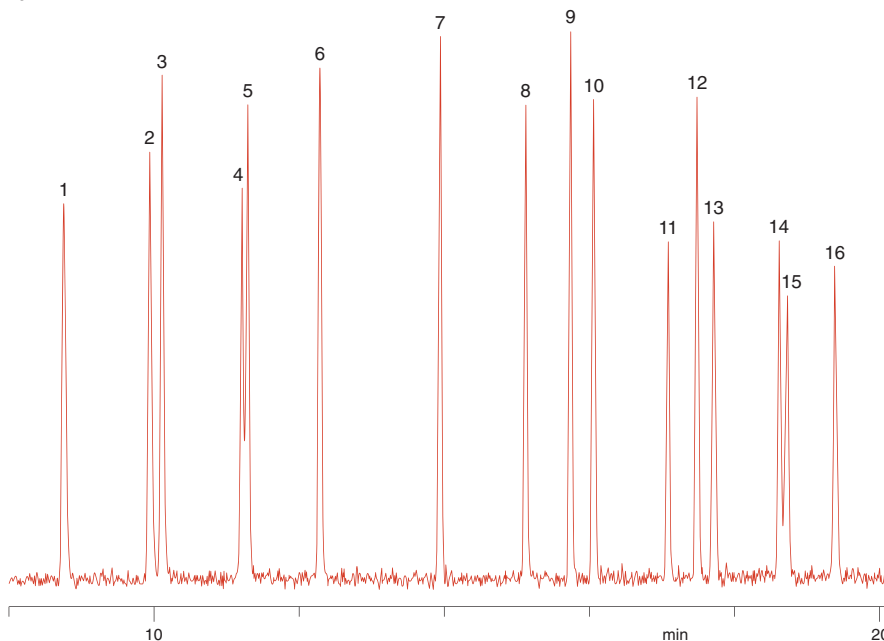
Detector: MSD

Peaks:

1. α-BHC
2. γ-BHC (lindane)
3. β-BHC
4. Heptachlor
5. δ-BHC
6. Aldrin

7. Heptachlor epoxide
8. Endosulfan I
9. 4,4'-DDE
10. Dieldrin
11. Endrin

12. 4,4'-DDD
13. Endosulfan II
14. 4,4'-DDT
15. Endrin aldehyde
16. Endosulfan sulphate



Analysis of organochlorine pesticides from cod liver oil MN Appl. No. 213270

Column: OPTIMA® δ-3, 60 m x 0.25 mm ID, 0.25 µm film, REF 726420.60,
max. temperature 340/360 °C

Sample: a) pesticide standard (150 pg/µl of each compound)
b) NP-HPLC group separation of NIST reference material SRM 1588 (semivolatile organohalogen compounds in cod liver oil) on 250 x 4 mm NUCLEOSIL® 100-10 NH₂ using 10 ml *n*-hexane (fraction 1) and 32 ml *n*-hexane – dichloromethane (90:10 v/v, fraction 2), flow rate 1 ml/min (see application 250510 for chromatogram of fraction 1)

Injection: 2 µl of fraction 2, on column

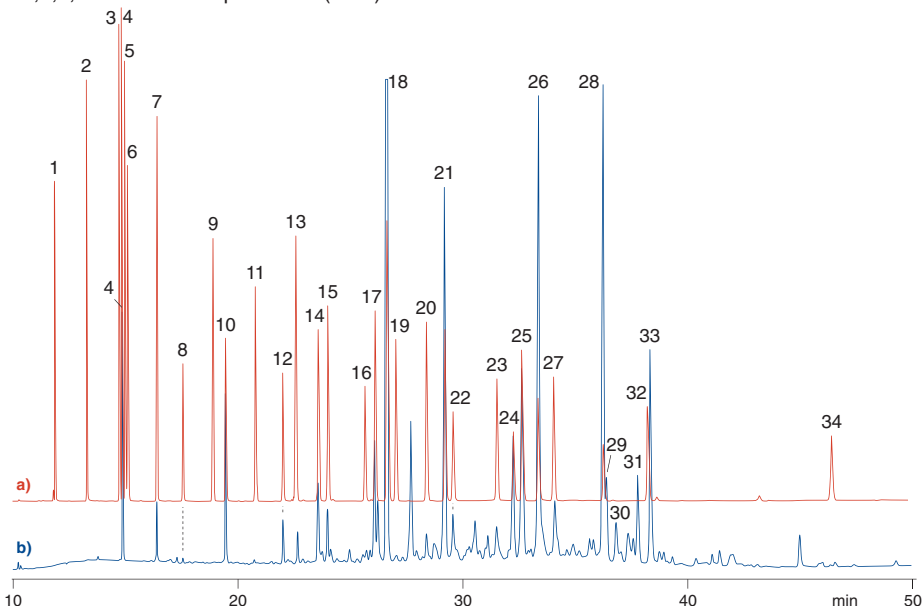
Carrier gas: 180 kPa H₂ $\xrightarrow{20\text{ °C/min}}$ 200 °C $\xrightarrow{1.5\text{ °C/min}}$ 290 °C (5 min)

Temperature: 80 °C (3 min) $\xrightarrow{20\text{ °C/min}}$ 200 °C $\xrightarrow{1.5\text{ °C/min}}$ 290 °C (5 min)

Detector: ECD 300 °C

Peaks:

- | | | |
|--|------------------------------------|---------------------------|
| 1. Pentachlorobenzene | 13. PCB-103 | 24. 2,4'-DDT |
| 2. 2,4,6-Tribromoanisole | 14. Oxychlordan | 25. <i>cis</i> -Nonachlor |
| 3. Tetrachloro-1,4-dimethoxybenzene | 15. <i>cis</i> -Heptachlor epoxide | 26. 4,4'-DDD |
| 4. α-BHC | 16. 2,4'-DDE | 27. Endosulfan 2 |
| 5. Pentachloroanisole | 17. <i>trans</i> -Chlordane | 28. 4,4'-DDT |
| 6. HCB | 18. <i>cis</i> -Chlordane | 29. Parlar 41 |
| 7. γ-BHC | + <i>trans</i> -nonachlor | 30. Parlar 40 |
| 8. β-BHC | 19. Endosulfan 1 | 31. Parlar 44 |
| 9. Heptachlor | 20. 4,4'-DDE | 32. Endosulfan sulphate |
| 10. ε-BHC | 21. Dieldrin | 33. Parlar 50 |
| 11. Aldrin | 22. 2,4'-DDD | 34. Mirex |
| 12. 1,2,3,4-Tetrachloronaphthalene (TCN) | 23. Endrin | |



R. Looser, K. Ballschmiter, J. Chromatogr. A 836 (1999) 271 – 284

Halogenated hydrocarbons



Analysis of organohalogen compounds incl. PCB from cod liver oil

MN Appl. No. 250510

Column: OPTIMA® δ-3, 60 m x 0.25 mm ID, 0.25 µm film, REF 726420.60,
max. temperature 340/360 °C

Sample: NP-HPLC group separation of NIST reference material SRM 1588 (semivolatile organohalogen compounds in cod liver oil) on 250 x 4 mm NUCLEOSIL® 100-10 NH₂ using 10 ml *n*-hexane (fraction 1) and 32 ml *n*-hexane – dichloromethane (90:10 v/v, fraction 2), flow rate 1 ml/min (see application 213270 for chromatogram of fraction 2)

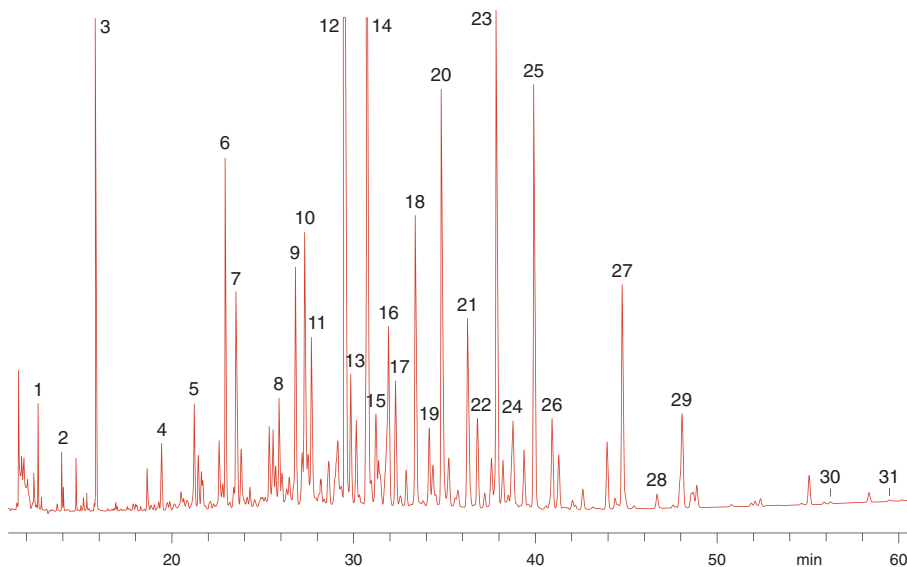
Injection: 2 µl of fraction 1, on column

Carrier gas: 180 kPa H₂ 20 °C/min 200 °C 1.5 °C/min 290 °C (5 min)

Detector: ECD 300 °C

Peaks:

- | | | | |
|---|---------------|---------------|--|
| 1. Pentachlorobenzene | 8. PCB-66 | 17. PCB-139 | 26. PCB-128 |
| 2. 2,4,6-Tribromoanisole | 9. 2,4'-DDE | 18. PCB-118 | 27. PCB-180 |
| 3. HCB | 10. PCB-101 | 19. PCB-146 | 28. 2,2',4,4'-Tetrabromodiphenyl ether |
| 4. PCB-28 | 11. PCB-99 | 20. PCB-153 | 29. Mirex |
| 5. PCB-52 | 12. 4,4'-DDE | 21. PCB-105 | 30. 2,2',4,4',6-Pentabromodiphenyl ether |
| 6. 1,2,3,4-Tetrachloro-naphthalene (TCN, recovery standard) | 13. PCB-87 | 22. PCB-137 | 31. 2,2',4,4',5-Pentabromodiphenyl ether |
| 7. PCB-103 (int. std.) | 14. Parlar 26 | 23. PCB-138 | |
| | 15. PCB-151 | 24. PCB-187 | |
| | 16. PCB-82 | 25. Parlar 50 | |



R. Looser, K. Ballschmiter, J. Chromatogr. A **836** (1999) 271 – 284

Environmental pollutants

Analysis of a pesticide mixture

MN Appl. No. 210710

Column: OPTIMA® δ-6,
50 m x 0.20 mm ID, 0.20 μm
film, REF 726465.50,
max. temperature 340/360 °C

Injection: 0.03 – 0.5 μg/ml, depending
on component, splitless,
purge time 0.8 min, 270 °C

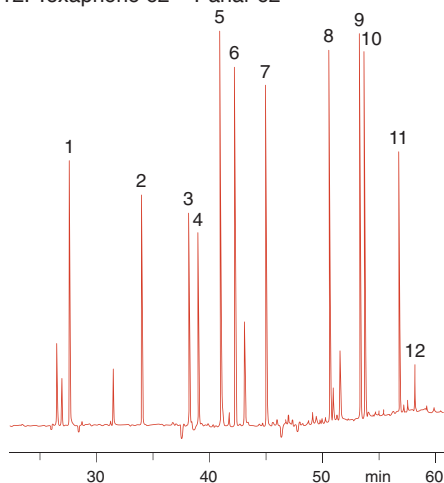
Carrier gas: 2.3 ml/min He

Temperature: 130 °C (1 min) $\xrightarrow{25\text{ °C/min}}$
180 °C (5 min) $\xrightarrow{2\text{ °C/min}}$ 260 °C
 $\xrightarrow{6\text{ °C/min}}$ 310 °C (14 min)

Detector: ECD 300 °C

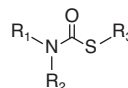
Peaks:

1. Musk xylene
2. Musk ketone
3. *cis*-Heptachlor epoxide
4. *trans*-Heptachlor epoxide
5. *trans*-Chlordane
6. α-Endosulfan
7. Toxaphene 26 = Parlar 26
8. β-Endosulfan
9. Toxaphene 50 = Parlar 50
10. Endosulfan sulphate
11. Ketoendrin
12. Toxaphene 62 = Parlar 62

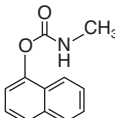
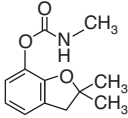
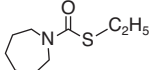
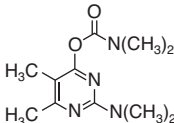


Courtesy of Mr. Grabher, CLUA, Sigmaringen,
Germany

Pesticide structures: carbamates



Compound	R ₁	R ₂	R ₃
Butylate	<i>i</i> -C ₄ H ₉	<i>i</i> -C ₄ H ₉	C ₂ H ₅
Cycloate	C ₂ H ₅	cyclohexyl	C ₂ H ₅
EPTC	<i>n</i> -C ₃ H ₇	<i>n</i> -C ₃ H ₇	C ₂ H ₅
Prothiocarb	H	(CH ₂) ₃ – N(CH ₃) ₂	C ₂ H ₅
Pebulate	C ₂ H ₅	<i>n</i> -C ₄ H ₉	<i>n</i> -C ₃ H ₇
Vernolate	<i>n</i> -C ₃ H ₇	<i>n</i> -C ₃ H ₇	<i>n</i> -C ₃ H ₇
Thiobencarb	C ₂ H ₅	C ₂ H ₅	

Compound	Structure
Carbaryl	
Carbofuran	
Molinate	
Pirimicarb	

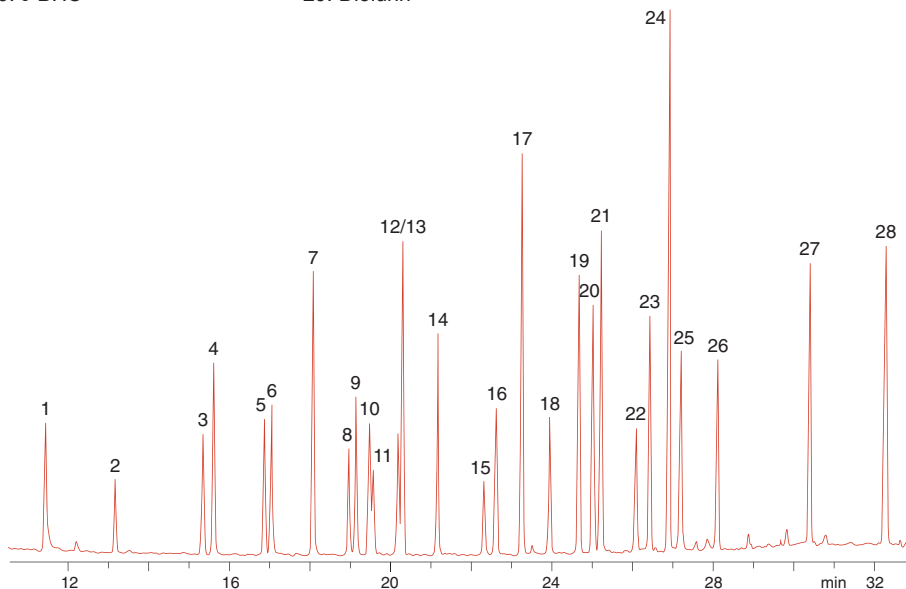


Analysis of 28 pesticides (Ökotex 100) MN Appl. No. 210270

Column: OPTIMA® δ -3, 30 m x 0.25 mm ID, 0.25 μ m film, REF726420.30,
max. temperature 340/360 °C
Injection: 250 °C, splitless, purge time 1 min
Carrier gas: He
Pressure: 21.4 kPa (1 min) $\xrightarrow{9 \text{ kPa/min}}$ 42.9 kPa $\xrightarrow{1.1 \text{ kPa/min}}$ 72 kPa (20 min)
Temperature: 60 °C (1 min) $\xrightarrow{40 \text{ °C/min}}$ 150 °C $\xrightarrow{4.8 \text{ °C/min}}$ 280 °C (20 min)
Detector: MSD 320 °C

Peaks:

- | | | |
|---------------------|--------------------------------------|-------------------------|
| 1. Carbaryl | 11. Parathion-methyl | 21. <i>o,p'</i> -DDD |
| 2. Trifluralin | 12. Malathion | 22. Endrin |
| 3. α -BHC | 13. Aldrin | 23. <i>o,p'</i> -DDT |
| 4. HCB | 14. Parathion | 24. <i>p,p'</i> -DDD |
| 5. Lindane | 15. <i>cis</i> -Heptachlor epoxide | 25. β -Endosulfan |
| 6. Quintozene | 16. <i>trans</i> -Heptachlor epoxide | 26. <i>p,p'</i> -DDT |
| 7. β -BHC | 17. <i>o,p'</i> -DDE | 27. Methoxychlor |
| 8. Heptachlor | 18. α -Endosulfan | 28. Mirex |
| 9. δ -BHC | 19. <i>p,p'</i> -DDE | |
| 10. ϵ -BHC | 20. Dieldrin | |



Courtesy of K. Friedrichs, Chem. Untersuchungsamt, Bielefeld, Germany

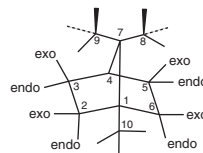
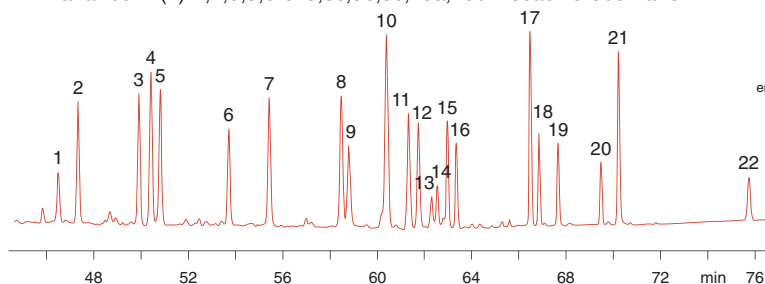
Analysis of toxaphenes

MN Appl. No. 211920

Column: OPTIMA® 17, 50 m x 0.25 mm ID, 0.25 µm film, REF 726022.50,
max. temperature 320/340 °C
Injection: 1 – 2 µl, 1.0 min splitless, 240 °C
Carrier gas: He
Temperature: 70 °C (2 min) $\xrightarrow{15\text{ °C/min}}$ 180 °C $\xrightarrow{1.5\text{ °C/min}}$ 250 °C (5 min) $\xrightarrow{10\text{ °C/min}}$ 290 °C (15 min)
Detector: MSD

Peaks:

1. Parlar 11 (±)-2,2,3-exo-Trichloro,5,5-bis(chloromethyl),6-(E)-chloromethylene-8,9,10-trinorbornane
2. Parlar 12 (±)-5-exo,6-endo-Dichloro,2-endo-chloromethyl,3-(E)-chloromethylene-8,9,10-trinorbornane
3. Parlar 15 (±)-5-exo,6-endo,7-anti-Trichloro,2,2-bis(chloromethyl),3-(E)-chloromethylene-8,9,10-trinorbornane
4. Parlar 21 (±)-2,2,5,5-9c,10a,10b-Heptachlorobornane
5. Parlar 26 (±)-2-endo,3-exo,5-endo,6-exo,8b,8c,10a,10c-Octachlorobornane
6. Parlar 25 (±)-2,2,3-exo-Trichloro,5-endo-chloromethyl,6-(E)-chloromethylene-5-dichloromethyl,8,9,10-trinorbornane
7. Parlar 31 (±)-2,2,3-exo-Trichloro,6-(E)-chloromethylene,5,5-bis(dichloromethyl),8,9,10-trinorbornane
8. Parlar 32 (±)-2,2,5-endo,6-exo,8c,9b,10a-Heptachlorobornane
9. Parlar 38 (±)-2,2,5,5,9b,9c,10a,10b-Octachlorobornane
10. Parlar 39 (±)-2,2,3-exo,5-endo,6-exo,8c,9b,10a-Octachlorobornane
11. Parlar 41 (±)-2-exo,3-endo,5-exo,8c,9b,9c,10a,10b-Octachlorobornane
12. Parlar 40 (±)-2-endo,3-exo,5-endo,6-exo,8b,9c,10a,10c-Octachlorobornane
13. Parlar 42a (±)-2,2,5-endo,6-exo,8b,8c,9c,10a-Octachlorobornane
14. Parlar 42b (±)-2,2,5-endo,6-exo,8c,9b,9c,10a-Octachlorobornane
15. Parlar 44 (±)-2-exo,5,5,8c,9b,9c,10a,10b-Octachlorobornane
16. Parlar 50 (±)-2-endo,3-exo,5-endo,6-exo,8b,8c,9c,10a,10c-Nonachlorobornane
17. Parlar 51 (±)-2,2,5,5,8c,9b,10a,10b-Octachlorobornane
- + Parlar 56 (±)-2,2,5-endo,6-exo,8b,8c,9c,10a,10b-Nonachlorobornane
18. Parlar 58 (±)-2,2,3-exo,5,5,8c,9b,10a,10b-Nonachlorobornane
19. Parlar 59 (±)-2,2,5,endo,6-exo,8c,9b,9c,10a,10b-Nonachlorobornane
20. Parlar 62 (±)-2,2,5,5,8c,9b,9c,10a,10b-Nonachlorobornane
21. Parlar 63 (±)-2-exo,3-endo,5-exo,6-exo,8b,8c,9c,10a,10c-Nonachlorobornane
22. Parlar 69 (±)-2,2,5,5,6-exo,8c,9b,9c,10a,10b-Decachlorobornane



M. Kaltenecker, Dissertation, Universität Wuppertal, Germany and M. Kaltenecker, K.-H. Schwind, K.-H. Ueberschär, H. Hecht, M. Petz, Organohalogen Compounds **35** (1998) 281 – 285

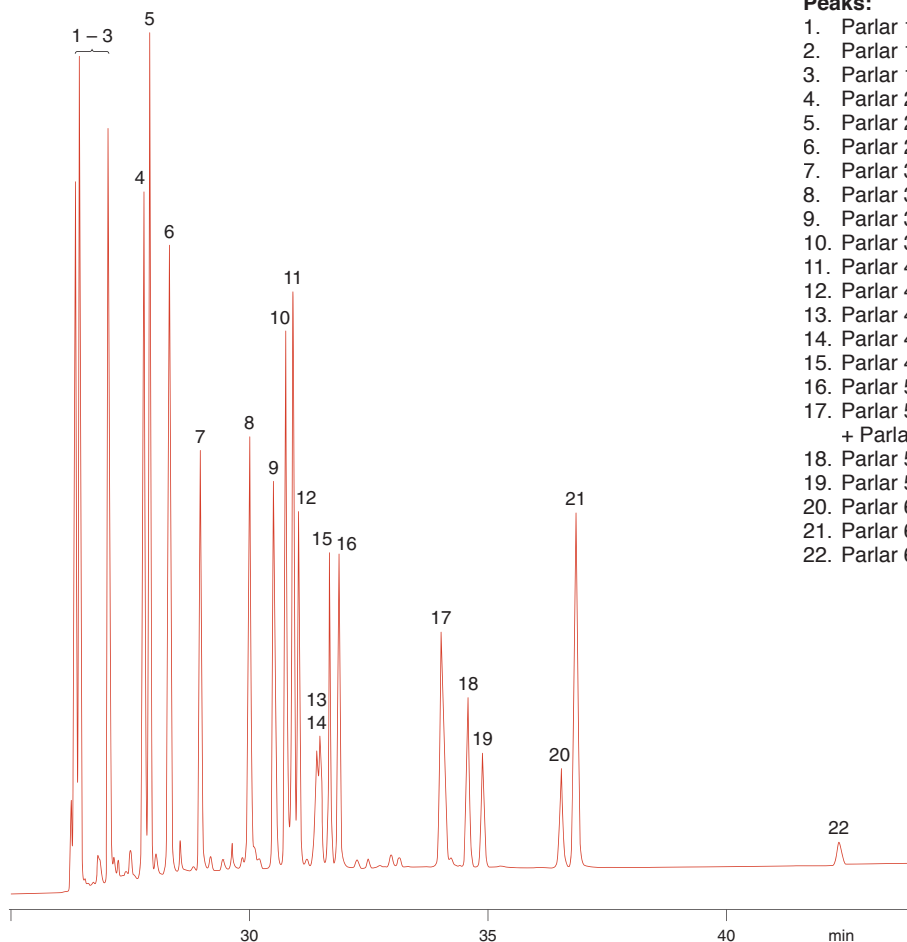


Analysis of a complex toxaphene mixture MN Appl. No. 250130

Column: OPTIMA® δ -3, 40 m x 0.20 mm ID, 0.35 μ m film, custom-made column
we recommend REF 726400.50, max. temperature 340/360 °C
Injection: 1 μ l, splitless
Carrier gas: 1.3 ml/min He
Temperature: 100 °C $\xrightarrow{7\text{ }^{\circ}\text{C/min}}$ 270 °C
Detector: ECD

Peaks:

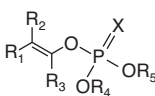
1. Parlar 11
2. Parlar 12
3. Parlar 15
4. Parlar 21
5. Parlar 26
6. Parlar 25
7. Parlar 31
8. Parlar 32
9. Parlar 38
10. Parlar 39
11. Parlar 41
12. Parlar 40
13. Parlar 42a
14. Parlar 42b
15. Parlar 44
16. Parlar 50
17. Parlar 51
+ Parlar 56
18. Parlar 58
19. Parlar 59
20. Parlar 62
21. Parlar 63
22. Parlar 69

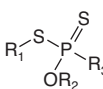
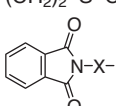
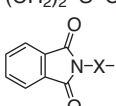
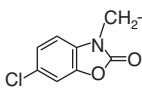
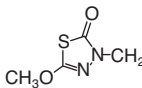
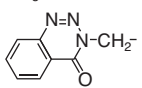
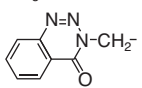


Courtesy of R. Baycan-Keller, M. Oehme, Inst. Org. Anal. Chem., University of Basel, Switzerland

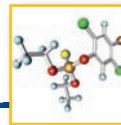
Environmental pollutants

Structures of organophosphorus pesticides

Structure	Compound	X	R ₁	R ₂	R ₃	R ₄	R ₅
	Chlorfenvinphos	O	Cl	H		C ₂ H ₅	C ₂ H ₅
	Crotoxyphos	O	CO-O-CH(C ₆ H ₅)-CH ₃	H	CH ₃	CH ₃	CH ₃
	Dichlorphos	O	Cl	Cl	H	CH ₃	CH ₃
	Dicrotophos	O	CO-N(CH ₃) ₂	H	CH ₃	CH ₃	CH ₃
	Mevinphos	O	CO-O-CH ₃	H	CH ₃	CH ₃	CH ₃
	Monocrotophos	O	CO-NH-CH ₃	H	CH ₃	CH ₃	CH ₃
	Phosphamidon	O	CO-N(C ₂ H ₅) ₂	Cl	CH ₃	CH ₃	CH ₃
	Tetrachlorvinphos	O	Cl	H		CH ₃	CH ₃
	Methacrifos	S	CO-OCH ₃	CH ₃	H	C ₂ H ₅	C ₂ H ₅

Structure	Compound	R ₁	R ₂	R ₃
	Chlormephos	CH ₂ Cl	C ₂ H ₅	OC ₂ H ₅
	Fonofos	C ₆ H ₅	C ₂ H ₅	C ₂ H ₅
	Malathion	CH ₂ -CO-OC ₂ H ₅ CH ₂ -CO-OC ₂ H ₅	CH ₃	OCH ₃
	Dimethoate	CH ₂ -CO-NH-CH ₃	CH ₃	OCH ₃
	Amidithion	CH ₂ -CO-NH-(CH ₂) ₂ -OCH ₃	CH ₃	OCH ₃
	Mecarbam	CH ₂ -CO-N(CH ₃)-CO-OC ₂ H ₅	C ₂ H ₅	OC ₂ H ₅
	Phorate	CH ₂ -S-C ₂ H ₅	C ₂ H ₅	OC ₂ H ₅
	Terbufos	CH ₂ -S-C(CH ₃) ₃	C ₂ H ₅	OC ₂ H ₅
	Carbophenothion	CH ₂ -S-4-C ₆ H ₄ Cl	C ₂ H ₅	OC ₂ H ₅
	Phenkapton	CH ₂ -S-2,5-C ₆ H ₄ Cl ₂	C ₂ H ₅	OC ₂ H ₅
	Disulfoton	(CH ₂) ₂ -S-C ₂ H ₅	C ₂ H ₅	OC ₂ H ₅
	Thiometon	(CH ₂) ₂ -S-C ₂ H ₅	CH ₃	OCH ₃
	Phosmet		X = CH ₂	CH ₃
	Dialifos		X = CH-CH ₂ Cl	C ₂ H ₅
	Phosalone		C ₂ H ₅	OC ₂ H ₅
	Methidathion		CH ₃	OCH ₃
	Azinphos-methyl		CH ₃	OCH ₃
	Azinphos-ethyl		C ₂ H ₅	OC ₂ H ₅

Organophosphorus pesticides



Structures of organophosphorus pesticides

Structure	Compound	R ₁	R ₂	R ₃
	Acephate	NH-CO-CH ₃	OCH ₃	SCH ₃
	Dimefox	N(CH ₃) ₂	N(CH ₃) ₂	F
	Edifenphos	OC ₂ H ₅	SC ₆ H ₅	SC ₆ H ₅
	Ethoprophos	OC ₂ H ₅	S- <i>n</i> -C ₃ H ₇	S- <i>n</i> -C ₃ H ₇
	Heptenophos		OCH ₃	OCH ₃
	Naled	OCHBr-CBrCl ₂	OCH ₃	OCH ₃
	Trichlorfon	CH(OH)-CCl ₃	OCH ₃	OCH ₃
	Vamidothion	S-(CH ₂) ₂ -S-CH(CH ₃)-CO-NH-CH ₃	OCH ₃	OCH ₃
	Coumaphos		C ₂ H ₅	C ₂ H ₅
	Chlorpyrifos		C ₂ H ₅	C ₂ H ₅
	Diazinon	X ₁ = CH ₃ , X ₂ = <i>i</i> -C ₃ H ₇	C ₂ H ₅	C ₂ H ₅
	Etrifos	X ₁ = OC ₂ H ₅ , X ₂ = C ₂ H ₅	CH ₃	CH ₃
	Pirimiphos-ethyl	X ₁ = CH ₃ , X ₂ = N(C ₂ H ₅) ₂	C ₂ H ₅	C ₂ H ₅
	Pirimiphos-methyl	X ₁ = CH ₃ , X ₂ = N(C ₂ H ₅) ₂	CH ₃	CH ₃
	Pyridaphenthion		C ₂ H ₅	C ₂ H ₅
	Quinalphos		C ₂ H ₅	C ₂ H ₅
	Thionazin		C ₂ H ₅	C ₂ H ₅
	Isazofos	X ₁ = <i>i</i> -C ₃ H ₇ , X ₂ = Cl	C ₂ H ₅	C ₂ H ₅
	Triazophos	X ₁ = C ₆ H ₇ , X ₂ = H	C ₂ H ₅	C ₂ H ₅
	Demeton-O	O	S	C ₂ H ₅
	Demeton-S	S	O	C ₂ H ₅
	Demeton-O-methyl	O	S	CH ₃
	Demeton-S-methyl	S	O	CH ₃
	Demephion-O	O	S	CH ₃
	Demephion-S	S	O	CH ₃
	Sulfotep	O	S	C ₂ H ₅
	TEPP	O	O	C ₂ H ₅
	Ethion	S-CH ₂ -S	S	C ₂ H ₅
	Dioxathion		S	C ₂ H ₅
				-

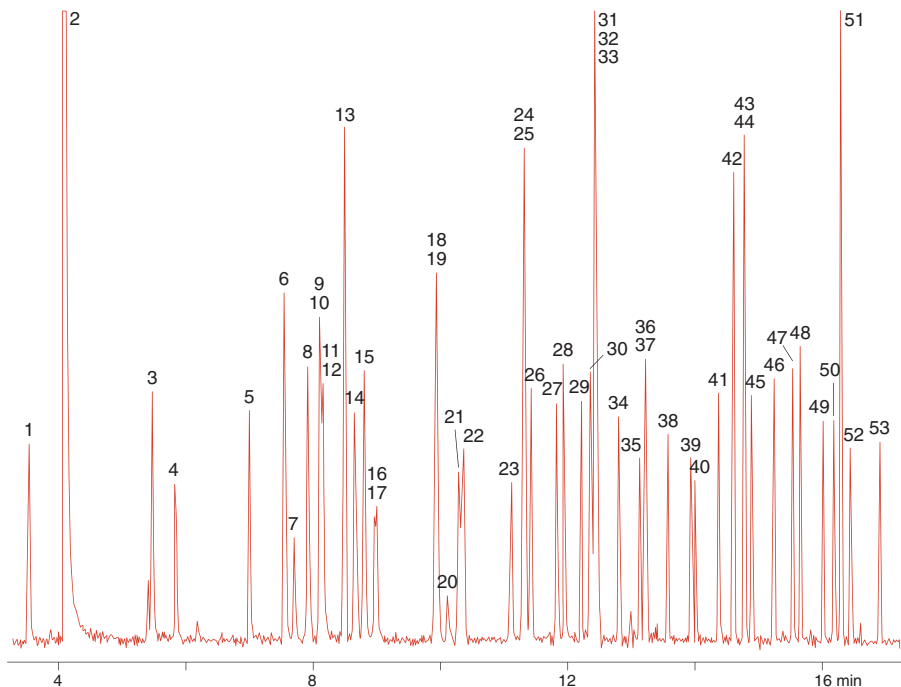
Environmental pollutants

Analysis of organophosphorus pesticides (EPA 8140 / 8141 / 8141 A) MN Appl. No. 213030

Column: OPTIMA® 1 MS Accent, 30 m x 0.32 mm ID, 0.50 µm film, REF 725807.30
Sample: 0.2 µg/ml in hexane,
8140/8141 OP pesticides calibration mix A and 8141 OP pesticides calibration mix B; IS triphenyl phosphate and tributyl phosphate
Injection: splitless (hold 1 min), 250 °C
Carrier gas: He, 1 ml/min, constant pressure
Temperature: 100 °C $\xrightarrow{10\text{ °C/min}}$ 180 °C (2 min) $\xrightarrow{18\text{ °C/min}}$ 300 °C (3 min)
Detector: FPD (Flame Photometric Detector), 280 °C

Peaks:

1. Dichlorvos, 2. Hexamethylphosphoramide, 3. Mevinphos, 4. Trichlorfon, 5. TEPP, 6. Thionazin, 7. Demeton-O, 8. Ethoprop 9. Tributyl phosphate (IS), 10. Dicrotophos, 11. Monocrotophos, 12. Naled, 13. Sulfotepp, 14. Phorate, 15. Dimethoate, 16. Demeton-S, 17. Dioxathion, 18. Terbufos, 19. Fonofos, 20. Phosphamidon isomer, 21. Diazinon, 22. Disulfoton, 23. Phosphamidon, 24. Dichlofenthion, 25. Parathion-methyl, 26. Chlorpyrifos methyl, 27. Ronnel, 28. Fenitrothion, 29. Malathion, 30. Fenthion, 31. Chlordane, 32. Parathion-ethyl, 33. Chlorpyrifos, 34. Trichloronate, 35. Chlorfenvinphos, 36. Merphos, 37. Crotoxyphos, 38. Stirofos, 39. Tokuthion (prothiophos), 40. Merphos oxidation product, 41. Fensulfothion, 42. Famphur, 43. Ethion, 44. Bolstar, 45. Carbophenothion, 46. Triphenyl phosphate (IS), 47. Phosmet, 48. EPN, 49. Azinphos-methyl, 50. Leptophos, 51. Tri-*o*-cresyl phosphate, 52. Azinphos-ethyl, 53. Coumaphos



Organophosphorus pesticides



Separation of organophosphorus pesticides (EPA 8140 / 8141)

MN Appl. No. 250420

Column: OPTIMA® δ-6, 50 m x 0.20 mm ID, 0.2 μm film, REF 726465.50,
max. temperature 340/360 °C

Sample: EPA 8140 OP pesticide calibration mix (Restek),
200 μg/ml each in hexane – acetone (95:5, v/v)

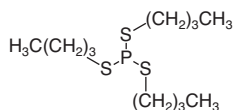
Injection: 1 μl, split 1:30

Carrier gas: 2.0 bar He

Temperature: 150 °C $\xrightarrow{2.5\text{ °C/min}}$ 300 °C (10 min)

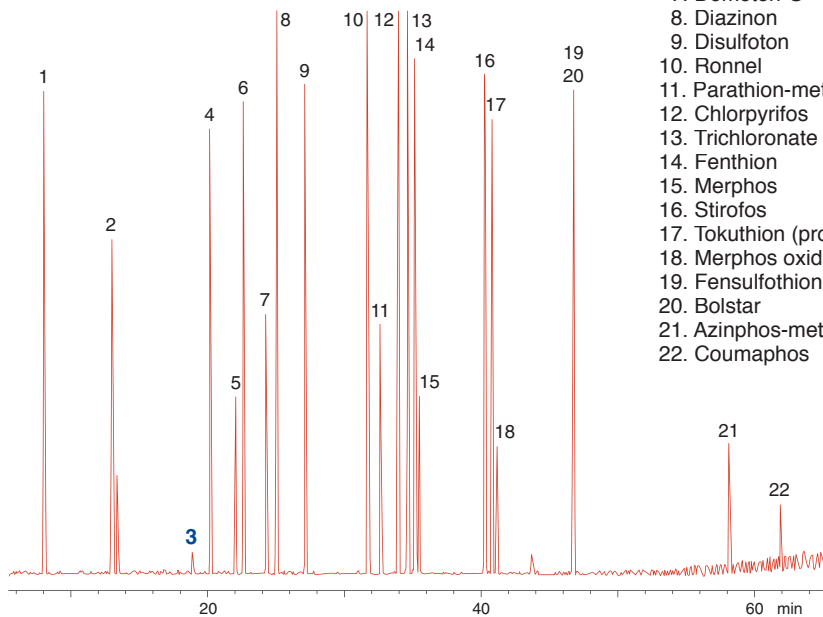
Detector: MSD

Merphos:



Peaks:

1. Dichlorvos
2. Mevinphos
3. Demeton-S
4. Ethoprop
5. Naled
6. Phorate
7. Demeton-O
8. Diazinon
9. Disulfoton
10. Ronnel
11. Parathion-methyl
12. Chlorpyrifos
13. Trichloronate
14. Fenthion
15. Merphos
16. Stirofos
17. Tokuthion (prothiophos)
18. Merphos oxidation product
19. Fensulfothion
20. Bolstar
21. Azinphos-methyl
22. Coumaphos



Environmental pollutants

Analysis of organophosphorus insecticides

MN Appl. No. 200940

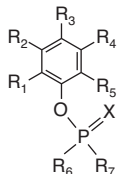
Column: OPTIMA® 1301, 25 m x 0.32 mm ID, 0.25 µm film, REF 726777.25,
max. temperature 300/320 °C
Injection: 0.5 µl on column
Carrier gas: 80 kPa H₂
Temperature: 80 $\xrightarrow{30\text{ °C/min}}$ 220 °C $\xrightarrow{4\text{ °C/min}}$ 270 °C
Detector: PND

Peaks: [ng]

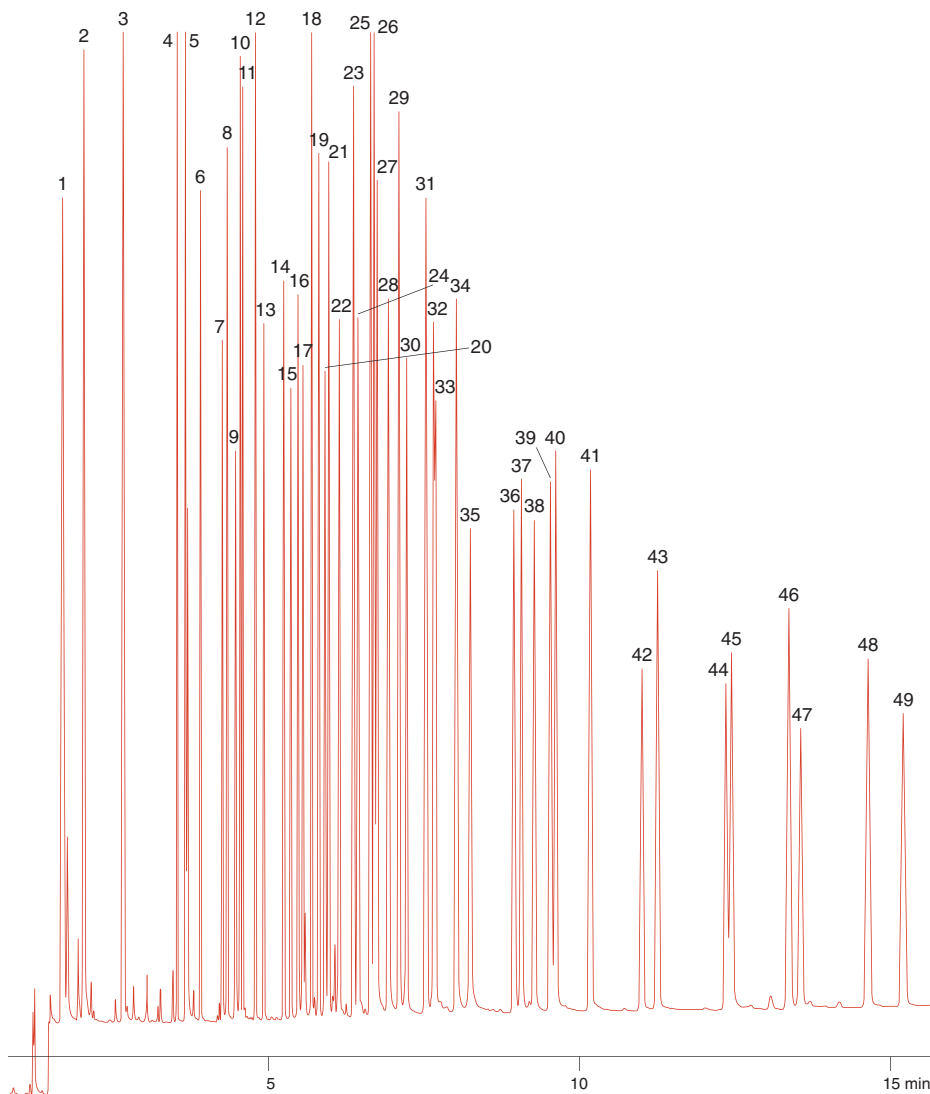
- | | | |
|---|-----------------------------|----------------------------------|
| 1. Dimefox [0.25] | 18. Dichlofenthion [0.5] | 34. Fenamiphos [0.5] |
| 2. Triethyl phosphate [0.25] | 19. Dursban-methyl = | 35. Vamidothion [0.5] |
| 3. Dichlorvos [0.25] | Chlorpyrifos-methyl [0.5] | 36. Ethion [0.25] |
| 4. Chlormephos [0.25] | 20. Tolclofos-methyl [0.25] | 37. Sulprofos [0.5] |
| 5. Mevinphos [0.5] | 21. Fenchlorphos [0.5] | 38. Carbophenothion [0.5] |
| 6. Methacrifos [0.25] | 22. Pirimiphos-methyl [0.5] | 39. Edifenphos [0.5] |
| 7. Demephion [0.5] | 23. Dursban-ethyl = | 40. Triazophos [0.5] |
| 8. Heptenophos [0.5] | Chlorpyrifos-ethyl [0.5] | 41. Triphenyl phosphate [0.5] |
| 9. TEPP [0.25] | 24. Malathion [0.5] | 42. Phenkapton [0.5] |
| 10. Ethoprophos [0.25] | 25. Pirimiphos-ethyl [1.0] | 43. Pyridaphenthion [0.5] |
| 11. Tributyl phosphate [0.25] | 26. Parathion-ethyl [0.5] | 44. Azinphos-methyl [0.5] |
| 12. Sulfotep [0.25] | 27. Amidithion [0.5] | 45. Phosalone [0.5] |
| 13. Thiometon [0.25] | 28. Crufomate [0.5] | 46. Azinphos-ethyl [0.5] |
| 14. Diazinon [0.25] | 29. Quinalphos [0.5] | 47. Dialifos [0.5] |
| 15. Disulfoton [0.25] | 30. Bromophos-ethyl [0.5] | 48. Tri-p-cresyl phosphate [0.5] |
| 16. Tris-2-chloroethyl phosphate [0.25] | 31. Methidathion [0.5] | 49. Coumaphos [0.5] |
| 17. Isazofos [0.25] | 32. Iodofenphos [0.5] | |
| | 33. Prothiophos [0.5] | |

Structures of organophosphorus pesticides

Structure	Compound	X	R ₁	R ₂	R ₃	R ₄	R ₅	R ₆	R ₇
	Bromophos-ethyl	S	Cl	H	Br	Cl	H	OC ₂ H ₅	OC ₂ H ₅
	Chlorthion	S	H	Cl	NO ₂	H	H	OCH ₃	OCH ₃
	Crufomate	O	Cl	H	C(CH ₃) ₃	H	H	OCH ₃	NH-CH ₃
	Dichlofenthion	S	Cl	H	Cl	H	H	OC ₂ H ₅	OC ₂ H ₅
	EPN	S	H	H	NO ₂	H	H	OC ₂ H ₅	C ₆ H ₅
	Famphur	S	H	H	SO ₂ -N(CH ₃) ₂	H	H	OCH ₃	OCH ₃
	Fenamiphos	O	H	CH ₃	SCH ₃	H	H	OC ₂ H ₅	NH- <i>i</i> -C ₃ H ₇
	Fenchlorphos	S	Cl	H	Cl	Cl	H	OCH ₃	OCH ₃
	Fenitrothion	S	H	CH ₃	NO ₂	H	H	OCH ₃	OCH ₃
	Fensulfothion	S	H	H	SO-CH ₃	H	H	OC ₂ H ₅	OC ₂ H ₅
	Fenthion	S	H	CH ₃	SCH ₃	H	H	OCH ₃	OCH ₃
	Iodofenphos	S	Cl	H	I	Cl	H	OCH ₃	OCH ₃
	Leptophos	S	Cl	H	Br	Cl	H	OCH ₃	C ₆ H ₅
	Parathion-ethyl	S	H	H	NO ₂	H	H	OC ₂ H ₅	OC ₂ H ₅
	Profenofos	O	Cl	H	Br	H	H	OC ₂ H ₅	S- <i>n</i> -C ₃ H ₇
	Prothiofos	S	Cl	H	Cl	H	H	OC ₂ H ₅	S- <i>n</i> -C ₃ H ₇
	Sulprofos	S	H	H	SCH ₃	H	H	OC ₂ H ₅	S- <i>n</i> -C ₃ H ₇
	Tolclofos-methyl	S	Cl	H	CH ₃	H	Cl	OCH ₃	OCH ₃
	Trichloronat	S	Cl	H	Cl	Cl	H	OC ₂ H ₅	C ₂ H ₅



Organophosphorus pesticides



Courtesy of Mr. Lembacher, Hipp N  hrmittel, Pfaffenhofen, Germany

Environmental pollutants

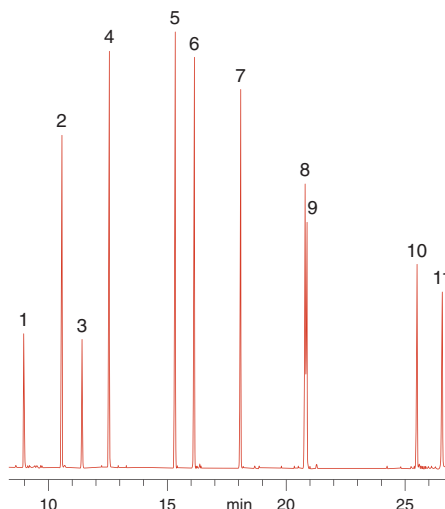
Analysis of organophosphorus pesticides

MN Appl. No. 213180

Column: OPTIMA® 35 MS, 30 m x 0.25 mm ID, 0.25 µm film, REF 726154.30,
max. temperature 360/370 °C
Injection: 1.0 µl, 250 °C, split 1:25
Carrier gas: 1.5 ml/min He
Temperature: 130 °C $\xrightarrow{6\text{ °C/min}}$ 300 °C
Detector: MSD

Peaks:

1. Demeton-O
2. Phorate
3. Demeton-S
4. Disulfoton
5. Trichloronate
6. Fenthion
7. Prothiophos
8. Sulprofos
9. Fensulfothion
10. Azinphos-methyl
11. Coumaphos



Pesticide structures: dinitroaniline herbicides

Structure	Compound	R ₁	R ₂	R ₃	R ₄
	Ethalfuralin	C ₂ H ₅	CH ₂ -C(CH ₃)=CH ₂	H	CF ₃
	Fluchloralin	C ₂ H ₄ Cl	<i>n</i> -C ₃ H ₇	H	CF ₃
	Isopropalin	<i>n</i> -C ₃ H ₇	<i>n</i> -C ₃ H ₇	H	CH(CH ₃) ₂
	Pendimethalin	H	CH(C ₂ H ₅) ₂	CH ₃	CH ₃
	Profluralin	<i>n</i> -C ₃ H ₇	cyclopropylmethyl	H	CF ₃
	Trifluralin	<i>n</i> -C ₃ H ₇	<i>n</i> -C ₃ H ₇	H	CF ₃

Pesticide structures: triazines

Structure	Compound	R ₁	R ₂	R ₃
	Isomethiozin	C(CH ₃) ₃	SCH ₃	N=CH-CH(CH ₃) ₂
	Metamitron	C ₆ H ₅	CH ₃	NH ₂
	Metribuzin	C(CH ₃) ₃	SCH ₃	NH ₂

	Hexazinone	—	—	—
--	------------	---	---	---

For more triazine structures see page 102.

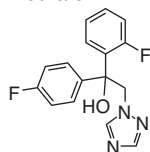
Pesticides



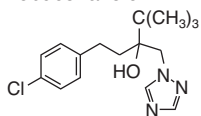
Pesticide structures: triazole and imidazole derivatives, conazoles

Structure	Compound	R ₁	R ₂	R ₃	R ₄
	Cyproconazole	H	OH	1-cyclopropylethyl	H
	Diclobutrazole	CH(OH)-C(CH ₃) ₃	H	H	Cl
	Hexaconazole	H	OH	<i>n</i> -C ₄ H ₉	Cl
	Myclobutanil	H	CN	<i>n</i> -C ₄ H ₉	H
	Penconazole	H	H	<i>n</i> -C ₃ H ₇	Cl
	Bitertanol	CH(OH)-C(CH ₃) ₃	C ₆ H ₅	–	–
	Triadimefon	CO-C(CH ₃) ₃	Cl	–	–
	Triadimenol	CH(OH)-C(CH ₃) ₃	Cl	–	–

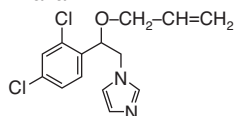
Flutriafol



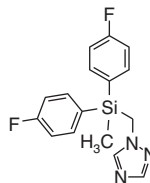
Tebuconazole



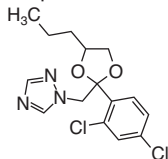
Imazalil



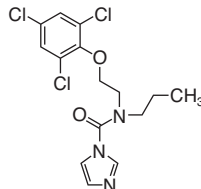
Flusilazole



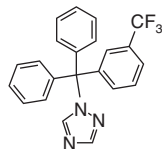
Propiconazole



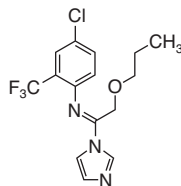
Prochloraz



Fluotrimazole



Triflumizole



Pesticide structures: urea derivatives

Structure	Compound	R ₁	R ₂	R ₃
	Fluometuron	F ₃ C-	H	CH ₃
	Tebuthiuron	(H ₃ C) ₃ C-	CH ₃	H

Environmental pollutants

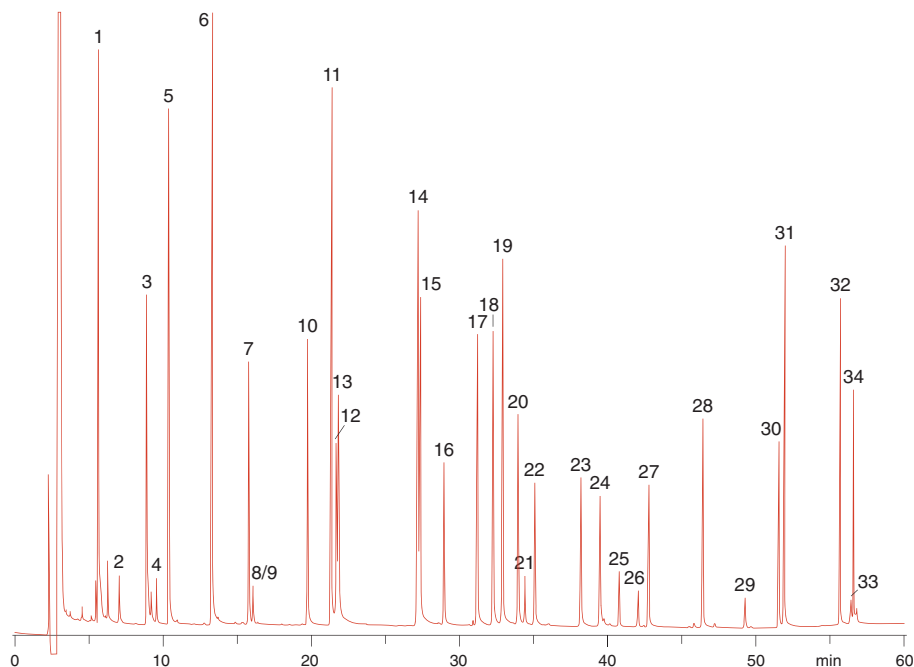
Analysis of a pesticide mixture

MN Appl. No. 210740

Column: OPTIMA® δ-6, 50 m x 0.20 mm ID, 0.20 µm film, REF 726465.50,
max. temperature 340/360 °C
Injection: 0.03 – 0.5 µg/ml depending on component, splitless, purge time 0.8 min, 270 °C
Carrier gas: 2.3 ml/min He
Temperature: 130 °C (1 min) $\xrightarrow{25\text{ °C/min}}$ 180 °C (5 min) $\xrightarrow{2\text{ °C/min}}$ 260 °C $\xrightarrow{6\text{ °C/min}}$ 310 °C (4 min)
Detector: PND 300 °C

Peaks:

- | | | |
|---|-----------------------------|---------------------|
| 1. Dichlorvos | 11. Dioxathion | 23. Methidathion |
| 2. Degradation product of ethion/
dioxathion | 12. Dimethoate | 24. Imazalil |
| 3. Mevinphos | 13. Etrinfos | 25. Flusilazole |
| 4. Propham | 14. Fenchlorphos/pirimiphos | 26. Myclobutanil |
| 5. Methacrifos | 15. Tolclofos-methyl | 27. Ethion |
| 6. Heptenophos | 16. Malathion | 28. Carbophenothion |
| 7. Sulfotep/
degradation product of diazinon | 17. Fenthion | 29. Tebuconazole |
| 8. Diphenylamine | 18. Bromophos-methyl | 30. Pyridaphenthion |
| 9. Chlorpropham | 19. Chlorthion | 31. Phenkapton |
| 10. Diazinon | 20. Chlorfenvinphos | 32. Azinphos-methyl |
| | 21. Penconazole | 33. Bitertanol |
| | 22. Quinalphos | 34. Azinphos-ethyl |



Courtesy of Mr. Grabher, CLUA, Sigmaringen, Germany



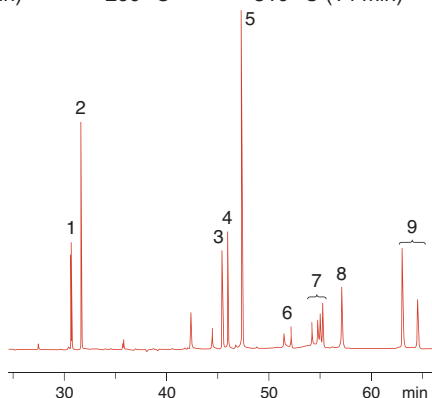
Analysis of pyrethroid insecticides

MN Appl. No. 210720

Column: OPTIMA® δ-6, 50 m x 0.20 mm ID, 0.20 µm film, REF 726465.50,
max. temperature 340/360 °C
Injection: 0.03 – 0.5 µg/ml depending on component, splitless, purge time: 0.8 min, 270 °C
Carrier gas: 2.3 ml/min He
Temperature: 130 °C (1 min) $\xrightarrow{25\text{ °C/min}}$ 180 °C (5 min) $\xrightarrow{2\text{ °C/min}}$ 260 °C $\xrightarrow{6\text{ °C/min}}$ 310 °C (14 min)
Detector: ECD 300 °C

Peaks:

1. Bioallethrin isomers
2. Chlorthion
3. Fenpropathrin + tetramethrin isomer 1
4. Tetramethrin isomer 2
5. Cyhalothrin
6. Permethrin isomers
7. Cyfluthrin isomers
8. Alphamethrin = α-Cypermethrin
9. Fenvalerate isomers



Courtesy of Mr. Grabher, CLUA, Sigmaringen, Germany

Pesticide structures: pyrethroids

Structure	Compound	R ₁	R ₂	R ₃	R ₄
	Cyfluthrin	CH=CCl ₂	H	CN	F
	Cyhalothrin	CH=CCl-CF ₃	H	CN	H
	α-Cypermethrin	CH=CCl ₂	H	CN	H
	Fenpropathrin	CH ₃	CH ₃	CN	H
	Permethrin	CH=CCl ₂	H	H	H
	Fenvalerate	4-Cl-phenyl	—	—	—
	Flucythrinate	4-CHF ₂ O-phenyl	—	—	—
	Fluvalinate	4-CF ₃ ,2-Cl-phenyl-NH	—	—	—
	Bioallethrin		—	—	—
	Tetramethrin		—	—	—

Analysis of acaricides and fungicides for peel-treatment MN Appl. No. 210210

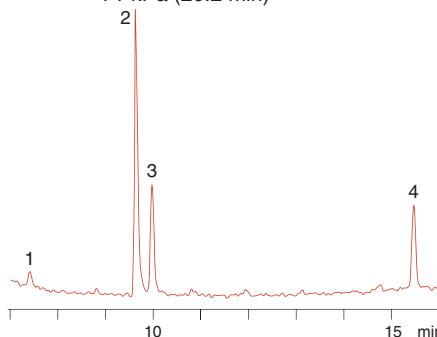
Column: OPTIMA® δ-3, 30 m x 0.25 mm ID, 0.25 µm film, REF 726420.30,
max. temperature 340/360 °C
Injection: 1.0 µl, 250 °C, 1 min splitless, then 1:25
Carrier gas: He
Pressure: 26.5 kPa (1 min) $\xrightarrow{7 \text{ kPa/min}}$ 42.9 kPa $\xrightarrow{1.1 \text{ kPa/min}}$ 71 kPa (20.2 min)
Temperature: 80 °C (1 min) $\xrightarrow{30 \text{ °C/min}}$ 150 °C
 $\xrightarrow{4.8 \text{ °C/min}}$ 280 °C (20 min)

Detector: MSD

Peaks:

1. Mecarbam
2. Imazalil
3. Chlorfenson
4. Tetradifon

Courtesy of K. Friedrichs,
Chem. Untersuchungsamt, Bielefeld, Germany



Pesticide structures: carboxylic acid amides

Structure	Compound	R ₁	R ₂	R ₃	R ₄
	Acetochlor	CH ₂ Cl	CH ₂ -OC ₂ H ₅	CH ₃	C ₂ H ₅
	Alachlor	CH ₂ Cl	CH ₂ -OCH ₃	C ₂ H ₅	C ₂ H ₅
	Butachlor	CH ₂ Cl	CH ₂ -O- <i>n</i> -C ₄ H ₉	C ₂ H ₅	C ₂ H ₅
	Fenfuram	<i>o</i> -methylfuranyl	H	H	H
	Metaxyl	CH ₂ -OCH ₃	CH(CH ₃)-CO-OCH ₃	CH ₃	CH ₃
	Metazachlor	CH ₂ Cl	CH ₂ -N $\begin{smallmatrix} \diagup \\ \diagdown \end{smallmatrix}$	CH ₃	CH ₃
	Methfuroxam	trimethylfuranyl	H	H	H
	Metolachlor	CH ₂ Cl	CH(CH ₃)-CH ₂ -OCH ₃	CH ₃	C ₂ H ₅
	Ofurace	CH ₂ Cl	γ -butyrolactone	CH ₃	CH ₃
	Propachlor	CH ₂ Cl	CH(CH ₃) ₂	H	H
	Chlorpropham	O- <i>i</i> -C ₃ H ₇	H	Cl	H
	Flamprop	C ₆ H ₅	CH(CH ₃)-COOH	Cl	F
	Pentanochlor	CH(CH ₃)- <i>i</i> -C ₃ H ₇	H	Cl	CH ₃
	Propanil	C ₂ H ₅	H	Cl	Cl
	Propham	O- <i>i</i> -C ₃ H ₇	H	H	H
	Diethyltoluamide	C ₂ H ₅	C ₂ H ₅	CH ₃	H
	Propyzamide	C(CH ₃) ₂ -C $\equiv$ CH	H	Cl	Cl
	Diphenamid	CH ₃	CH ₃	CH(C ₆ H ₅) ₂	
	Isocarbamid	CH ₂ -CH(CH ₃) ₂	H		
	Napropamide	C ₂ H ₅	C ₂ H ₅	CH(CH ₃)-O-1-naphthyl	



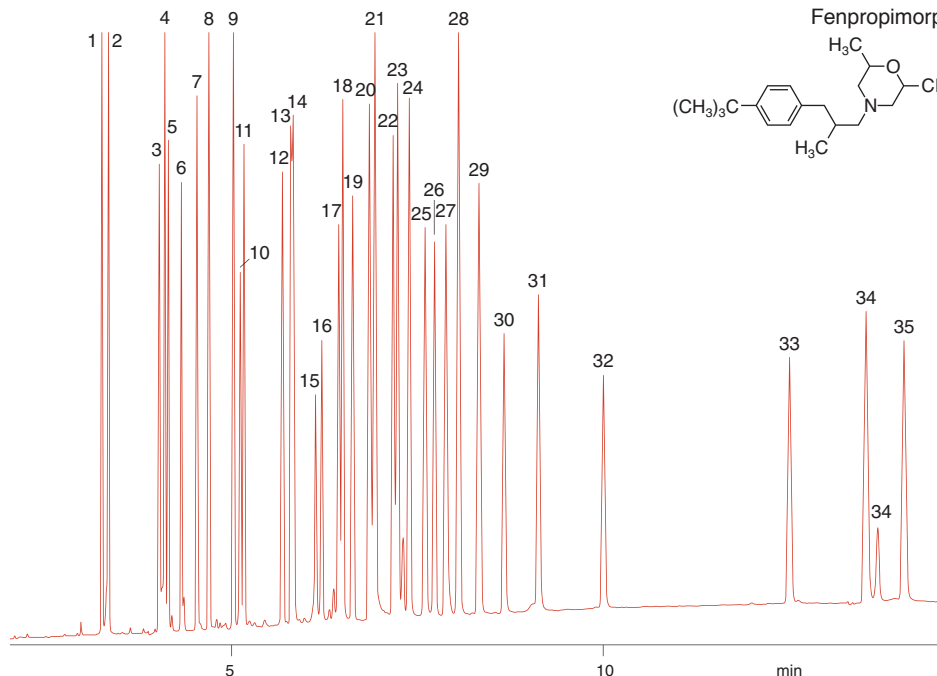
Analysis of fungicides

MN Appl. No. 200910

Column: OPTIMA® 1, 25 m x 0.32 mm ID, 0.25 µm film, REF 726302.25,
max. temperature 340/360 °C
Injection: 1 µl on column
Carrier gas: 50 kPa H₂
Temperature: 80 °C $\xrightarrow{40\text{ °C/min}}$ 200 °C $\xrightarrow{6\text{ °C/min}}$ 280 °C
Detector: PND 300 °C

Peaks: [ng]

1. Etridiazole [2]	13. Alachlor [5]	25. Flutriafol [2]
2. Prothiocard [2.5]	14. Metalaxyl [5]	26. Flurodifen [5]
3. Diphenylamine [2]	15. Dichlofluanid [2]	27. Hexaconazole [2]
4. Azobenzene [2]	16. Thiobencarb [2.5]	28. Myclobutanil [2]
5. Cycloate [2.5]	17. Methfuroxam [4]	29. Diclobutrazol [2]
6. Ethalfluralin [2]	18. Triadimefon [2]	30. Thioquinox [2]
7. Dicloran [2.5]	19. Fenpropimorph [5]	31. Ofurace [5]
8. Ethoxyquin [5]	20. Thiabendazole [2.5]	32. Nuairimol [2]
9. Benzo[h]quinoline [2]	21. Penconazole [2]	33. Fenarimol [2.5]
10. Fenfuram [2.5]	22. Triadimenol [2.5]	34. Bitertanol [2.5]
11. Fluchloralin [2.5]	23. Quinomethionate [2.5]	35. Prochloraz [2]
12. Vinclozolin [5]	24. Triflumizole [2.5]	



Courtesy of Mr. Lembacher, Hipp-Nährmittel, Pfaffenhofen, Germany

Environmental pollutants

Analysis of herbicides

MN Appl. No. 200770

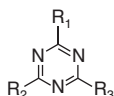
Column: OPTIMA® 1301, 25 m x 0.32 mm ID, 0.25 µm film, REF 726777.25,
max. temperature 300/320 °C
Injection: 0.5 µl, splitless, 280 °C
Carrier gas: 50 kPa H₂
Temperature: 80 $\xrightarrow{30\text{ °C/min}}$ 200 °C $\xrightarrow{4\text{ °C/min}}$ 270 °C
Detector: PND 310 °C

Peaks: [ng]

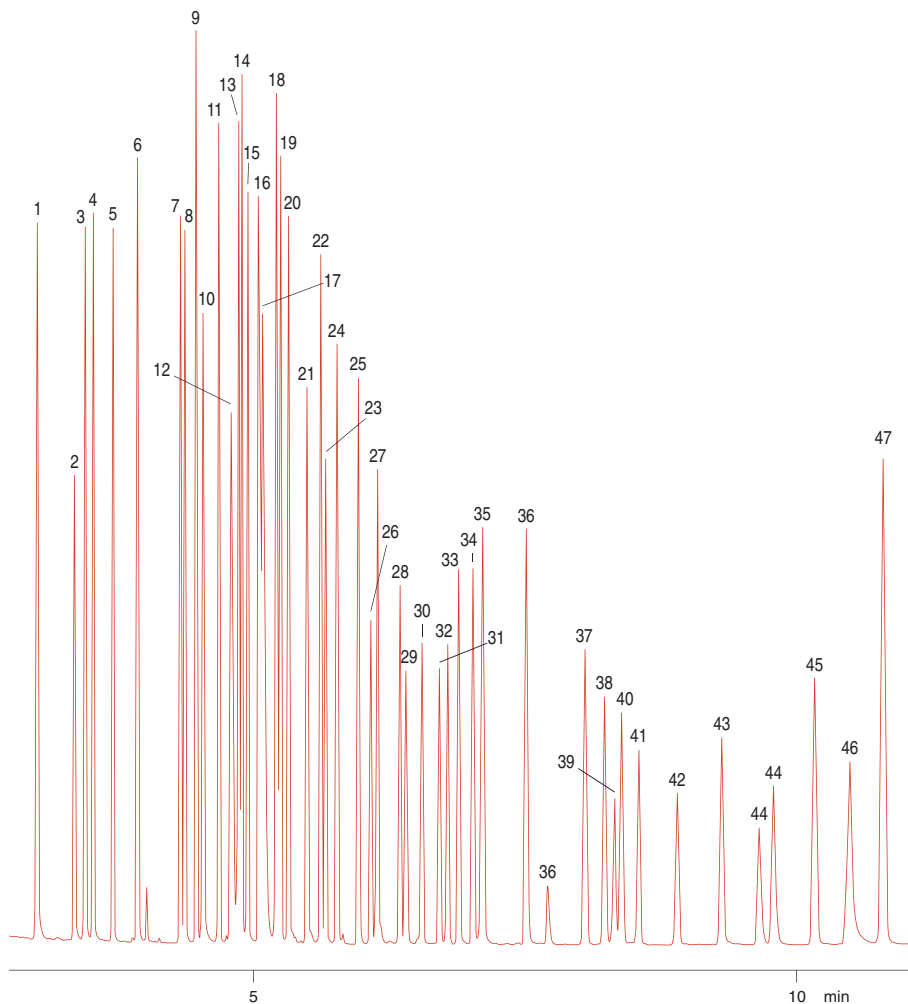
1. EPTC [5]	17. Prometon [2.5]	33. Dimethametryn [1.25]
2. Butylate [5]	18. Trietazine [1.25]	34. Metazachlor [2.5]
3. Vernolate [5]	19. Atrazine [1.25]	35. Bromacil [5]
4. Pebulate [5]	20. Sebuthylazine desethyl [1.25]	36. Triadimenol [5]
5. Propham [5]	21. Isocarbamid [2.5]	37. Oxadiazon [5]
6. Molinate [5]	22. Fluchloralin [5]	38. Methoprotryn [1.25]
7. Azobenzene [2.5]	23. Sebuthylazine [1.25]	39. Fluazifop-butyl [5]
8. Cycloate [5]	24. Desmetryn [1.25]	40. Thioquinox [2.5]
9. Diphenylamine [5]	25. Simetryn [1.25]	41. Fluorodifen [5]
10. Ethalfuralin [5]	26. Metalaxyl [5]	42. Flamprop-isopropyl [5]
11. Terbumeton desethyl [5]	27. Terbutryn [1.25]	43. Metamitron [2.5]
12. Desethyl-desisopropylatrazine [1.25]	28. Isomethiozin [1.25]	44. Propiconazole [5]
13. Desisopropylatrazine [1.25]	29. Metolachlor [5]	45. Fluotrimazole [5]
14. Desethylatrazine [1.25]	30. Propanil [5]	46. Lenacil [5]
15. Terbutylazine desethyl [1.25]	31. Pentanochlor [5]	47. Hexazinone [5]
16. Atraton [2.5]	32. Isopropalin [2.5]	

Pesticide structures: triazines

Compound	R ₁	R ₂	R ₃
Atrazine	Cl	NH-C ₂ H ₅	NH-CH(CH ₃) ₂
Cyanazine	Cl	NH-C ₂ H ₅	NH-C(CH ₃) ₂ -CN
Propazine	Cl	NH-CH(CH ₃) ₂	NH-CH(CH ₃) ₂
Sebuthylazine	Cl	NH-C ₂ H ₅	NH-(CH ₂) ₃ -CH ₃
Simazin	Cl	NH-C ₂ H ₅	NH-C ₂ H ₅
Terbutylazine	Cl	NH-C ₂ H ₅	NH-C(CH ₃) ₃
Trietazine	Cl	NH-C ₂ H ₅	N(C ₂ H ₅) ₂
Anilanzine	Cl	Cl	NH-(o-Cl-C ₆ H ₄)
Atraton	O-CH ₃	NH-C ₂ H ₅	NH-CH(CH ₃) ₂
Prometon	O-CH ₃	NH-CH(CH ₃) ₂	NH-CH(CH ₃) ₂
Secbumeton	O-CH ₃	NH-C ₂ H ₅	NH-CH(CH ₃)-C ₂ H ₅
Terbumeton	O-CH ₃	NH-C ₂ H ₅	NH-C(CH ₃) ₃
Ametryn	S-CH ₃	NH-C ₂ H ₅	NH-CH(CH ₃) ₂
Desmetryn	S-CH ₃	NH-CH ₃	NH-CH(CH ₃) ₂
Dimethametryn	S-CH ₃	NH-C ₂ H ₅	NH-CH(CH ₃)-CH(CH ₃) ₂
Methoprotryn	S-CH ₃	NH-CH(CH ₃) ₂	NH-(CH ₂) ₃ -OCH ₃
Prometryn	S-CH ₃	NH-CH(CH ₃) ₂	NH-CH(CH ₃) ₂
Simetryn	S-CH ₃	NH-C ₂ H ₅	NH-C ₂ H ₅
Terbutryn	S-CH ₃	NH-C ₂ H ₅	NH-C(CH ₃) ₃



Herbicides



Courtesy of Mr. Lembacher, Hipp N  hmittel, Pfaffenhofen, Germany

Environmental pollutants

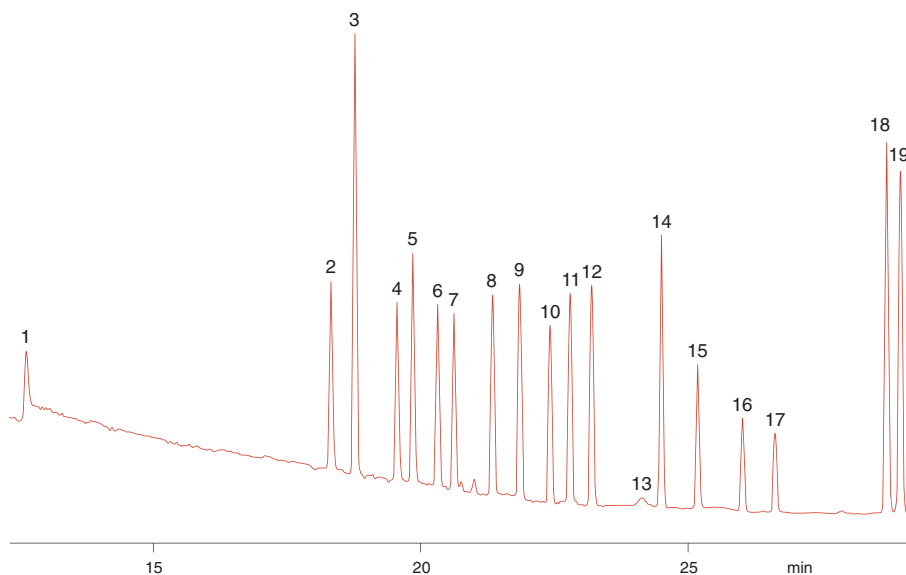
Analysis of triazine herbicides

MN Appl. No. 200820

Column: PERMABOND® FFAP, 25 m x 0.25 mm ID, 0.1 µm film, REF 723936.25,
max. temperature 220/240 °C
Injection: 2.5 µl, 1 min splitless, then split 1:50
Carrier gas: 250 kPa He
Temperature: 100 °C $\xrightarrow{5\text{ °C/min}}$ 240 °C
Detector: PND 250 °C

Peaks: [ng/ml]

- | | |
|------------------------|--------------------------------|
| 1. Azobenzene [4768] | 11. Simazin [856] |
| 2. Prometon [716] | 12. Ametryn [764] |
| 3. Terbumeton [1646] | 13. Cyanazine [728] |
| 4. Atraton [740] | 14. Simetryn [836] |
| 5. Propazine [820] | 15. Desethylatrazine [872] |
| 6. Terbutylazine [780] | 16. Metribuzin [744] |
| 7. Secbumeton [748] | 17. Desisopropylatrazine [784] |
| 8. Atrazine [828] | 18. Anilazine [3488] |
| 9. Prometryn [792] | 19. Methoprotryn [1544] |
| 10. Terbutryn [784] | |



Courtesy of J. Heering, Diplomarbeit FH Köln, Germany



Analysis of triazine herbicides

MN Appl. No. 210410

Column: OPTIMA® 5 Amine, 25 m x 0.20 mm ID, 0.35 µm film, REF 726355.25,
max. temperature 300/320 °C

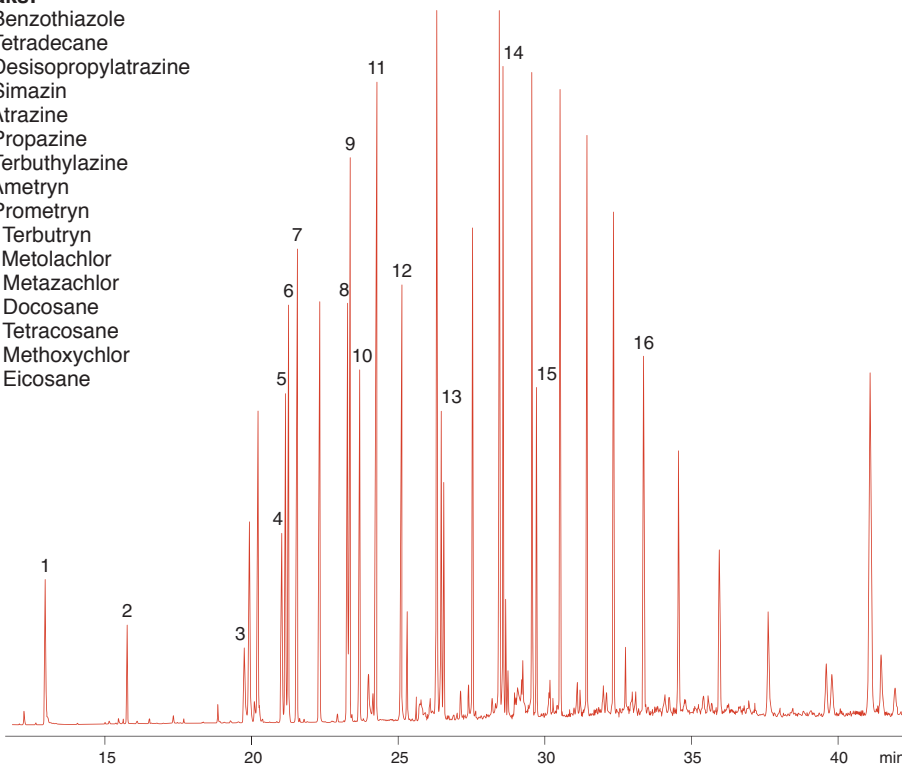
Carriergas: 24 cm/s He $\xrightarrow{8\text{ °C/min}}$ 300 °C

Temperature: 70 °C (3 min) $\xrightarrow{8\text{ °C/min}}$ 300 °C

Detector: MSD

Peaks:

1. Benzothiazole
2. Tetradecane
3. Desisopropylatrazine
4. Simazin
5. Atrazine
6. Propazine
7. Terbutylazine
8. Ametryn
9. Prometryn
10. Terbutryn
11. Metolachlor
12. Metazachlor
13. Docosane
14. Tetracosane
15. Methoxychlor
16. Eicosane



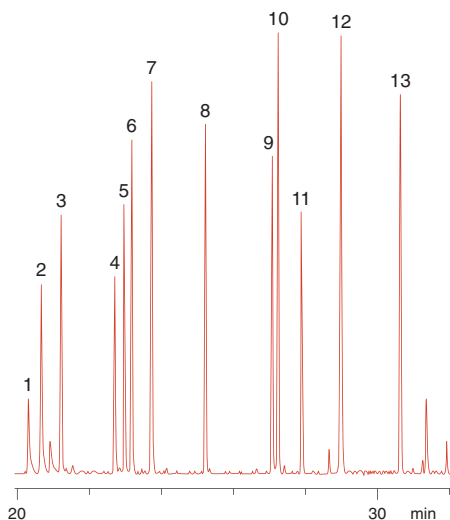
Environmental pollutants

Analysis of triazines MN App. No. 250210

Column: OPTIMA® 5 MS,
50 m x 0.25 mm ID,
0.25 µm film,
REF 726220.50,
max. temperature 340/360 °C
Injection: splitless
Carrier gas: 25 cm/s He $\xrightarrow{4\text{ °C/min}}$ 280 °C
Temperature: 120 °C (1 min) $\xrightarrow{4\text{ °C/min}}$ 280 °C
Detector: MSD

Peaks: [ppm]

1. Desisopropylatrazine [10.0]
2. Desethylatrazine [11.4]
3. Desethylterbutylazine [10.9]
4. Simazin [9.8]
5. Atrazine [10.2]
6. Propazine [10.8]
7. Terbutylazine [12.2]
8. Sebuthylazine [12.1]
9. Ametryn [9.7]
10. Prometryn [12.1]
11. Terbutryn [9.3]
12. Metolachlor [14.8]
13. Metazachlor [12.6]



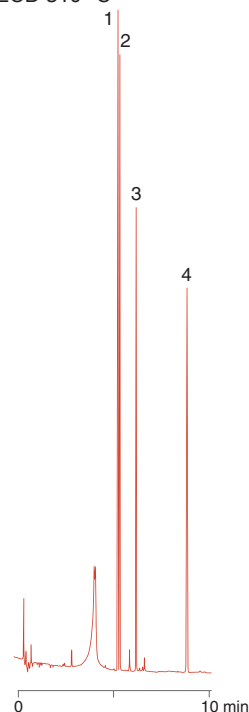
Analysis of dinitroaniline herbicides (EPA 627)

MN Appl. No. 200740

Column: OPTIMA® 1301,
25 m x 0.32 mm ID,
0.25 µm film,
REF 726777.25,
max. temperature 300/320 °C
Injection: 0.5 µl on column, 100 pg each
Carrier gas: 50 kPa H₂
Temperature: 80 °C $\xrightarrow{30\text{ °C/min}}$ 200 °C $\xrightarrow{4\text{ °C/min}}$ 270 °C
Detector: ECD 310 °C

Peaks:

1. Trifluralin
2. Ethalfluralin
3. Profluralin
4. Isopropalin



Courtesy of Mr. Lembacher, Hipp Nahrungsmittel,
Pfaffenhofen, Germany



Analysis of organo-nitrogen pesticides (EPA 633)

MN Appl. No. 250110

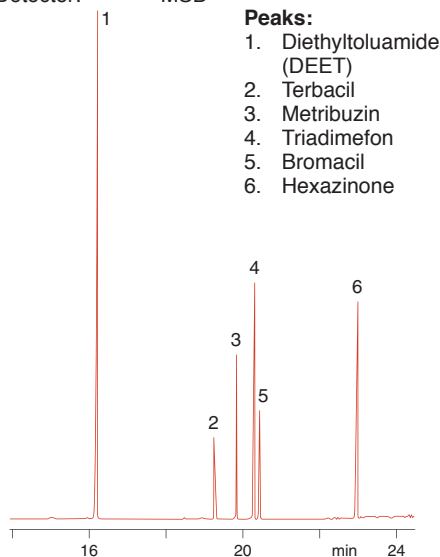
Column: OPTIMA® δ-3,
50 m x 0.25 mm ID,
0.25 μm film,
REF 726420.50,
max. temperature 340/360 °C

Injection: 1.0 μl, 3 s splitless

Carrier gas: 1.3 bar He

Temperature: 50 °C $\xrightarrow{10\text{ °C/min}}$ 220 °C $\xrightarrow{20\text{ °C/min}}$ 320 °C

Detector: MSD



Analysis of carbamate/urea pesticides (EPA 632)

MN Appl. No. 200860

Column: OPTIMA® 5,
25 m x 0.20 mm ID,
0.20 μm film,
REF 726857.25,
max. temperature 340/360 °C

Injection: 1.0 μl carbamate/urea pesticides (EPA 632), 10 μg/ml each in MeOH, 3 s splitless

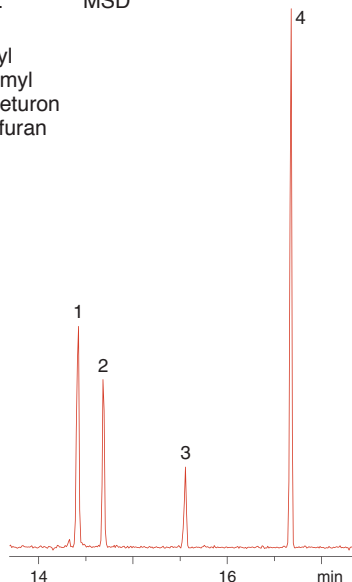
Carrier gas: 25 cm/s He

Temperature: 50 °C (5 min) $\xrightarrow{10\text{ °C/min}}$ 220 °C $\xrightarrow{20\text{ °C/min}}$ 330 °C

Detector: MSD

Peaks:

1. Oxamyl
2. Methomyl
3. Fluometuron
4. Carbofuran



Pesticide structures: uracil derivatives

Structure	Compound	R ₁	R ₂
	Bromacil	Br	CH(CH ₃)-C ₂ H ₅
	Terbacil	Cl	C(CH ₃) ₃
	Lenacil	-	-

Pesticide structures: oxime carbamates

Structure	Compound	R ₁
	Methomyl	CH ₃
	Oxamyl	CO - N(CH ₃) ₂

Environmental pollutants

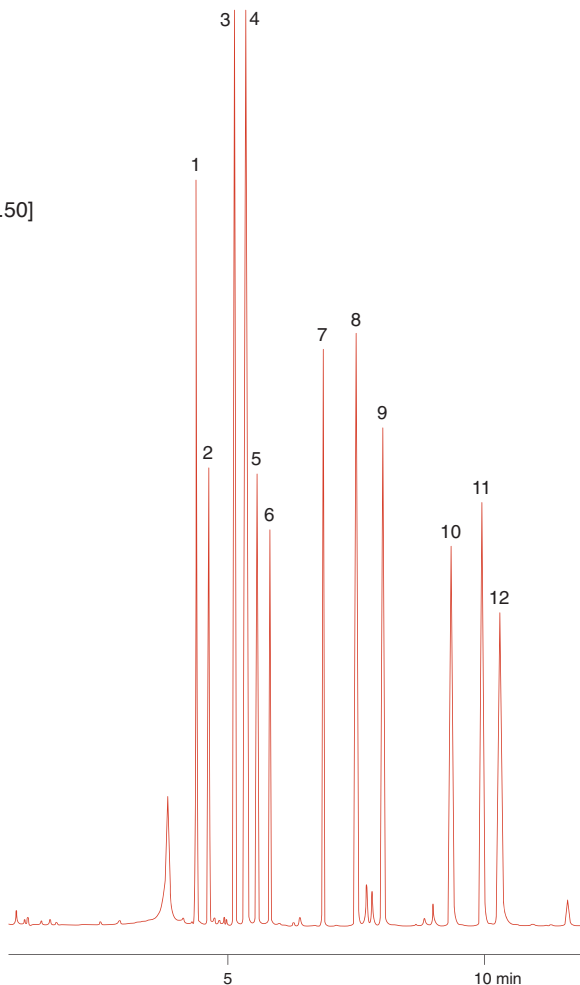
Analysis of phenoxy-carboxylic acid herbicides

MN Appl. No. 201060

Column: OPTIMA® 1301, 25 m x 0.32 mm ID, 0.25 µm film, REF 726777.25,
max. temperature 300/320 °C
Injection: 0.5 µl on column
Carrier gas: 70 kPa H₂
Temperature: 80 $\xrightarrow{30\text{ °C/min}}$ 220 °C $\xrightarrow{4\text{ °C/min}}$ 270 °C
Detector: ECD 310 °C

Peaks: [ng]

1. Chlorfenprop-methyl [0.25]
2. 2,4-D methyl ester [0.25]
3. 2,4,5-TP methyl ester [0.125]
4. 2,4,5-T methyl ester [0.125]
5. 2,4-D isobutyl ester [0.25]
6. 2,4-D 1-butyl ester [0.25]
7. Alodan (int. std.)[0.05]
8. Dichlorprop 2-ethyl 1-hexyl ester [0.50]
9. 2,4-D isooctyl ester [0.50]
10. 2,4,5-T isooctyl ester [0.125]
11. Dichlorprop methyl ester [0.50]
12. 2,4,5-T octyl ester [0.50]



Courtesy of Mr. Lembacher, Hipp N  hrmittel, Pfaffenhofen, Germany

Phenoxycarboxylic acid pesticides



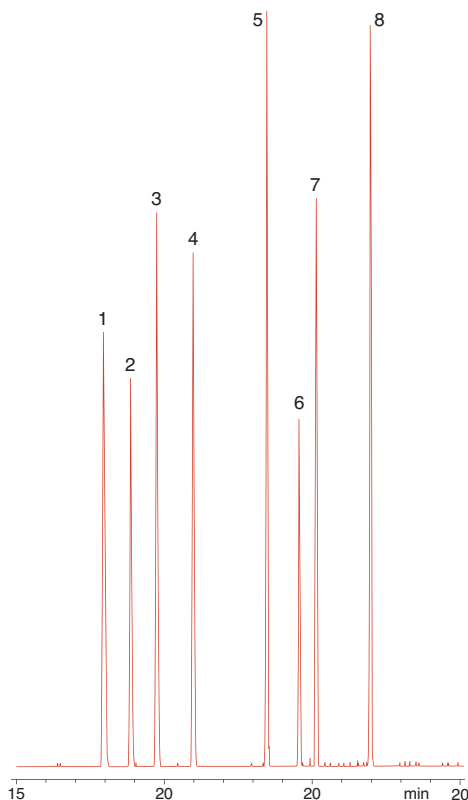
Analysis of phenoxycarboxylic acid herbicides (DIN 38407-14)

MN Appl. No. 213280

Column: OPTIMA® 35 MS,
30 m x 0.25 mm ID,
0.25 µm film, REF 726154.30,
max. temperature 360/370 °C
Injection: 2 µl, 280 °C, analytes deriva-
tised with TMSH
split 10 ml/min
Carrier gas: 1 bar He
Temperature: 145 °C $\xrightarrow{3\text{ °C/min}}$ 220 °C
Detector: MSD

Peaks: methyl esters of

- | | |
|----------------|-------------|
| 1. Mecoprop | 5. Fenoprop |
| 2. MCPA | 6. MCPB |
| 3. Dichlorprop | 7. 2,4,5-T |
| 4. 2,4-D | 8. 2,4-DB |



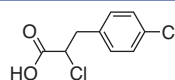
Pesticide structures: phenoxycarboxylic acids

Compound	R ₁	R ₂	R ₃
Phenoxyacetic acid derivatives			
2,4-D	Cl	Cl	H
2,4,5-T	Cl	Cl	Cl
MCPA	CH ₃	Cl	H

Phenoxypropionic acid derivatives

2,4-DP = Dichlorprop	Cl	Cl	H
2,4,5-TP = Fenoprop	Cl	Cl	Cl
MCPPP = Mecoprop	CH ₃	Cl	H
Fluazifop	H		H

Chlorfenprop



Phenoxybutyric acid derivatives

2,4-DB	Cl	Cl	—
MCPB	CH ₃	Cl	—

Environmental pollutants

Analysis of a pesticide mixture

MN Appl. No. 210680

Column: OPTIMA® δ-3,
50 m x 0.20 mm ID,
0.20 μm film,
REF 726400.50,
max. temperature 340/360 °C

Injection: 2 μl, 8 ng/μl, splitless,
KAS 60 °C $\xrightarrow{12\text{ °C/s}}$ 250 °C
(2 min)

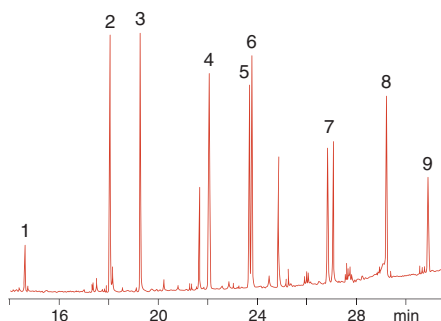
Carrier gas: 50 ml/min He, 35 psi

Temperature: 100 °C (12 min) $\xrightarrow{15\text{ °C/min}}$
220 °C (1 min) $\xrightarrow{5\text{ °C/min}}$ 340 °C
(10 min)

Detector: MSD

Peaks:

1. Acephate
2. Pirimicarb
3. Metalaxyl
4. Metazachlor
5. α-Endosulfan
6. Profenofos
7. Propiconazole
8. Methoxychlor
9. Phosalone



Courtesy of Mr. Heinzler, Staatl. Medizinal-,
Lebensmittel- und Veterinär-Untersuchungsamt,
Kassel, Germany

Analysis of organochlorine pesticides

MN Appl. No. 210840

Column: OPTIMA® δ-6,
30 m x 0.25 mm ID,
0.25 μm film, REF 726470.30,
max. temperature 340/360 °C

Injection: 10 μg/ml in acetone

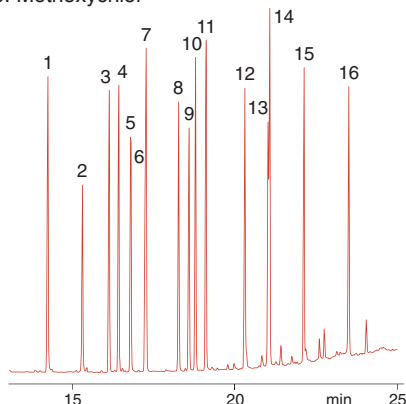
Carrier gas: H₂

Temperature: 60 °C (2 min) $\xrightarrow{30\text{ °C/min}}$ 150 °C
 $\xrightarrow{8\text{ °C/min}}$ 300 °C (6 min)

Detector: MSD (ion trap)

Peaks:

1. α-BHC
2. Quintozene
3. β-BHC
4. Heptachlor
5. δ-BHC
6. Chlorothalonil
7. Aldrin
8. Isodrin
9. *cis*-Heptachlor epoxide
10. *trans*-Heptachlor epoxide
11. *p,p'*-DDE
12. Dieldrin
13. Endrin
14. *o,p'*-DDT
15. *p,p'*-DDT
16. Methoxychlor



Courtesy of Mrs. Geilen, Berg. Wasser- u.
Umweltlabor, Wuppertal Germany

Pesticides



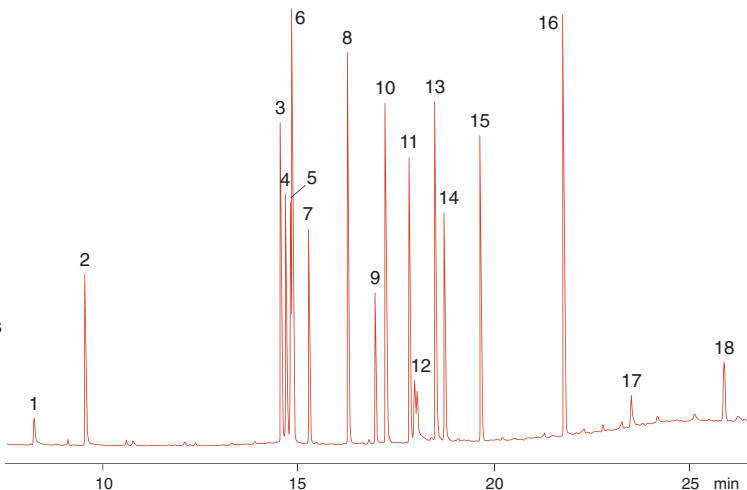
Column and conditions as in application 210840

MN Appl. No. 210830

Plant-protective pesticides

Peaks:

1. Carbofuran
2. Dichlobenil
3. Propazine
4. Atrazine
5. Simazin
6. Terbutylazine
7. γ -BHC
8. Alachlor
9. Parathion-methyl
10. Metolachlor
11. Parathion-ethyl
12. Bromacil
13. Chlorfenvinphos
14. Metazachlor
15. α -Endosulfan
16. β -Endosulfan
17. Chloridazon
18. Azinphos-ethyl

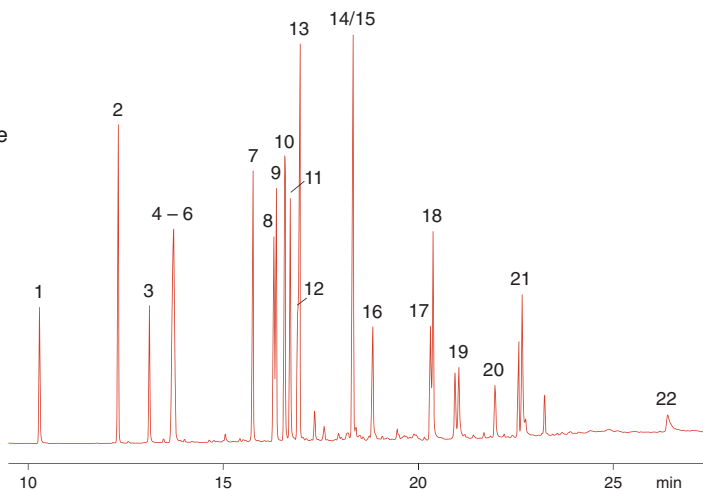


MN Appl. No. 210850

Organonitrogen pesticides

Peaks:

1. Protham
2. Trifluralin
3. Chlorpropham
4. Desethylatrazine
5. Desethylterbutylazine
6. Desisopropylatrazine
7. Sebutylazine
8. Vinclozolin
9. Desmetryn
10. Prometryn
11. Ametryn
12. Metribuzin
13. Terbutryn
14. Cyanazine
15. Pendimethalin
16. Triadimenol
17. Flusilazole
18. Methoprotryn
19. Cyproconazole
20. Metamitron
21. Tebuconazole
22. Prochloraz



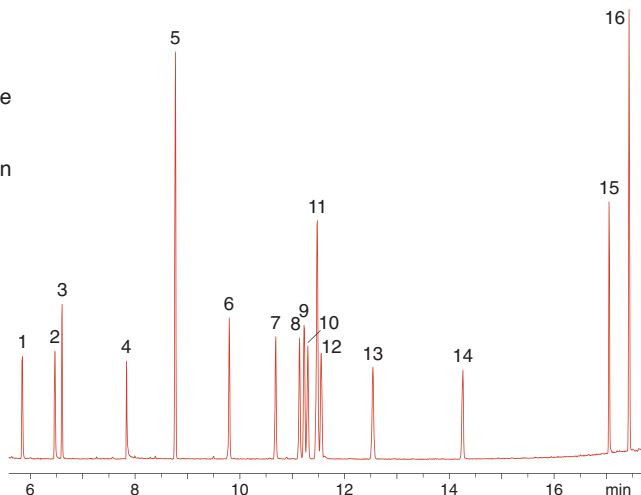
Courtesy of Mrs. Geilen, Berg. Wasser- u. Umweltlabor, Wuppertal Germany

Separation of pesticides and aromatic hydrocarbons (EPA 526 mix) MN Appl. No. 212820

Column: OPTIMA® 5 MS Accent, 30 m x 0.25 mm ID, 0.25 µm film, REF 725820.30, max. temperature 340/360 °C
Sample: EPA method 526 mix, 10 ppm (20 ppm internal standard)
Injection: 1 µl splitless, hold 0.3 min, 300 °C
Carrier gas: 0.8 ml/min He
Temperature: 50 °C (1 min) $\xrightarrow{20\text{ °C/min}}$ 200 °C (5 min) $\xrightarrow{30\text{ °C/min}}$ 310 °C (3 min)
Detector: MSD 280 °C

Peaks:

1. Nitrobenzene
2. 2,4-Dichlorophenol
3. 1,3-Dimethyl-2-nitrobenzene
4. 2,4,6-Trichlorophenol
5. Acenaphthene-d₁₀ (int. st.)
6. Azobenzene (decomposition product of 1,2-diphenylhydrazine)
7. Prometon
8. Terbufos
9. Diazinon
10. Fonofos
11. Phenanthrene-d₁₀ (int. st.)
12. Disulfoton
13. Acetochlor
14. Cyanazine
15. Triphenyl phosphate
16. Chrysene-d₁₂



Pesticide structures: miscellaneous heterocyclic aromatic compounds

Structure	Compound	R ₁	R ₂	R ₃	R ₄	R ₅
	Fluorochloridone		H	CF ₃	H	H
	Iprodione		H	Cl	H	Cl
	Oxadiazon	(CH ₃) ₃ C-C(=O)-O-N=N-	Cl	H	Cl	O-CH(CH ₃) ₂
	Procymidone		H	Cl	H	Cl
	Vinclozolin		H	Cl	H	Cl



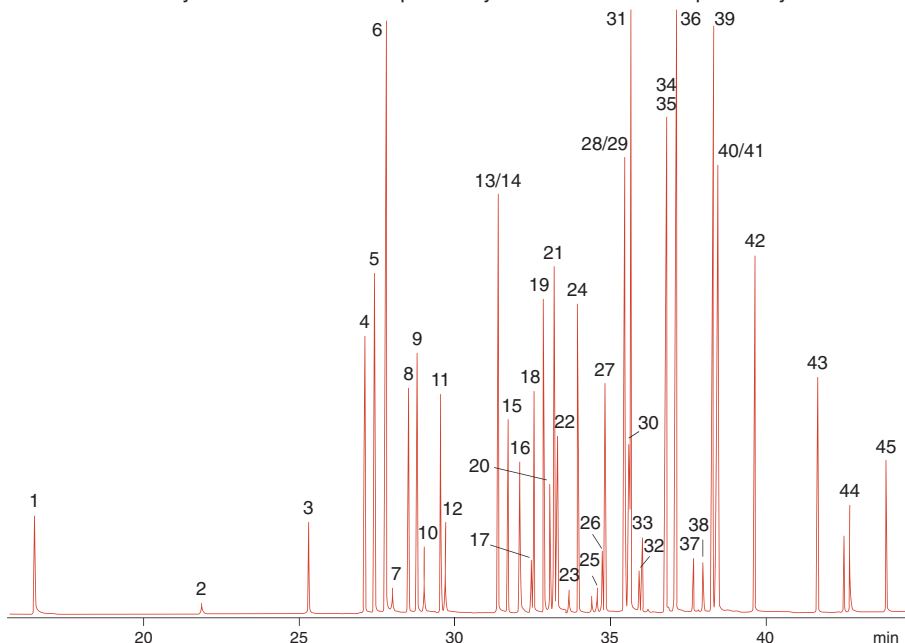
On-column analysis of a pesticide mixture

MN Appl. No. 210430

Column: OPTIMA® 5, 50 m x 0.25 mm ID, 0.25 µm film, REF 726056.50,
max. temperature 340/360 °C
Retention gap: Me-Sil deactivated, 1.6 m x 0.25 mm ID, REF 723706.10 (10 m)
Injection: 2 µl on column
Flow: 0.8 bar → 2.0 bar, 0.025 bar/min
Temperature: 60 °C (1 min) $\xrightarrow{5\text{ °C/min}}$ 270 °C $\xrightarrow{20\text{ °C/min}}$ 310 °C (8 min)
Detector: MSD

Peaks:

- | | | |
|----------------------------|--------------------------------------|---------------------------|
| 1. Dichlorvos | 16. Heptachlor | 31. <i>o,p'</i> -DDE |
| 2. Trichlorfon | 17. Fenitrothion | 32. α -Endosulfan |
| 3. Tecnazene | 18. Pirimiphos-methyl | 33. <i>cis</i> -Chlordane |
| 4. Sulfotep | 19. Malathion | 34. <i>p,p'</i> -DDE |
| 5. α -BHC | 20. Aldrin | 35. Dieldrin |
| 6. HCB | 21. Fenthion | 36. <i>o,p'</i> -DDD |
| 7. Dimethoate | 22. Chlorpyrifos-ethyl | 37. Endrin |
| 8. β -BHC | 23. Chlorthion | 38. β -Endosulfan |
| 9. γ -BHC | 24. Bromophos-methyl | 39. <i>p,p'</i> -DDD |
| 10. Quintozene | 25. <i>trans</i> -Heptachlor epoxide | 40. <i>o,p'</i> -DDT |
| 11. Diazinon | 26. <i>cis</i> -Heptachlor epoxide | 41. Ethion |
| 12. Disulfoton | 27. Chlorfenvinphos | 42. <i>p,p'</i> -DDT |
| 13. Chlorpyrifos-methyl | 28. Methidathion | 43. Methoxychlor |
| 14. Parathion-methyl | 29. <i>trans</i> -Chlordane | 44. Azinphos-methyl |
| 15. Demeton-S-methylsulfon | 30. Bromophos-ethyl | 45. Azinphos-ethyl |



Environmental pollutants

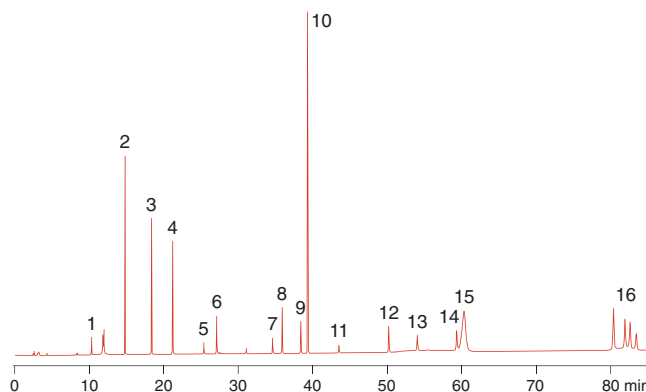
Analysis of pesticide mixtures

Column: OPTIMA® 1301, 60 m x 0.25 mm ID, 0.25 µm film, REF 726771.60,
max. temperature 300/320 °C
Injection: 3 µl, 0.1 ng/µl, 80 °C (1 min) → 250 °C (1 min), pulsed splitless
Carrier gas: 54 ml/min He
Temperature: 80 °C (2 min) $\xrightarrow{20\text{ °C/min}}$ 190 °C (12 min) $\xrightarrow{2\text{ °C/min}}$ 240 °C (23 min) $\xrightarrow{10\text{ °C/min}}$ 260 °C
(20 min)
Detector: ECD

MN Appl. No. 210630

Peaks:

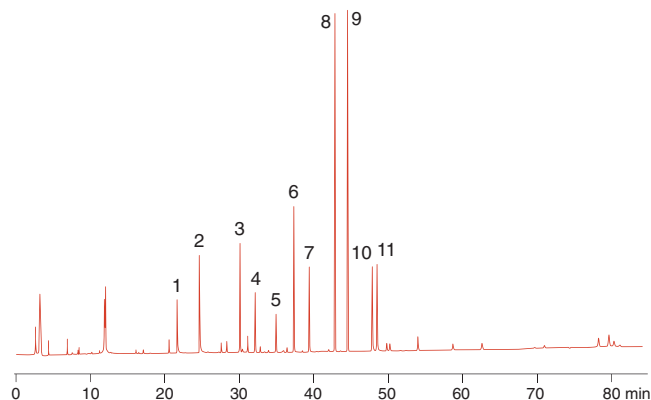
1. Phosphamidon
2. Tecnazene
3. α-BHC
4. Lindane
5. Chlorothalonil
6. Fenclorphos
7. Metazachlor
8. Bromophos-ethyl
9. Prothiophos
10. Dieldrin
- + profenofos
11. o,p'-DDT
12. Nuairimol
13. Bromopropylate
14. Bifenox
15. Tetradifon
16. – 19. Cypermethrin,
4 stereoisomers



MN Appl. No. 210640

Peaks:

1. Dimethoate
2. Propyzamide
3. Demeton-S-methyl
+ chlorpyrifos-ethyl
4. Parathion-ethyl
5. Quinalphos
- + penconazole
6. Procymidone
- + methidathion
7. Dieldrin
- + profenofos
8. Myclobutanil
9. PCB-128
- + β-endosulfan
- + ethion
10. PCB-153
11. Propiconazole



Courtesy of Mrs. Goda, Lebensmitteluntersuchungsamt, Braunschweig, Germany



Analysis of a pesticide mixture

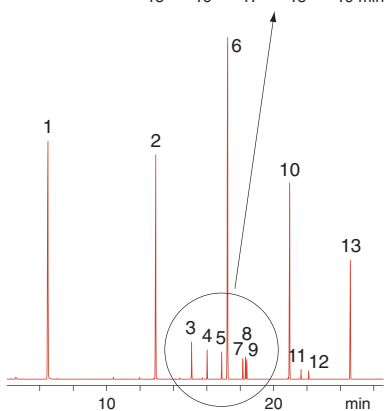
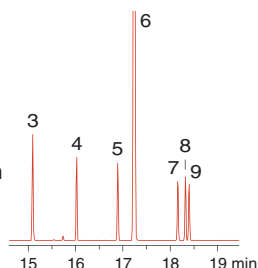
Sample: pesticide standard of Kanton Laboratory Schaffhausen (Switzerland)
 0.1 mg/ml or 0.01 mg/ml each
 Injection: 1.0 μ l, 3 s splitless
 Carrier gas: 25 cm/s He
 Temperature: 100 °C (3 min) $\xrightarrow{8\text{ °C/min}}$ 250 °C $\xrightarrow{10\text{ °C/min}}$ 320 °C
 Detector: MSD

MN Appl. No. 200920

Column: OPTIMA® 5,
 25 m x 0.20 mm ID,
 0.20 μ m film, REF 726857.25,
 max. temperature 340/360 °C

Peaks:

1. Dichlorvos
2. Naled
3. Chlorothalonil
4. Vinclozolin
5. Dichlofluanid
6. Chlorpyrifos
7. Captan
8. Folpet
9. Procymidone
10. Carbophenothion
11. Captafol
12. Iprodione
13. Coumaphos

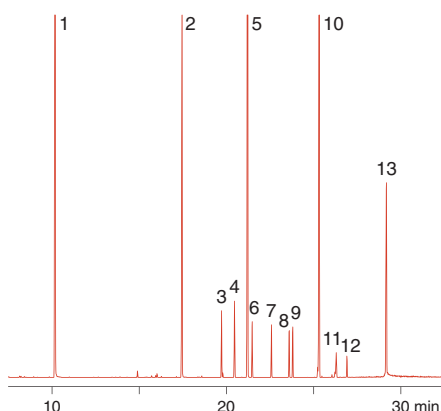


MN Appl. No. 200930

Column: OPTIMA® 17,
 25 m x 0.20 mm ID,
 0.20 μ m film, REF 726065.25,
 max. temperature: 320/340 °C

Peaks:

1. Dichlorvos
2. Naled
3. Vinclozolin
4. Chlorothalonil
5. Chlorpyrifos
6. Dichlofluanid
7. Procymidone
8. Captan
9. Folpet
10. Carbophenothion
11. Iprodione
12. Captafol
13. Coumaphos



Pesticide structures: phthalimide derivatives

Structure	Compound	R	Structure	Compound
	Captan	CCl ₃		Folpet
	Captafol	CCl ₂ -CCl ₂ H		

Environmental pollutants

Miscellaneous pesticide structures

Structure	Compound	R ₁	R ₂	Structure	Compound
	Chloridazon Norflurazon	H CH ₃	H CF ₃		Etridiazole
	Quinomethionat Thioquinox	CH ₃ H	O S		Fluridone
	Carboxin	—	—		Thiabendazole
	Ethoxyquin	—	—		Tricyclazole

Analysis of halogenated acetic acids (EPA 552.2)

MN Appl. No. 213210

Column: OPTIMA® 35 MS, 30 m x 0.25 mm ID, 0.25 µm film, REF 726154.30,
max. temperature 360/370 °C

Injection: 1 µl, split 1:20

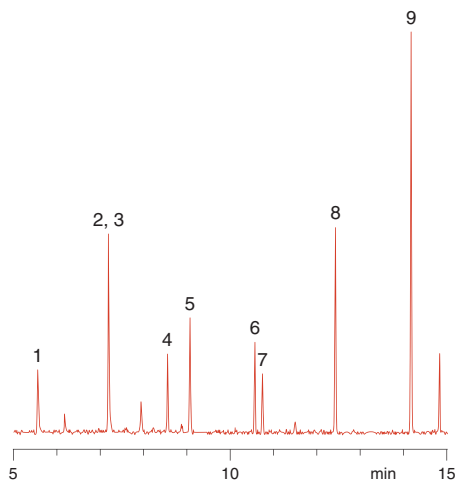
Carrier gas: 1 ml/min He

Temperature: 50 °C (3 min) $\xrightarrow{10\text{ °C/min}}$ 200 °C (10 min)

Detector: MSD

Peaks:

1. Chloroacetic acid methyl ester
2. Dichloroacetic acid methyl ester
3. Bromoacetic acid methyl ester
4. Trichloroacetic acid methyl ester
5. Bromochloroacetic acid methyl ester
6. Bromodichloroacetic acid methyl ester
7. Dibromoacetic acid methyl ester
8. Dibromochloroacetic acid methyl ester
9. Tribromoacetic acid methyl ester



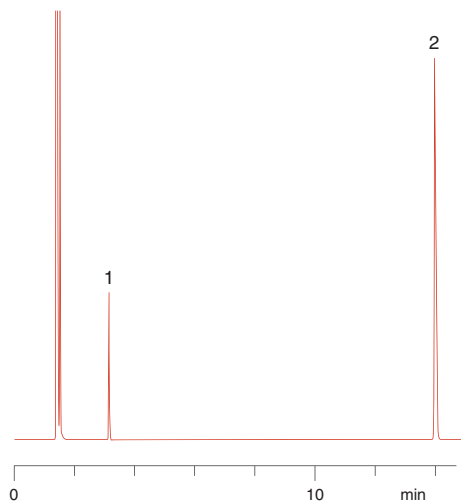


Analysis of acetic acid and trifluoroacetic acid MN Appl. No. 213250

Column: OPTIMA® FFAP,
25 m x 0.25 mm ID,
0.25 µm film, REF 726116.25,
max. temperature 250/260 °C
Injection: 1 µl, split 1:40
Carrier gas: 0.95 bar He
Temperature: 40 °C (5 min) $\xrightarrow{10\text{ °C/min}}$ 120 °C
Detector: FID 250 °C

Peaks:

1. Trifluoroacetic acid (TFA)
2. Acetic acid

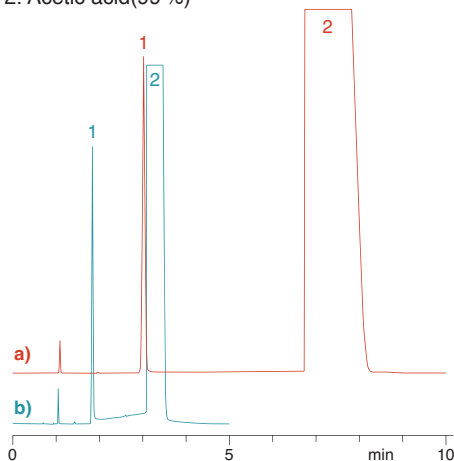


Analysis of acetic acid and acetic acid anhydride MN Appl. No. 250620

Column: PERMABOND® FFAP,
25 m x 0.53 mm ID,
1.0 µm film, REF 723555.25,
max. temperature 200/220 °C
Injection: 0.6 µl, split 1:100
Carrier gas: 0.4 bar He
Temperature: a) 100 °C, b) 130 °C
Detector: FID 200 °C

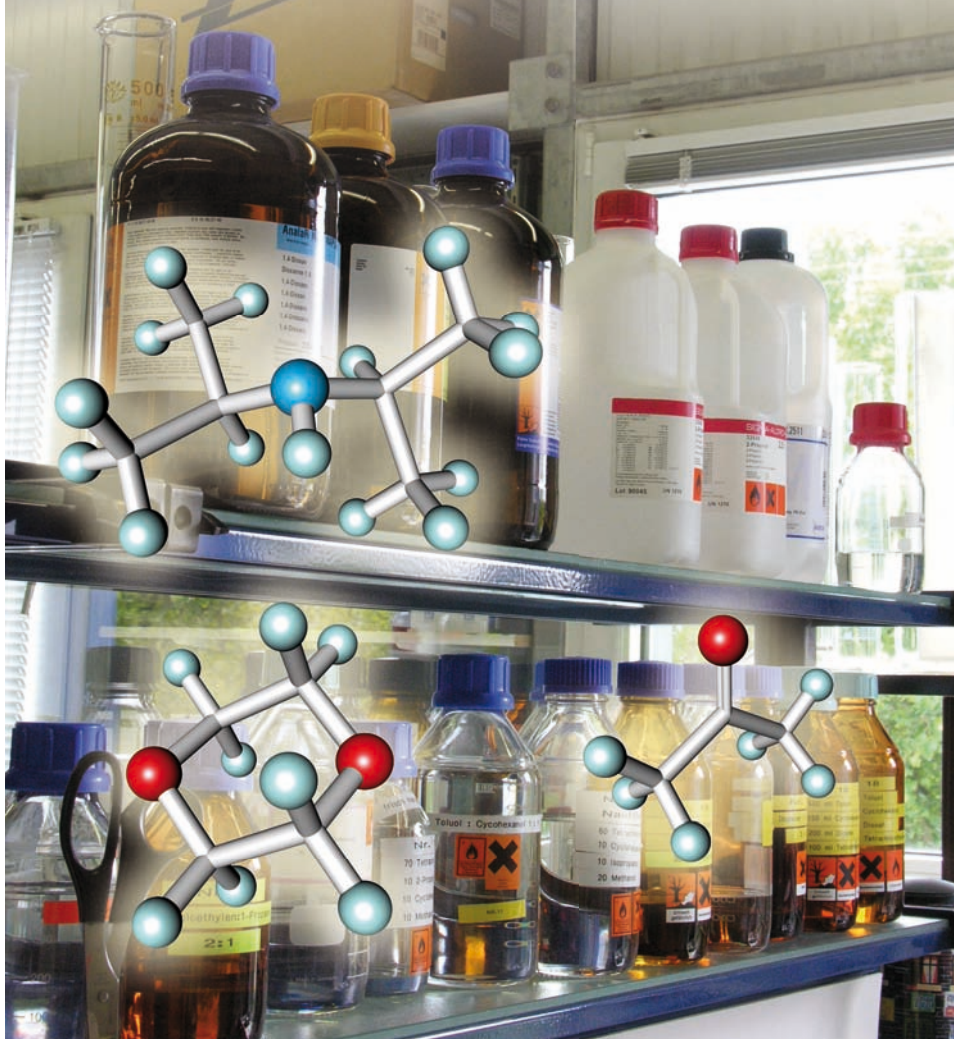
Peaks:

1. Acetic acid anhydride (1 %)
2. Acetic acid(99 %)



Although retention times are drastically decreased at higher temperatures, the title compounds are still perfectly separated at 130 °C with an even better peak shape.

Solvents Chemicals



Aromatic hydrocarbons



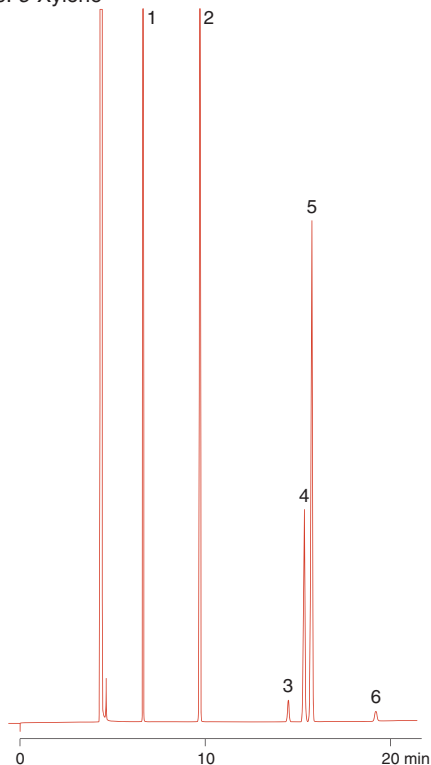
Analysis of aromatic hydrocarbons (BTX)

MN Appl. No. 200230

Column: OPTIMA® 210,
50 m x 0.25 mm ID,
0.5 µm film, REF 726874.50,
max. temperature 260/280 °C
Injection: 0.5 µl, split 105 ml/min
Carrier gas: 130 kPa N₂ (1.1 ml/min)
Temperature: 50 °C
Detector: FID 250 °C

Peaks:

1. Benzene
2. Toluene
3. Ethylbenzene
4. *p*-Xylene
5. *m*-Xylene
6. *o*-Xylene



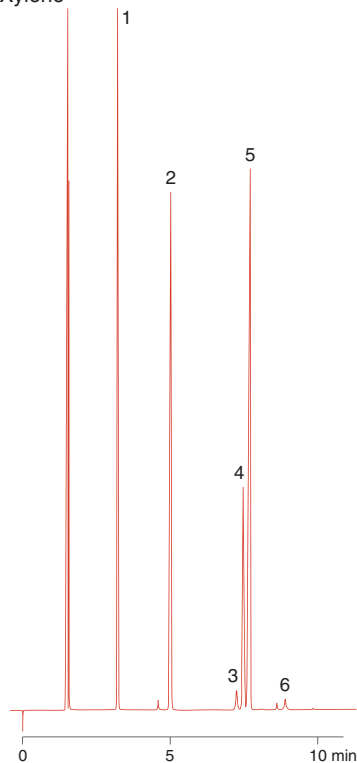
Analysis of aromatic hydrocarbons (BTX)

MN Appl. No. 200210

Column: PERMABOND® CW 20 M,
25 m x 0.25 mm ID,
0.25 µm film, REF 723060.25,
max. temperature 220/240 °C
Injection: 0.2 µl, split 1:50
Carrier gas: 0.8 bar N₂
Temperature: 40 °C (3 min) $\xrightarrow{5\text{ °C/min}}$ 80 °C
Detector: FID 260 °C

Peaks:

1. Benzene
2. Toluene
3. Ethylbenzene
4. *p*-Xylene
5. *m*-Xylene
6. *o*-Xylene



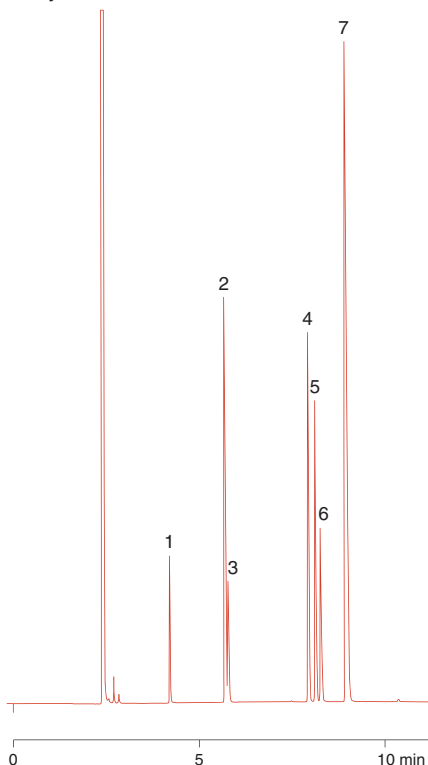
Analysis of aromatic hydrocarbons (BTX with toluene- d_8)

MN Appl. No. 200260

Column: FS-HYDRODEX β -PM,
50 m x 0.25 mm ID,
REF 723370.50,
max. temperature 230/250 °C
Split: 185 ml/min
Carrier gas: 120 kPa H_2 (2 ml/min)
Temperature: 60 °C $\xrightarrow{4\text{ °C/min}}$ 100 °C
Detector: FID 250 °C

Peaks in *n*-hexane:

1. Benzene
2. Toluene- d_8
3. Toluene
4. *p*-Xylene
5. *m*-Xylene
6. Ethylbenzene
7. *o*-Xylene



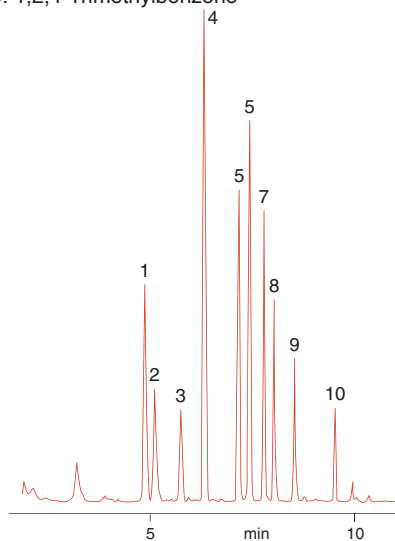
Analysis of aromatic hydrocarbons

MN Appl. No. 210590

Column: OPTIMA® δ -3,
30 m x 0.25 mm ID,
0.25 μ m film, REF 726420.30,
max. temperature 340/360 °C
Injection: 0.5 μ l, 250 °C, split 1:10
Carrier gas: He $\xrightarrow{12\text{ °C/min}}$ 200 °C
Temperature: 50 °C $\xrightarrow{12\text{ °C/min}}$ 200 °C
Detector: MSD

Peaks:

1. Ethylbenzene
2. *p*-Xylene
3. *o*-Xylene
4. Cumene
5. *n*-Propylbenzene
6. 4-Ethyltoluene
7. 1,3,5-Trimethylbenzene
8. 2-Ethyltoluene
9. 1,2,3-Trimethylbenzene
10. 1,2,4-Trimethylbenzene



Courtesy of Mr. B. Sievers, Riedel-de Haen AG,
Seelze, Germany

Aromatic hydrocarbons



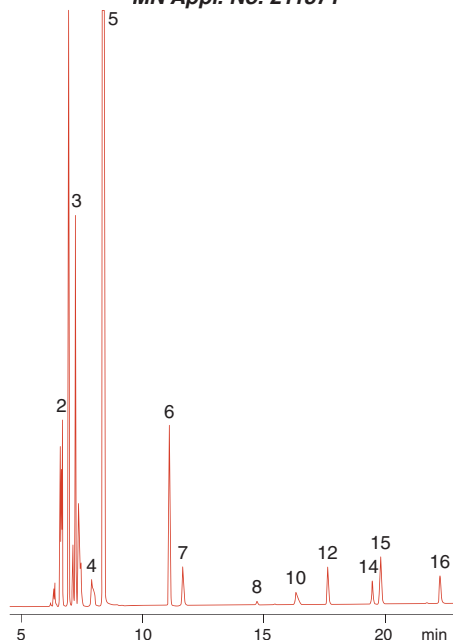
Analysis of aromatic solvents

Column: OPTIMA® 210,
50 m x 0.25 mm ID,
0.25 µm film, REF 726871.50,
max. temperature 260/280 °C
Injection: 0.2 µl, split 89 ml/min
Carrier gas: 60 kPa N₂
Temperature: 50 °C (10 min) $\xrightarrow{2\text{ °C/min}}$ 180 °C
(20 min)
Detector: FID 260 °C

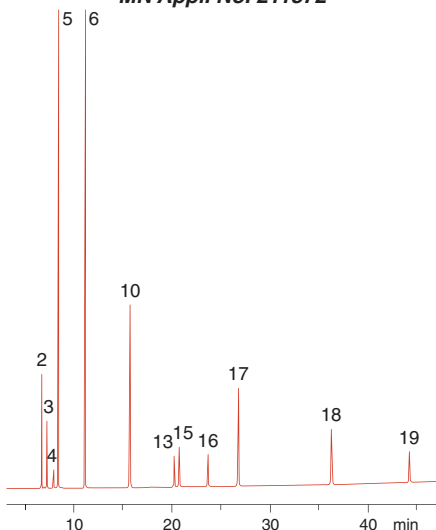
MN Appl. No. 211871

Peaks:

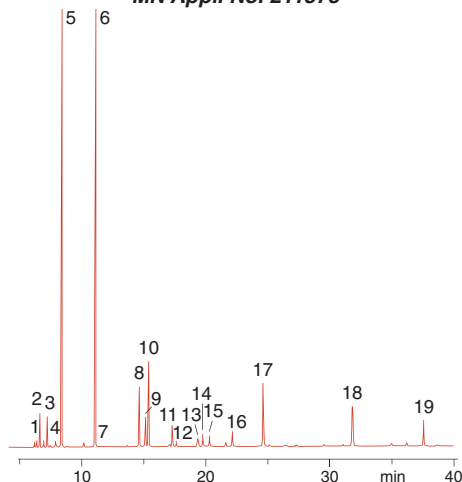
- | | |
|-----------------------------|---------------------------------|
| 1. <i>n</i> -Butane | 13. <i>n</i> -Propylbenzene |
| 2. <i>n</i> -Hexane | 14. <i>p</i> -Ethyltoluene |
| 3. Cyclohexane | 15. <i>m</i> -Ethyltoluene |
| 4. Methylcyclohexane | 16. 1,2,4-Trimethyl-
benzene |
| 5. Benzene | 17. Indane |
| 6. Toluene | 18. Tetralin |
| 7. <i>n</i> -Nonane | 19. Naphthalene |
| 8. Ethylbenzene | |
| 9. <i>p</i> -Xylene | |
| 10. <i>m</i> -Xylene | |
| 11. <i>o</i> -Xylene | |
| 12. <i>i</i> -Propylbenzene | |



MN Appl. No. 211872



MN Appl. No. 211873



Courtesy of Aral Aromatics GmbH, Gelsenkirchen, Germany



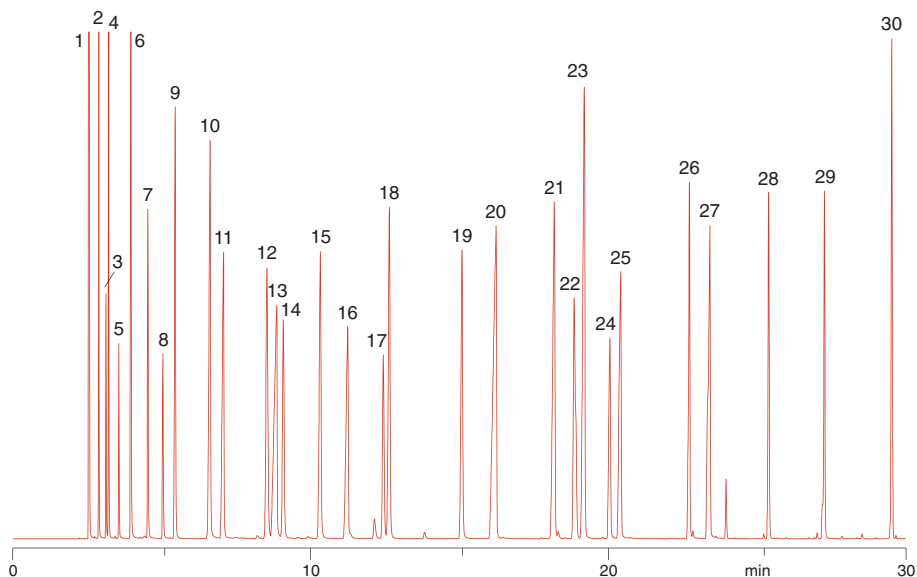
Analysis of solvents

MN Appl. No. 201390

Column: OPTIMA® 1, 60 m x 0.32 mm ID, 1.0 µm film, REF 726323.60,
max. temperature 340/360 °C
Sample: solvent mixture, courtesy of J. Lutz, Alcan Rorschach, Switzerland
Injection: 0.4 µl, split 1:60
Carrier gas: 120 kPa H₂
Temperature: 50 °C (9 min) $\xrightarrow{4\text{ °C/min}}$ 90 °C $\xrightarrow{14\text{ °C/min}}$ 280 °C (2 min)
Detector: FID 300 °C

Peaks:

- | | |
|----------------------------|------------------------------------|
| 1. Methanol | 16. 1-Ethoxy-2-propanol |
| 2. Ethanol | 17. Toluene |
| 3. Acetone | 18. <i>i</i> -Butyl acetate |
| 4. Propanol-2 | 19. <i>n</i> -Butyl acetate |
| 5. Methyl acetate | 20. 4-Hydroxy-4-methyl-2-pentanone |
| 6. Propanol-1 | 21. 1-Methoxy-2-propyl acetate |
| 7. Methyl ethyl ketone | 22. Xylene |
| 8. Ethyl acetate | 23. Cyclohexanone |
| 9. Isobutanol | 24. Ethyl glycol acetate |
| 10. Butanol-1 | 25. Butyl glycol |
| 11. 1-Methoxy-2-propanol | 26. Heptanol |
| 12. Isooctane | 27. Ethyldiglycol |
| 13. Ethylglycol | 28. Butyldiglycol |
| 14. Isoheptane | 29. Butyl glycol acetate |
| 15. Methyl isobutyl ketone | 30. Butyldiglycol acetate |





Analysis of solvents

MN Appl. No. 213150

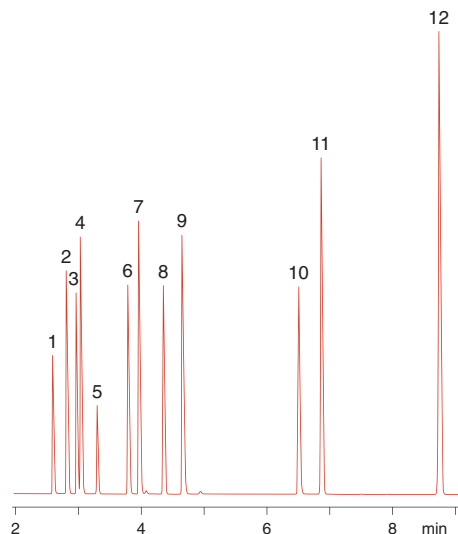
Column: OPTIMA® 1 MS Accent,
60 m x 0.32 mm ID,
0.5 µm film,
REF 725807.60,
max. temperature 340/360 °C
Injection: 1 µl, split 1:150
Carrier gas: 1 bar He
Temperature: 40 °C (5 min) $\xrightarrow{10\text{ °C/min}}$ 60 °C
(2 min)
Detector: FID 280 °C

Peaks:

1. Methanol
2. Ethanol
3. Acetone
4. Propanol-2
5. Dichloromethane
6. *t*-Butyl ethyl ether
7. Methyl ethyl ketone
8. Hexane
9. Tetrahydrofuran (THF)
10. Dioxane



11. Heptane
12. Toluene



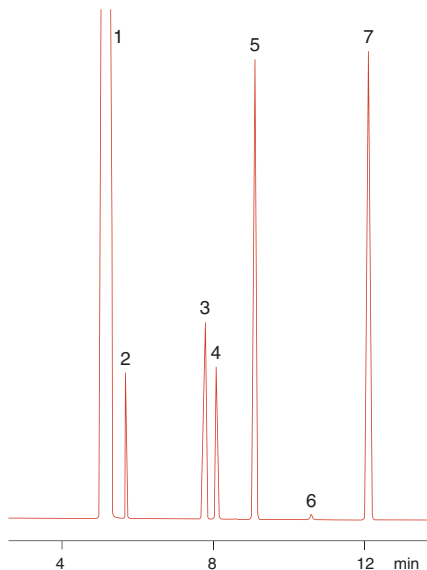
Analysis of volatile solvents

MN Appl. No. 210300

Column: OPTIMA® δ-3,
60 m x 0.32 mm ID,
1 µm film, REF 726442.60,
max. temperature 340/360 °C
Injection: 1 µl, split 1:75
Carrier gas: 1.2 bar He
Temperature: 34 °C (1 min) $\xrightarrow{4\text{ °C/min}}$
50 °C $\xrightarrow{6\text{ °C/min}}$ 100 °C $\xrightarrow{8\text{ °C/min}}$
200 °C (5 min)
Detector: FID

Peaks:

1. Methanol
2. Methyl formate
3. Methyl acetate
4. Acetonitrile
5. *n*-Hexane
6. Ethyl acetate
7. Cyclohexane



Analysis of solvents MN Appl. No. 210260

Column: OPTIMA® δ-3,
60 m x 0.32 mm ID,
1 μm film, REF 726442.60,
max. temperature 340/360 °C

Injection: 0.5 μl, split 1:100

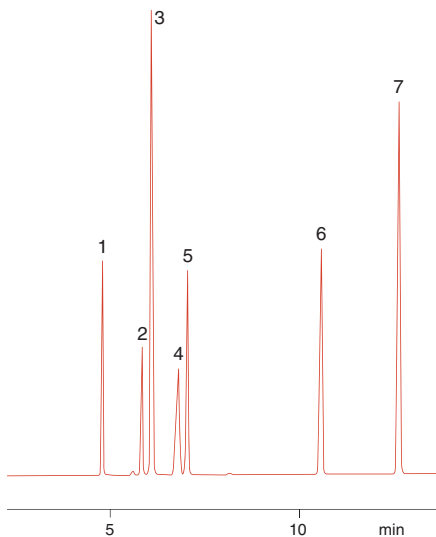
Carrier gas: 1.2 bar He $\xrightarrow{6\text{ °C/min}}$

Temperature: 60 °C (1 min) $\xrightarrow{240\text{ °C}}$
(5 min)

Detector: FID

Peaks:

1. Methanol
2. Ethanol
3. Pentane
4. Propanol-2
5. Acetone
6. Ethyl acetate
7. Benzene



Analysis of solvents by headspace GC MN Appl. No. 212960

Column: Optima® WAX,
30 m x 0.25 mm ID,
0.25 μm film, REF 726600.30,
max. temperature 250/260 °C

Sample: headspace: heat 1 ml water +
20 μl of each solvent to 80 °C
in a headspace vial for 15 min

Injection: 240 °C, split 3.64:1

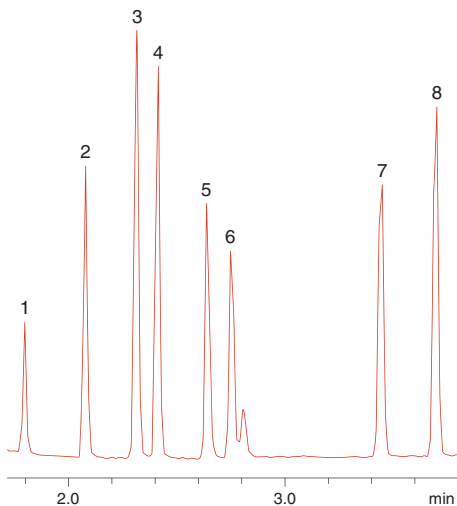
Carrier gas: 7.35 psi He, 1 ml/min

Temperature: 45 °C (1 min) $\xrightarrow{10\text{ °C/min}}$ 180 °C
(2 min) $\xrightarrow{30\text{ °C/min}}$ 240 °C (3 min)

Detector: MSD

Peaks:

1. Cyclohexane
2. Acetone
3. Tetrahydrofuran
4. Ethyl acetate
5. Dichloromethane
6. Propanol-2
7. Trichloromethane
8. Toluene



Courtesy of V. Cirimele, Laboratoire ChemTox,
Illkirch Graffenstaden, France

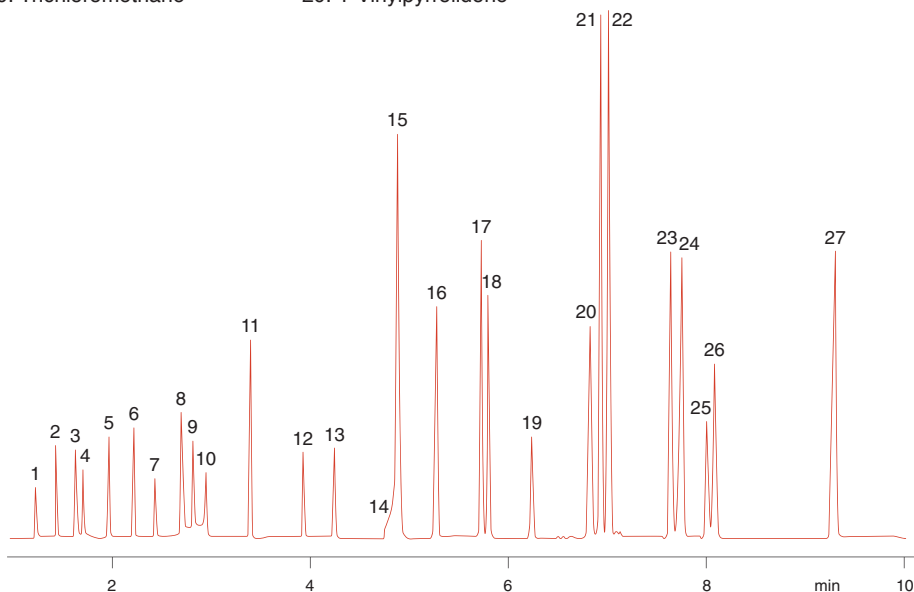


Rapid analysis of a complex solvent mixture MN Appl. No. 250500

Column: OPTIMA® δ -3, 30 m x 0.25 mm ID, 0.25 μ m film, REF 726420.30,
max. temperature 340/360 °C
Injection: 1.0 μ l, split 1:80
Carrier gas: 100 kPa He
Temperature: 35 °C (1.8 min) $\xrightarrow{25\text{ °C/min}}$ 50 °C $\xrightarrow{70\text{ °C/min}}$ 200 °C (15 min)
Detector: FID

Peaks:

- | | | |
|----------------------|------------------------|----------------------------|
| 1. Methanol | 11. Benzene | 21. Menthofuran + menthone |
| 2. Ethanol | 12. Trichloroethene | 22. Isomenthone + pulegone |
| 3. Propanol-2 | 13. 1,4-Dioxane | 23. Menthyl acetate |
| 4. Acetone | 14. Propylene glycol | 24. Carvone |
| 5. Dichloromethane | 15. Toluene + pentenol | 25. Anethole |
| 6. <i>n</i> -Hexane | 16. Heptanol-1 | 26. Anisaldehyde |
| 7. Vinyl acetate | 17. Limonene | 27. Dodecanol-1 |
| 8. Butanol-2 | 18. Cineole | |
| 9. Ethyl acetate | 19. Linalool | |
| 10. Trichloromethane | 20. 1-Vinylpyrrolidone | |



Courtesy of M. Bürkler, Solco Basel AG, Birsfelden, Switzerland

Solvents · Chemicals

Analysis of solvents and semi-volatiles

MN Appl. No. 212520

Column: OPTIMA® 624 LB, 30 m x 0.32 mm ID, 1.8 µm film, REF 726786.30
max. temperature 280/300 °C, retention gap Phe-Sil 0.5 m x 0.53 mm,
REF 723711.10

Injection: 1 µl, cold on-column, 10 ppm per substance in acetone

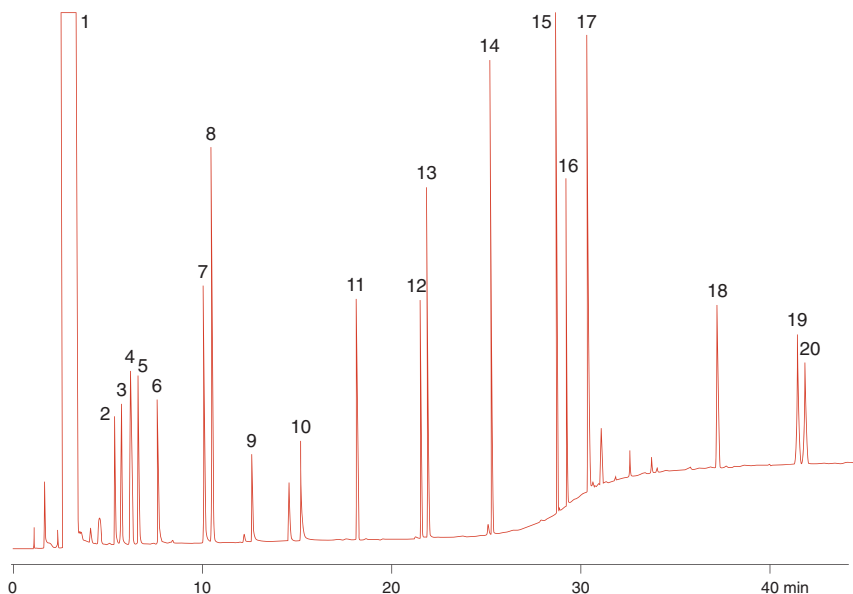
Carrier gas: 1.1 bar He

Temperature: 45 °C (3 min) $\xrightarrow{6\text{ °C/min}}$ 150 °C $\xrightarrow{18\text{ °C/min}}$ 300 °C (20 min)

Detection: FID 280 °C

Peaks:

- | | |
|-----------------------|---------------------------------------|
| 1. Acetone | 11. Decane |
| 2. Ethyl acetate | 12. Octanol-1 |
| 3. Tetrahydrofuran | 13. Acetophenone |
| 4. Cyclohexane | 14. Butyrophenone |
| 5. 2-Methylbutanol-2 | 15. Heptanophenone |
| 6. Butanol-1 | 16. Methoxy-5-indole |
| 7. Pyridine | 17. Dibenzylamine |
| 8. Toluene | 18. Methyl eicosanoate |
| 9. Dimethylformamide | 19. Methyl <i>cis</i> -13-docosenoate |
| 10. Dimethylsulfoxide | 20. Methyl docosanoate |



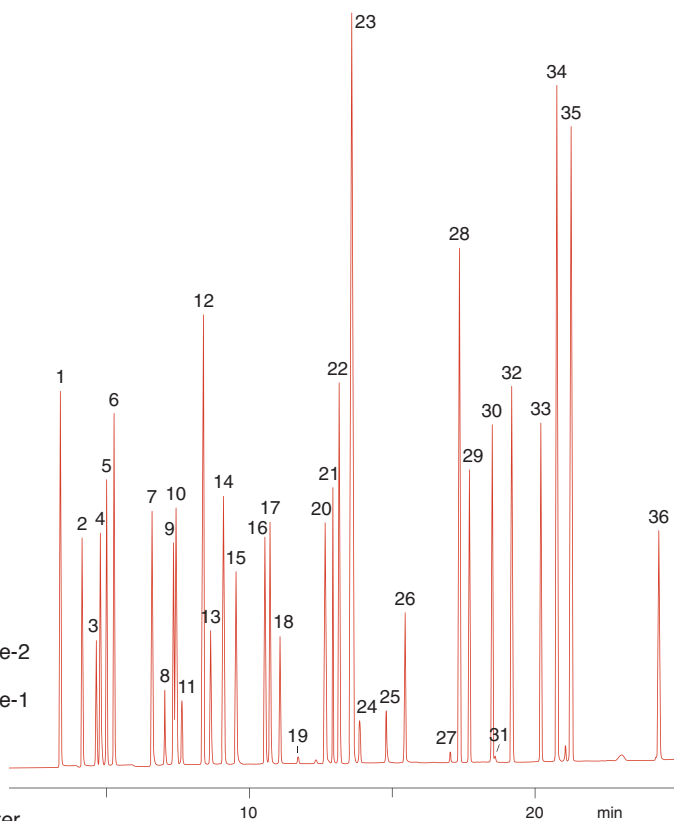


Analysis of solvents MN Appl. No. 201410

Column: OPTIMA® 1701, 50 m x 0.32 mm ID, 1.0 µm film, REF 726929.50,
max. temperature 300/320 °C
Injection: 0.1 µl, split 147 ml/min
Carrier gas: 150 kPa He (3 ml/min)
Temperature: 50 °C (5 min) $\xrightarrow{5\text{ °C/min}}$ 160 °C
Detector: FID 240 °C

Peaks:

1. Methanol
2. Ethanol
3. Acetone
4. Propanol-2
+ methyl acetate
5. Dichloromethane
6. *t*-Butanol
7. Propanol-1
8. Ethyl acetate
9. Butanone (MEK)
10. Trichloroethane
11. Trichloromethane
12. Benzene
13. *i*-Propyl acetate
14. *i*-Butanol
15. Methyl glycol
16. 1-Methoxypropanol-2
17. Butanol-1
18. *n*-Propyl acetate
19. Glycolic acid
20. Ethyl glycol
21. Toluene
22. *i*-Butyl methyl ketone
23. *i*-Butyl acetate
24. Tetrachloroethene
25. Ethoxypropanol
26. *n*-Butyl acetate
27. Ethylbenzene
28. *m*-Xylene + *p*-xylene
29. Methyl glycol acetate
30. 1-Methoxypropyl acetate-2
31. *o*-Xylene
32. 2-Methoxypropyl acetate-1
33. Ethyl glycol acetate
34. Butyl glycol
+ ethoxypropyl acetate
35. Cyclohexanone
36. Glycolic acid *n*-butyl ester



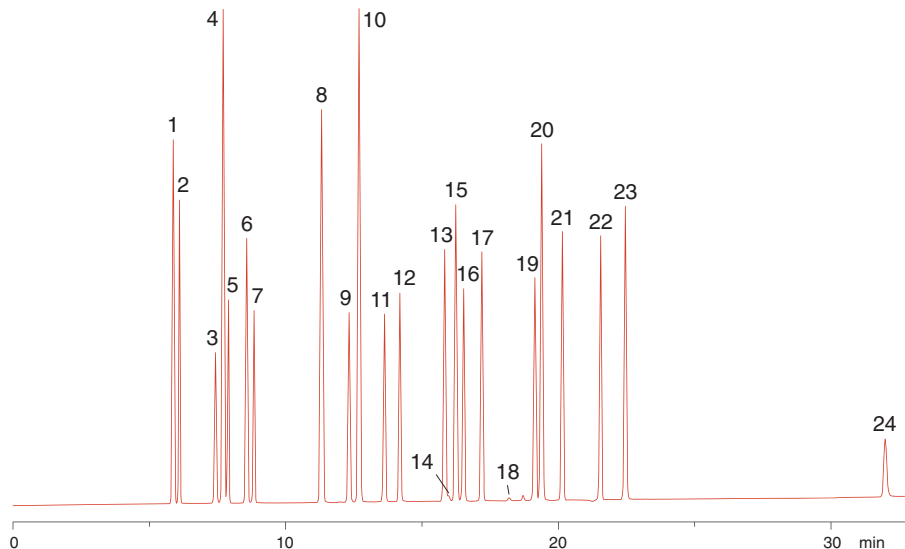
Analysis of solvents

MN Appl. No. 201420

Column: PERMABOND® CW 20 M, 50 m x 0.32 mm ID, 0.5 µm film, REF 723296.50,
max. temperature 220/240 °C
Injection: 0.2 µl, split 90 ml/min
Carrier gas: 1 bar N₂
Temperature: 60 °C (5 min) $\xrightarrow{5\text{ °C/min}}$ 210 °C
Detector: FID 250 °C

Peaks:

- | | |
|---------------------------------|----------------------------------|
| 1. Acetone | 13. 1-Methoxypropanol-2 |
| 2. Methyl acetate | 14. Ethylbenzene |
| 3. Ethyl acetate | 15. Butanol-1 + <i>p</i> -xylene |
| 4. Methanol + isopropyl acetate | 16. <i>m</i> -Xylene |
| 5. Butanone (MEK) | 17. Ethoxypropanol |
| 6. Propanol-2 | 18. <i>o</i> -Xylene |
| 7. Ethanol | 19. 1-Methoxypropyl acetate-2 |
| 8. Isobutyl methyl ketone | 20. Ethyl glycol |
| 9. Propanol-1 | 21. 2-Methoxypropyl acetate-1 |
| 10. Toluene | 22. Ethyl glycol acetate |
| 11. <i>n</i> -Butyl acetate | 23. Cyclohexanone |
| 12. <i>i</i> -Butanol | 24. Ethylene glycol |





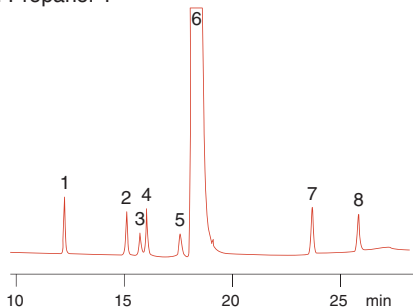
Analysis of solvent residues in ethanolic tinctures

MN Appl. No. 210880

Column: PERMABOND® CW 20 M,
60 m x 0.32 mm ID,
0.5 µm film, REF 723296.60,
max. temperature 220/240 °C
Injection: 1 µl, split 50 ml/min
Carrier gas: 50 kPa N₂
Temperature: 60 °C (19 min) $\xrightarrow{5\text{ °C/min}}$ 100 °C
Detector: FID 250 °C

Peaks:

1. Acetone
2. Ethyl acetate
3. Methanol
4. Methyl ethyl ketone
5. Propanol-2
6. Ethanol
7. Isobutyl methyl ketone
8. Propanol-1



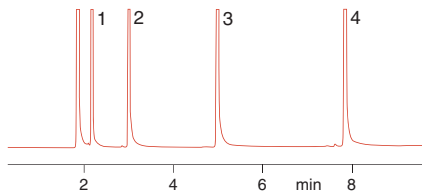
Analysis of glycols

MN Appl. No. 210950

Column: OPTIMA® δ-3,
30 m x 0.25 mm ID,
0.25 µm film, REF 726420.30,
max. temperature 340/360 °C
Injection: 1 µl, split 100 ml/min
Carrier gas: 70 kPa He
Temperature: 150 °C $\xrightarrow{10\text{ °C/min}}$ 250 °C
(15 min)
Detector: FID

Peaks:

1. Diethylene glycol
2. Triethylene glycol
3. Tetraethylene glycol
4. Pentaethylene glycol



Courtesy of Mr. Fischer, Pharmaz. Fabrik Evers
& Co. GmbH, Hamburg, Germany

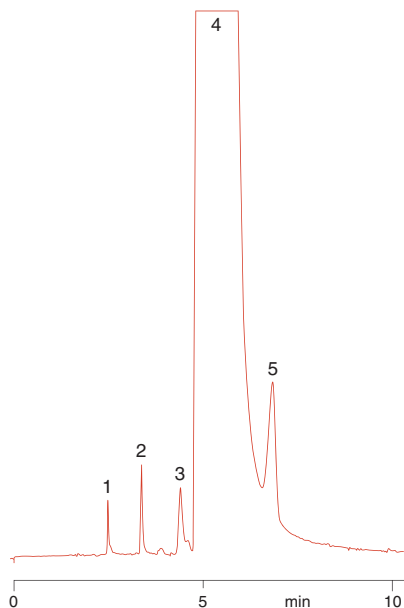
Analysis of alcohols in dichloromethane

MN Appl. No. 201460

Column: OPTIMA® 5,
25 m x 0.32 mm ID,
5.0 µm film,
REF 726934.25,
max. temperature 300/320 °C
Injection: 2 µl, split 30 ml/min
Carrier gas: 1 bar N₂
Temperature: 60 °C
Detector: FID 240 °C

Peaks:

1. Methanol
2. Ethanol
3. Propanol-2
4. Dichloromethane
5. Propanol-1



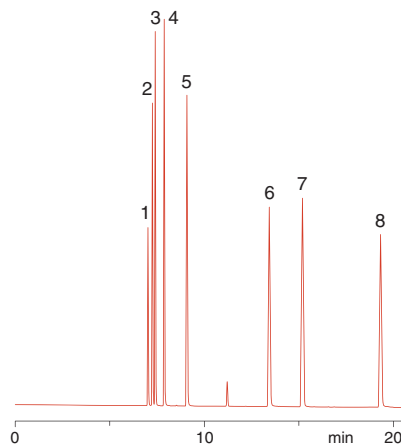
Analysis of alcohols

MN Appl. No. 201430

Column: OPTIMA® 17,
50 m x 0.32 mm ID,
0.5 µm film, REF 723744.50,
max. temperature 320/340 °C
Injection: 1 µl, split 1:50
Carrier gas: 50 kPa N₂
Temperature: 50 °C $\xrightarrow{4\text{ °C/min}}$ 200 °C
Detector: FID 250 °C

Peaks:

1. Methanol
2. Ethanol
3. Propanol-2
4. Propanol-1
5. Butanol-1
6. Hexanol-1
7. Cyclohexanol
8. Octanol-1



Alcohols



Headspace analysis of alcohols

MN Appl. No. 212630

Column: OPTIMA® 624,
60 m x 0.32 mm ID,
1.80 µm film, REF 726787.60,
max. temperature 280/300 °C

Injection: headspace 200 °C, sample
temperature 60 °C for 20 min,
needle temperature 70 °C,
transfer line 80 °C

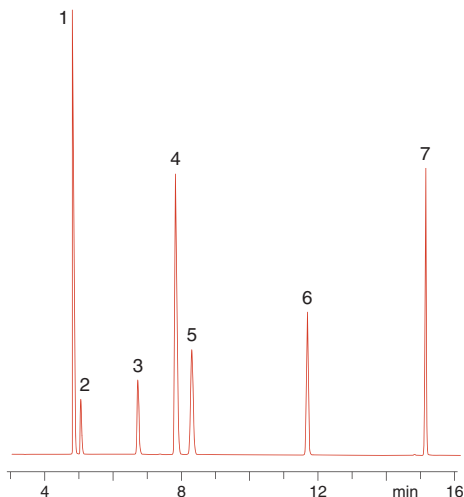
Carrier gas: 130 kPa He $\xrightarrow{20\text{ °C/min}}$ 200 °C

Temperature: 40 °C (5 min) $\xrightarrow{20\text{ °C/min}}$ 200 °C

Detector: FID 200 °C

Peaks:

1. Acetaldehyde
2. Methanol
3. Ethanol
4. Acetone
5. Propanol-2
6. Propanol-1
7. Butanol-1



Courtesy of Mr. Kress, INSTAND e.V., Düsseldorf, Germany

Headspace analysis of alcohols

MN Appl. No. 212940

Column: OPTIMA® WAX,
30 m x 0.25 mm ID,
0.25 µm film, REF 726600.30,
max. temperature 250/260 °C

Sample: headspace: heat 1 ml water +
20 µl of each alcohol to 80 °C
in a headspace vial for 15 min

Injection: 250 °C, split 30:1

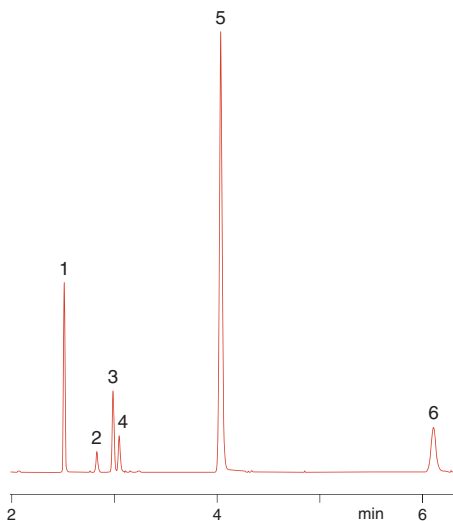
Carrier gas: 11.78 psi He, 1 ml/min

Temperature: 70 °C (10 min)

Detector: MSD

Peaks:

1. Acetone
2. Methanol
3. Propanol-2
4. Ethanol
5. Propanol-1
6. Butanol-1



Courtesy of V. Cirimele, Laboratoire ChemTox, Illkirch Graffenstaden, France

Headspace analysis of alcohols

MN Appl. No. 212950

Column: OPTIMA® WAX,
30 m x 0.25 mm ID,
0.25 µm film, REF 726600.30,
max. temperature 250/260 °C

Injection: headspace: heat 1 ml blood
+ 20 µl int. std. (50 mg/l
propanol-1) to 80 °C in a
headspace vial for 15 min;
240 °C, split 3.64:1

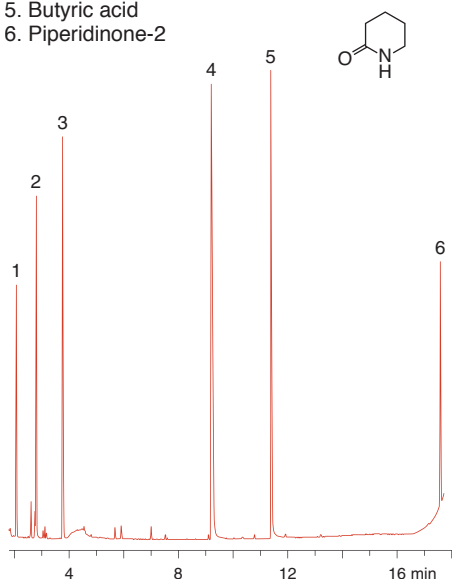
Carrier gas: 7.35 psi He, 1 ml/min

Temperature: 45 °C (1 min) $\xrightarrow{10\text{ °C/min}}$ 180 °C
(2 min) $\xrightarrow{30\text{ °C/min}}$ 240 °C (3 min)

Detector: MSD

Peaks:

1. Acetone
2. Ethanol
3. Propanol-1
4. Acetic acid
5. Butyric acid
6. Piperidinone-2



Courtesy of V. Cirimele, Laboratoire ChemTox,
Illkirch Graffenstaden, France



Analysis of a mixture of neutral and basic organic compounds

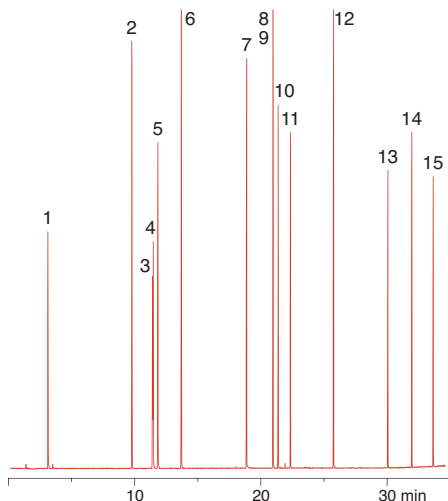
Sample: Base/Neutrals Mix 1, 200 µg/ml each in 2-propanol
 Injection: 1.0 µl, split 1:150
 Carrier gas: 25 cm/s He
 Detector: MSD

Peaks:

- | | |
|---------------------------------|---------------------------------|
| 1. N-Nitrosodimethylamine | 9. Diethyl phthalate |
| 2. s-Dichloroethyl ether | 10. N-Nitrosodiphenylamine |
| 3. Bis(1-chloroisopropyl) ether | 11. 4-Bromophenoxybenzene |
| 4. Bis(2-chloroisopropyl) ether | 12. Dibutyl phthalate |
| 5. N-Nitroso-di-n-propylamine | 13. Benzyl butyl phthalate |
| 6. Bis(2-chloroethoxy)methane | 14. Bis(2-ethylhexyl) phthalate |
| 7. Dimethyl phthalate | 15. Di-n-octyl phthalate |
| 8. 4-Chlorophenoxybenzene | |

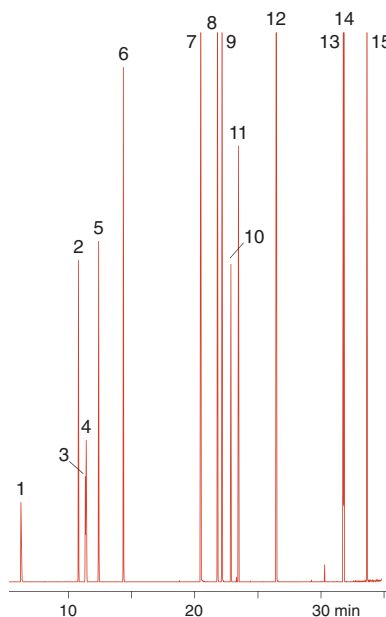
MN Appl. No. 201170

Column: OPTIMA® 5,
 25 m x 0.20 mm ID,
 0.20 µm film, REF 726857.25,
 max. temperature: 340/360 °C
 Temperature: 45 °C (5 min) $\xrightarrow{8\text{ °C/min}}$ 330 °C



MN Appl. No. 201180

Column: OPTIMA® 17,
 25 m x 0.20 mm ID,
 0.20 µm film, REF 726065.25,
 max. temperature 320/340 °C
 Temperature: 50 °C (3 min) $\xrightarrow{8\text{ °C/min}}$ 320 °C



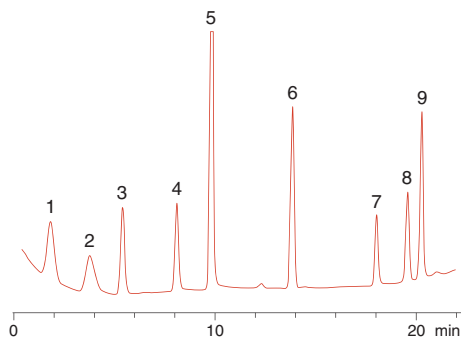
Analysis of organic pollutants from water-packaging materials

MN Appl. No. 212870

Column: OPTIMA® 17,
15 m x 0.53 mm ID,
1.0 µm film, REF 726747.15
max. temperature 300/320 °C
splitless, 270 °C
Injection: 10 ml/min N₂ 10 °C/min
Carrier gas: 50 °C (1 min) → 270 °C
Temperature: (10 min)
Detector: FID 300 °C

Peaks:

1. Chlorobenzene
2. Isopentyl acetate
3. Styrene
4. Phenol
5. Benzyl alcohol
6. Cyclohexylbenzene
7. Benzophenone
8. Diazinon
9. Dibutyl phthalate



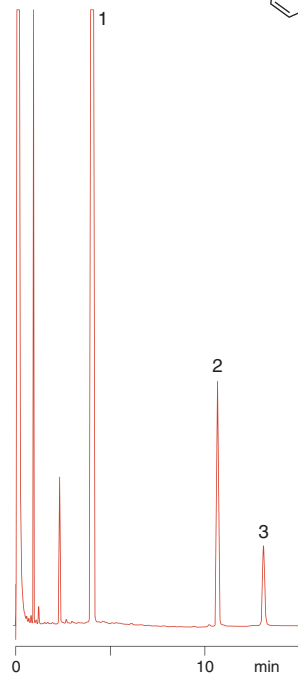
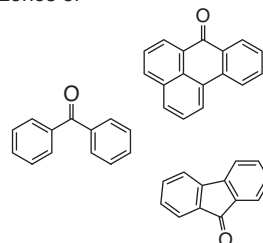
S. Guillot, M. T. Kelly, H. Fenet, M. Larroque,
J. Chromatogr. A **1101** (2006) 46 – 52

Analysis of 2,4-dinitrophenyl-hydrazones

MN Appl. No. 201250

Column: OPTIMA® 1,
10 m x 0.53 mm ID,
0.50 µm film,
REF 726519.10,
max. temperature 320/340 °C
Injection: 1 µl, split 1:50
Carrier gas: 0.3 bar H₂
Temperature: 60 °C → 30 °C/min → 160 °C → 8 °C/min → 300 °C
Detector: FID 320 °C
Peaks: DNP hydrazones of

1. Benzanthrone
2. Benzophenone
3. Fluorenone



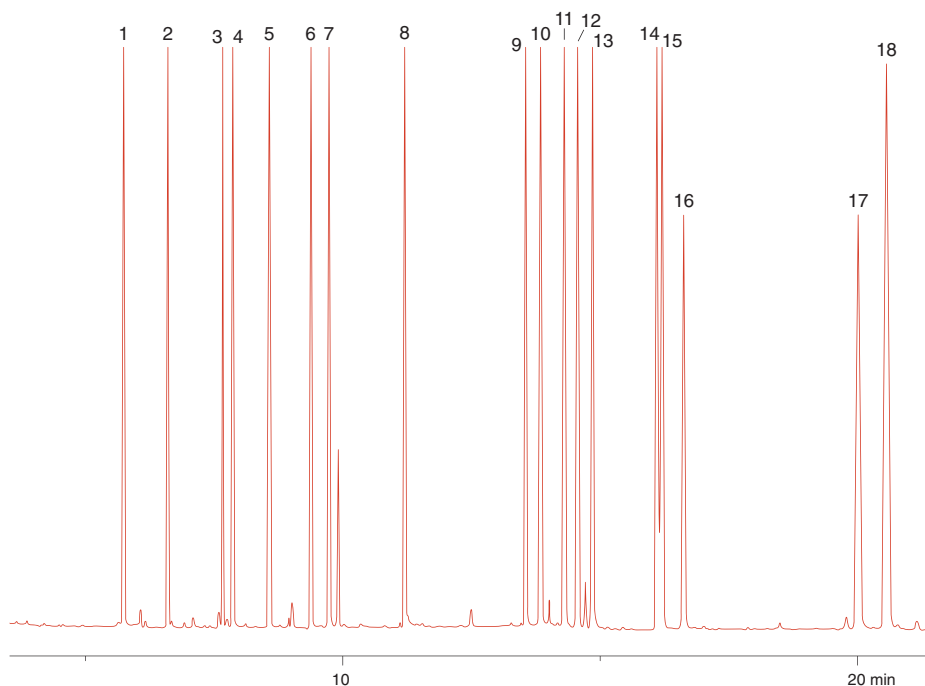


Analysis of plasticizers MN Appl. No. 201200

Column: OPTIMA® 5, 50 m x 0.25 mm ID, 0.25 µm film, REF 726056.50,
max. temperature 340/360 °C
Injection: 1 µl on column, 20 ng each
Carrier gas: 140 kPa H₂
Temperature: 80 $\xrightarrow{40\text{ °C/min}}$ 240 °C $\xrightarrow{6\text{ °C/min}}$ 280 °C
Detector: FID

Peaks:

- | | |
|----------------------------------|---------------------------------------|
| 1. Dimethyl phthalate | 10. Di(2-ethylhexyl) adipate |
| 2. Diethyl phthalate | 11. Triphenyl phosphate |
| 3. Diallyl phthalate | 12. Diphenyl-(2-ethylhexyl) phosphate |
| 4. Dipropyl phthalate | 13. Tri(2-ethylhexyl) phosphate |
| 5. Diisobutyl phthalate | 14. Dicyclohexyl phthalate |
| 6. Di- <i>n</i> -butyl phthalate | 15. Di(2-ethylhexyl) phthalate |
| 7. Methyl glycol phthalate | 16. Diphenyl phthalate |
| 8. Dibutyl sebacate | 17. Di- <i>n</i> -octyl phthalate |
| 9. Butyl benzyl phthalate | 18. Tri- <i>p</i> -cresyl phosphate |



Courtesy of Mr. Lembacher, Hipp Nahrungsmittel, Pfaffenhofen, Germany

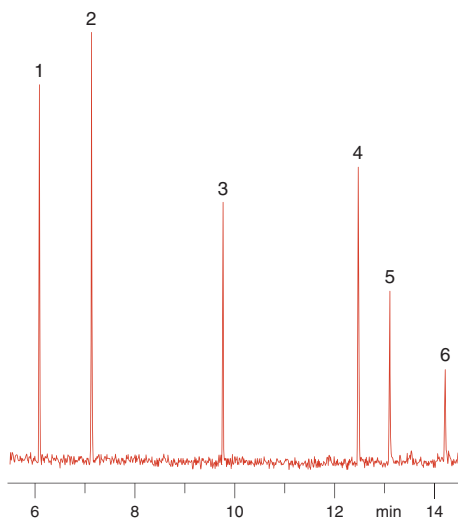
Analysis of phthalates (EPA 606)

MN Appl. No. 213160

Column: OPTIMA® δ-3,
30 m x 0.25 mm ID,
0.25 µm film, REF 726420.30,
max. temperature 340/360 °C
Injection: 1 µl, split 1:10
Carrier gas: 1.4 ml/min H₂
Temperature: 100 °C $\xrightarrow{15\text{ °C/min}}$ 320 °C
Detector: MSD

Peaks:

1. Dimethyl phthalate
2. Diethyl phthalate
3. Dibutyl phthalate
4. Butyl benzyl phthalate
5. Di(2-ethylhexyl) phthalate
6. Di-*n*-octyl phthalate



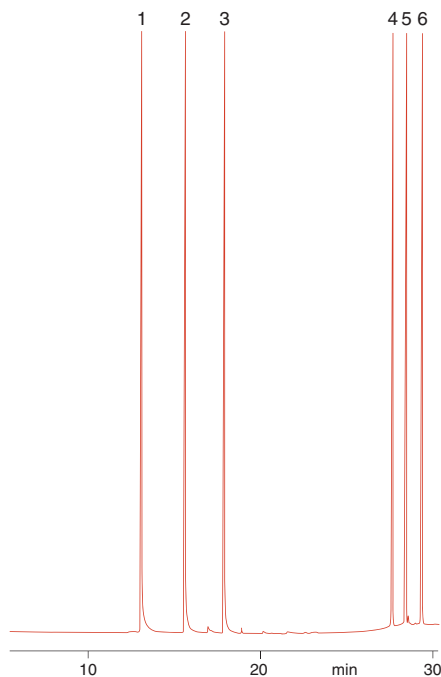
Analysis of phosphonic acid diethyl esters

MN Appl. No. 211860

Column: OPTIMA® δ-3,
30 m x 0.25 mm ID,
0.25 µm film, REF 726420.30,
max. temperature 340/350 °C
Injection: 0.5 µl, 250 °C, split 40 ml/min
Carrier gas: 80 kPa He
Temperature: 60 °C $\xrightarrow{5\text{ °C/min}}$ 150 °C $\xrightarrow{20\text{ °C/min}}$ 300 °C
Detector: FID 300 °C

Peaks:

1. Diethyl methylphosphonate
2. Diethyl ethylphosphonate
3. Diethyl allylphosphonate
4. Tetraethyl-ethylene-1,2-diphosphonate
5. Tetraethyl-propylene-1,3-diphosphonate
6. Tetraethyl-butylene-1,4-diphosphonate



VeZerf Laborsynthesen GmbH, Idar-Oberstein,
Germany

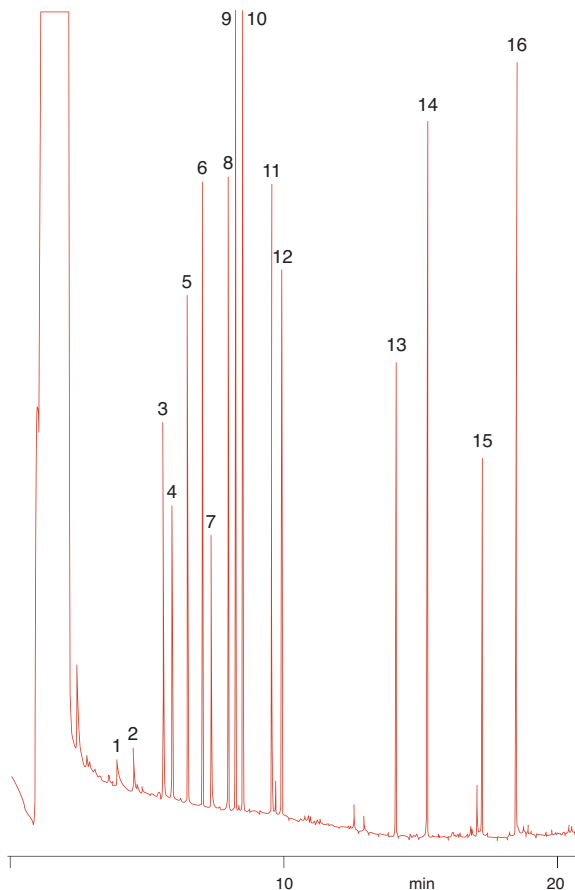


Analysis of halogeno- and phosphororganic compounds MN Appl. No. 210340

Column: OPTIMA® 5, 30 m x 0.25 mm ID, 0.25 µm film, REF 726056.30,
max. temperature 340/360 °C
Injection: 1.0 µl, splitless
Carrier gas: 1.5 bar H₂
Temperature: 39 °C (1 min) $\xrightarrow{10\text{ °C/min}}$ 120 °C $\xrightarrow{12\text{ °C/min}}$ 280 °C (10 min)
Detector: MSD

Peaks:

1. Dimethyl methylphosphonate
2. Trimethyl phosphate
3. 4-Fluorophenol
4. 1,2-Dichlorobenzene
5. 1-Octanol
6. 2,6-Dimethylphenol
7. Triethyl phosphate
8. 2,6-Dimethylbenzylamine
9. Naphthalene
10. *n*-Dodecane
11. 1-Dodecanol
12. 5-Chloro-2-methylbenzylamine
13. Tributyl phosphate
14. Dibenzothiophene
15. Malathion
16. Methyl octadecanoate



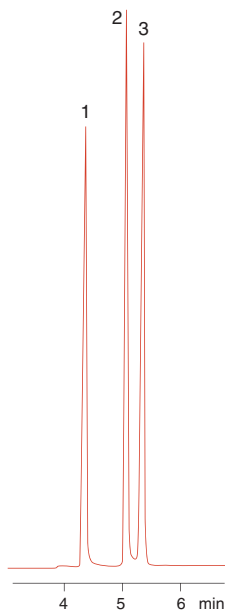
Courtesy of Mr. Kremer, Wehrwissenschaftl. Inst. für Schutztechnologien, Munster, Germany

Analysis of methylamines MN Appl. No. 210490

Column: OPTIMA® 5,
50 m x 0.32 mm ID,
5.0 μ m film, REF 726934.50,
max. temperature 300/320 °C
Split: 108 ml/min
Carrier gas: 2.2 ml/min He
Temperature: 85 °C
Detector: FID

Peaks: 20 ng/peak

1. Methylamine
2. Dimethylamine
3. Trimethylamine

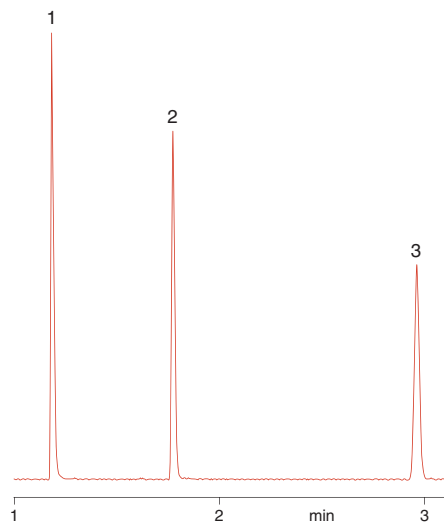


Analysis of ethylamines MN Appl. No. 250050

Column: OPTIMA® 5 Amine,
30 m x 0.32 mm ID,
1.0 μ m film, REF 726353.30,
max. temperature 300/320 °C
Injection: 1 μ l, split 1:70
Carrier gas: 0.5 bar H₂
Temperature: 40 °C $\xrightarrow{5\text{ }^{\circ}\text{C/min}}$ 100 °C
Detector: FID

Peaks:

1. Ethylamine
2. Diethylamine
3. Triethylamine





Analysis of amines

MN Appl. No. 201520

Column: FS-CW 20 M-AM,
25 m x 0.25 mm ID,
0.25 μ m film, REF 733110.25,
max. temperature 220/240 °C

Injection: 0.2 μ l, split 1:100

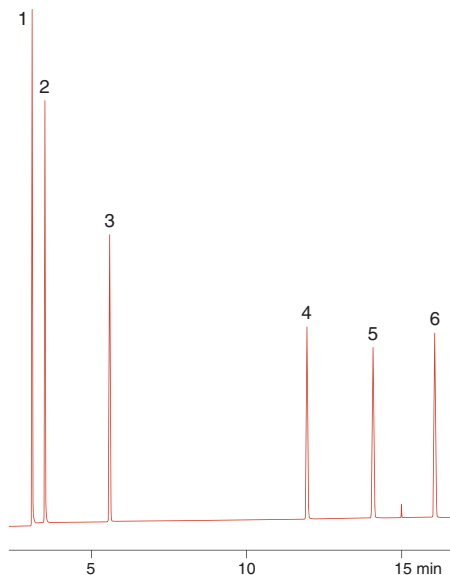
Carrier gas: 0.45 bar N₂ $\xrightarrow{8\text{ °C/min}}$ 180 °C

Temperature: 80 °C (5 min) $\xrightarrow{8\text{ °C/min}}$ 180 °C

Detector: FID 250 °C

Peaks:

1. Butylamine
2. Di-*i*-propylamine
3. Dibutylamine
4. Nonylamine
5. Decylamine
6. Undecylamine



Analysis of amines

MN Appl. No. 201530

Column: FS-CW 20 M-AM,
25 m x 0.32 mm ID,
0.25 μ m film, REF 733299.25,
max. temperature 220/240 °C

Injection: 0.5 μ l

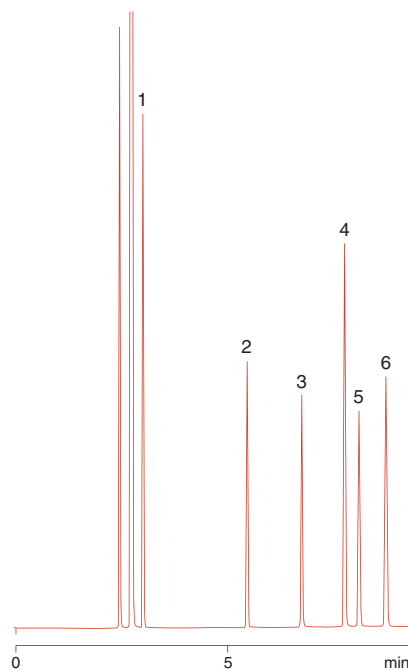
Carrier gas: 0.5 bar N₂ $\xrightarrow{10\text{ °C/min}}$ 190 °C

Temperature: 80 °C $\xrightarrow{10\text{ °C/min}}$ 190 °C

Detector: FID 240 °C

Peaks:

1. Dibutylamine
2. *n*-Nonylamine
3. *n*-Decylamine
4. Benzylamine
5. *n*-Undecylamine
6. Dicyclohexylamine



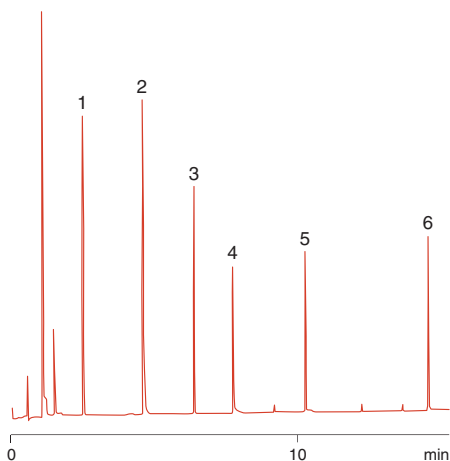
Separation of the OPTIMA® Amine test mixture (REF 722317)

MN Appl. No. 250020

Column: OPTIMA® 5 Amine,
30 m x 0.32 mm ID,
1.0 μ m film, REF 726353.30,
max. temperature 300/320 °C
Injection: 1 μ l, split 1:40
Carrier gas: 0.6 bar H₂
Temperature: 100 °C $\xrightarrow{10\text{ °C/min}}$ 280 °C
Detector: FID

Peaks:

1. Di-*i*-butylamine
2. Diethanolamine
3. 2,6-Dimethylaniline
4. *o*-Propanol-pyridine
5. Dicyclohexylamine
6. Dibenzylamine



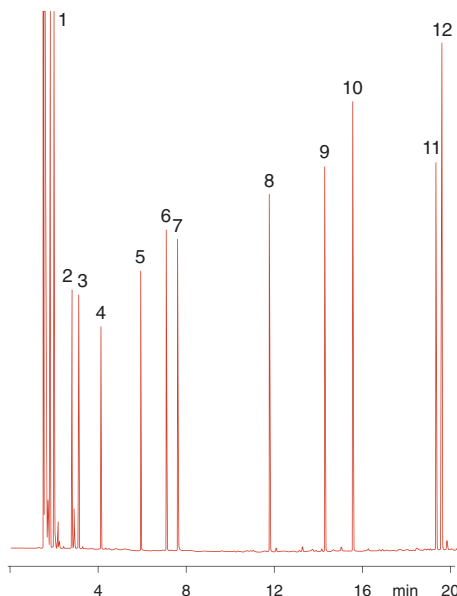
Analysis of secondary and tertiary amines

MN Appl. No. 210280

Column: OPTIMA® 5 Amine,
30 m x 0.25 mm ID,
1.0 μ m film, REF 726358.30,
max. temperature 300/320 °C
Injection: 1.0 μ l, 0.01% in pentane
split 1:100
Carrier gas: 0.55 bar H₂
Temperature: 50 °C (3 min) $\xrightarrow{10\text{ °C/min}}$ 280 °C
(10 min)
Detector: FID 280 °C

Peaks:

1. Diethylamine
2. Di-*i*-propylamine
3. Triethylamine
4. Di-*n*-propylamine
5. Di-*n*-butylamine
6. Tri-*n*-propylamine
7. Di-*i*-butylamine
8. Tri-*n*-butylamine
9. Di-*i*-hexylamine
10. Dicyclohexylamine
11. Dibenzylamine
12. Tri-*n*-hexylamine





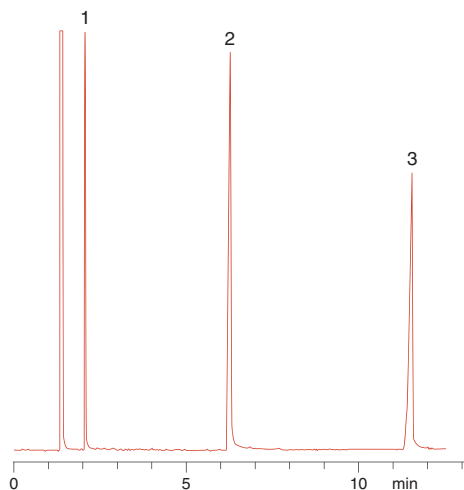
Analysis of ethanolamines

MN Appl. No. 250040

Column: OPTIMA® 5 Amine,
30 m x 0.32 mm ID,
1.0 μ m film, REF 726353.30,
max. temperature 300/320 °C
Injection: 1 μ l, 20 % in methanol
split 1:30
Carrier gas: 0.6 bar H₂
Temperature: 80 °C $\xrightarrow{10\text{ °C/min}}$ 200 °C
Detector: FID

Peaks:

1. Ethanolamine
2. Diethanolamine
3. Triethanolamine



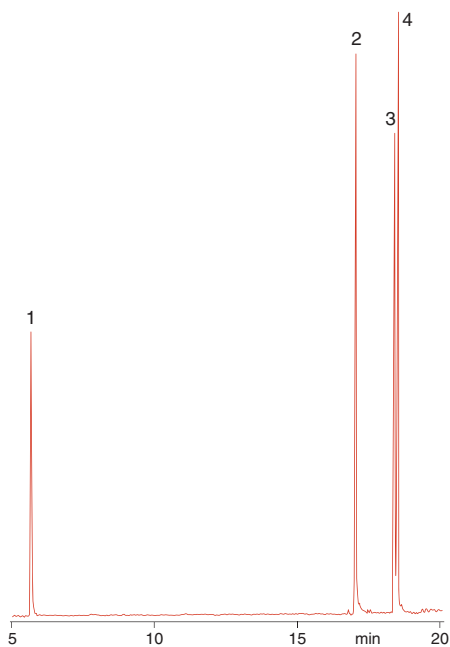
Analysis of pyridine compounds

MN Appl. No. 210360

Column: OPTIMA® 5 Amine,
25 m x 0.20 mm ID,
0.35 μ m film, REF 726355.25,
max. temperature 300/320 °C
Injection: 310 °C
Carrier gas: 25 cm/s He $\xrightarrow{8\text{ °C/min}}$ 290 °C
Temperature: 50 °C (3 min) $\xrightarrow{8\text{ °C/min}}$ 290 °C
Detector: MSD

Peaks:

1. Pyridine
2. *o*-Propanolpyridine
3. *m*-Propanolpyridine
4. *p*-Propanolpyridine

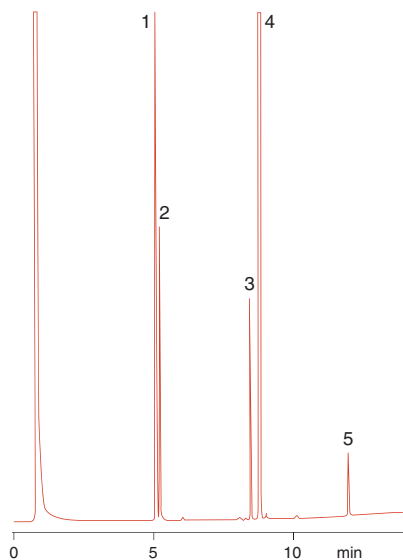


Analysis of an oxazole derivative of *N,N*-dimethylformamide *MN Appl. No. 210520*

Column: OPTIMA® 240,
25 m x 0.25 mm ID,
0.25 µm film, REF 726089.25,
max. temperature 260/280 °C
Injection: 0.5 µl, split 100 ml/min
Carrier gas: 0.8 bar He
Temperature: 50 °C $\xrightarrow{7\text{ °C/min}}$ 200 °C
Detector: FID 280 °C

Peaks:

1. *N,N*-Dimethylformamide
2. 5-Cyano-4-methyloxazole
3. Methyl decanoate
4. *N*-Methylpyrrolidone
5. 4-Methyl-5-oxazole-carboxylic acid amide



Courtesy of Hoffmann-La Roche AG, Basel, Switzerland

N-Methylpyrrolidone



Oxazole

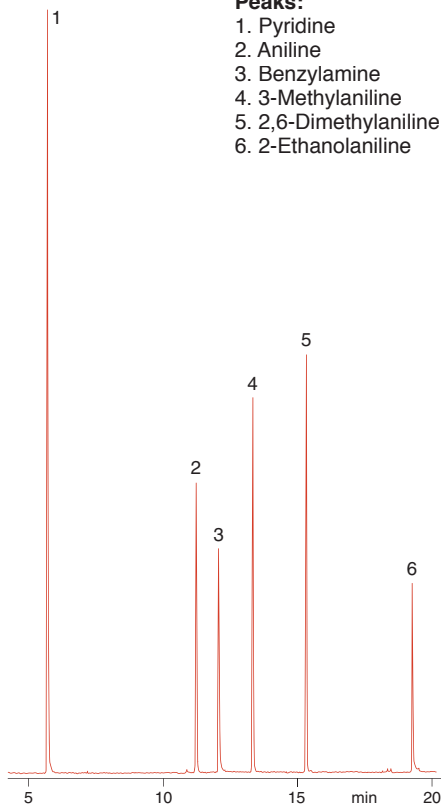


Analysis of aniline derivatives *MN Appl. No. 210380*

Column: OPTIMA® 5 Amine,
25 m x 0.20 mm ID,
0.35 µm film, REF 726355.25,
max. temperature 300/320 °C
Injection: 310 °C
Carrier gas: 25 cm/s He $\xrightarrow{8\text{ °C/min}}$ 290 °C
Temperature: 50 °C (3 min) $\xrightarrow{8\text{ °C/min}}$ 290 °C
Detector: MSD

Peaks:

1. Pyridine
2. Aniline
3. Benzylamine
4. 3-Methylaniline
5. 2,6-Dimethylaniline
6. 2-Ethanolaniline





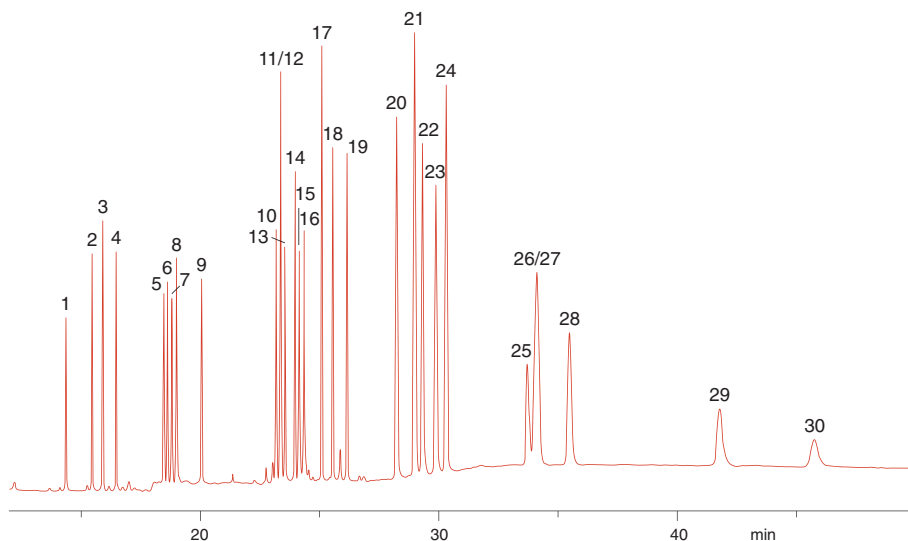
Analysis of substituted anilines

MN Appl. No. 201560

Column: FS-CW 20 M-AM, 25 m x 0.25 mm ID, 0.25 μ m film, REF 733110.25,
max. temperature 220/240 °C
Injection: 2 μ l, split 1:20
Carrier gas: 80 kPa He
Temperature: 80 °C (5 min) $\xrightarrow{5\text{ }^{\circ}\text{C/min}}$ 210 °C (10 min)
Detector: MSD

Peaks:

- | | | |
|---|------------------------------|---------------------------------|
| 1. Aniline | 11. 2,4,6-Trichloroaniline | 21. 3,4-Dichloroaniline |
| 2. <i>o</i> -Toluidine | 12. <i>m</i> -Chloroaniline | 22. 2,4,5-Trichloroaniline |
| 3. <i>p</i> -Toluidine | 13. 3-Chloro-2-methylaniline | 23. 1-Naphthylamine (int. std.) |
| 4. <i>m</i> -Toluidine | 14. 4-Chloro-2-methylaniline | 24. 2,3,4-Trichloroaniline |
| 5. <i>o</i> -Chloroaniline | 15. 5-Chloro-2-methylaniline | 25. 5-Chloro-2-nitroaniline |
| 6. 2-Chloro-6-methylaniline | 16. <i>m</i> -Phenetidine | 26. 2-Chloro-5-nitroaniline |
| 7. 2,6-Dichloroaniline | 17. 2,4-Dichloroaniline | 27. 4-Chloro-2-nitroaniline |
| 8. 4- <i>i</i> -Propylaniline (int. std.) | 18. 2,5-Dichloroaniline | 28. 3,4,5-Trichloroaniline |
| 9. 2-Chloro-4-methylaniline | 19. 2,3-Dichloroaniline | 29. 2-Chloro-4-nitroaniline |
| 10. <i>p</i> -Chloroaniline | 20. 3,5-Dichloroaniline | 30. 4-Chloro-3-nitroaniline |



B. Scholz, N. Palauschek, Z. Anal. Chem. **331** (1988) 282 – 289

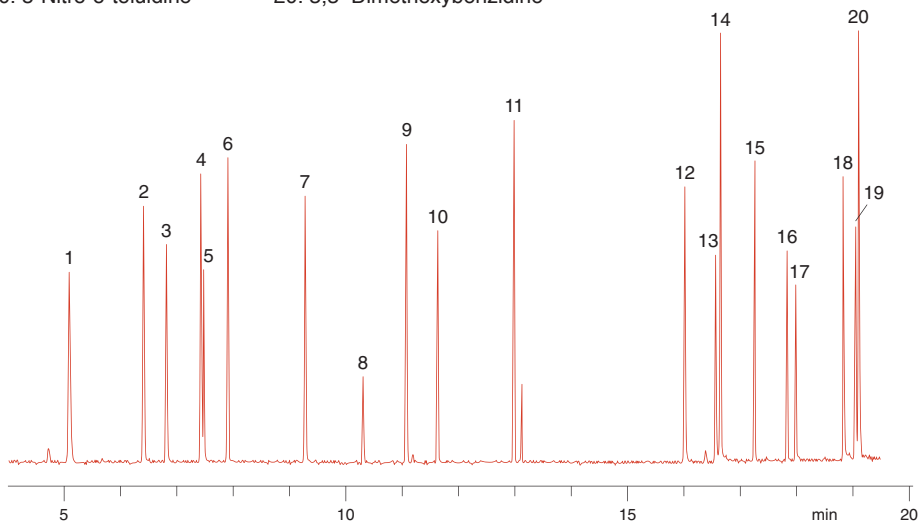
Analysis of aromatic amines

MN Appl. No. 213230

Column: OPTIMA® 35 MS, 30 m x 0.25 mm ID, 0.25 µm film, REF 726154.30,
max. temperature 360/370 °C
Injection: 1 µl, split 1:20
Carrier gas: 1.0 ml/min He
Temperature: 80 °C $\xrightarrow{10\text{ °C/min}}$ 320 °C (10 min)
Detector: MSD

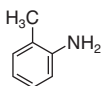
Peaks:

- | | |
|---------------------------|---|
| 1. o-Toluidine | 11. 4-Aminobiphenyl |
| 2. o-Anisidine | 12. 4-Aminoazobenzene |
| 3. 4-Chloroaniline | 13. 4,4'-Oxydianiline |
| 4. p-Cresidine | 14. Benzidine + 4,4'-diaminodiphenylmethane |
| 5. 2,4,5-Trimethylaniline | 15. o-Aminoazotoluene + 3,3'-dimethyl-4,4'-diaminodiphenylmethane |
| 6. 4-Chloro-o-toluidine | 16. 3,3'-Dimethylbenzidine |
| 7. 2,4-Diaminotoluene | 17. 4,4'-Thiodianiline |
| 8. 2,4-Diaminoanisole | 18. 3,3'-Dichlorobenzidine |
| 9. 2-Naphthylamine | 19. 4,4'-Methylene-bis-(2-chloroaniline) |
| 10. 5-Nitro-o-toluidine | 20. 3,3'-Dimethoxybenzidine |

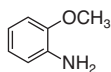


Structures of selected aromatic amines

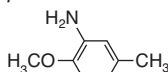
o-Toluidine



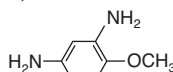
o-Anisidine



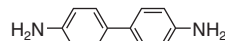
p-Cresidine



2,4-Diaminoanisole



Benzidine



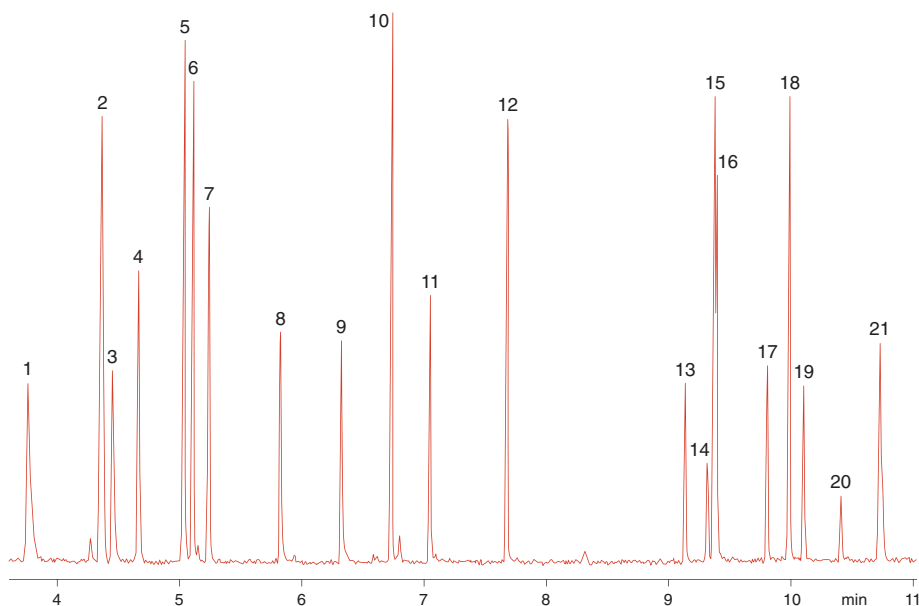


Analysis of azo-dyes and aromatic amines by fast GC MN Appl. No. 210820

Column: OPTIMA® δ -3, 10 m x 0.1 mm ID, 0.1 μ m film, REF 726410.10,
max. temperature 340/360 °C
Injection: 1.0 μ l, 250 °C
Carrier gas: 84 ml/min, splitless
Pressure: 580 kPa (1 min) $\xrightarrow{42 \text{ kPa/min}}$ 966 kPa (5 min)
Temperature: 50 °C (1 min) $\xrightarrow{25 \text{ °C/min}}$ 280 °C (5 min)
Detector: MSD 320 °C

Peaks:

- | | |
|----------------------------------|---|
| 1. <i>o</i> -Toluidine | 13. 4-Aminoazobenzene |
| 2. 2,4- and 2,6-Dimethylaniline | 14. 4,4'-Oxydianiline |
| 3. <i>o</i> -Anisidine | 15. 4,4'-Diaminodiphenylmethane |
| 4. 4-Chloroaniline | 16. Benzidine |
| 5. <i>p</i> -Cresidine | 17. <i>o</i> -Aminoazotoluene |
| 6. 2,4,5-Trimethylaniline | 18. 3,3'-Dimethyl-4,4'-diaminodiphenylmethane |
| 7. 4-Chloro- <i>o</i> -toluidine | 19. 3,3'-Dimethylbenzidine |
| 8. 2,4-Diaminotoluene | 20. 4,4'-Thiodianiline |
| 9. 2,4-Diaminoanisole | 21. 3,3'-Dimethoxybenzidine |
| 10. 2-Naphthylamine | + 4,4'-methylene-bis-(2-chloroaniline) |
| 11. 5-Nitro- <i>o</i> -toluidine | + 3,3'-dichlorobenzidine |
| 12. 4-Aminobiphenyl | |



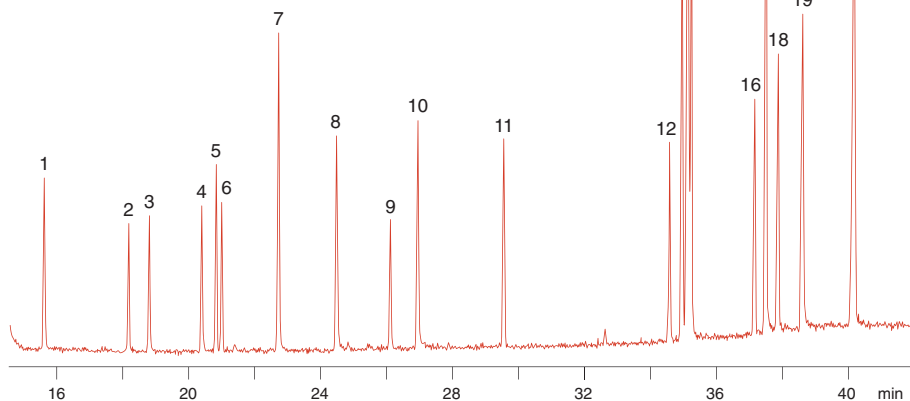
Courtesy of Mrs. K. Friedrichs, Chem. Untersuchungsamt, Bielefeld, Germany

Analysis of azo-dyes and aromatic amines MN Appl. No. 210170

Column: OPTIMA® 5 Amine, 30 m x 0.25 mm ID, 0.5 µm film, REF 726354.30,
max. temperature 300/320 °C
Injection: 2.0 µl
Carrier gas: He
Temperature: 50 °C $\xrightarrow{7\text{ °C/min}}$ 300 °C
Detector: MSD

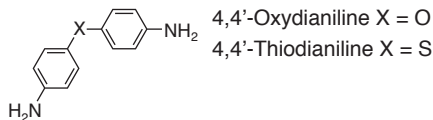
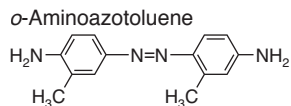
Peaks:

- | | |
|----------------------------------|-----------------------------------|
| 1. <i>o</i> -Toluidine | 13. 4,4'-Oxydianiline |
| 2. <i>o</i> -Anisidine | 14. Benzidine |
| 3. 4-Chloroaniline | 15. 4,4'-Diaminodiphenylmethane |
| 4. <i>p</i> -Cresidine | 16. <i>o</i> -Aminoazotoluene |
| 5. 2,4,5-Trimethylaniline | 17. 3,3'-Dimethyl-4,4'-diamino- |
| 6. 4-Chloro- <i>o</i> -toluidine | diphenylmethane |
| 7. 2,4-Diaminotoluene | 18. 3,3'-Dimethylbenzidine |
| 8. 2,4-Diaminoanisole | 19. 4,4'-Thiodianiline |
| 9. 2-Naphthylamine | 20. 4,4'-Methylene-bis-(2-chloro- |
| 10. 5-Nitro- <i>o</i> -toluidine | aniline) + 3,3'-dichlorobenzidine |
| 11. 4-Aminobiphenyl | + 3,3'-dimethoxybenzidine |
| 12. 4-Aminoazobenzene | |



Courtesy of Mrs. Friedrichs, Chem. Untersuchungsamt Bielefeld, Germany

Structures of selected aromatic amines





Analysis of organotin compounds

MN Appl. No. 210730

The sample is dissolved in *n*-hexane. After the Grignard reaction with pentylmagnesium bromide the residue is extracted with diethyl ether and dried with Na_2SO_4 .

Column: OPTIMA® 5 MS,
12 m x 0.20 mm ID,
0.35 μm film, REF 726215.12,
max. temperature 340/360 °C

Septum purge: 4.5 ml/min

Injection: 1 μl , 250 °C, splitless

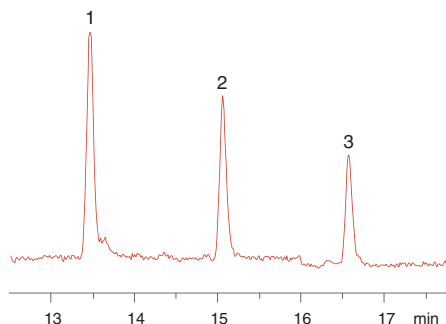
Carrier gas: 1 ml/min He (20 kPa)

Temperature: 90 °C $\xrightarrow{6\text{ °C/min}}$ 280 °C (3 min)

Detector: MSD 280 °C, EI, SIM

Peaks:

1. Tetraethyl tin (internal standard)
2. Triethyl pentyl tin
3. Diethyl dipentyl tin



Courtesy of Mr. Wohlfarth, Staatl. Med. Leb-
ensm, Vet. UA, Wiesbaden, Germany

Qualitative determination of organotin compounds

MN Appl. No. 212140

Column: OPTIMA® 1,
25 m x 0.2 mm ID,
0.35 μm film, REF 726837.25,
max. temperature 340/360 °C

Injection: 1 μl , split 15 ml/min

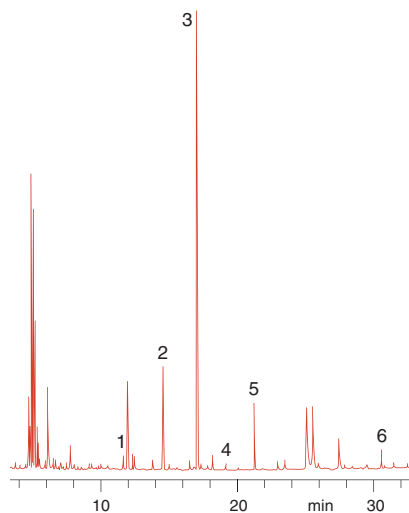
Carrier gas: He $\xrightarrow{6\text{ °C/min}}$ 300 °C

Temperature: 50 °C $\xrightarrow{6\text{ °C/min}}$ 300 °C

Detector: MSD, transfer line 300 °C

Peaks:

1. BuSnEt_3
2. Bu_2SnEt_2
3. Bu_3SnEt
4. Bu_4Sn
5. $i\text{-OcBu}_2\text{SnEt}$
6. Oc_3SnEt



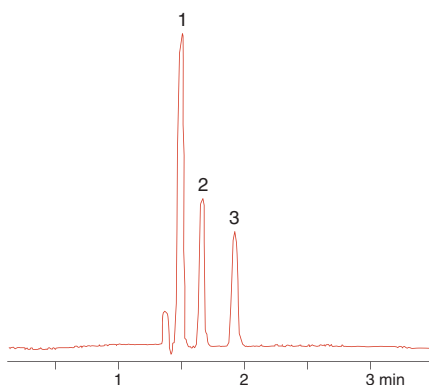
Analysis of methyl silanes

MN Appl. No. 200080

Column: PERMABOND® Silane,
50 m x 0.32 mm ID,
REF 723409.50,
max. temperature 260/280 °C
Injection: 1 ml gas
Carrier gas: 60 kPa He
Temperature: 35 °C
Detector: MSD

Peaks:

1. Methylsilane
2. Dimethylsilane
3. Trimethylsilane



Courtesy of Th. Goldschmidt AG, Essen,
Germany

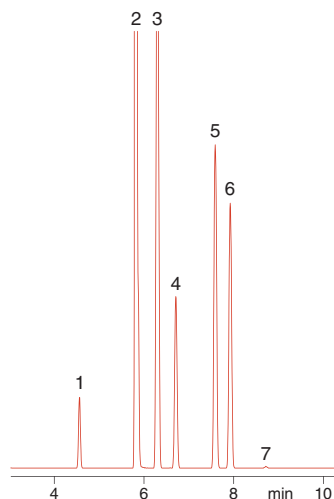
Analysis of chloromethyl silanes

MN Appl. No. 200090

Column: PERMABOND® Silane,
50 m x 0.32 mm ID,
REF 723409.50,
max. temperature 260/280 °C
Injection: 0.5 µl, split 80 ml/min
Carrier gas: 1 ml/min He (constant flow)
Temperature: 50 °C (1 min) $\xrightarrow{5\text{ °C/min}}$ 100 °C
Detector: MSD

Peaks:

1. Tetramethylsilane
2. Dichloromethane
3. Tetrachlorosilane
4. Chlorotrimethylsilane
5. Methyltrichlorosilane
6. Dichlorodimethylsilane
7. Hexamethyldisiloxane





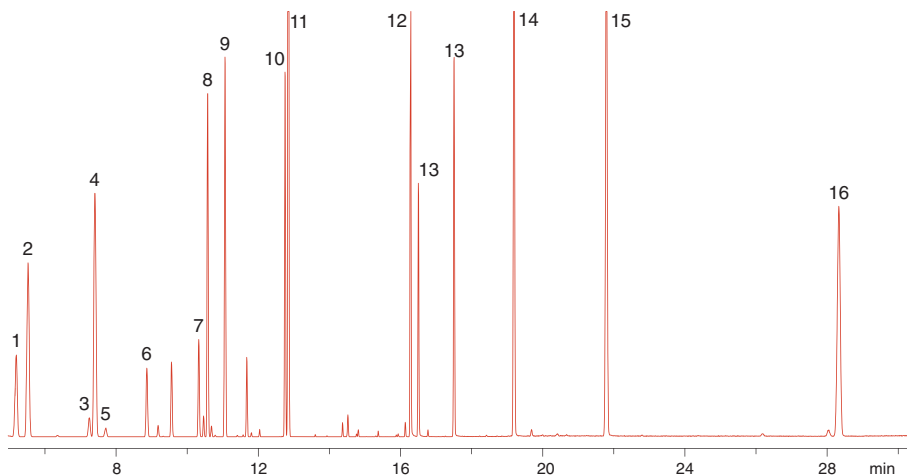
Analysis of silanes

MN Appl. No. 200100

Column: PERMABOND® Silane, 50 m x 0.32 mm ID, REF 723409.50,
max. temperature 260/280 °C
Injection: 0.5 µl, split 80 ml/min
Carrier gas: 1.5 ml/min He (constant flow)
Temperature: 60 °C (6 min) $\xrightarrow{20\text{ °C/min}}$ 250 °C (15 min)
Detector: MSD

Peaks:

- | | |
|----------------------------------|--|
| 1. Trimethylmethoxysilane | 9. 1,3-Divinyltetramethyldisiloxane |
| 2. Methylhydrodimethoxysilane | 10. 1,3-Divinyltetramethyldisilazane |
| 3. Vinyltrimethylmethoxysilane | 11. Octamethylcyclotetrasiloxane (D ₄) |
| 4. Dimethyldimethoxysilane | 12. (Cyanopropyl)methyldimethoxysilane |
| 5. Hexamethyldisiloxane | 13. (Phenylpropyl)methyldimethoxysilane |
| 6. 1,1,3,3-Tetramethyldisiloxane | 14. Diphenyldimethoxysilane |
| 7. Hexamethyldisilazane (HMDS) | 15. 1,3-Diphenyltetramethyldisilazane |
| 8. Vinyltrimethoxysilane | 16. 1,3-Dicyanopropyltetramethyldisiloxane |

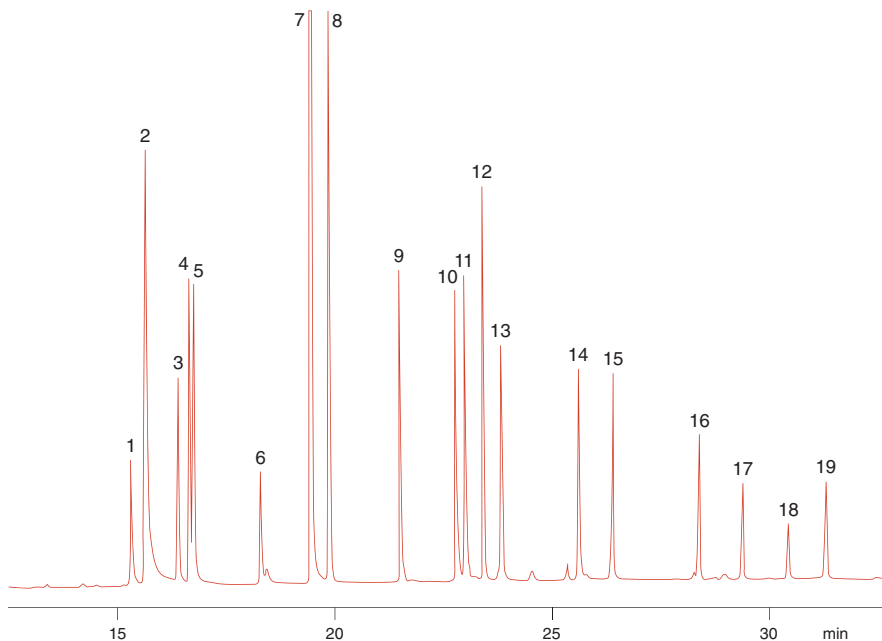


Analysis of bifunctional alkyl nitrates MN Appl. No. 210290

Column: OPTIMA® 1701, 50 m x 0.32 mm ID, 0.25 µm film, REF 726318.50,
max. temperature 300/320 °C
Carrier gas: H₂
Temperature: 40 °C (3 min) $\xrightarrow{5\text{ °C/min}}$ 200 °C (10 min)
Detector: ECD 260 °C

Peaks:

- | | |
|--|---|
| 1. 1-Nitrooxypropanol-2 | 11. 1,3-Propanediol dinitrate |
| 2. 2-Nitrooxyethanol | 12. 1,3-Butanediol dinitrate |
| 3. <i>RR/SS</i> -3-Nitrooxybutanol-2 | 13. 1,2-Pentanediol dinitrate |
| 4. 2-Nitrooxypropanol-1 | 14. 1-Hydroxy-2-nitrooxycyclohexane |
| 5. <i>RS/SR</i> -3-Nitrooxybutanol-2 | 15. 1,2-Hexanediol dinitrate |
| 6. 1-Nitrooxybutanol-2 | 16. <i>trans</i> -1,2-Cyclohexanediol dinitrate |
| 7. 1,2-Ethanediol dinitrate / 1,2-Propanediol
dinitrate / 2-Nitrooxybutanol-1 | 17. <i>cis</i> -1,2-Cyclohexanediol dinitrate |
| 8. 2,3-Butanediol dinitrate | 18. <i>cis/trans</i> -1,3-Cyclohexanediol dinitrate |
| 9. 1,2-Butanediol dinitrate | 19. <i>trans</i> -1,2-Cycloheptanediol dinitrate /
<i>cis/trans</i> -1,3-Cyclohexanediol dinitrate |
| 10. 1-Hydroxy-2-nitrooxycyclopentane | |



J. Kastler, K. Ballschmiter, Fresenius J Anal Chem **360** (1998) 812 – 816



Comparison of a separation on a 50 m standard capillary with separation on a 10 m fast GC column

MN Appl. No. 211260

A) Fast GC column

Column: OPTIMA® 5, 10 m x 0.1 mm ID,
0.1 μ m film, REF 726846.10,
max. temperature 340/360 °C
injection 1 μ l, split 1:40,
carrier gas 0.75 bar He

B) standard GC column

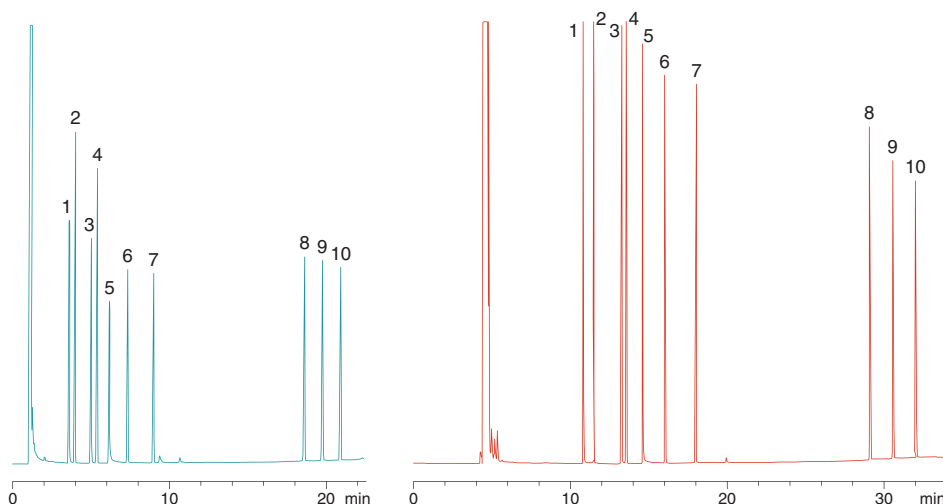
Column: OPTIMA® 5, 50 x 0.25 mm ID,
0.25 μ m film, REF 726056.50,
max. temperature 340/360 °C
injection 1 μ l, split 1:35,
carrier gas 1.5 bar He

both separations: temperature: 80 °C $\xrightarrow{8\text{ }^{\circ}\text{C/min}}$ 320 °C (10 min), detector: FID

While maintaining the temperature programme and halving the pressure a time saving of 30% results with identical separation efficiency.

Peaks:

1. Octanol
2. Undecane
3. Dimethylaniline
4. Dodecane
5. Decylamine
6. Methyl decanoate
7. Methyl undecanoate
8. Henicosane
9. Docosane
10. Tricosane



Food and Cosmetic Components

Fragrances

FAMES





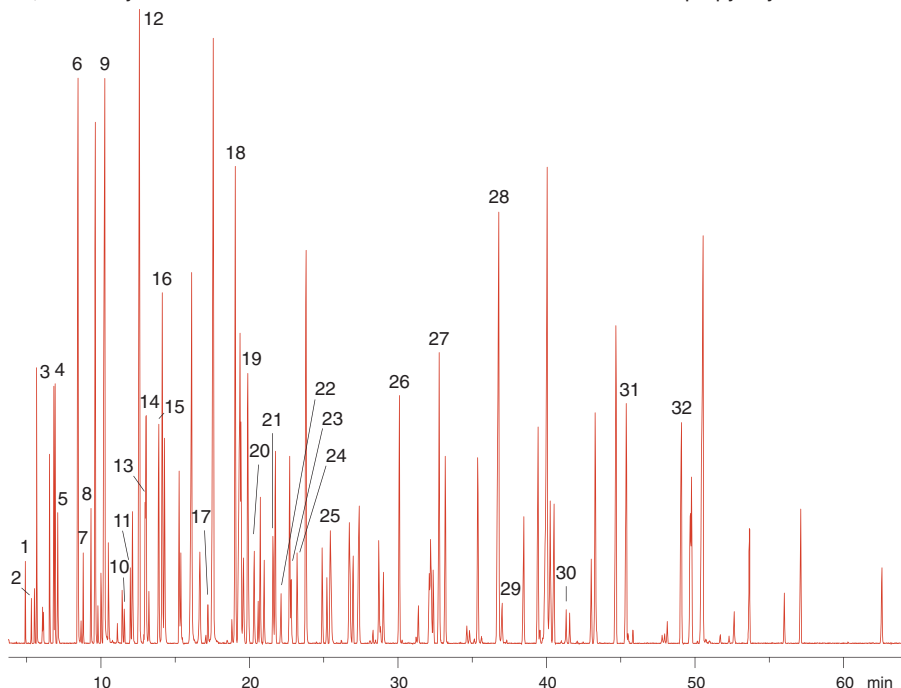
Analysis of a perfume standard

MN Appl. No. 201350

Column: OPTIMA® 5, 25 m x 0.20 mm ID, 0.20 µm film, REF 726857.25,
max. temperature 340/360 °C
Injection: 1.0 µl of a standard of perfume ingredients, 1:100 in EtOH, split 1:100
sample courtesy of Mr. Vogel, Luzi AG, Dietlikon, Switzerland
Carrier gas: 25 cm/s He
Temperature: 70 °C (3 min) $\xrightarrow{2\text{ °C/min}}$ 320 °C
Detector: MSD

Peaks:

- | | | |
|-----------------------------------|------------------------------------|--------------------------------------|
| 1. Benzaldehyde | 12. Phenylacetic acid methyl ester | 22. 2-Aminobenzoic acid methyl ester |
| 2. β-Pinene | 13. Menthol | 23. 2-Carene |
| 3. Limonene | 14. Acetic acid isononyl ester | 24. Eugenol |
| 4. Eucalyptol (Cineole) | 15. Linalyl propanoate | 25. Diphenyl ether |
| 5. Benzyl alcohol | 16. α-Methylbenzyl acetate | 26. α-Isomethyl ionone |
| 6. 2,6-Dimethyl-7-octen-2-ol | 17. 4-Methoxybenzaldehyde | 27. Lilial |
| 7. Hexanoic acid 2-propenyl ester | 18. Bornyl acetate | 28. Diethyl phthalate |
| 8. Benzoic acid methyl ester | 19. Nonanoic acid ethyl ester | 29. 2-Naphthyl methyl ketone |
| 9. Phenylethanol | 20. 3-Phenyl-2-propen-1-ol | 30. Methyl dihydrojasmonate |
| 10. p-Menth-1(7)-en-9-ol | 21. Piperonal | 31. Benzyl benzoate |
| 11. 3,7-Dimethyl-6-octenal | | 32. Isopropyl myristate |



Food and cosmetic components

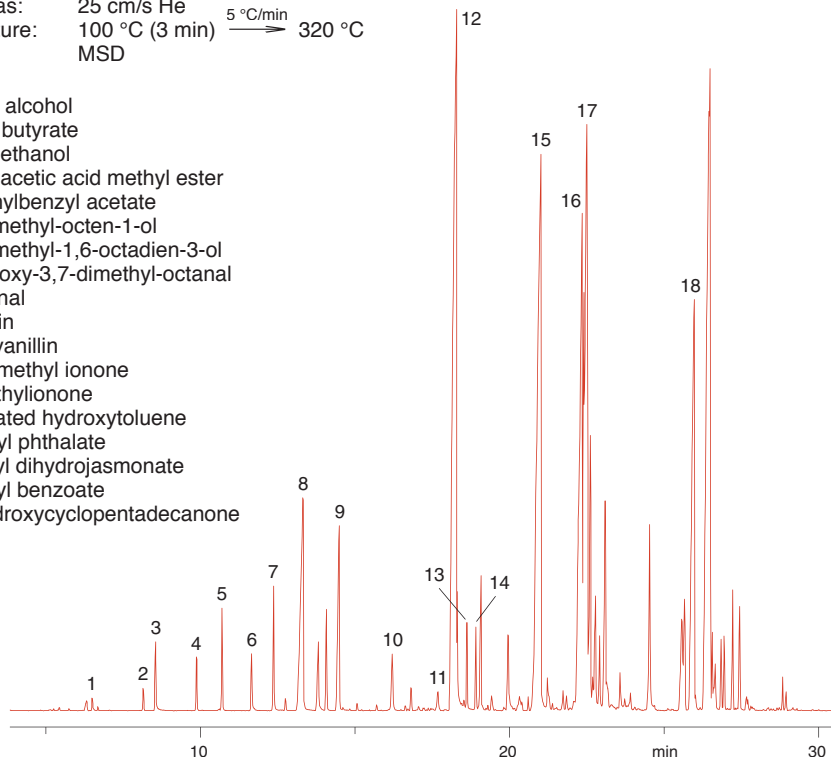
Analysis of commercial perfume

MN Appl. No. 201360

Column: OPTIMA® 5, 25 m x 0.20 mm ID, 0.20 µm film, REF 726857.25,
max. temperature 340/360 °C
Injection: 1.0 µl perfume sample, 1:100 in EtOH, 3 s splitless
Carrier gas: 25 cm/s He
Temperature: 100 °C (3 min) $\xrightarrow{5\text{ °C/min}}$ 320 °C
Detector: MSD

Peaks:

1. Benzyl alcohol
2. Linalyl butyrate
3. Phenylethanol
4. Phenylacetic acid methyl ester
5. α-Methylbenzyl acetate
6. 3,7-Dimethyl-octen-1-ol
7. 3,7-Dimethyl-1,6-octadien-3-ol
8. 7-Hydroxy-3,7-dimethyl-octanal
9. Piperonal
10. Vanillin
11. Ethylvanillin
12. α-Isomethyl ionone
13. β-Methylionone
14. Butylated hydroxytoluene
15. Diethyl phthalate
16. Methyl dihydrojasmonate
17. Benzyl benzoate
18. 2-Hydroxycyclopentadecanone



Structures of selected perfume ingredients

Structure	Compound	R	Structure	Compound
	Vanillin	CHO		β-Methylionone
	Vinylguaiacol	CH=CH ₂		
	Eugenol	CH ₂ -CH=CH ₂		
	Piperonal	CHO		Methyl dihydrojasmonate
	Safrole	CH ₂ -CH=CH ₂		
	Anisaldehyde	CHO		2-Carene
	Anethole	CH=CH-CH ₃		



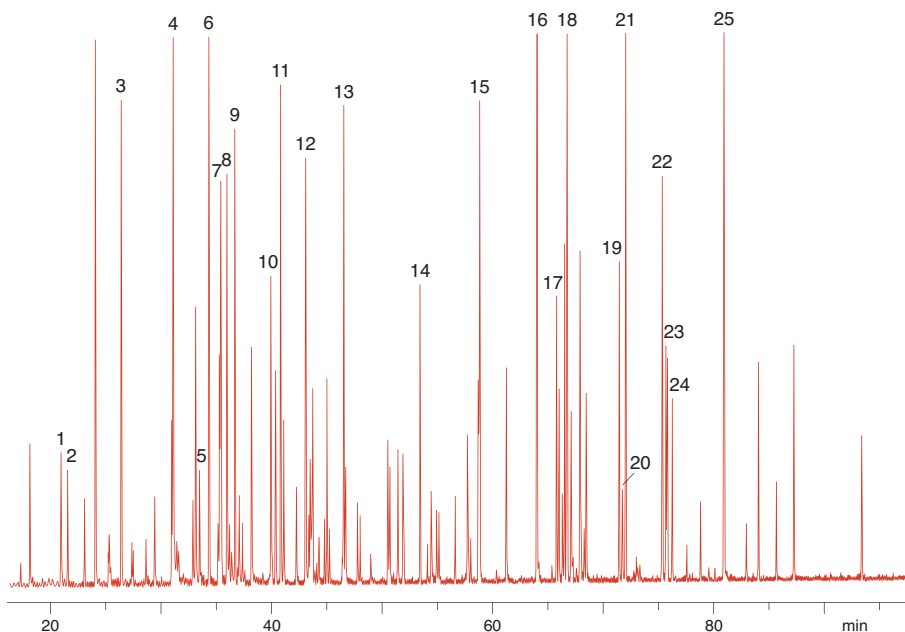
Analysis of a perfume standard

MN Appl. No. 250530

Column: OPTIMA® δ-6, 50 m x 0.20 mm ID, 0.20 µm film, REF 726465.50,
max. temperature 340/360 °C
Injection: 1.0 µl of a standard of perfume ingredients, 1:30 in EtOH, split 1:40
sample courtesy of Mr. Vogel, Luzi AG, Dietlikon, Switzerland
Carrier gas: 1.8 bar He
Temperature: 60 °C $\xrightarrow{2\text{ °C/min}}$ 280 °C (5 min)
Detector: MSD

Peaks:

- | | |
|---|---|
| 1. Limonene | 14. α-Isomethyl ionone |
| 2. Cineole | 15. Lilial |
| 3. Linalool | 16. Diethyl phthalate |
| 4. 2-Phenylethanol | 17. 3,5-Di- <i>t</i> -butylphenol |
| 5. Menthol | 18. Methyl dihydrojasmonate |
| 6. Benzyl acetate | 19. Isopropyl myristate |
| 7. Styralyl acetate (1-phenylethyl acetate) | 20. Methyl atratate |
| 8. 3,7-Dimethyl-6-octen-1-ol | (methyl 2,4-dihydroxy-3,6-dimethylbenzoate) |
| 9. Linalyl anthranilate | 21. Hexylcinnamic aldehyde |
| 10. Ethyl nonanoate | 22. Benzyl benzoate |
| 11. Bornyl acetate | 23. AETT (acetyltetramethyltetralin) |
| 12. Hydroxycitronellal | 24. Oxacyclohexadecan-2-one |
| 13. 4- <i>t</i> -Butylcyclohexyl acetate | 25. Benzyl salicylate |



Food and cosmetic components

Analysis of essential oils

MN Appl. No. 201340

Column: PERMABOND® CW 20 M, 50 m x 0.32 mm ID, 0.5 µm film, REF 723296.50, max. temperature 220/240 °C

Injection: 1 µl

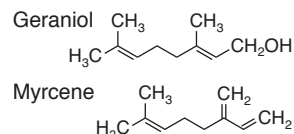
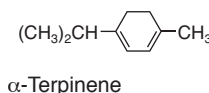
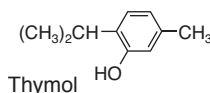
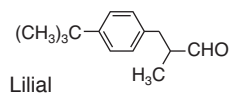
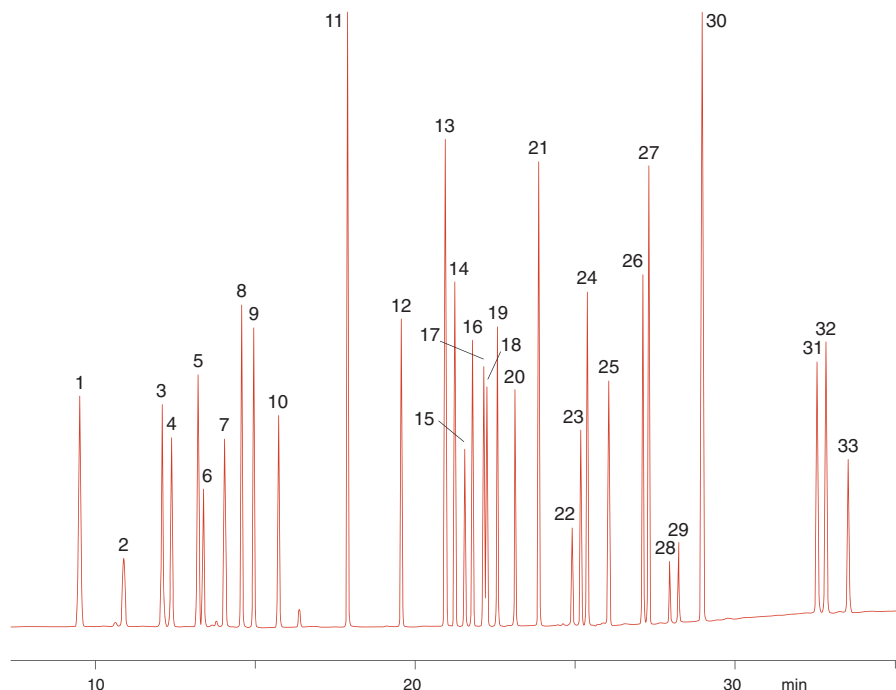
Carrier gas: 1.5 bar He

Temperature: 60 °C (5 min) $\xrightarrow{8\text{ °C/min}}$ 250 °C (10 min)

Detector: FID 250 °C

Peaks:

1. α-Pinene	8. Limonene	15. Isomenthone	22. α-Terpineol	28. Benzyl alcohol
2. Camphene	9. Cineole	16. Linalool	23. Borneol	29. Safrole
3. β-Pinene	10. γ-Terpinene	17. Camphor	24. Menthyl valerianate	30. Dodecanol
4. Sabinene	11. Hexanol-1	18. Linalyl acetate	25. Carvone	31. Thymol
5. Carene	12. Fenchone	19. Menthyl acetate	26. Geraniol	32. Eugenol
6. Myrcene	13. L-Menthone	20. Bornyl acetate	27. Anethole	33. α-Bisabolol
7. α-Terpinene	14. Menthofuran	21. Menthol		





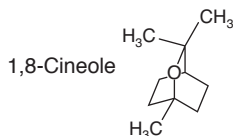
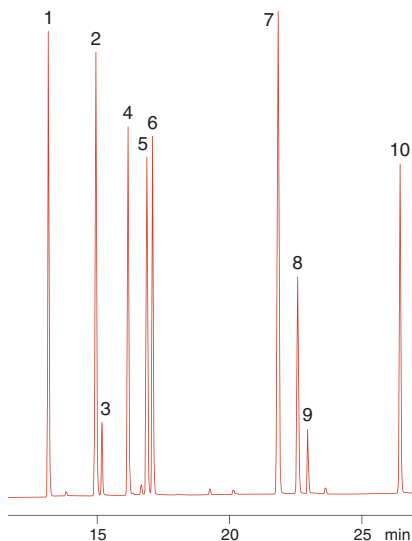
Analysis of essential oils

MN Appl. No. 201330

Column: OPTIMA® 5,
50 m x 0.25 mm ID,
0.35 µm film,
REF 726623.50,
max. temperature: 340/360 °C
Injection: 0.5 µl, split 1:50
Carrier gas: 1 bar N₂
Temperature: 95 °C (10 min) $\xrightarrow{6\text{ °C/min}}$ 200 °C
Detector: FID 280 °C

Peaks:

1. α-Pinene
2. β-Pinene
3. Int. std.
4. Δ³-Carene
5. Limonene
6. 1,8-Cineole
7. Camphor
8. Menthol
9. Int. std.
10. Bornyl acetate



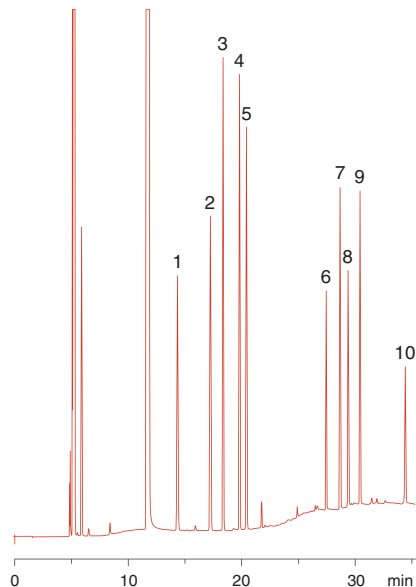
Analysis of essential oils

MN Appl. No. 201370

Column: PERMABOND® FFAP,
50 m x 0.32 mm ID,
0.5 µm film,
REF 723344.50,
max. temperature 220/240 °C
Injection: 0.5 µl, split 1:50
Carrier gas: 0.7 bar N₂
Temperature: 60 °C (6 min) $\xrightarrow{8\text{ °C/min}}$ 200 °C
Detector: FID 250 °C

Peaks:

1. α-Pinene
2. β-Pinene
3. Δ³-Carene
4. Limonene
5. Eucalyptol (Cineole)
6. Camphor
7. Linalool
8. Bornyl acetate
9. Menthol
10. Carvone



Food and cosmetic components

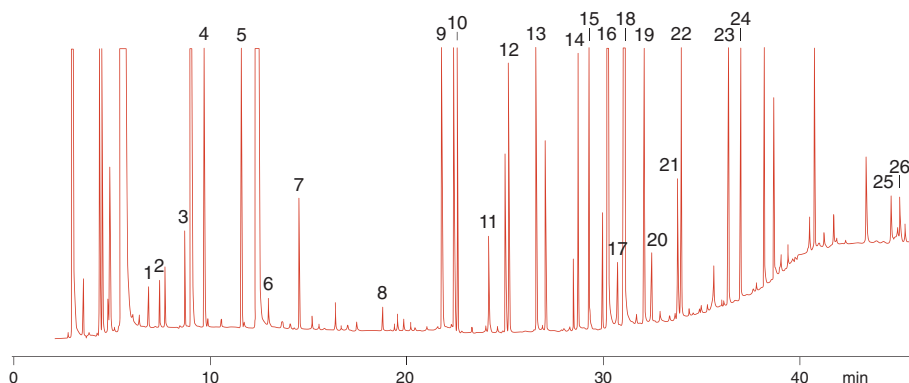
Key odorants in beer

MN Appl. No. 211880

Column: PERMABOND® FFAP, 60 m x 0.25 mm ID, 0.25 µm film, REF 723116.60,
max. temperature 220/240 °C
Injection: 1 µl, splitless (split open at 1.0 min)
Carrier gas: H₂ (5.0)
Temperature: 40 °C (2 min) $\xrightarrow{5\text{ °C/min}}$ 230 °C (40 min)
Detector: FID

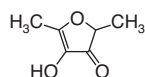
Peaks:

- | | | |
|---------------------------|---|-------------------------|
| 1. 2-Methylpropyl acetate | 10. 2-Methylpropionic acid | 18. 2-Phenylethanol |
| 2. Ethyl butanoate | 11. Butyric acid | 19. (E)-2-Hexenoic acid |
| 3. Methyl pentanoate | 12. 3-Methylbutyric acid | 20. Maltol |
| 4. 3-Methylbutyl acetate | 13. 3-Methylthiopropanol | 21. Furaneol |
| 5. Methyl hexanoate | 14. 2-Phenylethyl acetate | 22. Octanoic acid |
| 6. Ethyl hexanoate | 15. Hexanoic acid | 23. Eugenol |
| 7. Methyl heptanoate | 16. Benzyl alcohol | 24. 4-Vinylguaiacol |
| 8. Ethyl octanoate | 17. 3-Ethyl-2-hydroxy-2-cyclopenten-1-one | 25. Phenylacetic acid |
| 9. 2-Methylthioethanol | | 26. Vanillin |



B. Thum, W. Back, EBC Congress 1999

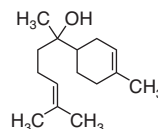
Furaneol



Maltol



Bisabolol



Alcoholic beverages



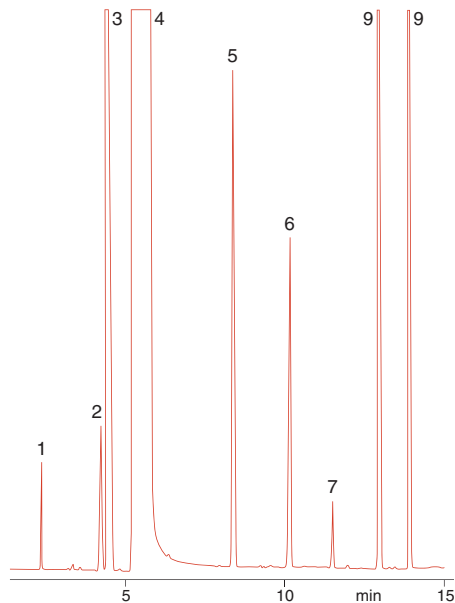
Analysis of a plum brandy

MN Appl. No. 210930

Column: PERMABOND® CW 20 M,
25 m x 0.32 mm ID, 0.5 µm
film, REF 723296.25,
max. temperature 220/240 °C
Injection: 1 ml, 150 °C, split 1:20
Carrier gas: N₂, 40 Kpa (1 – 2 ml/min)
Temperature: 50 °C (2 min) $\xrightarrow{2\text{ °C/min}}$ 60 °C
(5 min) $\xrightarrow{10\text{ °C/min}}$ 150 °C (9 min)
 $\xrightarrow{10\text{ °C/min}}$ 170 °C (5 min)
Detector: FID 180 °C

Peaks:

1. Acetaldehyde
2. Ethyl acetate
3. Methanol
4. Ethanol
5. Propanol-1
6. 2-Methylpropanol-1
7. Butanol-1
8. Isoamyl alcohol
9. Pentanol-1



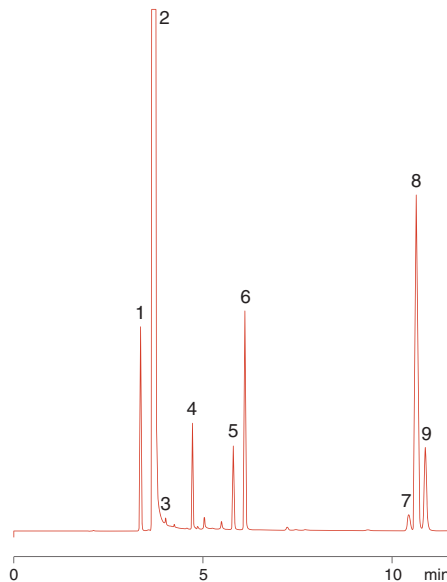
Analysis of brandy made from wine

MN Appl. No. 201480

Column: OPTIMA® 5,
50 m x 0.32 mm ID,
1.0 µm film,
REF 726325.50,
max. temperature 340/360 °C
Injection: 0.5 µl, split 30 ml/min
Carrier gas: 1 bar N₂
Temperature: 60 °C (10 min) $\xrightarrow{4\text{ °C/min}}$ 80 °C
Detector: FID 250 °C

Peaks:

1. Acetaldehyde/methanol
2. Ethanol
3. Acetone
4. Propanol-1
5. Ethyl acetate
6. *i*-Butanol
7. Diethyl acetal
8. 3-Methyl-1-butanol
9. 2-Methyl-1-butanol



Courtesy of Dr. Nilles, Wein- u. Bodenkulturbau,
Volkach, Germany

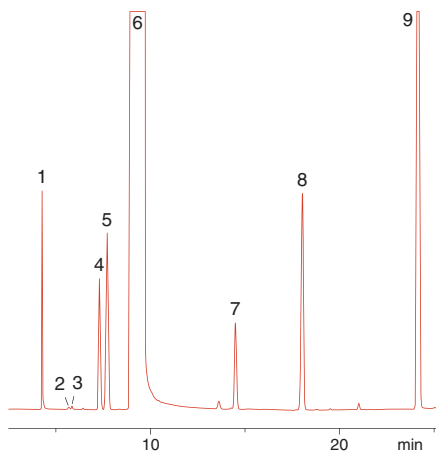
Food and cosmetic components

Analysis of brandy made from wine *MN Appl. No. 201490*

Column: PERMABOND® CW 20 M,
50 m x 0.32 mm ID,
0.5 µm film,
REF 723296.50,
max. temperature 220/240 °C
Injection: 1 µl, split 20 ml/min
Carrier gas: 1 bar N₂
Temperature: 60 °C (10 min) $\xrightarrow{4\text{ °C/min}}$ 140 °C
Detector: FID 250 °C

Peaks:

1. Acetaldehyde
2. Acetone
3. Methyl acetate
4. Ethyl acetate
5. Methanol
6. Ethanol
7. Propanol-1
8. *i*-Butanol
9. 2-Methyl-1-butanol/3-methyl-1-butanol

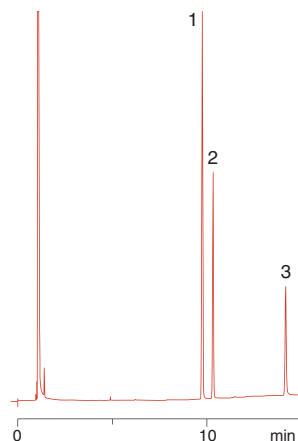


Analysis of DEG standard (5 g/l ethanol) for wine analysis *MN Appl. No. 201500*

Column: PERMABOND® CW 20 M-DEG
25 m x 0.25 mm ID,
0.25 µm film, REF 723063.25,
max. temperature 220/240 °C
Injection: 0.5 µl, split 1:40
Carrier gas: 1.2 bar N₂
Temperature: 80 °C $\xrightarrow{10\text{ °C/min}}$ 200 °C
Detector: FID 260 °C

Peaks: DEG standard

1. 1,4-Butanediol
2. Diethylene glycol (DEG)
3. Glycerol



Fatty acid methyl esters (FAME)



Summary of important saturated fatty acids

Code	Common name	Systematic name
C2:0	acetic acid	ethanoic acid
C3:0	propionic acid	propanoic acid
C4:0	butyric acid	butanoic acid
C4:0 iso	isobutyric acid	2-methylpropanoic acid
C5:0	valeric acid	pentanoic acid
C5:0 iso	isovaleric acid	3-methylbutanoic acid
C6:0	caproic acid	hexanoic acid
C6:0 iso	isocaproic acid	4-methylvaleric acid
C7:0	enanthic acid	heptanoic acid
C8:0	caprylic acid	octanoic acid
C9:0	pelargonic acid	nonanoic acid
C10:0	capric acid	decanoic acid
C12:0	lauric acid	dodecanoic acid
C14:0	myristic acid	tetradecanoic acid
C14:0 iso	isomyristic acid	12-methyltridecanoic acid
C15:0		pentadecanoic acid
C15:0 iso	13-methylmyristic acid	13-methyltetradecanoic acid
C15:0 anteiso	12-methylmyristic acid	12-methyltetradecanoic acid
C16:0	palmitic acid	hexadecanoic acid
C16:0 iso	isopalmitic acid	14-methylpentadecanoic acid
C17:0	margaric acid	heptadecanoic acid
C17:0 iso	isomargaric acid	15-methylhexadecanoic acid
	15-methylpalmitic acid	
C17:0 anteiso	anteisomargaric acid	14-methylhexadecanoic acid
	14-methylpalmitic acid	
C18:0	stearic acid	octadecanoic acid
C19:0 cyclo	methylene stearic acid	methylene octadecanoic acid
C20:0	arachidic acid	eicosanoic acid
C22:0	behenic acid	docosanoic acid
C23:0	tricosanoic acid	tricosanoic acid
C24:0	lignoceric acid	tetracosanoic acid
C26:0	cerotic acid	hexacosanoic acid

Food and cosmetic components

Summary of important unsaturated fatty acids

Code	Systematic name	ω reference system	Common name	ω -3 acid	ω -6 acid
C3:1	propenoic acid		acrylic acid		
C6:2	2,4-hexadienoic acid	C6:2n2	sorbic acid		
C14:1	<i>cis</i> -9-tetradecenoic acid	C14:1n5c	myristoleic acid		
	<i>trans</i> -9-tetradecenoic acid	C14:1n5t	myristelaidic acid		
C15:1	<i>cis</i> -10-pentadecenoic acid	C15:1n5			
C16:1	<i>cis</i> -9-hexadecenoic acid	C16:1n7c	palmitoleic acid		
	<i>trans</i> -9-hexadecenoic acid	C16:1n7t	palmitelaidic acid		
C17:1	<i>cis</i> -10-heptadecenoic acid	C17:1n7			
C18:1	<i>cis</i> -9-octadecenoic acid	C18:1n9c	oleic acid		
	<i>trans</i> -9-octadecenoic acid	C18:1n9t	elaidic acid		
	<i>cis</i> -11-octadecenoic acid	C18:1n7	vaccenic acid		
C18:2	<i>cis</i> -9,12-octadecadienoic acid	C18:2n6c	linoleic acid		x
	<i>trans</i> -9,12-octadecadienoic acid	C18:2n6t	linolelaidic acid		x
C18:3	<i>cis</i> -9,12,15-octadecatrienoic acid	C18:3n3	α -linolenic acid	x	
	<i>cis</i> -6,9,12-octadecatrienoic acid	C18:3n6	γ -linolenic acid		x
C18:4	<i>cis</i> -6,9,12,15-octadecatetraenoic acid	C18:4n3	stearidonic acid	x	
C20:1	<i>cis</i> -11-eicosenoic acid	C20:1n9	gondoic acid		
C20:2	<i>cis</i> -11,14-eicosadienoic acid	C20:2n6			x
C20:3	<i>cis</i> -5,8,11-eicosatrienoic acid	C20:3n9	mead acid		
	<i>cis</i> -8,11,14-eicosatrienoic acid	C20:3n6			x
	<i>cis</i> -11,14,17-eicosatrienoic acid	C20:3n3		x	
C20:4	<i>cis</i> -5,8,11,14-eicosatetraenoic acid	C20:4n6	arachidonic acid		x
	<i>cis</i> -8,11,14,17-eicosatetraenoic acid	C20:4n3		x	
C20:5	<i>cis</i> -5,8,11,14,17-eicosapentaenoic acid	C20:5n3		x	
C22:1	<i>cis</i> -13-docosenoic acid	C22:1n9	erucic acid		
C22:2	<i>cis</i> -13,16-docosadienoic acid	C22:2n6			x
C22:3	<i>cis</i> -13,16,19-docosatrienoic acid	C22:3n3		x	
C22:4	<i>cis</i> -7,10,13,16-docosatetraenoic acid	C22:4n6	adrenic acid		x
C22:5	<i>cis</i> -7,10,13,16,19-docosapentaenoic a.	C22:5n3	clupanodonic acid	x	
C22:6	<i>cis</i> -4,7,10,13,16,19-docosahexaenoic a.	C22:6n3		x	
C24:1	<i>cis</i> -15-tetracosenoic acid	C24:1n9	nervonic acid		



Determination of butyric acid content in edible fats

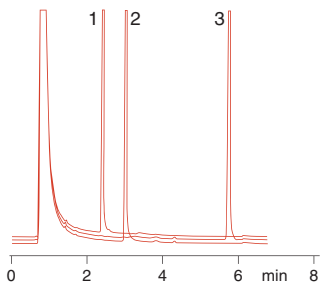
MN Appl. No. 210130

In so-called mixed spreads unsaturated vegetable oils and fats replace a certain percentage of milk fat (MF). The proportion of MF in the product has to be labelled to protect the consumer from fraudulent malpractice, since the price of MF is higher than that of other relevant raw materials. In order to check correct labelling of mixed spreads, food inspection authorities need a reliable analytical method to determine the percentage of MF in the spread. The most distinctive feature of MF, i.e. the unique occurrence of butyric acid, is most often determined by chromatography and used as an indicator for calculating the MF content in foodstuffs.

MF (100 mg) is weighed in a 10 ml HS vial, 1 ml sodium methoxide (0.5 % metallic sodium in methanol), 1 ml internal standard solution (isobutyric acid, valeric acid) are added, and the vial is sealed with a PTFE-lined crimp cap.

Column: OPTIMA® 5,
25 m x 0.32 mm ID,
1.0 µm film, REF 726325.25,
max. temperature 340/360 °C
Injection:
Carrier gas: 3 ml/min H₂, 50 kPa head-
pressure
Temperature: 50 °C (7 min) $\xrightarrow{4\text{ °C/min}}$ 60 °C
Detector: FID 250 °C

Peaks:
1. C4 iso
2. C4
3. C5

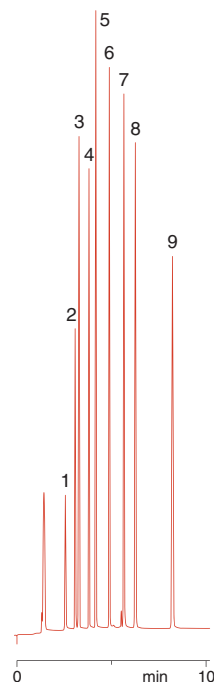


Analysis of free carboxylic acids C2 – C7

MN Appl. No. 201570

Column: PERMABOND® FFAP,
10 m x 0.25 mm ID,
0.25 µm film, REF 723116.10,
max. temperature 220/240 °C
Injection: 0.5 µl, split 1:50
Carrier gas: 0.3 bar N₂
Temperature: 120 °C $\xrightarrow{6\text{ °C/min}}$ 150 °C
(10 min)
Detector: FID 260 °C

Peaks:
1. C2
2. C3
3. C4 iso
4. C4
5. C5 iso
6. C5
7. C6 iso
8. C6
9. C7



F. Ulberth, Z. Lebensm. Unters. Forsch. A, 206
(1998) 305 – 307

Food and cosmetic components

Separation of C18 FAMES

MN Appl. No. 212640

Column: OPTIMA® δ -6,
50 m x 0.20 mm ID,
0.2 μ m film, REF 726465.50,
max. temperature 340/360 °C

Injection: 1 μ l, 0.1 % FAME in hexane
split 20 ml/min

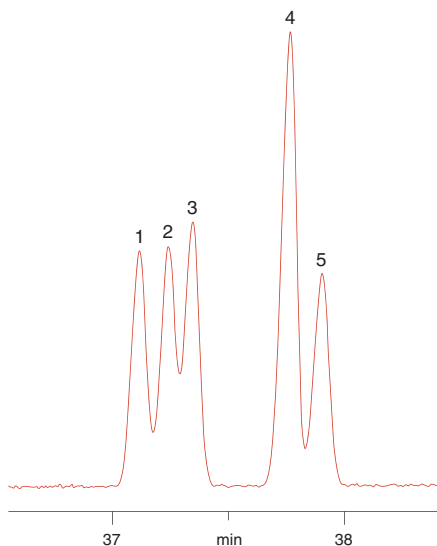
Carrier gas: 2.2 bar He

Temperature: 120 °C $\xrightarrow{6\text{ °C/min}}$ 180 °C
 $\xrightarrow{2\text{ °C/min}}$ 300 °C (5 min)

Detector: MSD 280 °C

Peaks:

1. C18:1n9c
2. C18:2n6c
3. C18:1n9t
4. C18:0
5. C18:3n3



Analysis of FAMES C10:0 – C18:2

MN Appl. No. 201700

Column: PERMABOND® FFAP,
25 m x 0.25 mm ID,
0.25 μ m film,
REF 723116.25,
max. temperature 220/240 °C

Injection: 0.5 μ l, split 1:50

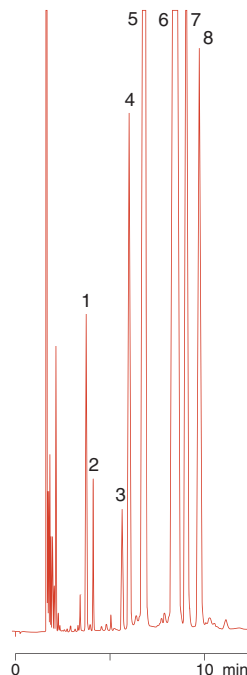
Carrier gas: 1 bar N₂

Temperature: 200 °C

Detector: FID 260 °C

Peaks:

1. C10:0
2. C16:0
3. C18:0
4. C18:1n9c
5. C18:2n6c
6. C18:2 (*cis, cis*, 9, 11)
7. C18:2 (*cis, trans*, 9, 11)
8. C18:2 (*trans, trans*, 9, 11)



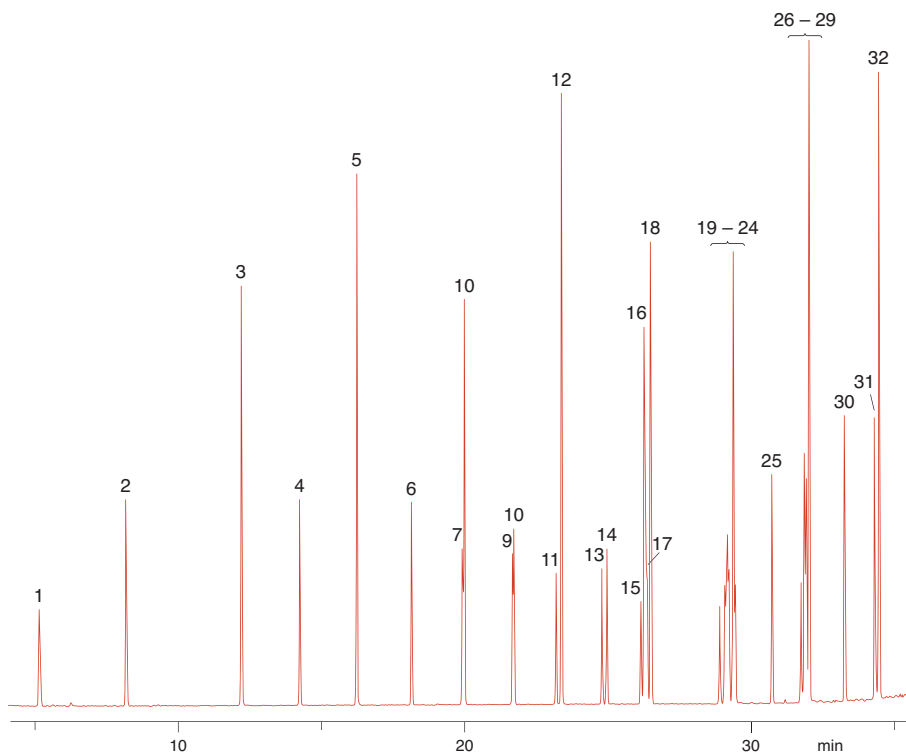


Analysis of FAMES MN Appl. No. 250120

Column: OPTIMA® δ-3, 25 m x 0.20 mm ID, 0.20 µm film, REF 726400.25,
max. temperature 340/360 °C
Injection: 1 µl, split 1:40
Carrier gas: 1.1 bar He
Temperature: 100 °C (1 min) $\xrightarrow{6\text{ °C/min}}$ 300 °C (5 min)
Detector: MSD

Peaks:

1. C6:0	9. C15:1	17. C18:0	25. C21:0
2. C8:0	10. C15:0	18. C18:2	26. C20:5
3. C10:0	11. C16:1	19. C20:4	27. C22:1
4. C11:0	12. C16:0	20. C20:3	28. C22:2
5. C12:0	13. C17:1	21. C20:1	29. C22:0
6. C13:0	14. C17:0	22. C20:2	30. C23:0
7. C14:1	15. C18:3	23. C20:0	31. C24:1
8. C14:0	16. C18:1	24. C20:3	32. C24:0



Food and cosmetic components

Analysis of FAMES

MN Appl. No. 250470

Column: OPTIMA® δ-6, 50 m x 0.20 mm ID, 0.20 µm film, REF 726465.50,
max. temperature 340/360 °C

Injection: 1 µl, 10 mg FAME mix per ml in CH₂Cl₂, split 1:50

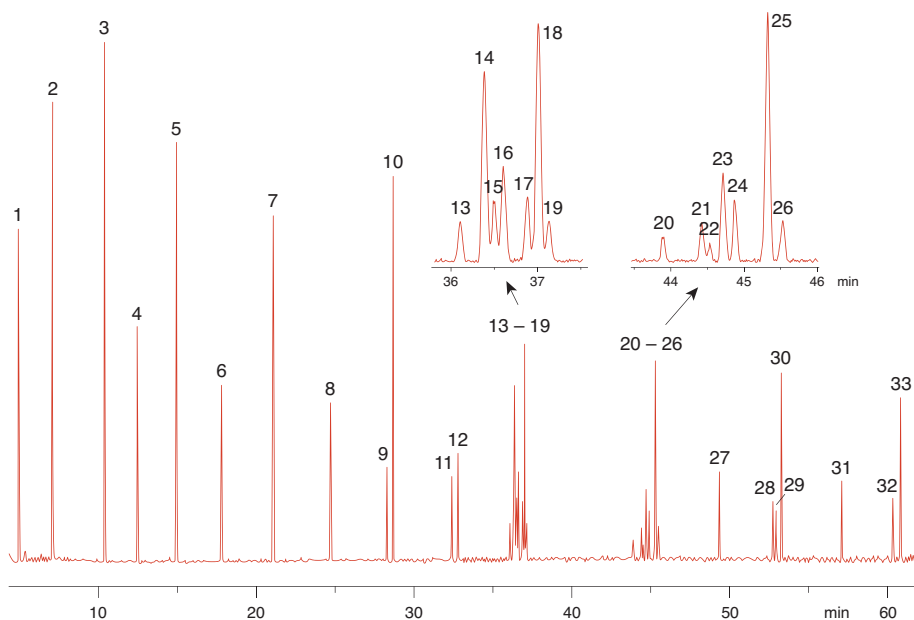
Carrier gas: 2.2 bar He

Temperature: 120 °C $\xrightarrow{6\text{ °C/min}}$ 180 °C $\xrightarrow{2\text{ °C/min}}$ 300 °C (5 min)

Detector: MSD

Peaks:

1. C6:0	10. C16:0	18. C18:0	26. C20:3n3
2. C8:0	11. C17:1n7	19. C18:3n3	27. C21:0
3. C10:0	12. C17:0	20. C20:4n6	28. C22:1n9
4. C11:0	13. C18:3n6	21. C20:3n6	29. C22:2n6
5. C12:0	14. C18:1n9c	22. C20:5n3	30. C22:0
6. C13:0	15. C18:2n6c	23. C20:1n9	31. C23:0
7. C14:0 + C14:1n5c	16. C18:1n9t	24. C20:2n6	32. C24:1n9
8. C15:0 + C15:1n5c	17. C18:2n6t	25. C20:0	33. C24:0
9. C16:1n7c			





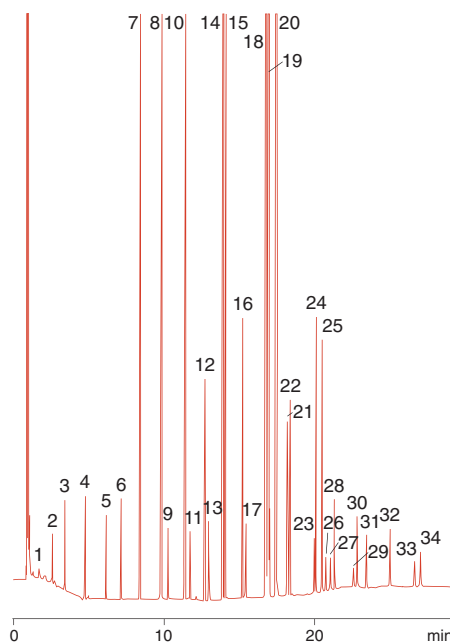
Analysis of FAMES in porcine fat MN Appl. No. 210060

Column: OPTIMA® 225, 25 m x 0.32 mm ID, 0.25 µm film, REF 726352.25,
max. temperature 260/280 °C
Injection: 1 µl, split 1:40
Carrier gas: 60 kPa H₂
Temperature: 50 °C (2 min) $\xrightarrow{30\text{ °C/min}}$ 125 °C $\xrightarrow{5\text{ °C/min}}$ 160 °C $\xrightarrow{20\text{ °C/min}}$ 180 °C $\xrightarrow{3\text{ °C/min}}$ 200 °C
 $\xrightarrow{20\text{ °C/min}}$ 220 °C (10 min)
Detector: FID 260 °C

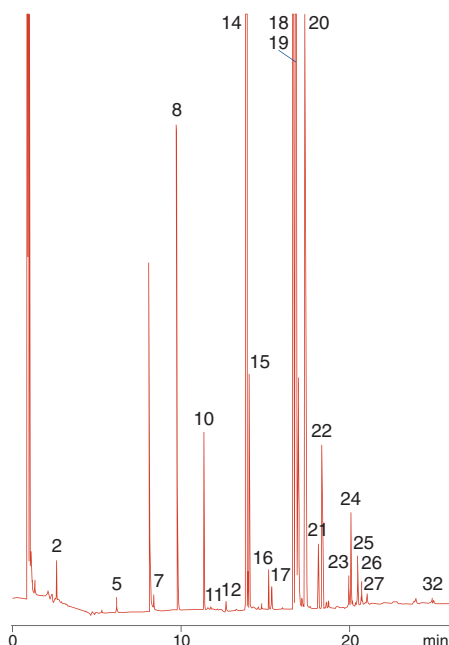
Peaks:

1. C4:0	6. C11:0	11. C14:1n5c	16. C17:0	21. C18:3	26. C20:4n6	31. C22:2
2. C5:0	7. C12:0	12. C15:0	17. C17:1	22. C19:0	27. C20:3	32. C22:6
3. C6:0	8. C13:0	13. C15:1	18. C18:0	23. C20:0	28. C20:5	33. C24:0
4. C8:0	9. C13:1	14. C16:0	19. C18:1n9c	24. C20:1	29. C22:0	34. C24:1n9
5. C10:0	10. C14:0	15. C16:1n7c	20. C18:2n6c	25. C20:2	30. C22:1	

FAME standard



FAMES in porcine fat



Courtesy of Dr. Bantleon, Mr. Leusche, Mr. Hagemann, VFG-Labor, Versmold, Germany

Food and cosmetic components

Analysis of FAMES from pork and salmon

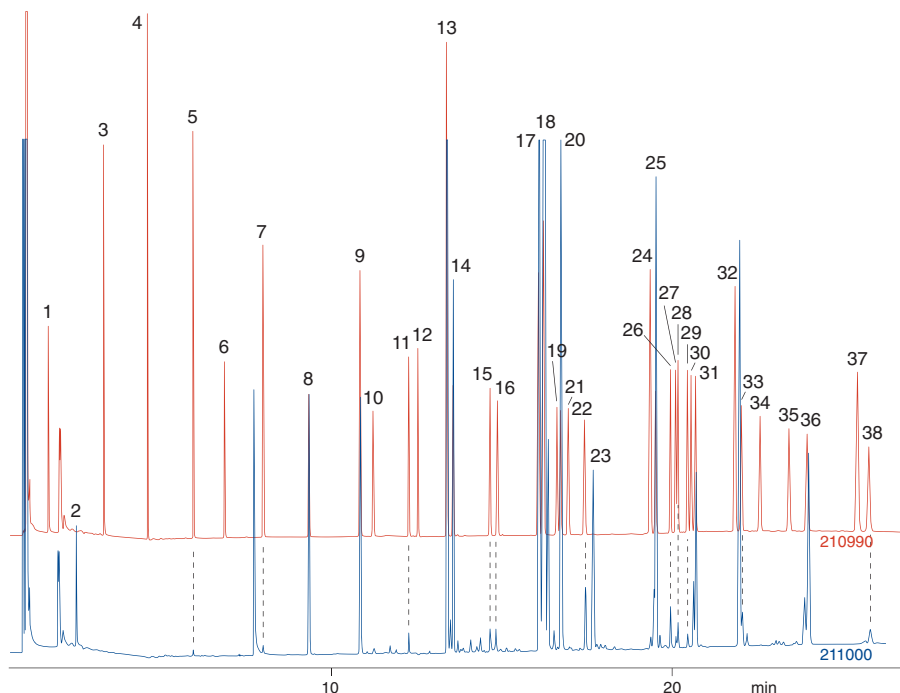
MN Appl. No. 210990 / 211000

Column: OPTIMA® 225, 25 m x 0.32 mm ID, 0.25 µm film, REF 726352.25,
max. temperature 260/280 °C
Injection: 1 µl, split 1:40
Carrier gas: 60 kPa H₂
Temperature: 50 °C (2 min) $\xrightarrow{30\text{ °C/min}}$ 125 °C $\xrightarrow{5\text{ °C/min}}$ 160 °C $\xrightarrow{20\text{ °C/min}}$ 180 °C $\xrightarrow{3\text{ °C/min}}$ 200 °C
 $\xrightarrow{20\text{ °C/min}}$ 220 °C (10 min)
Detector: FID 260 °C

Application 210990: FAME mixture · Application 211000: FAMES from pork and salmon

Peaks:

1. C4:0	9. C14:0	17. C18:0	25. C20:1n9	33. C22:1n9
2. C5:0	10. C14:1n5	18. C18:1n9c+t	26. C20:2n6	34. C22:2n6
3. C6:0	11. C15:0	19. C18:2n6t	27. C20:3n6	35. C23:0
4. C8:0	12. C15:1n5	20. C18:2n6c	28. C20:4n6	36. C22:6n3
5. C10:0	13. C16:0	21. C18:3n6	29. C20:3n3	37. C24:0
6. C11:0	14. C16:1n7	22. C18:3n3	30. C21:0	38. C24:1n9
7. C12:0	15. C17:0	23. C19:0	31. C20:5n3	
8. C13:0	16. C17:1n7	24. C20:0	32. C22:0	



Courtesy of Dr. Bantleon, Mr. Leusche, Mr. Hagemann, VFG-Labor, Versmold, Germany

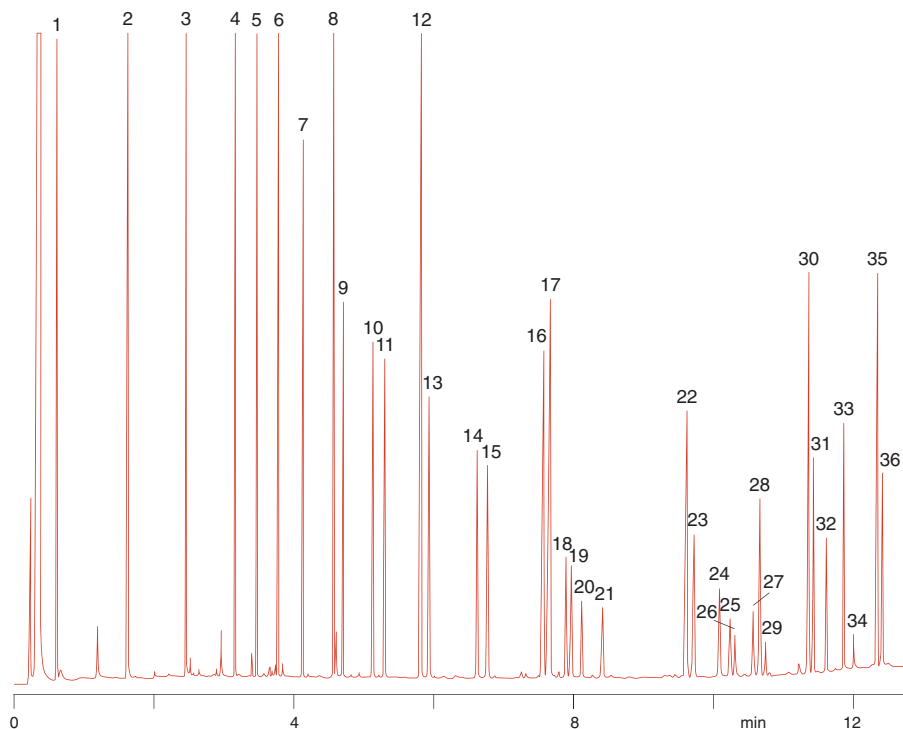


Fast analysis of a FAME mixture MN Appl. No. 211820

Column: OPTIMA® 225, 10 m x 0.1 mm ID, 0.1 µm film, REF 726080.10,
max. temperature 260/280 °C
Injection: 1 µl, split 1:250
Carrier gas: 280 kPa H₂
Temperature: 50 °C (1 min) $\xrightarrow{40\text{ °C/min}}$ 160 °C $\xrightarrow{6\text{ °C/min}}$ 200 °C $\xrightarrow{25\text{ °C/min}}$ 240 °C (2 min)
Detector: FID 260 °C

Peaks:

1. C4:0	10. C15:0	19. C18:2n6c	28. C21:0
2. C6:0	11. C15:1n5	20. C18:3n6	29. C20:5n3
3. C8:0	12. C16:0	21. C18:3n3	30. C22:0
4. C10:0	13. C16:1	22. C20:0	31. C22:1n9
5. C11:0	14. C17:0	23. C20:1n9	32. C22:2n6
6. C12:0	15. C17:1n7	24. C20:2n6	33. C23:0
7. C13:0	16. C18:0	25. C20:3n6	34. C22:6n3
8. C14:0	17. C18:1n9c+t	26. C20:4n6	35. C24:0
9. C14:1	18. C18:2n6t	27. C20:3n3	36. C24:1



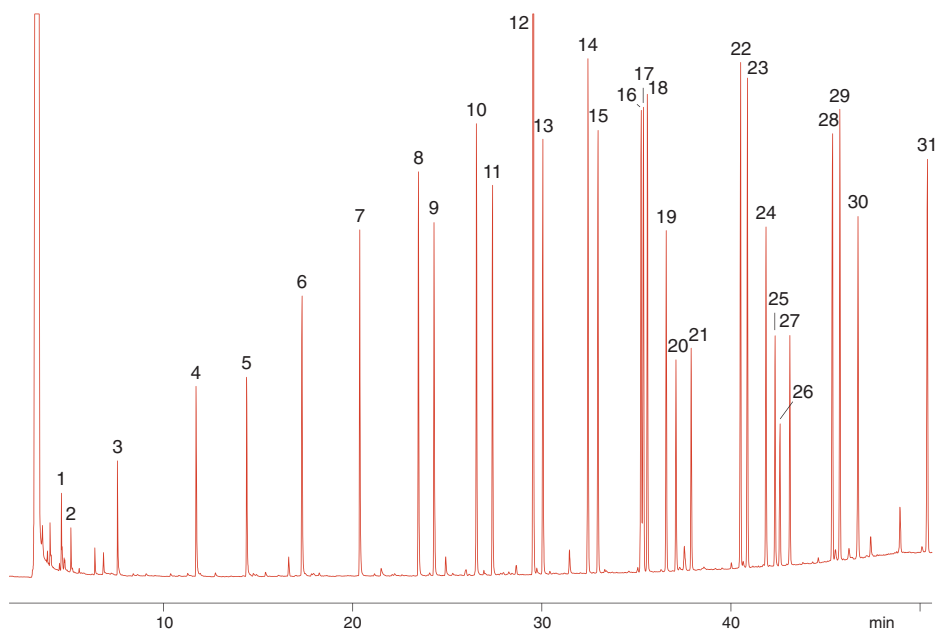
Food and cosmetic components

Analysis of FAMES incl. *cis/trans* C18:1 MN Appl. No. 201620

Column: OPTIMA® 240, 60 m x 0.25 mm ID, 0.25 µm film, REF 726089.60,
max. temperature 260/280 °C
Injection: 1.0 µl of a FAME mixture, split 1:25
Carrier gas: 150 kPa H₂
Temperature: 80 °C $\xrightarrow{20\text{ °C/min}}$ 120 °C $\xrightarrow{3\text{ °C/min}}$ 260 °C (10 min)
Detector: FID 280 °C

Peaks:

1. C4:0	9. C14:1	17. C18:1n9t	25. C20:3
2. C5:0	10. C15:0	18. C18:1n9c	26. C20:4
3. C8:0	11. C15:1	19. C18:2	27. C20:3
4. C10:0	12. C16:0	20. C18:3	28. C22:0
5. C11:0	13. C16:1	21. C18:3	29. C22:1
6. C12:0	14. C17:0	22. C20:0	30. C22:3
7. C13:0	15. C17:1	23. C20:1	31. C24:1
8. C14:0	16. C18:0	24. C20:2	





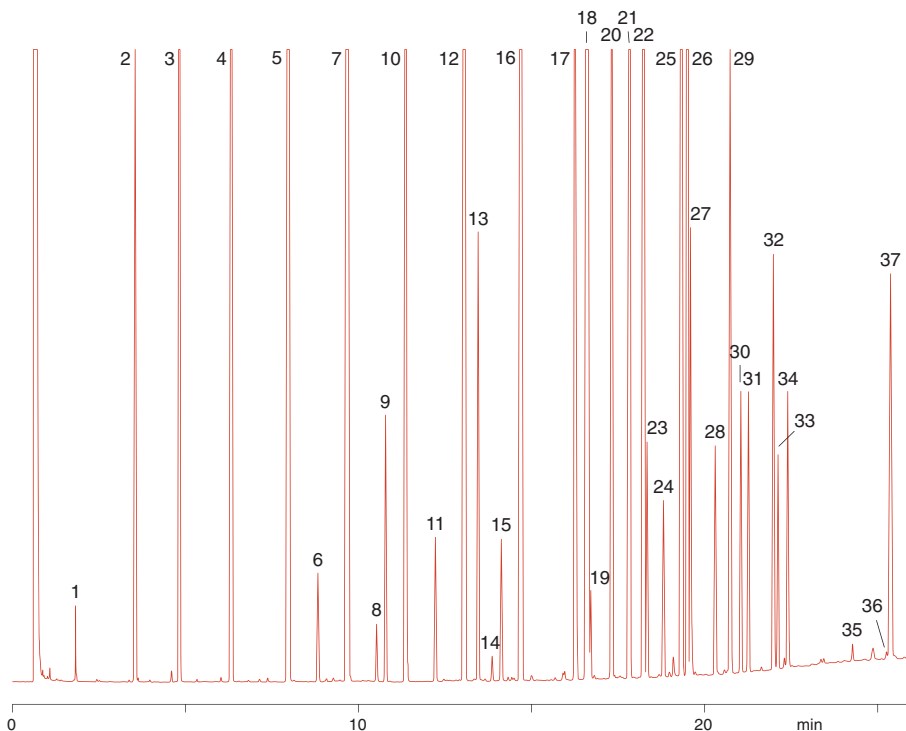
Analysis of FAMES

MN Appl. No. 201660

Column: PERMABOND® FFAP, 25 m x 0.25 mm ID, 0.1 µm film, REF 723936.25,
max. temperature 220/240 °C
Injection: 0.5 µl, split 120 ml/min
Carrier gas: 0.75 bar H₂
Temperature: 100 °C $\xrightarrow{6\text{ °C/min}}$ 240 °C (5 min)
Detector: FID 250 °C

Peaks:

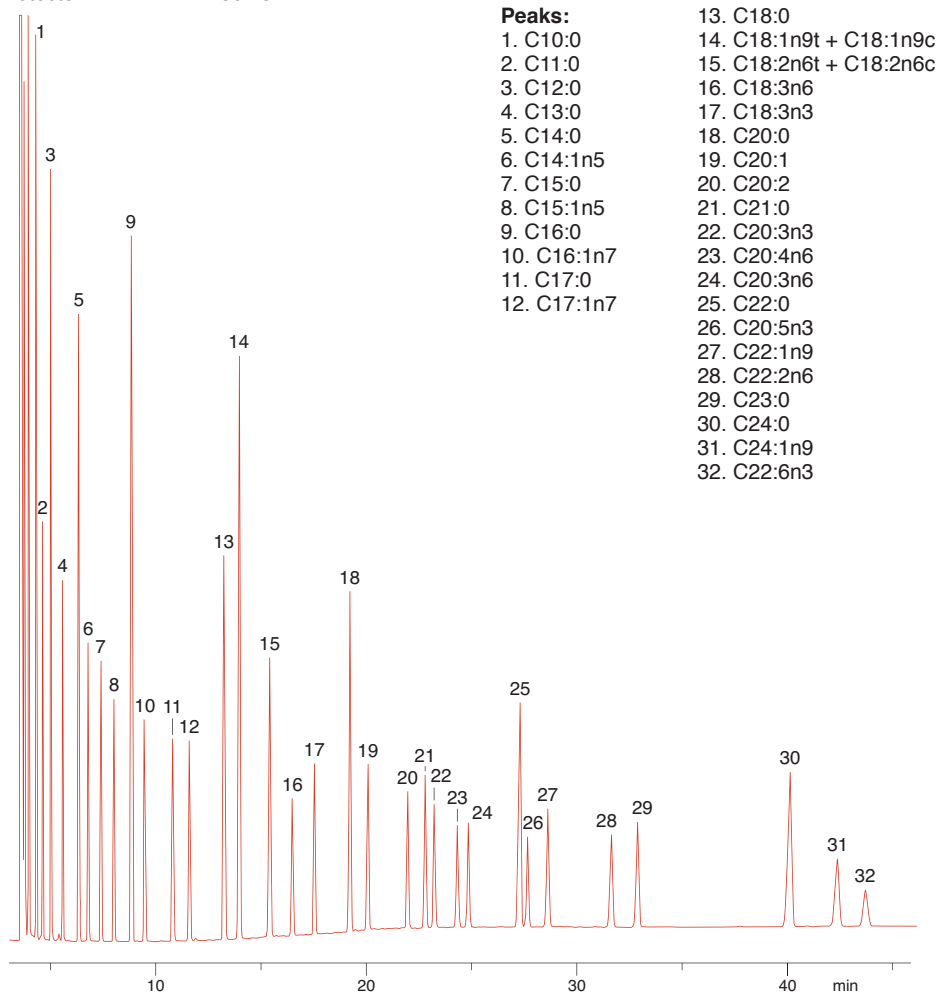
1. C8:0	11. C16:0 iso	21. C19:0 + C18:3n6	31. C20:3n3
2. C10:0	12. C16:0	22. C19:0 cyclo	32. C20:5
3. C11:0	13. C16:1	23. C18:3n3	33. C22:0
4. C12:0	14. C17:0 iso	24. C18:4	34. C22:1
5. C13:0	15. C17:0 anteiso	25. C20:0	35. C24:0
6. C14:0 iso	16. C17:0	26. C20:1	36. C24:1
7. C14:0	17. C18:0	27. C20:1	37. C22:6
8. C15:0 iso	18. C18:1n9c	28. C20:2	
9. C15:0 anteiso	19. C18:1n9t	29. C21:0	
10. C15:0	20. C18:2	30. C20:4	



Food and cosmetic components

Analysis of a FAME standard MN Appl. No. 212750

Column: PERMABOND® FFAP, 50 m x 0.25 mm ID, 0.25 µm film, REF 723116.50,
max. temperature 220/240 °C
Injection: 1.0 µl, FAME standard, 250 °C
Carrier gas: 80 kPa H₂
Temperature: 200 °C (10 min) $\xrightarrow{2\text{ °C/min}}$ 220 °C
Detector: FID 250 °C



Courtesy of Dr. E. Most, Institut für Tierernährung und Ernährungsphysiologie,
Justus-Liebig-Universität, Gießen, Germany



Analysis of FAMES in rapeseed oil by fast GC

MN Appl. No. 212010

Column: PERMABOND® FFAP,
20 m x 0.10 mm ID,
0.1 µm film, REF 723180.20,
max. temperature 220/240 °C

Injection: 0.5 µl

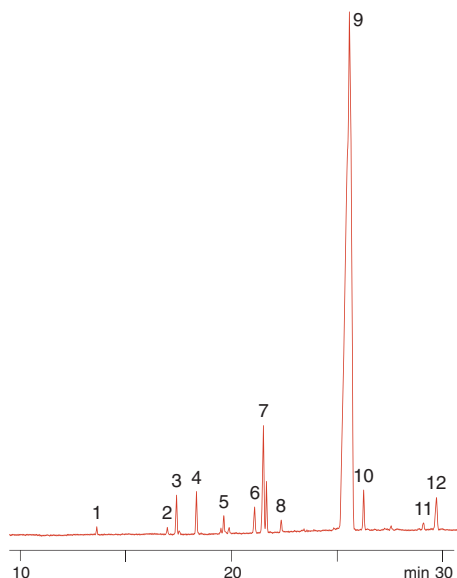
Carrier gas: 170 kPa

Temperature: 60 °C $\xrightarrow{15\text{ °C/min}}$ 120 °C $\xrightarrow{10\text{ °C/min}}$
210 °C (5 min) $\xrightarrow{15\text{ °C/min}}$ 240 °C
(30 min)

Detector: FID 300 °C

Peaks:

- | | |
|-------------|--------------------|
| 1. C16:0 | 7. C20:1 |
| 2. C18:0 | 8. C20:2 |
| 3. C18:1n9c | 9. C22:0 + C22:1n9 |
| 4. C18:2 | 10. C22:2 |
| 5. C18:3 | 11. C24:0 |
| 6. C20:0 | 12. C24:1 |



Analysis of FAMES in rapeseed oil by fast GC

MN Appl. No. 212150

Column: PERMABOND® FFAP,
10 m x 0.10 mm ID,
0.25 µm film, REF 723181.10,
max. temperature 220/240 °C

Injection: 1.0 µl, 250 °C

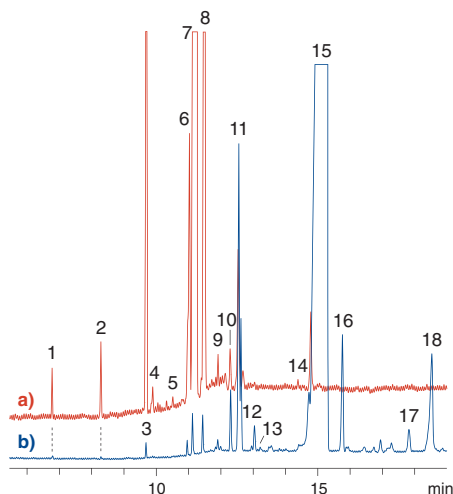
Carrier gas: He

Temperature: 60 °C $\xrightarrow{15\text{ °C/min}}$ 230 °C
(13.67 min)

Detector: FID 300 °C

Peaks:

- a) fraction rich in oleic acid
- b) fraction of rapeseed oil rich in erucic acid
- | | | |
|----------|-------------|-------------|
| 1. C12:0 | 7. C18:1n9c | 13. C21:0 |
| 2. C14:0 | 8. C18:2 | 14. C22:0 |
| 3. C16:0 | 9. C18:3 | 15. C22:1n9 |
| 4. C16:1 | 10. C20:0 | 16. C22:2 |
| 5. C17:0 | 11. C20:1 | 17. C24 |
| 6. C18:0 | 12. C20:2 | 18. C24:1 |



Food and cosmetic components

Analysis of FAMES from olive oil

MN Appl. No. 210550

Column: PERMABOND® FFAP,
25 m x 0.32 mm ID,
0.25 μ m film, REF 723341.25,
max. temperature 220/240 °C

Injection: 1 μ l, split 110 ml/min

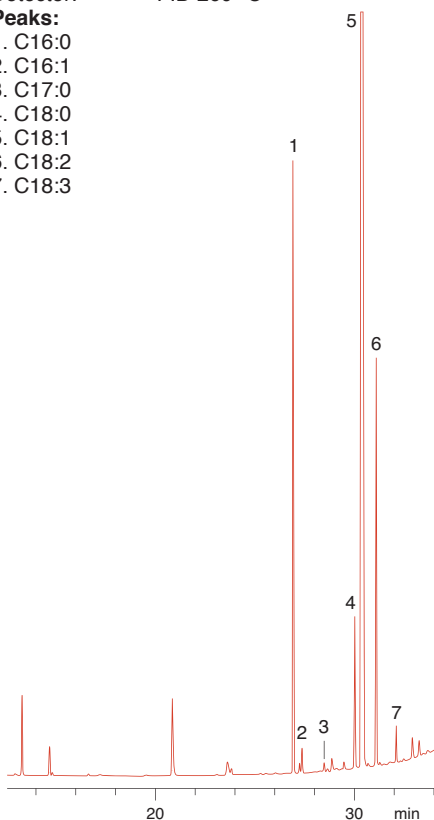
Carrier gas: 3.3 ml/min H₂

Temperature: 50 °C (4 min) $\xrightarrow{6\text{ °C/min}}$ 240 °C

Detector: FID 260 °C

Peaks:

1. C16:0
2. C16:1
3. C17:0
4. C18:0
5. C18:1
6. C18:2
7. C18:3



Analysis of ethoxylated fatty alcohols C12 and C14 (TMS)

MN Appl. No. 201760

Column: OPTIMA® 5,
25 m x 0.20 mm ID,
0.2 μ m film, REF 726857.25,
max. temperature 340/360 °C

Sample: ethoxylated fatty alcohols
R – (O – CH₂ – CH₂)_n – O –
TMS, silylated with BSTFA,
R = C12 and C14

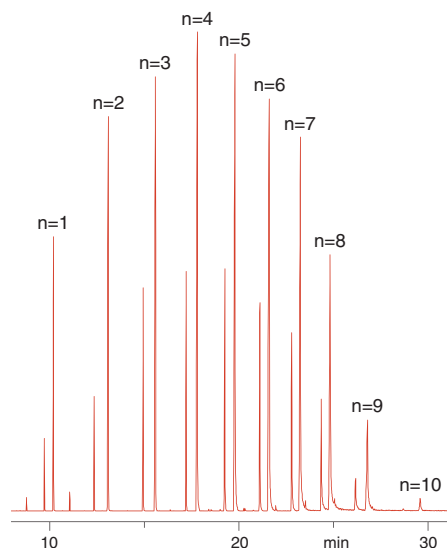
Injection: 0.5 μ l, 300 °C, split 1:150

Carrier gas: 25 cm/s He

Temperature: 120 °C (3 min) $\xrightarrow{10\text{ °C/min}}$ 330 °C

Detector: MSD

n = degree of ethoxylation



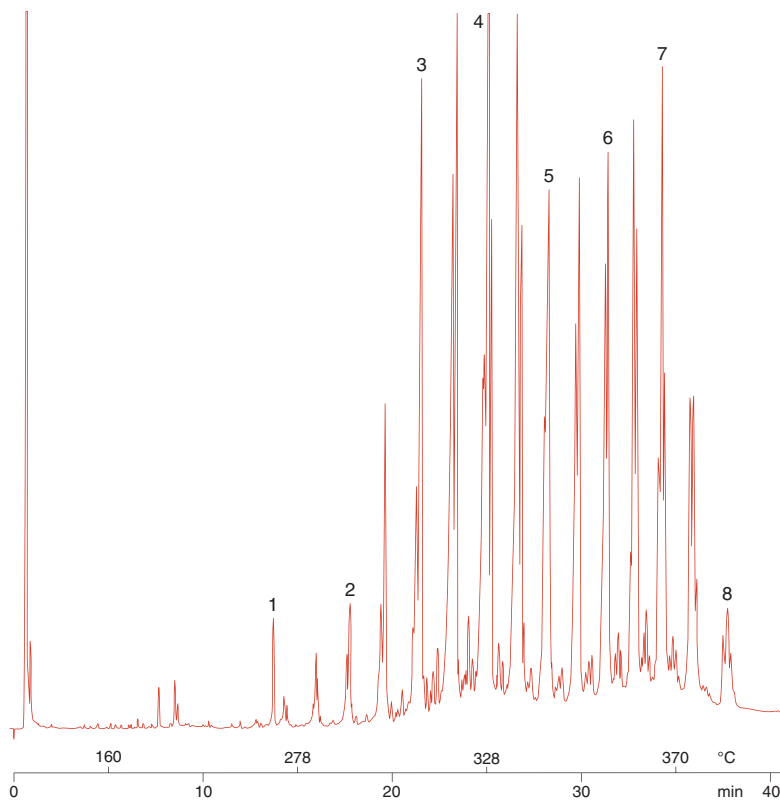


Analysis of triglycerides from butter MN Appl. No. 201790

Column: OPTIMA® 1-TG, 25 m x 0.32 mm ID, REF 726132.25, max. temperature 370 °C
Injection: 0.5 µl
Carrier gas: 80 kPa H₂
Temperature: 80 °C (1 min) $\xrightarrow{20\text{ °C/min}}$ 250 °C $\xrightarrow{5\text{ °C/min}}$ 370 °C (10 min)
Detector: FID 380 °C

Peaks:

1. Cholesterol
2. T-30
3. T-34
4. T-38
5. T-42
6. T-46
7. T-50
8. T-54



Food and cosmetic components

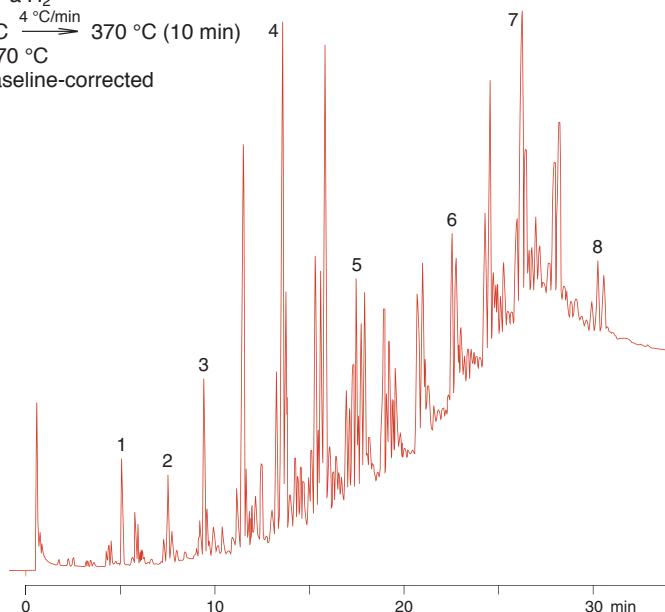
Analysis of triglycerides from butter

MN Appl. No. 201800

Column: OPTIMA® 17-TG, 25 m x 0.25 mm ID, REF 726130.25, max. temperature 370 °C
 Injection: 1 µl
 Carrier gas: 100 kPa H₂
 Temperature: 260 °C $\xrightarrow{4\text{ °C/min}}$ 370 °C (10 min)
 Detector: FID 370 °C
 not baseline-corrected

Peaks:

1. T-26
2. Cholesterol
3. T-30
4. T-36
5. T-40
6. T-46
7. T-50
8. T-54



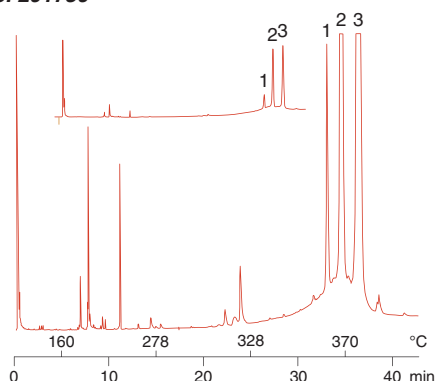
Analysis of triglycerides from olive oil

MN Appl. No. 201780

Column: OPTIMA® 1-TG, 25 m x 0.32 mm ID, REF 726132.25, max. temperature 370 °C
 Injection: 0.5 µl
 Carrier gas: 80 kPa H₂
 Temperature: 80 °C (1 min) $\xrightarrow{20\text{ °C/min}}$ 250 °C
 $\xrightarrow{5\text{ °C/min}}$ 370 °C (10 min)
 Detector: FID 380 °C

Peaks:

1. T-50
2. T-52
3. T-54





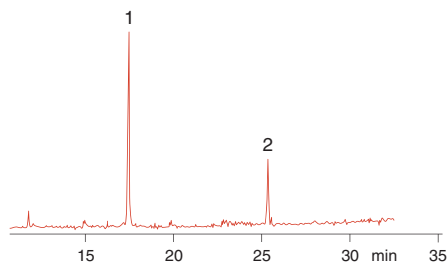
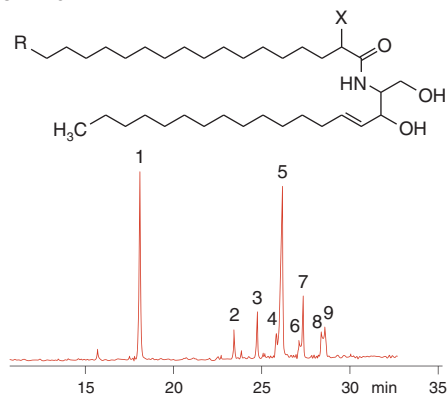
Ceramide analysis by GC/MS

MN Appl. No. 211980

Column: OPTIMA® 5,
30 m x 0.32 mm ID,
0.1 µm film, REF 726313.30,
max. temperature 340/360 °C
Injection: 0.5 µl, 320 °C, splitless
Carrier gas: He, 0.6 bar He
Temperature: 260 °C $\xrightarrow{3\text{ °C/min}}$ 360 °C
(20 min)
Detector: MSD

Peaks:

- | | |
|--|--|
| a) Ceramide IV
(2-hydroxy fatty
acid ceramides,
X = OH) | b) Ceramide III (non-
hydroxy fatty acid
ceramides, X = H) |
| 1. 18:0 | 1. 18:0 |
| 2. 22:0 | 2. 24:1 |
| 3. 23:0 | |
| 4. 24:1 | |
| 5. 24:0 | |



K. Raith, J. Darius, R.H.H. Neubert,
J. Chromatography A, **876** (2000) 229 – 233

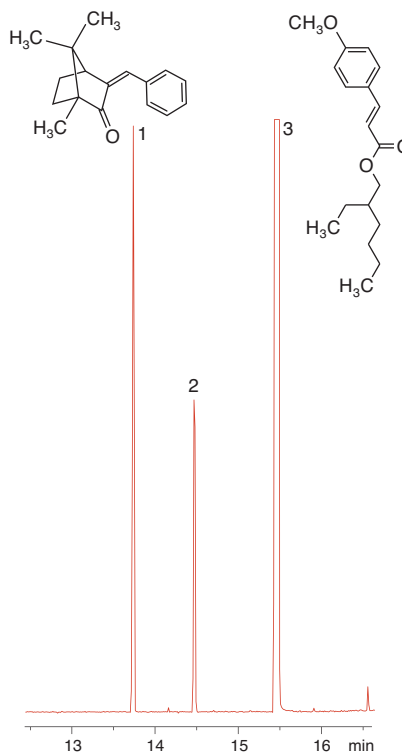
Analysis of parsols

MN Appl. No. 201240

Column: OPTIMA® 5,
25 m x 0.20 mm ID,
0.20 µm film, REF 726857.25,
max. temperature 340/360 °C
Injection: 1.0 µl parsol mixture,
1:1000 in MeOH, split 1:150
Carrier gas: 25 cm/s He
Temperature: 120 °C (3 min) $\xrightarrow{10\text{ °C/min}}$ 220 °C
 $\xrightarrow{20\text{ °C/min}}$ 330 °C
Detector: MSD

Peaks:

- Parsol 5000 =
3-(4'-Methylbenzylidene)camphor
- Parsol MCX = *p*-Methoxycinnamic acid
2-ethylhexyl ester (*cis* isomer)
- Parsol MCX (*trans* isomer)



Food and cosmetic components

Determination of benzimidazole anthelmintics in meat samples

MN Appl. No. 212040

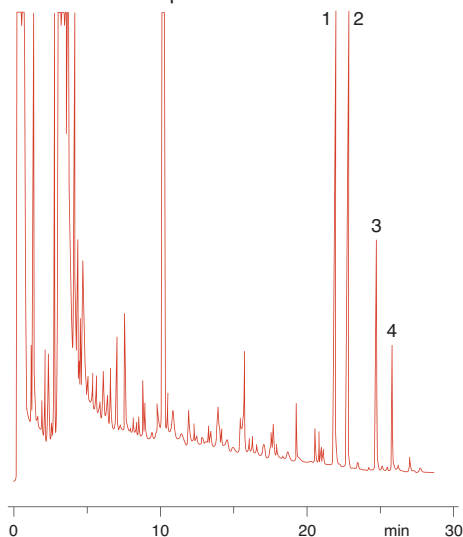
Column: OPTIMA® 1, 10 m x 0.25 mm ID, 0.25 µm film, REF 726050.10,
max. temperature 340/360 °C
Injection: 1 µl, 270 °C, splitless
Carrier gas: 1 ml/min He
Temperature: 60 °C (0.5 min) $\xrightarrow{30\text{ °C/min}}$ 150 °C $\xrightarrow{6\text{ °C/min}}$ 300 °C
Detector: PND 350 °C

Chromatogram a)

extract of a kidney sample from a pig treated with fenbendazole, derivatized with pentafluorobenzyl bromide

Peaks:

1. and 2. PFB derivatives of fenbendazole
3. and 4. PFB derivatives of the metabolite fenbendazole sulphone

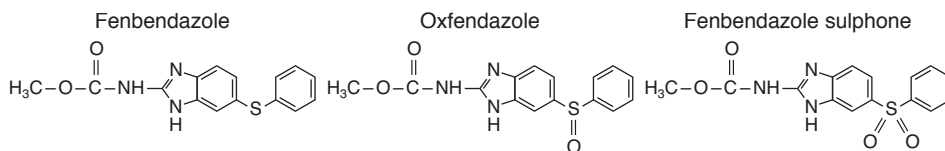
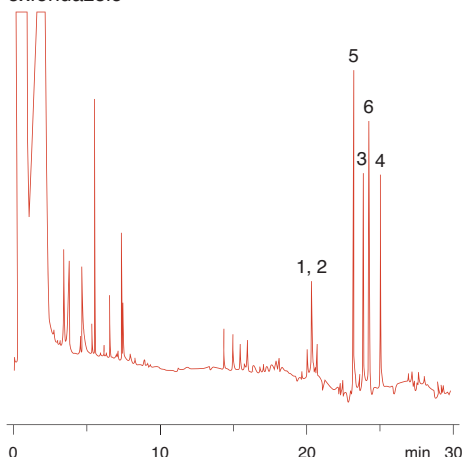


Chromatogram b)

extract of a blood sample from a pig treated with fenbendazole, derivatized with methyl iodide

Peaks:

1. and 2. Me derivatives of fenbendazole
3. and 4. Me derivatives of the metabolite fenbendazole sulphone
5. and 6. Me derivatives of the metabolite oxfendazole



A.M. Marti, A.E. Mooser, H. Koch, J. Chromatography **498** (1990) 145 – 157



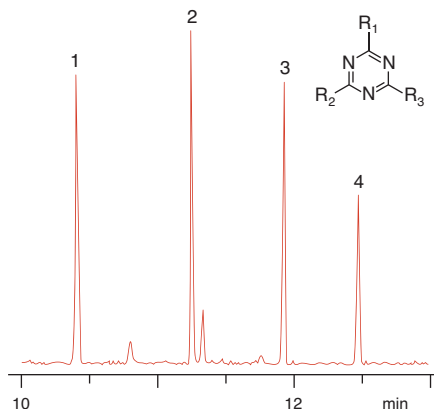
Determination of melamine, ammeline, ammeline and cyanuric acid in accordance with FDA regulations

MN Appl. No. 213300

Column: OPTIMA® 5 MS,
30 m x 0.25 mm ID,
0.25 µm film, REF 726220.30,
max. temperature 340/360 °C
Injection: 1 µl, 250 °C, split 10 ml/min
Carrier gas: 0.5 ml/min He
Temperature: 75 °C (1 min) $\xrightarrow{15\text{ °C/min}}$ 320 °C
Detector: MSD

Peaks:

1. Cyanuric acid ($R_1 = R_2 = R_3 = \text{OH}$)
2. Ammelide ($R_1 = \text{NH}_2$, $R_2 = R_3 = \text{OH}$)
3. Ammeline ($R_1 = R_2 = \text{NH}_2$, $R_3 = \text{OH}$)
4. Melamine ($R_1 = R_2 = R_3 = \text{NH}_2$)



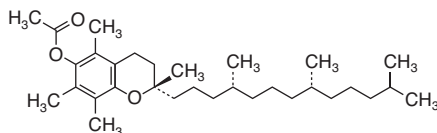
Analysis of D, L-α-tocopherol acetate (vitamin E acetate)

MN Appl. No. 210440

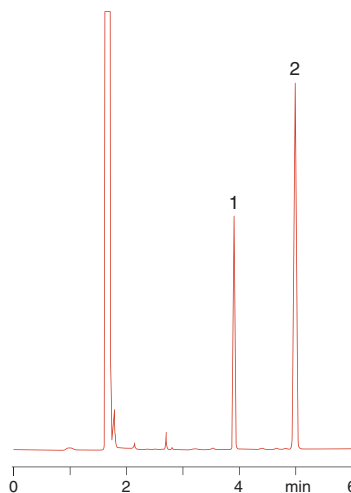
Column: OPTIMA® 1,
25 m x 0.25 mm ID,
0.25 µm film, REF 726050.25,
max. temperature 340/360 °C
Injection: 1 µl, 320 °C, split 1:100
Carrier gas: 10 psi N₂
Temperature: 50 °C (3 min) $\xrightarrow{8\text{ °C/min}}$ 290 °C
Detector: FID 320 °C

Peaks:

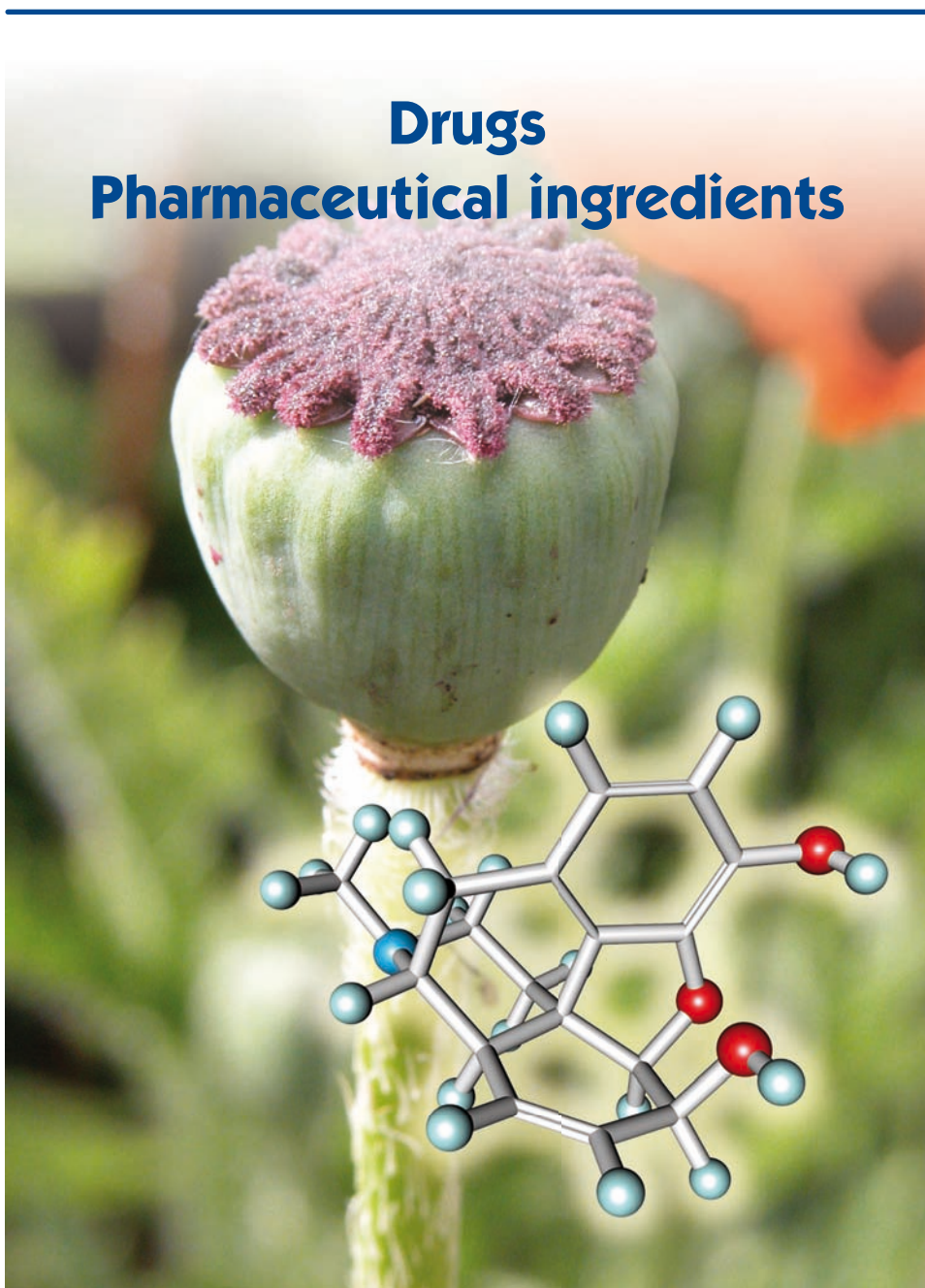
1. D,L-α-Tocopherol acetate [2.5 ng/ml]



2. Tetratriacontane [5.0 ng/ml]



Drugs Pharmaceutical ingredients





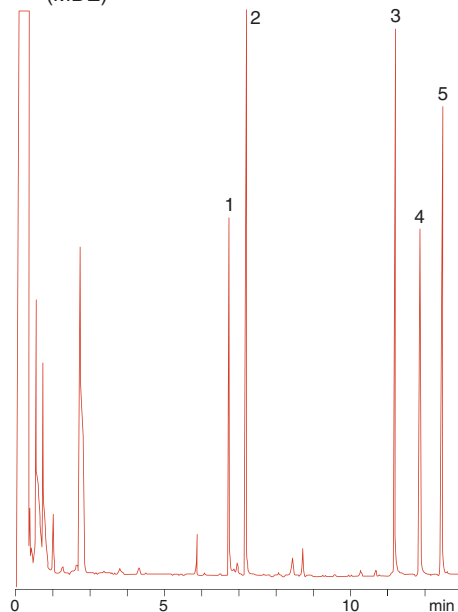
Analysis of amphetamines

MN App. No. 210460

Column: OPTIMA® δ-3,
30 m x 0.25 mm ID,
0.25 µm film,
REF 726420.30,
max. temperature 340/360 °C
Injection: 1.0 µl, 20 µg/ml in methanol,
250 °C, split 10 ml/min
Carrier gas: 2 ml/min He (5.0)
Temperature: 60 °C (2 min) $\xrightarrow{20\text{ °C/min}}$ 150 °C
 $\xrightarrow{6\text{ °C/min}}$ 250 °C (5 min)
Detector: MSD (Ion Trap)

Peaks:

1. Amphetamine
2. Methamphetamine
3. 3,4-Methylenedioxyamphetamine (MDA)
4. 3,4-Methylenedioxymethamphetamine (MDMA)
5. 3,4-Methylenedioxyethylamphetamine (MDE)

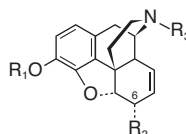


Courtesy of Mr. S. Motsch, Mr. P. Stein, Zoll-kriminalamt, Köln, Germany

Structures of selected drugs

Compound	Structure
Alprazolam	Midazolam
Noscapine = Narcotine	Papaverine
Meconin	Methadone

Morphine alkaloids



Substance	R ₁	R ₂	R ₃
Morphine	H	OH	CH ₃
Nalorphine	H	OH	CH ₂ -CH=CH ₂
Codeine	CH ₃	OH	CH ₃
Heroin	COCH ₃	OCOCH ₃	CH ₃

Drugs • Pharmaceutical ingredients

Compound	Structure	Compound	Structure
Clomethiazole		Nicotine	
Diphenhydramine		Caffeine	
Tilidine		Methaqualone	
Thioridazine		Thiothixene	
Butaperazine		Quinine	
Lidoflazine			



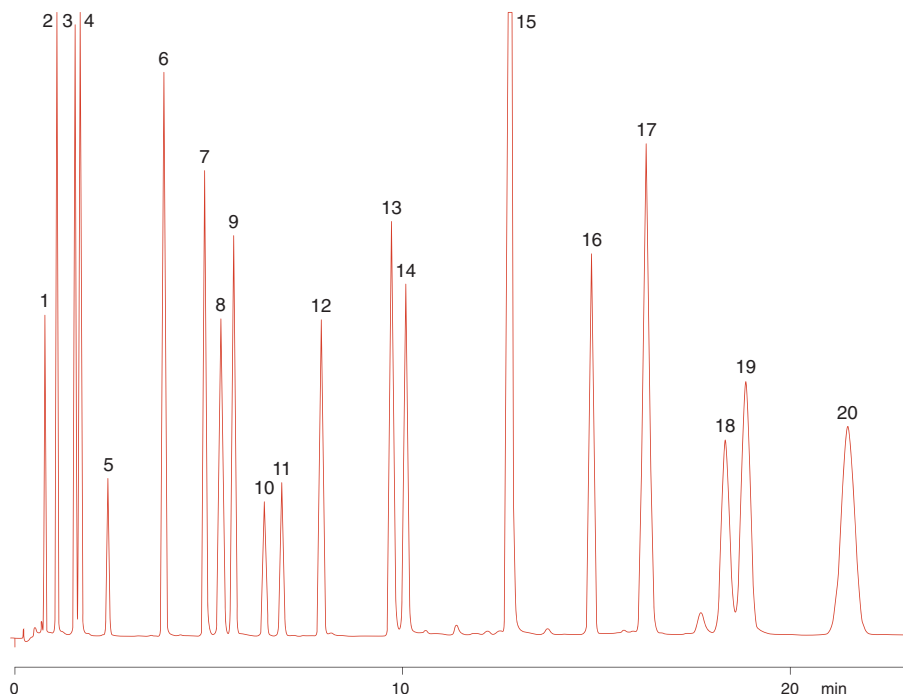
Drug screening analysis

MN Appl. No. 201300

Column: OPTIMA® 1, 15 m x 0.53 mm ID, 2.0 µm film, REF 726521.15,
max. temperature 320/340 °C
Injection: 1 µl in methanol, 0.5 min splitless, 270 °C
Carrier gas: 10 ml/min H₂ (constant flow)
Temperature: 134 °C $\xrightarrow{10\text{ °C/min}}$ 277 °C
Detector: N-FID 300 °C

Peaks: [ng]

- | | |
|-------------------------|---|
| 1. Amphetamine [50] | 11. 5-Chloro-2-aminobenzophenone (int. std.) [50] |
| 2. Clomethiazole [50] | 12. Methaqualone [50] |
| 3. Nicotine [20] | 13. Codeine [100] |
| 4. Ephedrine [100] | 14. Morphine [100] |
| 5. Barbitol [100] | 15. Quinine [200] |
| 6. Phenacetin [100] | 16. Thioridazine (int. std.) [100] |
| 7. Caffeine [20] | 17. Butaperazine [1006] |
| 8. Diphenhydramine [50] | 18. Thiothixene |
| 9. Tilidine [100] | 19. Thiothixene artefact } [100] |
| 10. Cyclobarbitol (200) | 20. Lidoflazine [100] |



Courtesy of Institut für Rechtsmedizin der Heinrich-Heine-Universität, Düsseldorf, Germany

Drugs • Pharmaceutical ingredients

Analysis of drugs of abuse

MN Appl. No. 211900

Column: OPTIMA® 5 MS,
12 m x 0.2 mm ID,
0.2 µm film, REF 726210.12,
max. temperature 340/360 °C

Injection: OCI/PTV, split ratio 28

Carrier gas: 1.0 ml/min, 57.9 cm/s

Total flow: 30.0 ml/min

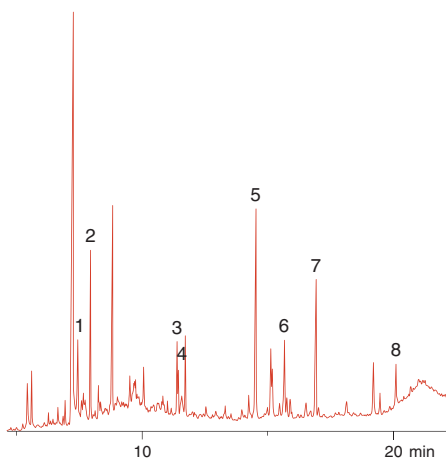
Pressure: 58.7 kPa (3 min) $\xrightarrow{7 \text{ kPa/min}}$
108 kPa $\xrightarrow{2.6 \text{ kPa/min}}$ 133 kPa
8 kPa/min $\xrightarrow{150 \text{ kPa (4.7 min)}}$

Temperature: 70 °C (3 min) $\xrightarrow{20 \text{ °C/min}}$
200 °C $\xrightarrow{7 \text{ °C/min}}$ 270 °C
20 °C/min $\xrightarrow{300 \text{ °C (4.7 min)}}$

Detector: MSD 280 °C

Peaks:

1. Amphetamine
2. Methamphetamine
3. Methadone
4. Methaqualone
5. Codeine
6. Morphine
7. Nalorphine
8. Alprazolam



Courtesy of Mr. Szigan, Laboratorium Lembke-Lempfrid, Köln, Germany

Analysis of cocaine

MN Appl. No. 210020

Column: OPTIMA® 1,
30 m x 0.25 mm ID, 0.25 µm film,
REF 726050.30,
max. temperature 340/360 °C

Sample: 4 mg of drug mixture dissolved
in 600 µl CH₂Cl₂ and derivatised
with MSTFA (REF 701270.201).

Injection: 1 µl

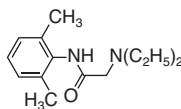
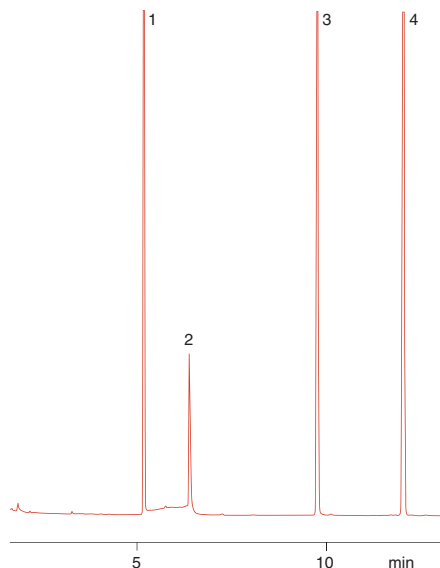
Carrier gas: N₂ $\xrightarrow{7.5 \text{ °C/min}}$

Temperature: 170 °C $\xrightarrow{280 \text{ °C (4 min)}}$

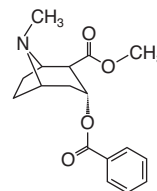
Detector: FID 330 °C

Peaks:

1. H-Lidocaine, MSTFA derivative
2. N-Lidocaine, MSTFA derivative
3. Cocaine
4. Internal standard



Lidocain



Cocaine



Analysis of heroin

MN Appl. No. 210010

Column: OPTIMA® 1,
30 m x 0.25 mm ID,
0.25 µm film, REF 726050.30,
max. temperature 340/360 °C

Sample preparation: 4 mg of a drug mixture are dissolved in 600 µl CH₂Cl₂ and derivatised with MSTFA (REF 701270.201).

Injection: 1 µl

Carrier gas: N₂

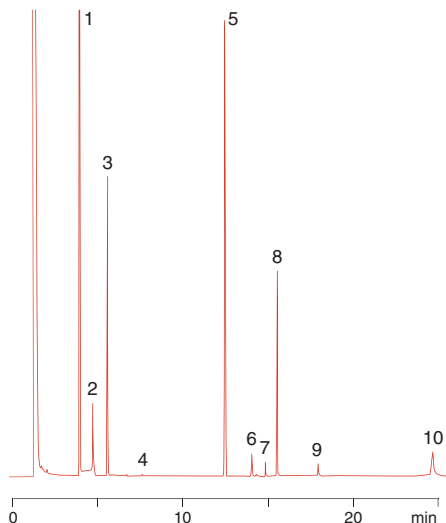
Temperature: 170 °C $\xrightarrow{7.5\text{ °C/min}}$ 280 °C

(9.5 min)

Detector: FID 330 °C

Peaks:

1. Diparacetamol (MSTFA derivative)
2. Monoparacetamol ((MSTFA derivative)
3. Caffeine
4. Glucose (MSTFA derivative)
5. Internal standard
6. Acetylcodeine
7. Acetylmorphine (MSTFA derivative)
8. Diacetylmorphine (heroin)
9. Papaverine
10. Narcotine (MSTFA derivative)



Analysis of commercial heroin

MN Appl. No. 250360

Column: OPTIMA® 5,
30 m x 0.20 mm ID,
0.35 µm film, REF 726860.30,
max. temperature 340/360 °C

Injection: 1 µl, 30 s splitless

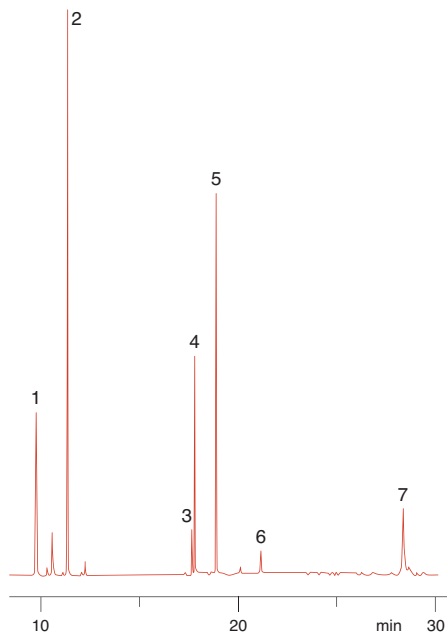
Carrier gas: 30 cm/s He

Temperature: 60 °C (1 min) $\xrightarrow{30\text{ °C/min}}$ 120 °C
 $\xrightarrow{10\text{ °C/min}}$ 280 °C

Detector: MSD

Peaks:

1. Paracetamol
2. Caffeine
3. Acetylcodeine
4. Monoacetylmorphine
5. Diacetylmorphine (heroin)
6. Papaverine
7. Narcotine



Courtesy of A. Jeger, Gerichtschemisches Institut, Basel, Switzerland

Drugs • Pharmaceutical ingredients

Analysis of heroin (street quality)

MN Appl. No. 210470

Column: OPTIMA® δ-3,
30 m x 0.25 mm ID,
0.25 μm film, REF 726420.30,
max. temperature 340/360 °C

Sample: 5 mg heroin (street quality) +
1 mg tetracosane + 1 ml CHCl₃ +
200 μl pyridine + 150 μl MSTFA

Injection: 1.0 μl, 250 °C, split 30 ml/min

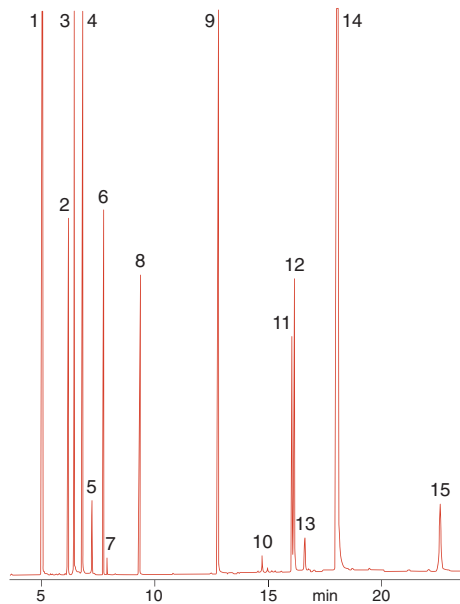
Carrier gas: 2 ml/min He (5.0)

Temperature: 150 °C $\xrightarrow{9\text{ °C/min}}$ 280 °C (25 min)

Detector: MSD (ion trap)

Peaks:

- | | |
|-----------------------|----------------------------------|
| 1. Paracetamol di-TMS | 10. Morphine di-TMS |
| 2. Glucose TMS peak 1 | 11. Monoacetyl-
morphine TMS |
| 3. Citric acid 4-TMS | 12. Acetylcodeine |
| 4. Glucose TMS peak 2 | 13. Acetylthebaol |
| 5. Paracetamol TMS | 14. Diacetylmorphine
(heroin) |
| 6. Glucose TMS peak 3 | 15. Papaverine |
| 7. Meconin | |
| 8. Caffeine | |
| 9. Tetracosane | |



Courtesy of Mr. S. Motsch, Mr. P. Stein,
Zollkriminalamt, Köln, Germany

Analysis of hashish

MN Appl. No. 212072

Column: OPTIMA® δ-3,
60 m x 0.25 mm ID,
0.25 μm film, REF 726420.60,
max. temperature 340/360 °C

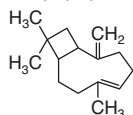
Carrier gas: 180 kPa (26 psi) He

Temperature: 100 °C $\xrightarrow{10\text{ °C/min}}$ 360 °C

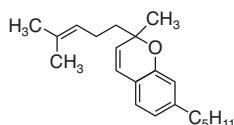
Detector: MSD

Peaks:

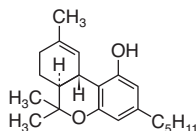
1. Caryophyllene



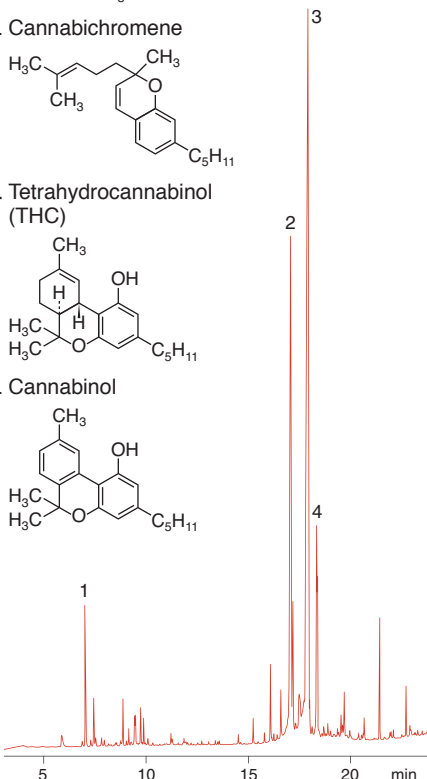
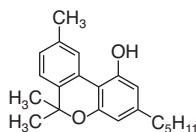
2. Cannabichromene



3. Tetrahydrocannabinol (THC)



4. Cannabinol



N. Bertram, LTA Labor f. Toxikol. u. Analytik,
Königswinter, Germany



Analysis of barbiturates

MN Appl. No. 201290

Column: OPTIMA® 5,
25 m x 0.25 mm ID,
0.25 µm film, REF 726056.25,
max. temperature 340/360 °C

Injection: 1 µl, split 1:10

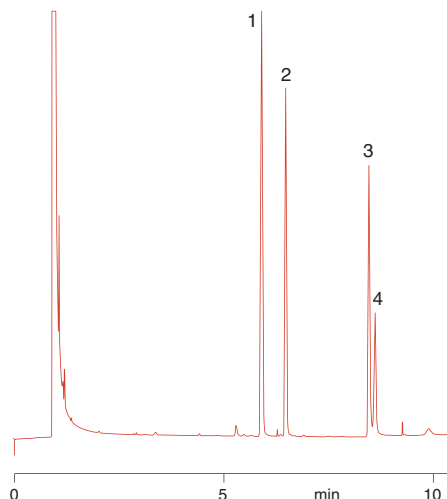
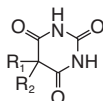
Carrier gas: 1.8 bar N₂

Temperature: 150 °C $\xrightarrow{8\text{ °C/min}}$ 300 °C

Detector: FID 280 °C

Peaks:

1. Pentobarbital
2. Secobarbital
3. Phenobarbital
4. Cyclobarbital



Compound	R ₁	R ₂
Barbital	C ₂ H ₅	C ₂ H ₅
Pentobarbital	C ₂ H ₅	CH(CH ₃)-C ₃ H ₇
Secobarbital	CH ₂ -CH=CH ₂	CH(CH ₃)-C ₃ H ₇
Phenobarbital	C ₂ H ₅	C ₆ H ₅
Cyclobarbital	C ₂ H ₅	

Analysis of narcotic analgesic drugs in urine

MN Appl. No. 201270

Column: OPTIMA® 17,
25 m x 0.20 mm ID,
0.20 µm film, REF 726065.25,
max. temperature 320/340 °C

Injection: 1.0 µl of a liquid extract of a
urine sample, 1:10 in MeOH,
concentration of peaks 2 – 5
approximately 0.5 µg/ml,
3 s splitless

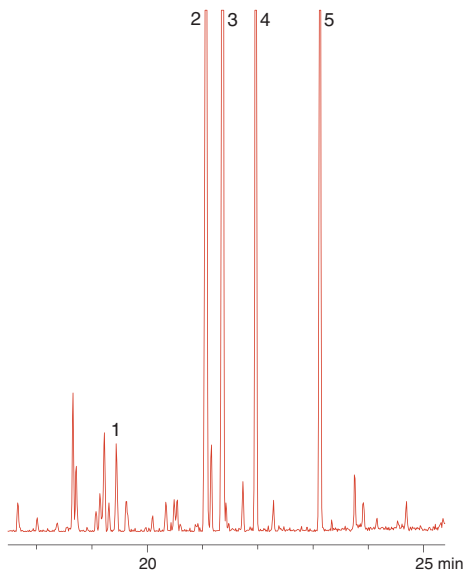
Carrier gas: 25 cm/s He

Temperature: 130 °C (3 min) $\xrightarrow{8\text{ °C/min}}$ 320 °C

Detector: MSD

Peaks:

1. Levomethorphan
2. Dextrophan
3. unknown metabolite
4. Di(2-ethylhexyl) phthalate
5. unknown metabolite



Dextromethorphan and metabolites courtesy of
Dr. Baumann, Hôpital Prilly, Switzerland

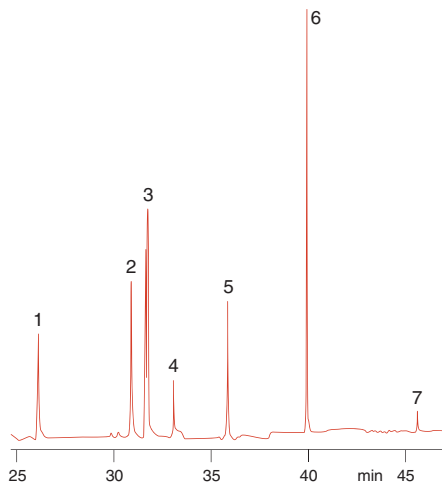
Drugs • Pharmaceutical ingredients

Analysis of β -agonists from urine MN Appl. No. 210670

Column: OPTIMA® δ -3, 30 m x 0.25 mm ID, 0.25 μ m film, REF 726420.30,
max. temperature 340/360 °C
Injection: 1.0 μ l, 10 μ g/l from urine, derivatised with HMDS/TMCS 9:1, 280 °C, splitless
Temperature: 100 °C (1 min) $\xrightarrow{5\text{ }^{\circ}\text{C/min}}$ 300 °C
Detector: MSD 300 °C

Peaks:

1. Mabuterol-TMS
2. Salbutamol-TMS
3. Clenbuterol-TMS + Clenbuterol-D₆-TMS
4. Cimaterol-TMS
5. Hydroxymethyl-clenbuterol-TMS
6. Isoxsuprine-TMS
7. Ractopamine-TMS



Courtesy of Mr. Wohlfarth, Mr. Rosensprung, Staatl. Untersuchungsamt, Wiesbaden, Germany

Structure	Compound	R ₁	R ₂	R ₃	R ₄
	Mabuterol	CF ₃	NH ₂	Cl	C(CH ₃) ₃
	Clenbuterol	Cl	NH ₂	Cl	C(CH ₃) ₃
	Cimaterol	CN	NH ₂	H	CH(CH ₃) ₂
	Salbutamol	CH ₂ OH	OH	H	C(CH ₃) ₃
	Isoxsuprine	CH ₃	O-C ₆ H ₅	–	–
	Ractopamine	H	CH ₂ - <i>p</i> -C ₆ H ₄ -OH	–	–



Analysis of acidic drugs from waste water

MN Appl. No. 212000

Sample preparation: 100 μ l sample is derivatised with 30 μ l of a 0.2 M methanolic TMSH solution (REF 701520.101).

Column: OPTIMA[®] δ -3, 30 m x 0.25 mm ID, 0.25 μ m film, REF 726420.30,
max. temperature 340/360 $^{\circ}$ C

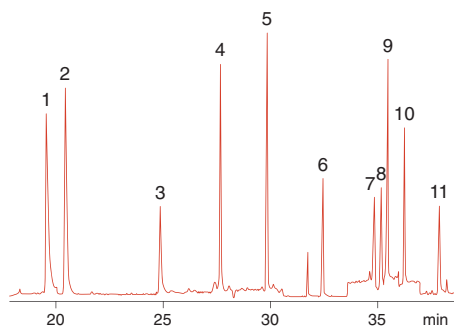
Injection: 2 μ l (5 μ l/s), 275 $^{\circ}$ C, splitless

Temperature: 70 $^{\circ}$ C (1 min) $\xrightarrow{15\text{ }^{\circ}\text{C/min}}$ 280 $^{\circ}$ C $\xrightarrow{5\text{ }^{\circ}\text{C/min}}$ 300 $^{\circ}$ C (10 min)

Detector: MSD

Peaks:

1. Clofibric acid
2. Ibuprofen
3. 2,3-Dichlorophenoxyacetic acid (int. std.)
4. Gemfibrozil
5. Fenoprofen
6. Naproxen
7. Ketoprofen
8. Indomethacin
9. Tolfenamic acid
10. Diclofenac
11. Meclofenamic acid



Courtesy of Staatliches Umweltamt Aachen, Germany

Structure	Compound	R ₁	R ₂	R ₃	R ₄
	Clofibric acid	H	Cl	–	C(CH ₃) ₂ –COOH
	Gemfibrozil	CH ₃	CH ₃	–	(CH ₂) ₃ –C(CH ₃) ₂ –COOH
	Diclofenac	Cl	H	Cl	CH ₂ –COOH
	Meclofenamic acid	Cl	CH ₃	Cl	COOH
	Tolfenamic acid	H	Cl	CH ₃	COOH
	Naproxen	–	–	–	–

Other formulas see structure index from page 291.

Drugs • Pharmaceutical ingredients

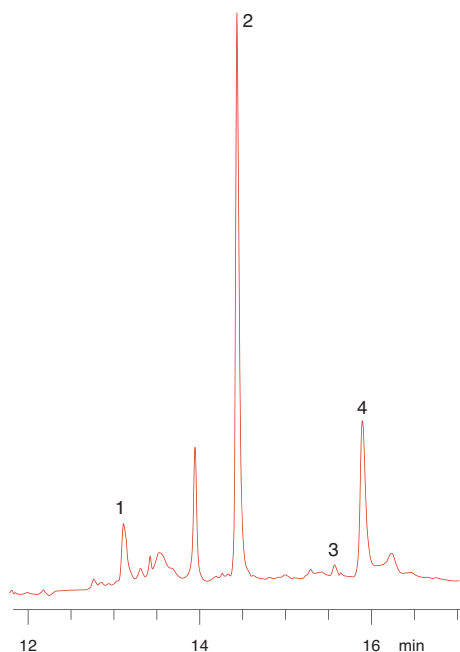
Determination of picogram levels of midazolam and metabolites in human plasma

MN Appl. No. 212580

Column: OPTIMA® 1,
25 m x 0.25 mm ID,
0.25 µm film, REF 726050.25,
max. temperature 340/360 °C
Injection: 3 µl, 300 °C, pulsed splitless
Carrier gas: 1.1 ml/min He
(8.6 psi, 43 cm/s)
Temperature: 85 °C (1 min) $\xrightarrow{30\text{ °C/min}}$ 200 °C
 $\xrightarrow{10\text{ °C/min}}$ 310 °C (1 min)
Detector: MSD

Peaks:

1. Midazolam
2. *N*-Ethyloxazepam (int. std.)
3. 4-Hydroxymidazolam
4. 1-Hydroxymidazolam



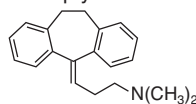
Analysis of amitriptyline

MN Appl. No. 210370

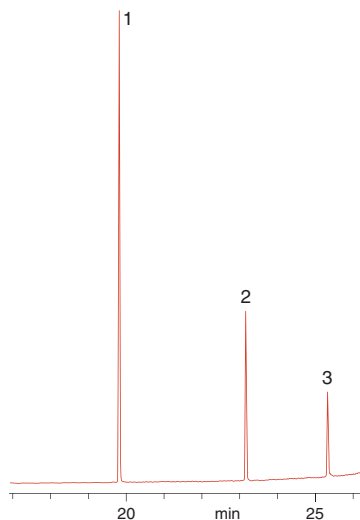
Column: OPTIMA® 5 Amine,
25 m x 0.20 mm ID,
0.35 µm film, REF 726355.25,
max. temperature 300/320 °C
Injection: 310 °C
Carrier gas: 25 cm/s He
Temperature: 70 °C (3 min) $\xrightarrow{10\text{ °C/min}}$ 290 °C
Detector: MSD

Peaks:

1. Diphenhydramine
2. Amitriptyline



3. 5-(*p*-Methylphenyl)-5-phenylhydantoin (MPPH)



C. B. Eap et al., J. Chromatography B, **802**
(2004) 339 – 345



Analysis of *Fusarium* mycotoxins

MN Appl. No. 210660

Column: OPTIMA® 1701,
25 m x 0.32 mm ID,
0.35 μ m film, REF 726824.25,
max. temperature 300/320 °C

Injection: 1 μ l, 250 °C, splitless

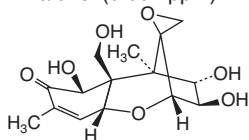
Carrier gas: 30 cm/s N₂, linear velocity at
160 °C

Temperature: 160 °C (3 min) $\xrightarrow{6\text{ °C/min}}$ 240 °C
 $\xrightarrow{30\text{ °C/min}}$ 270 °C (3 min)

Detector: ⁶³Ni-ECD 320 °C

Peaks:

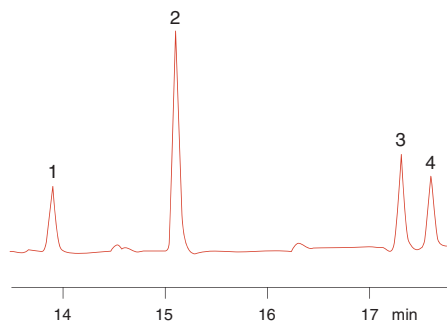
1. Nivalenol (0.007 ppm)



2. Deoxynivalenol (0.006 ppm)

3. 15- α -Acetyl-4-deoxynivalenol (0.006 ppm)

4. 3-Acetyldeoxynivalenol (0.005 ppm)



F. Walker, B. Meier, Journal of AOAC Int. Vol.
81, No. 4 (1998) 741 – 748

Analysis of free steroids

MN Appl. No. 201310

Column: OPTIMA® 17,
25 m x 0.53 mm ID,
1.0 μ m film, REF 726747.25,
max. temperature 300/320 °C

Injection: 1 μ l (0.1 % in CH₂Cl₂)
split 1:30

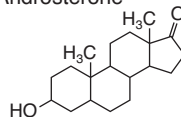
Carrier gas: 40 kPa N₂

Temperature: 260 °C

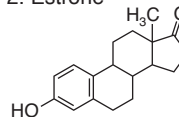
Detector: FID 280 °C

Peaks:

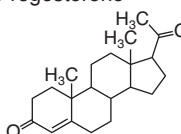
1. Androsterone



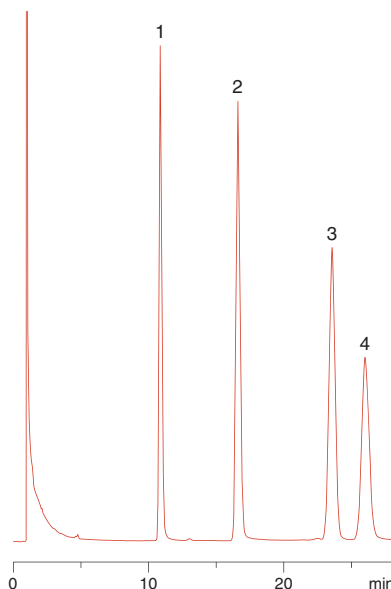
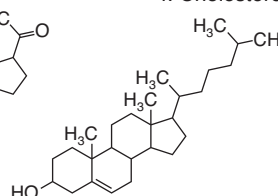
2. Estrone



3. Progesterone



4. Cholesterol



Drugs • Pharmaceutical ingredients

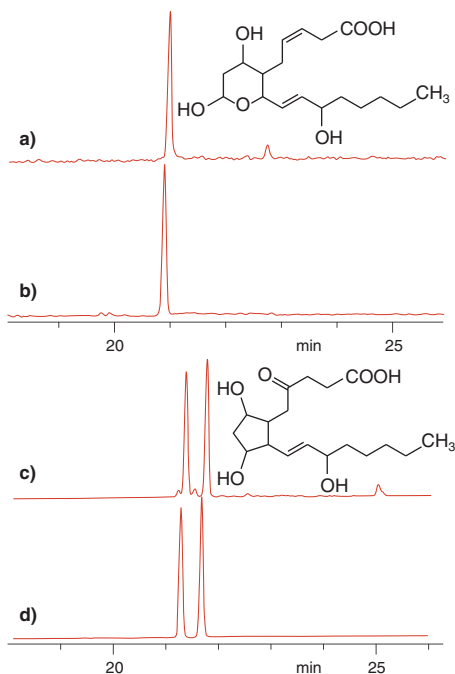
Quantification of 2,3-dinor-thromboxane B₂ and 2,3-dinor-6-oxo-prostaglandin F_{1α} in human urine

MN Appl. No. 211950

Column: OPTIMA® 17,
30 m x 0.25 mm ID,
0.25 μm film, REF 726022.30,
max. temperature 320/340 °C
Injection: 280 °C, splitless
Carrier gas: 55 kPa He
Temperature: 70 °C (2 min) $\xrightarrow{25\text{ °C/min}}$ 280 °C
 $\xrightarrow{4\text{ °C/min}}$ 320 °C
Detector: MSD (quadrupole)

Chromatograms:

- endogenous 2,3-dinorthromboxane B₂
- tetradeluterated 2,3-dinorthromboxane B₂
- endogenous 2,3-dinor-6-oxoprostaglandin F_{1α}
- tetradeluterated 2,3-dinor-6-oxoprostaglandin F_{1α}



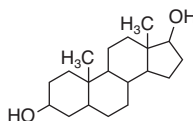
Analysis of steroids

MN Appl. No. 250300

Column: OPTIMA® 5,
30 m x 0.20 mm ID,
0.35 μm film, REF 726860.30,
max. temperature 340/360 °C
Carrier gas: He
Temperature: 200 °C (2 min) $\xrightarrow{3\text{ °C/min}}$ 320 °C
Detector: MSD

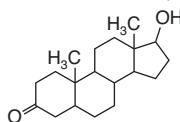
Peaks:

- Androstanediol

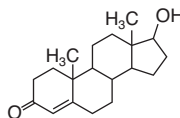


- Androstanolone (1st isomer)

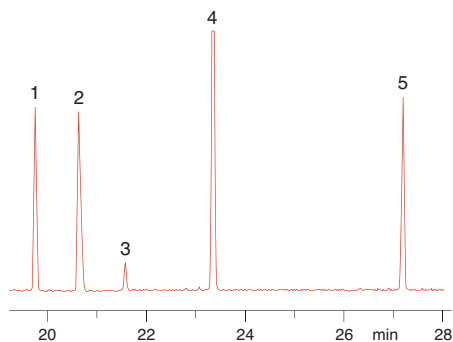
- Androstanolone (2nd isomer)



- Testosterone



- Progesterone



D. Tsikas, F.M. Böhme, I. Fuchs, J. Fröhlich, J. Chromatography A, **885** (2000) 351 – 359



Separation of endocrinic compounds

MN Appl. No. 212620

Column: OPTIMA® δ-6, 30 m x 0.32 mm ID, 0.35 µm film, REF 726481.30,
max. temperature 340/360 °C

Injection: 5.0 µl standards 10 mg/l, nonylphenol (NP) 20 mg/l in methanol, splitless

Carrier gas: 5.0 ml/min H₂, constant flow

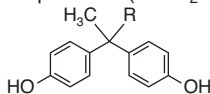
Temperature: 80 °C $\xrightarrow{12\text{ °C/min}}$ 305 °C $\xrightarrow{60\text{ °C/min}}$ 340 °C

Detector: FID 300 °C

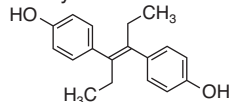
Peaks:

1. 4-*t*-Octylphenol
2. tech. 4-Nonylphenol (NP)
3. Bisphenol A (R = CH₃)

4. Bisphenol B (R = C₂H₅)



5. Diethylstilbestrol

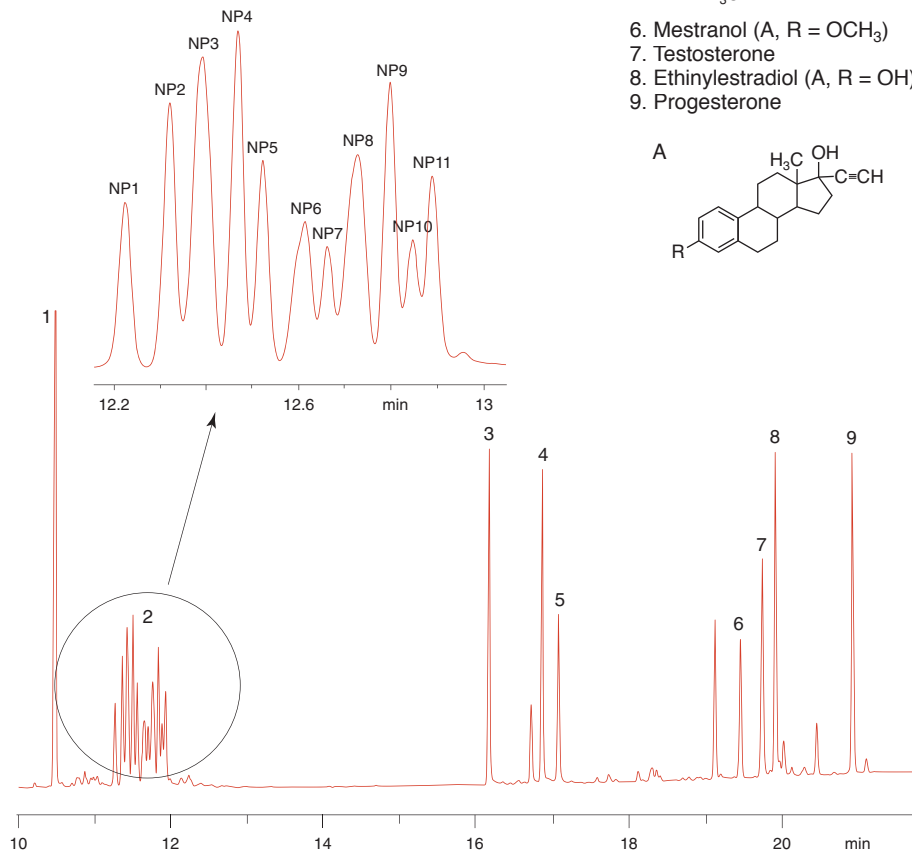
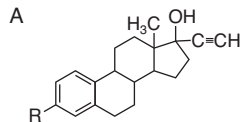


6. Mestranol (A, R = OCH₃)

7. Testosterone

8. Ethinylestradiol (A, R = OH)

9. Progesterone



Courtesy of Mr. Stollenwerk, Inst. f. Chemische Verfahrenstechnik, RWTH Aachen, Germany

Drugs · Pharmaceutical ingredients

Separation of dicarboxylic acids for profiling of urinary acidic metabolites

MN Appl. No. 212590

Column: OPTIMA® δ-6, 30 m x 0.25 mm ID, 0.25 µm film, REF 726470.30,
max. temperature 340/360 °C

Injection: 2 µl, 220 °C, split 1:10

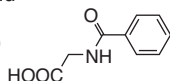
Carrier gas: 80 kPa H₂

Temperature: 80 °C $\xrightarrow{10\text{ °C/min}}$ 310 °C

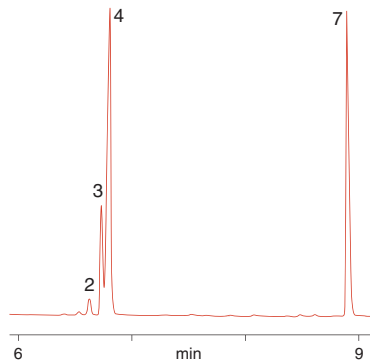
Detector: FID 280 °C

Peaks:

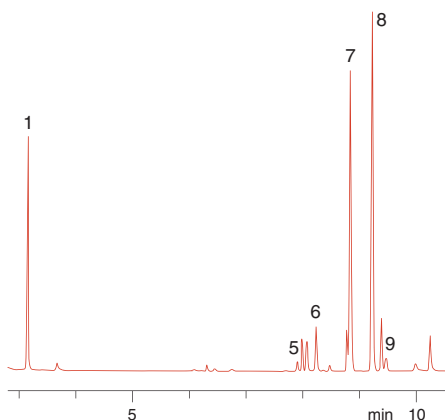
1. 3-Hydroxyisovaleric acid
(CH₃)₂C(OH)—CH₂—COOH
2. Lactic acid [H₃C)—CH(OH)—COOH]
3. Succinic acid [HOOC—(CH₂)₂—COOH]
4. Fumaric acid [HOOC—CH=CH—COOH]
5. Glutaric acid [HOOC—(CH₂)₃—COOH]
6. 3-Methylglutaric acid
7. 2-Phenylbutyric acid (int. std.)
8. 3-Hydroxy-3-methylglutaric acid
9. Adipic acid [HOOC—(CH₂)₄—COOH]
10. 3-Hydroxyglutaric acid
11. 2-Oxoglutaric acid
12. Hippuric acid
(benzoylglycine)



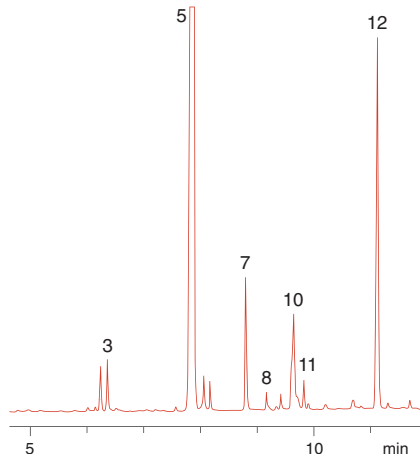
Sample: urine of a patient with fumarase enzyme impaired function, derivatised with ethyl chloroformate



Sample: urine of a patient with 3-hydroxy-3-methylglutaric aciduria, derivatised with ethyl chloroformate



Sample: urine of a patient with glutaric aciduria of the I. type, derivatised with ethyl chloroformate



P. Hušek et al., Chromatographia **58** (2003) 623 – 630



Petrochemical products



© christian42 / www.fotolia.com

Petrochemical products

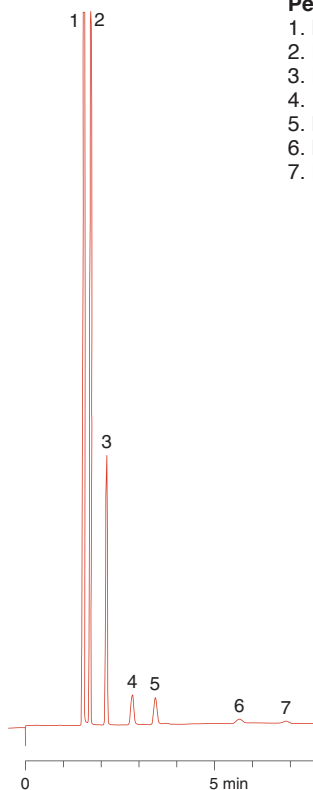
Analysis of hydrocarbons C₁ – C₅

MN Appl. No. 200030

Column: OPTIMA® 5,
15 m x 0.32 mm ID,
5.0 µm film, REF 726934.15,
max. temperature 300/320 °C
Injection: 100 µl, split 60 ml/min
Carrier gas: 0.08 bar N₂
Temperature: 36 °C
Detector: FID

Peaks:

1. Methane
2. Ethane
3. Propane
4. *i*-Butane
5. Butane
6. Methylbutane
7. Pentane



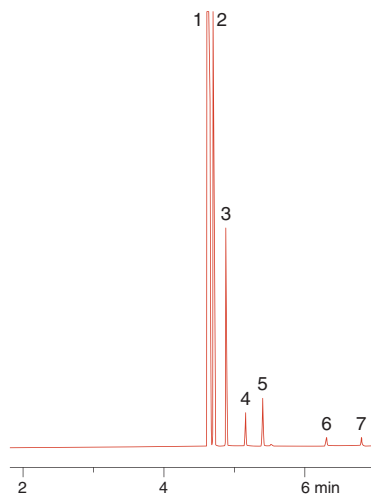
Analysis of hydrocarbons C₁ – C₅

MN Appl. No. 200041

Column: PERMABOND® P-100,
100 m x 0.25 mm ID,
0.5 µm film, REF 723890.100,
max. temperature 300/320 °C
Injection: 100 µl, split 200 ml/min
Carrier gas: 200 kPa H₂ (2.3 ml/min)
Temperature: 31 °C
Detector: FID 250 °C

Peaks:

1. Methane
2. Ethane
3. Propane
4. *i*-Butane
5. Butane
6. Methylbutane
7. Pentane





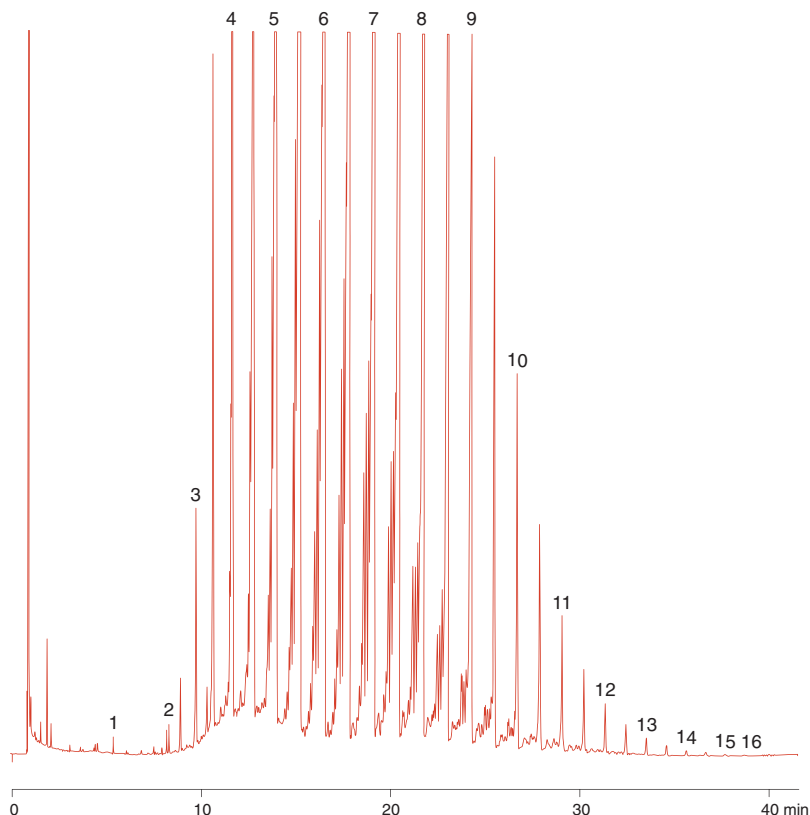
Analysis of tea warmer candle

MN Appl. No. 200050

Column: OPTIMA® 1-TG, 25 m x 0.32 mm ID, REF 726132.25, max. temperature: 370 °C
Injection: 1.0 µl, split 100 ml/min
Carrier gas: 50 kPa H₂ (3.5 ml/min)
Temperature: 80 °C (1 min) $\xrightarrow{20\text{ °C/min}}$ 200 °C $\xrightarrow{5\text{ °C/min}}$ 370 °C (5 min)
Detector: FID 380 °C, baseline corrected

Peaks:

1. C14
2. C18
3. C20
4. C22
5. C24
6. C26
7. C28
8. C30
9. C32
10. C34
11. C36
12. C38
13. C40
14. C42
15. C44
16. C45



Petrochemical products

Analysis of hydrocarbons acc. to ISO DIS 9377-4 (H-53)

MN Appl. No. 210600

Sample: standard (1 mg/l) extracted with 50:900 ml *n*-hexane:H₂O, pH 2 with H₂SO₄, (extraction solvent *n*-hexane with *n*-C₁₀ and *n*-C₄₀ standard), then evaporated with N₂ to 1 ml

Column: OPTIMA® 1, 25 m x 0.32 mm ID, 0.25 µm film, REF 726302.25, max. temperature 340/360 °C

Injection: 1.0 µl on column

Carrier gas: 3 ml/min H₂ $\xrightarrow{15\text{ °C/min}}$ 350 °C (19 min)

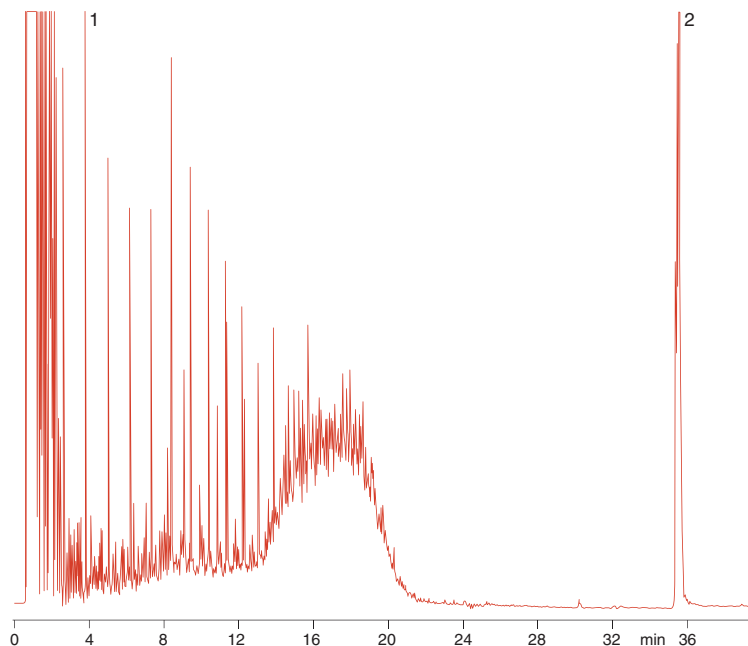
Temperature: 60 °C (1 min) $\xrightarrow{15\text{ °C/min}}$ 350 °C (19 min)

Detector: FID (30:300:40 ml/min H₂:O₂:He)

Peaks:

1. *n*-C₁₀

2. *n*-C₄₀



For a sample chromatogram of a mixture of diesel and heavy oil see application 210600 at www.mn-net.com

Courtesy of Mr. W. Elling, INFU, Dortmund, Germany

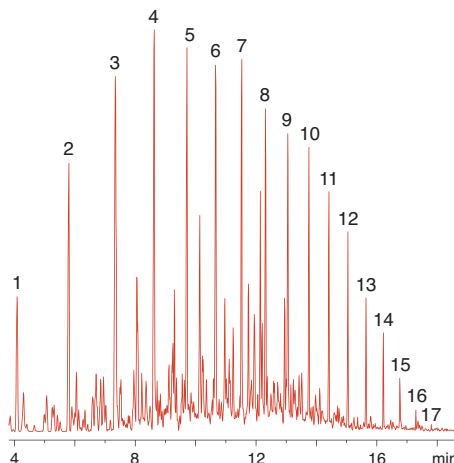


Analysis of commercial gas oil MN Appl. No. 200060

Column: OPTIMA® 17,
25 m x 0.20 mm ID,
0.2 µm film, REF 726065.25,
max. temperature 320/340 °C
Injection: 2.0 µl gas oil, diluted 1:100 in
MeOH, split 1:120
Carrier gas: 25 cm/s He
Temperature: 80 °C (4 min) $\xrightarrow{15\text{ °C/min}}$ 320 °C
Detector: MSD

Peaks:

- | | |
|--------|---------|
| 1. C10 | 10. C19 |
| 2. C11 | 11. C20 |
| 3. C12 | 12. C21 |
| 4. C13 | 13. C22 |
| 5. C14 | 14. C23 |
| 6. C15 | 15. C24 |
| 7. C16 | 16. C25 |
| 8. C17 | 17. C26 |
| 9. C18 | |

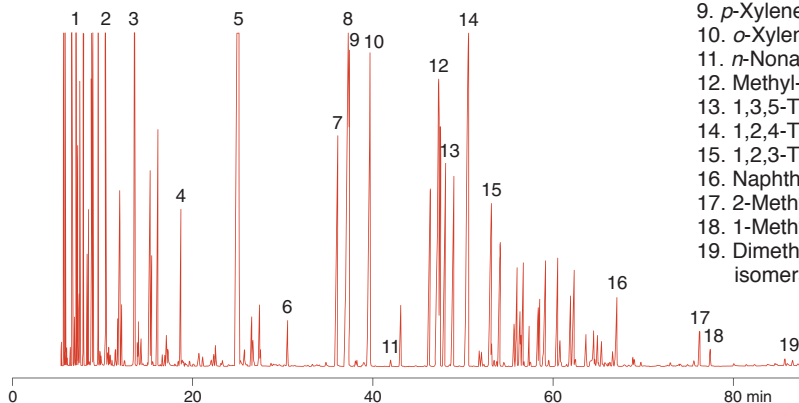


Analysis of unleaded gasoline MN Appl. No. 200071

Column: PERMABOND® P-100, 100 m x 0.25 mm ID,
REF 726890.100, max. temperature 300/320 °C
Injection: 0.5 µl, split 70 ml/min
Carrier gas: 200 kPa H₂ (2.3 ml/min)
Temperature: 35 °C (15 min) $\xrightarrow{2\text{ °C/min}}$ 200 °C (5 min)
Detector: FID 250 °C

Peaks:

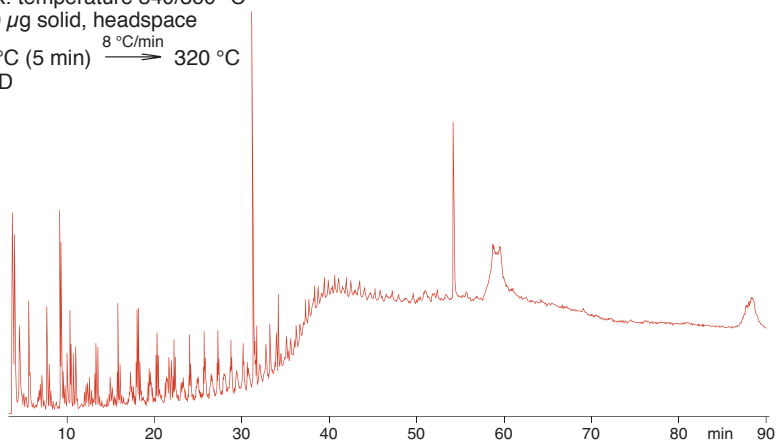
1. *n*-Pentane
2. *n*-Hexane
3. Benzene
4. *n*-Heptane
5. Toluene
6. *n*-Octane
7. Ethylbenzene
8. *m*-Xylene
9. *p*-Xylene
10. *o*-Xylene
11. *n*-Nonane
12. Methyl-3-ethylbenzene
13. 1,3,5-Trimethylbenzene
14. 1,2,4-Trimethylbenzene
15. 1,2,3-Trimethylbenzene
16. Naphthalene
17. 2-Methylnaphthalene
18. 1-Methylnaphthalene
19. Dimethylnaphthalene isomers



Petrochemical products

Pyrolysis GC of polyethylene and polypropylene *MN Appl. No. 210030*

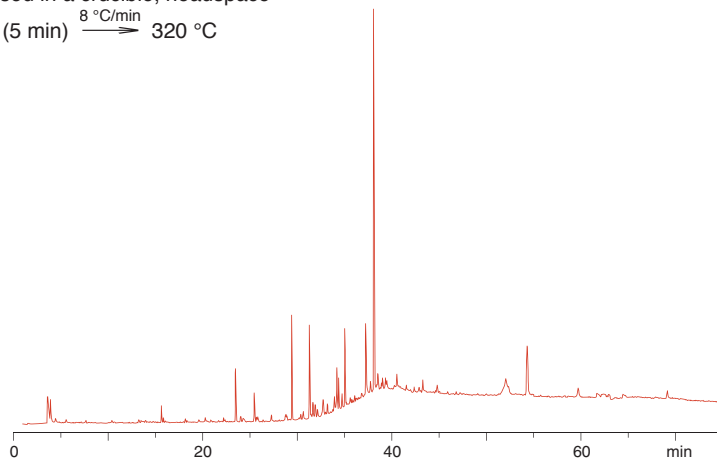
Column: OPTIMA® 5 MS, 60 m x 0.32 mm ID, 1 µm film, REF 726212.60,
max. temperature 340/360 °C
Injection: 270 µg solid, headspace
Temperature: 60 °C (5 min) $\xrightarrow{8\text{ °C/min}}$ 320 °C
Detector: MSD



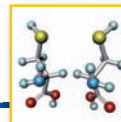
Courtesy of Mrs. Engel, VW, Wolfsburg, Germany

Pyrolysis GC of chloroprene rubber *MN Appl. No. 210040*

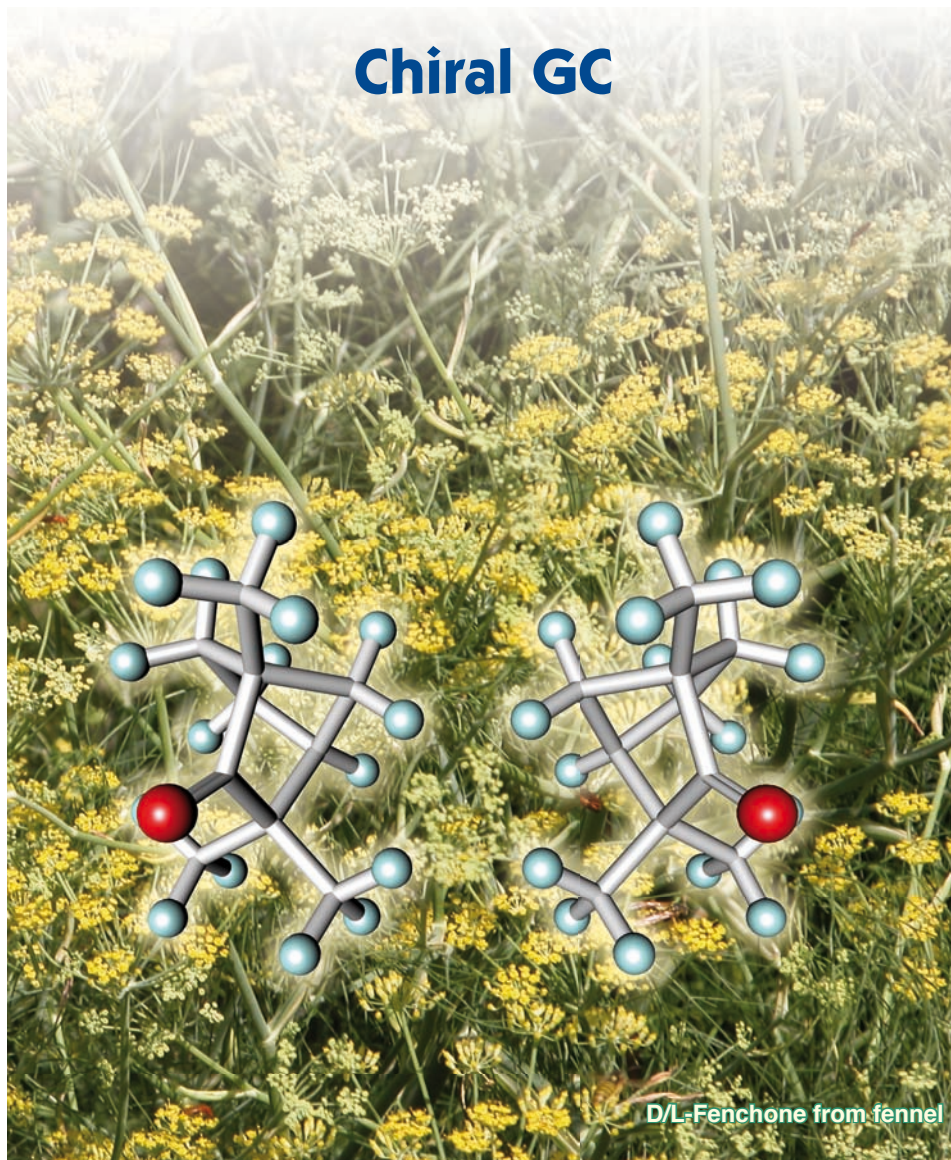
Column: OPTIMA® 5 MS, 60 m x 0.32 mm ID, 1 µm film, REF 726212.60,
max. temperature 340/360 °C
Injection: 1 g solid is dissolved in 10 ml of hexane or methanol, 100 µl of this solution are
pyrolysed in a crucible, headspace
Temperature: 60 °C (5 min) $\xrightarrow{8\text{ °C/min}}$ 320 °C
Detector: MSD



Courtesy of Mrs. Engel, VW, Wolfsburg, Germany



Chiral GC



Please note: enantiomer separations have been arranged by increasing molecular size. Several chromatograms show separations of homologous series; they are listed according to the smallest molecule of the mixture. Related molecules or families of compounds may be found on earlier or later pages depending on size. Specific compounds can be found in the chromatogram index from page 276.

Chiral separations

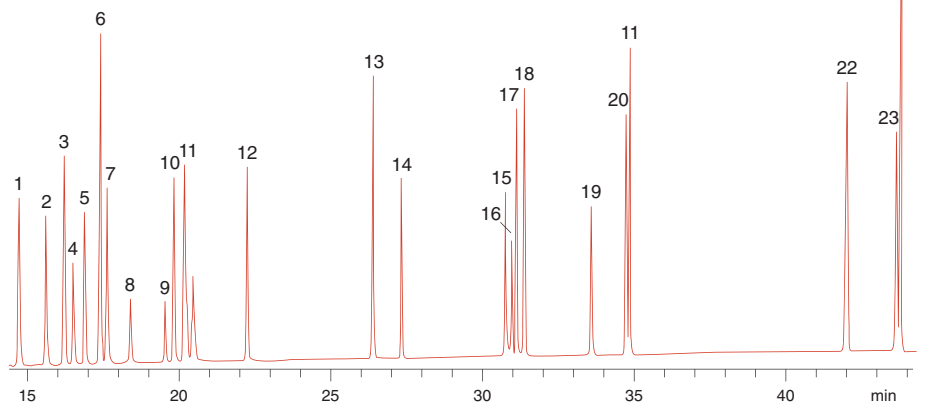
Analysis of amino acid enantiomers

MN Appl. No. 212761

Column: FS-LIPODEX® E, 25 m x 0.25 mm ID, REF 723368.25, max. temp. 200/220 °C
 Injection: 250 °C
 Carrier gas: 1.2 ml/min He, purge flow 65.9 ml/min
 Temperature: 60 °C (5 min) $\xrightarrow{3\text{ °C/min}}$ 185 °C (15 min)
 Detector: MSD 280 °C

Peaks: (N-pentafluoropropionyl 2-propyl esters)

- | | | | |
|-------------------------|------------------|---------------------|---------------------|
| 1. Aminoisobutyric acid | 7. L-Valine | 13. L-Proline | 19. D-Phenylalanine |
| 2. L-Isovaline | 8. D-Isoleucine | 14. D-Proline | 20. D-Glutamic acid |
| 3. D-Isovaline | 9. D-Leucine | 15. D-Aspartic acid | 21. L-Glutamic acid |
| 4. D-Valine | 10. L-Isoleucine | 16. D-Methionine | 22. D-Ornithine |
| 5. D-Alanine | 11. L-Leucine | 17. L-Methionine | 23. D-Lysine |
| 6. L-Alanine | 12. Glycine | 18. L-Aspartic acid | 24. L-Lysine |

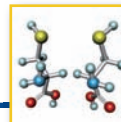


O. Vandenabeele-Trambouze et al., *Chromatographia* **53** (2001) Suppl., S-332 – S-339

Summary of important amino acids

Formula	Common name	Systematic name	Structure
C ₂ H ₅ NO ₂	glycine	aminoethanoic acid	NH ₂ -CH ₂ -COOH
C ₃ H ₇ NO ₂	alanine	2-aminopropanoic acid	CH ₃ -CH(NH ₂)-COOH
C ₄ H ₇ NO ₄	aspartic acid	2-aminobutanedioic acid	HOOC-CH ₂ -CH(NH ₂)-COOH
C ₄ H ₉ NO ₂	aminoisobutyric acid	2-amino-2-methylpropanoic acid	(CH ₃) ₂ C(NH ₂)-COOH
C ₅ H ₉ NO ₂	proline	pyrrolidine-2-carboxylic acid	
C ₅ H ₉ NO ₄	glutamic acid	2-aminopentanedioic acid	HOOC-(CH ₂) ₂ -CH(NH ₂)-COOH
C ₅ H ₁₁ NO ₂	valine	2-amino-3-methylbutanoic acid	(CH ₃) ₂ CH-CH(NH ₂)-COOH
C ₅ H ₁₁ NO ₂	isovaline	2-amino-2-methylbutanoic acid	H ₃ C ₂ -C(CH ₃)(NH ₂)-COOH
C ₅ H ₁₁ NO ₂ S	methionine	2-amino-4-(methylsulfanyl)-butanoic acid	CH ₃ -S-(CH ₂) ₂ -CH(NH ₂)-COOH
C ₅ H ₁₂ N ₂ O ₂	ornithine	2,5-diaminopentanoic acid	H ₂ N-(CH ₂) ₃ -CH(NH ₂)-COOH
C ₆ H ₁₃ NO ₂	leucine	2-amino-4-methylpentanoic acid	(CH ₃) ₂ CH-CH ₂ -CH(NH ₂)-COOH
C ₆ H ₁₃ NO ₂	isoleucine	2-amino-3-methylpentanoic acid	C ₂ H ₅ -CH(CH ₃)-CH(NH ₂)-COOH
C ₆ H ₁₄ N ₂ O ₂	lysine	2,6-diaminohexanoic acid	H ₂ N-(CH ₂) ₄ -CH(NH ₂)-COOH
C ₉ H ₁₁ NO ₂	phenylalanine	2-amino-3-phenyl-propanoic acid	C ₆ H ₅ -CH ₂ -CH(NH ₂)-COOH

Compounds C₂ and greater



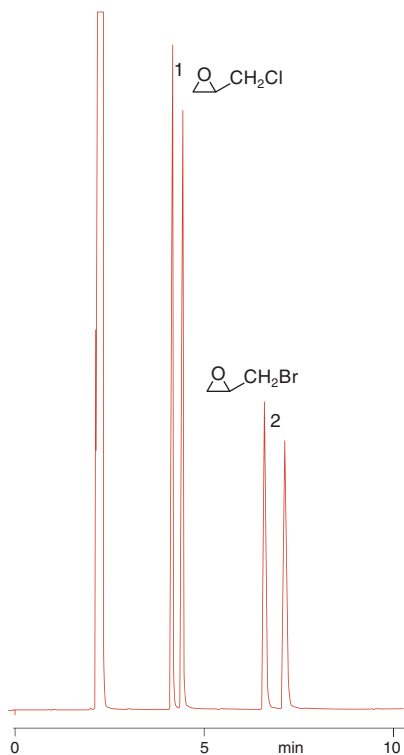
Enantiomer separation of epichloro- and epibromohydrin

MN Appl. No. 202621

Column: FS-LIPODEX® A,
50 m x 0.25 mm ID,
REF 723360.50,
max. temperature 200/220 °C
Injection: 0.5 µl (1 % in CH₂Cl₂)
split 245 ml/min
Carrier gas: 110 kPa H₂ (2.3 ml/min)
Temperature: 65 °C
Detector: FID 240 °C

Peaks:

1. Epichlorohydrin (C₃H₅ClO)
2. Epibromohydrin (C₃H₅BrO)



Enantiomer separation of epichlorohydrin

MN Appl. No. 212880

Column: FS-LIPODEX® A,
50 m x 0.25 mm ID,
REF 723360.50,
max. temperature 200/220 °C
Sample: compounds dissolved in
MeOH (1:100)
Injection: 0.2 µl sample + 0.3 µl air,
split 80 ml/min, 230 °C
Carrier gas: 85 kPa N₂
Temperature: 50 °C
Detector: FID 250 °C

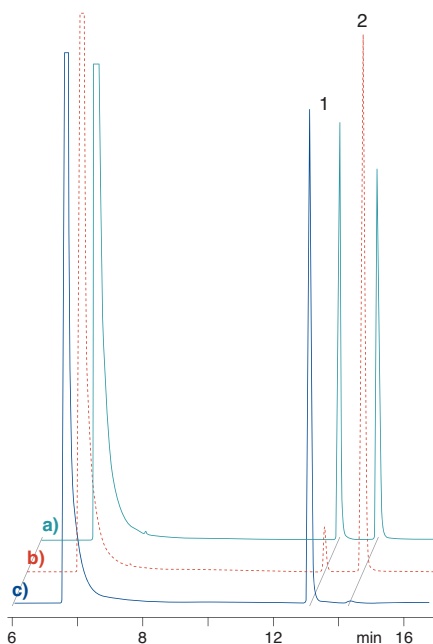
a) racemate (S/R 50:50)

b) S/R 2:98

c) S/R 99:1

Peaks:

1. S-Epichlorohydrin (C₃H₅ClO)
2. R-Epichlorohydrin



Courtesy of B. Mischke, Schulung und Chromatographie, Berlin (Germany)

Chiral separations

Enantiomer separation of propane-1,2-diol and butane-1,3-diol (TFA)

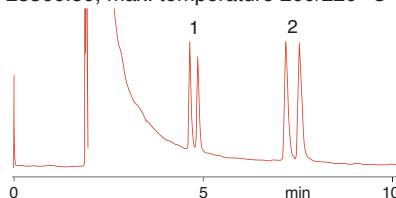
MN Appl. No. 202881

Column: LIPODEX® D, 25 m x 0.25 mm ID, REF 723366.50, max. temperature 200/220 °C
 Carrier gas: 0.7 bar H₂
 Temperature: 105 °C
 Detector: FID

Peaks:

1. Propane-1,2-diol (C₃H₈O₂)
2. Butane-1,3-diol (C₄H₁₀O₂)

W.A. König et al., HRC 11 (1988) 506 – 509



Enantiomer separation of chiral amines and amino alcohols (N,O-TFA)

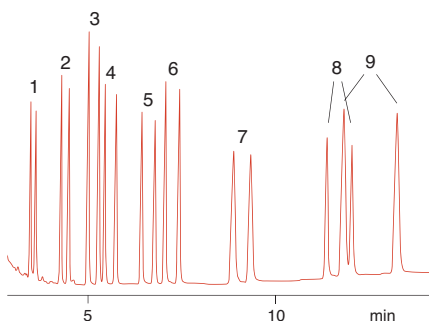
MN Appl. No. 202421

Column: LIPODEX® D, 50 m x 0.25 mm ID, REF 723366.50, max. temperature 200/220 °C

Carrier gas: 1 bar H₂
 Temperature: 140 °C $\xrightarrow{2\text{ °C/min}}$ 170 °C
 Detector: FID

Peaks: R enantiomers are eluted first.

1. 2-Aminopentane (C₅H₁₃N)
2. 2-Aminohexane (C₆H₁₅N)
3. 2-Amino-5-methylhexane (C₇H₁₇N)
4. 2-Aminoheptane (C₇H₁₇N)
5. 2-Amino-6-methylheptane (C₈H₁₉N)
6. 2-Aminooctane (C₈H₁₉N)
7. Valinol (C₅H₁₃NO)
8. Phenylethylamine (C₈H₁₁N)
9. Alaninol (C₃H₉NO)



W.A. König et al., HRC 11 (1988) 506 – 509

Enantiomer separation of amino alcohols (N,O-TFA)

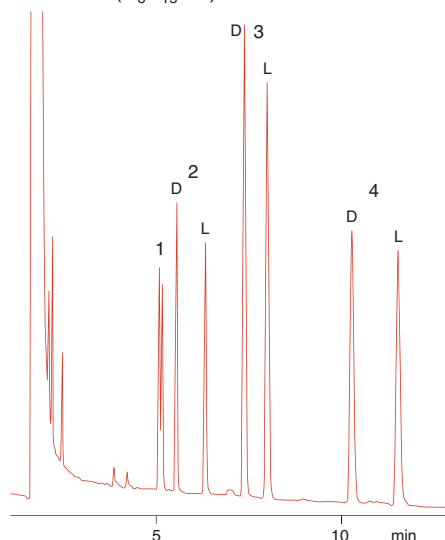
MN Appl. No. 202431

Column: LIPODEX® B, 50 m x 0.25 mm ID, REF 723362.50, max. temperature 200/ 220 °C

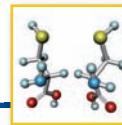
Carrier gas: H₂
 Temperature: 150 °C
 Detector: FID

Peaks:

1. 1-Aminopropan-2-ol (C₃H₉NO)
2. Alaninol = 2-aminopropan-1-ol (C₃H₉NO)
3. 2-Aminobutanol (C₄H₁₁NO)
4. Leucinol (C₆H₁₅NO)



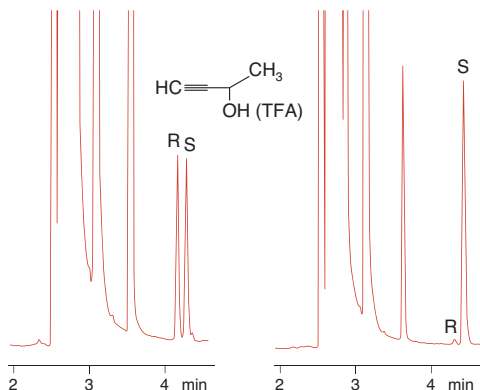
Courtesy of Prof. W.A. König, Hamburg, Germany



Enantiomer separation of 1-butyn-3-ol (TFA)

MN Appl. No. 202751

Column: LIPODEX® C,
50 m x 0.25 mm ID,
REF 723364.50,
max. temperature 200/220 °C
Carrier gas: 1 bar H₂
Temperature: 20 °C
Detector: FID
C₄H₆O



W.A. König et al., HRC 12 (1989) 35 – 39

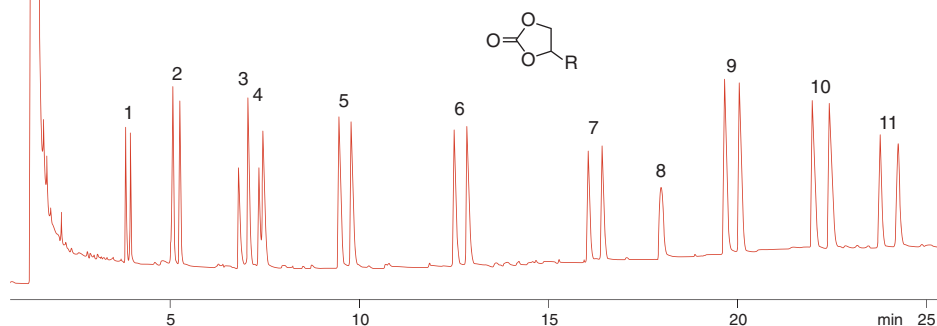
Enantiomer analysis of 1,2-diols (cyclic carbonates)

MN Appl. No. 202861

Column: LIPODEX® B, 50 m x 0.25 mm ID, REF 723362.50, max. temperature 200/220 °C
Carrier gas: 1 bar H₂
Temperature: 165 °C $\xrightarrow{2\text{ °C/min}}$ 210 °C
Detector: FID

Peaks:

1. 3,3-Dimethylbutane-1,2-diol (C₇H₁₂O₃, R = t-C₄H₉)
2. 3-Methylbutane-1,2-diol (C₆H₁₀O₃, R = CH(CH₃)₂)
3. Propane-1,2-diol (C₄H₈O₃, R = CH₃)
4. Butane-1,2-diol (C₅H₈O₃, R = C₂H₅)
5. Pentane-1,2-diol (C₆H₁₀O₃, R = C₃H₇)
6. Hexane-1,2-diol (C₇H₁₂O₃, R = C₄H₉)
7. Heptane-1,2-diol (C₈H₁₄O₃, R = C₅H₁₁)
8. Phenylglycol (C₉H₈O₃, R = C₆H₅)
9. Octane-1,2-diol (C₉H₁₆O₃, R = C₆H₁₃)
10. 1-Octene-7,8-diol (C₉H₁₄O₃, R = (CH₂)₄-CH=CH₂)
11. Nonane-1,2-diol (C₁₀H₁₈O₃, R = C₇H₁₅)



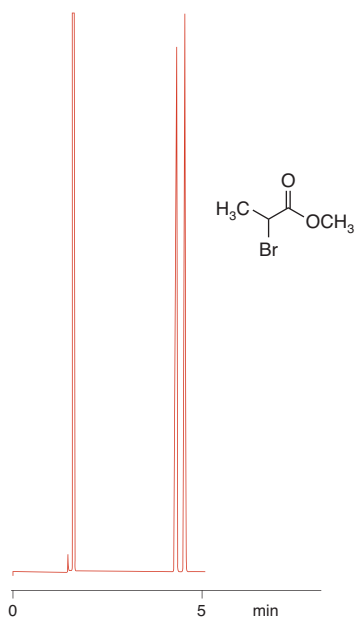
W.A. König et al., HRC 11 (1988) 621 – 625

Chiral separations

Enantiomer separation of 2-bromopropionic acid methyl ester

MN Appl. No. 210580

Column: FS-LIPODEX® A, 25 m x 0.25 mm ID, REF 723360.25, max. temperature 200/220 °C
 Injection: 1 µl (1% in CH₂Cl₂) split 200 ml/min
 Carrier gas: 11 psi He
 Temperature: 60 °C
 Detector: FID 250 °C
 C₄H₇BrO₂



Courtesy of Mr. B. Sievers, Riedel-de Haen AG, Seelze, Germany

Enantiomer separation of 2-chloropropionic acid methyl ester

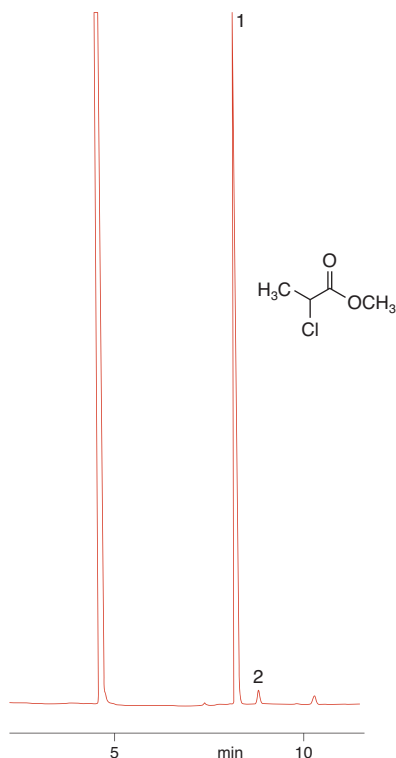
MN Appl. No. 202512

Column: FS-LIPODEX® C, 50 m x 0.25 mm ID, REF 723364.50, max. temperature 200/220 °C
 Split: 200 ml/min
 Carrier gas: 150 kPa He
 Temperature: 70 °C
 Detector: FID 200 °C
 C₄H₇ClO₂

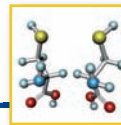
Peaks:

1. L-2-Chloropropionic acid methyl ester
 2. D-2-Chloropropionic acid methyl ester
- } ee = 96%

C₄H₇ClO₂



Compounds C₄ – C₈



Enantiomer separation of 1,2-epoxyalkanes

MN Appl. No. 202611

Column: FS-LIPODEX® E,
25 m x 0.25 mm ID,
REF 723368.25,
max. temperature
200/220 °C

Injection: 0.5 µl
split 135 ml/min

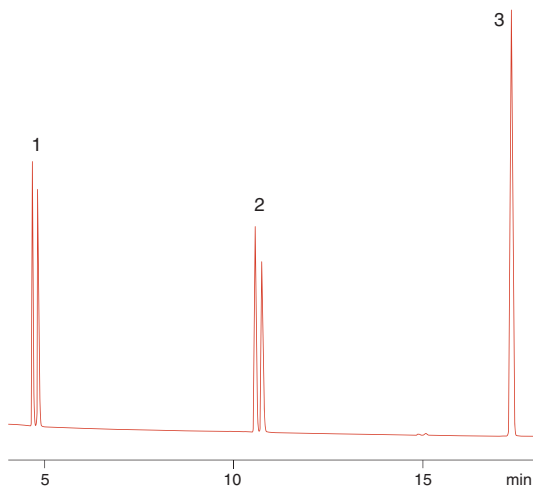
Carrier gas: 50 kPa H₂
(1.2 ml/min)

Temperature: 40 °C (3 min) $\xrightarrow{4\text{ °C/min}}$
120 °C

Detector: FID 250 °C

Peaks:

- 1,2-Epoxybutane (C₄H₈O)
- 1,2-Epoxyhexane (C₆H₁₂O)
- 1,2-Epoxyoctane (C₈H₁₆O)



Enantiomer separation of aldols (TFA)

MN Appl. No. 202961

Column: LIPODEX® B, 50 m x 0.25 mm ID, REF 723362.50, max. temperature 200/220 °C

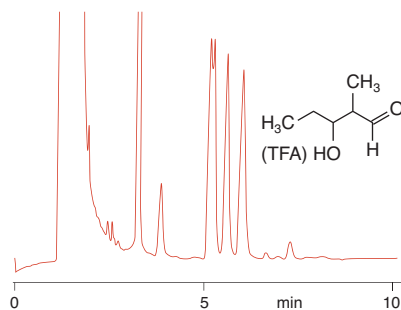
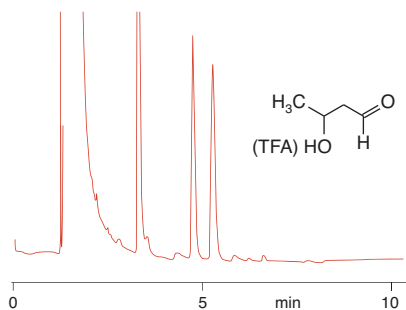
Carrier gas: 1 bar H₂

Temperature: 85 °C

Detector: FID

3-Hydroxybutyraldehyde (C₄H₈O₂)

2-Methyl-3-hydroxyvaleraldehyde (C₆H₁₂O₂)



W.A. König et al., HRC 11 (1988) 621 – 625

Chiral separations

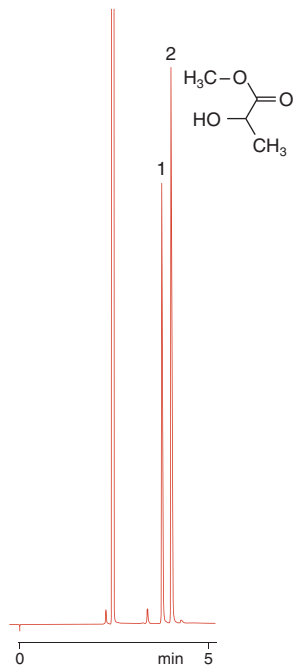
Enantiomer separation of methyl lactate

MN Appl. No. 202762

Column: FS-LIPODEX® A,
50 m x 0.25 mm ID,
REF 723360.50,
max. temperature 200/220 °C
Injection: 0.1 µl (1 % in CH₂Cl₂)
split 320 ml/min
Carrier gas: 120 kPa H₂ (2.2 ml/min)
Temperature: 80 °C
Detector: FID 250 °C

Peaks: (C₄H₈O₃)

1. S-(–)
2. R-(+)

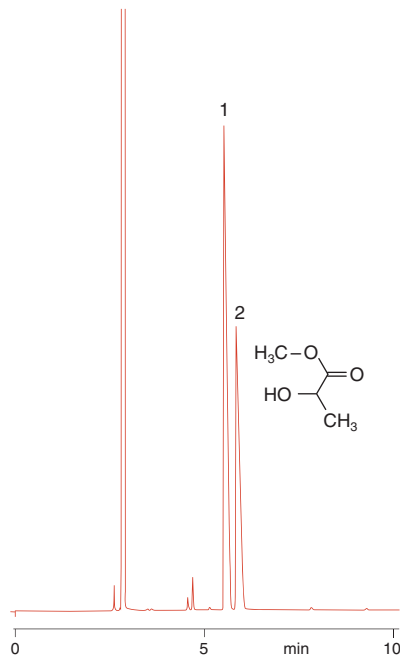


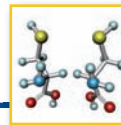
MN Appl. No. 202772

Column: FS-HYDRODEX β-PM,
50 m x 0.25 mm ID,
REF 723370.50,
max. temperature 230/250 °C
Injection: 0.1 µl (1 % in CH₂Cl₂)
split 150 ml/min
Carrier gas: 120 kPa H₂ (1.7 ml/min)
Temperature: 90 °C
Detector: FID 250 °C

Peaks: (C₄H₈O₃)

1. R-(+)
2. S-(–)





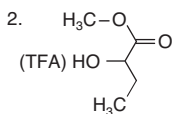
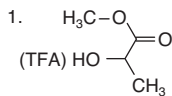
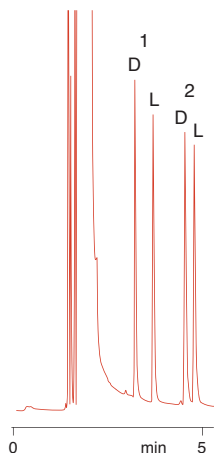
Enantiomer separation of lactic and 2-hydroxybutyric acid methyl esters (TFA)

MN Appl. No. 202752

Column: LIPODEX® A, 50 m x 0.25 mm ID, REF 723360.50, max. temperature 200/220 °C
Carrier gas: H₂
Temperature: 50 °C
Detector: FID

Peaks:

1. Lactic acid methyl ester (C₄H₈O₃)
2. 2-Hydroxybutyric acid methyl ester (C₅H₁₀O₃)



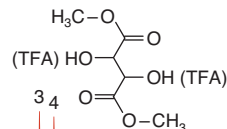
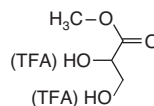
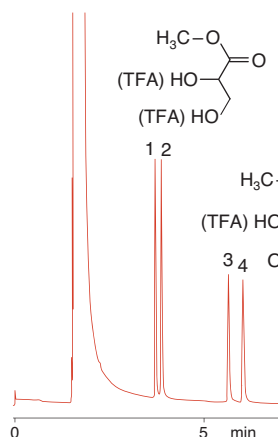
Enantiomer separation of glyceric and tartaric acid methyl esters (TFA)

MN Appl. No. 202732

Column: LIPODEX® A, 50 m x 0.25 mm ID, REF 723360.50, max. temperature 200/220 °C
Carrier gas: 1 bar H₂
Temperature: 90 °C
Detector: FID

Peaks:

1. D-Glyceric acid methyl ester (C₄H₈O₄)
2. L-Glyceric acid methyl ester
3. L-Tartaric acid dimethyl ester (C₆H₁₀O₆)
4. D-Tartaric acid dimethyl ester



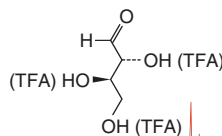
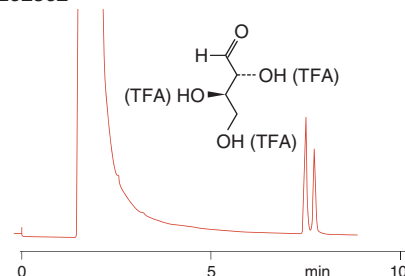
Courtesy of Prof. W.A. König, Hamburg, Germany

W.A. König, S. Lutz, G. Wenz, Angew. Chem. Int. Ed. Engl. **27** (1988) 979-980

Enantiomer separation of erythrose (TFA)

MN Appl. No. 202362

Column: LIPODEX® A, 50 m x 0.25 mm ID, REF 723360.50, max. temperature 200/220 °C
Carrier gas: H₂
Temperature: 80 °C
Detector: FID
C₄H₈O₄



Courtesy of Prof. W.A. König, Hamburg, Germany

Chiral separations

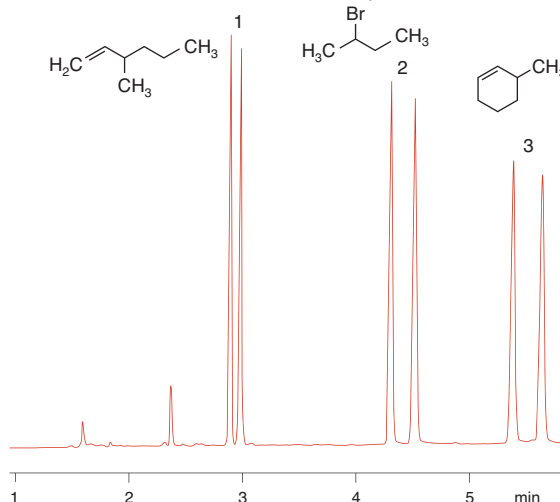
Enantiomer separation of 3-methyl-1-hexene, 2-bromobutane and 3-methylcyclohexene

MN Appl. No. 201880

Column: LIPODEX® C, 50 m x 0.25 mm ID, REF 723364.50, max. temperature 200/220 °C
 Injection: headspace
 Carrier gas: 1 bar H₂
 Temperature: 30 °C
 Detector: FID

Peaks:

1. 3-Methyl-1-hexene (C₇H₁₄)
2. 2-Bromobutane (C₄H₉Br)
3. 3-Methylcyclohexene (C₇H₁₂)



W.A. König et al., HRC 12
 (1989) 35 – 39

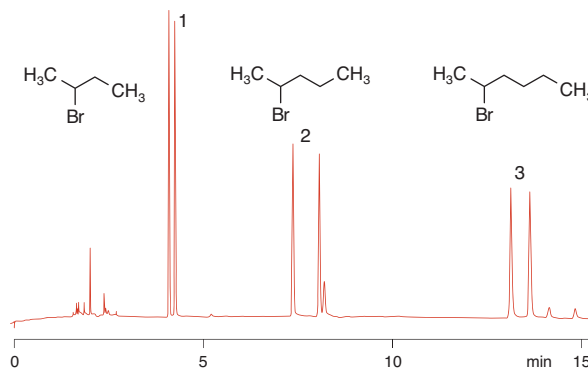
Enantiomer separation of chiral alkyl bromides

MN Appl. No. 202250

Column: LIPODEX® C, 50 m x 0.25 mm ID, REF 723364.50, max. temperature 200/220 °C
 Carrier gas: H₂
 Temperature: 50 °C
 Detector: FID

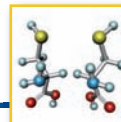
Peaks:

1. 2-Bromobutane (C₄H₉Br)
2. 2-Bromopentane (C₅H₁₁Br)
3. 2-Bromohexane (C₆H₁₃Br)



Courtesy of Prof. W.A. König,
 Hamburg, Germany

Compounds C₄ – C₇



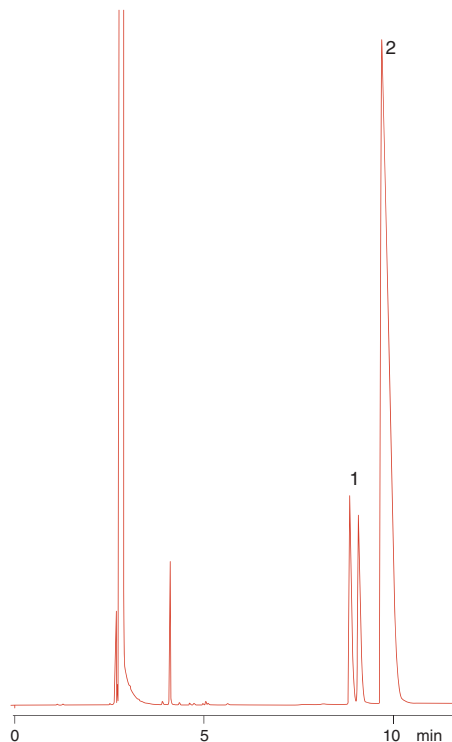
Enantiomer separation of butane-2,3-diol

MN Appl. No. 202901

Column: FS-HYDRODEX β-PM,
50 m x 0.25 mm ID
REF 723370.50,
max. temperature 230/250 °C
Injection: 1 µl (1 % in CH₂Cl₂)
split 150 ml/min
Carrier gas: 120 kPa H₂ (1.7 ml/min)
Temperature: 100 °C
Detector: FID 250 °C

Peaks: (C₄H₁₀O₂)

1. RR/SS
2. meso



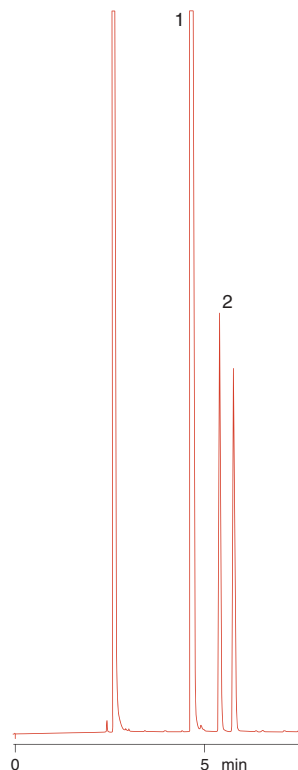
Enantiomer separation of butane-2,3-diol (TFA)

MN Appl. No. 202891

Column: FS-LIPODEX® E,
50 m x 0.25 mm ID,
REF 723368.50,
max. temperature 200/220 °C
Injection: 0.5 µl (1 % in CH₂Cl₂)
split 200 ml/min
Carrier gas: 120 kPa H₂ (2.1 ml/min)
Temperature: 80 °C
Detector: FID 250 °C

Peaks: (C₄H₁₀O₂)

1. meso
2. RR/SS



Chiral separations

Enantiomer separation of butane-1,3-diol MN Appl. No. 211320

Columns:	all 25 m x 0.25 mm ID:		
	a) FS-HYDRODEX β -6TBDM (REF 723381.25),		
	b) FS-LIPODEX [®] E (REF 723368.25),		
	c) FS-HYDRODEX β -PM (REF 723370.25)		
Injection:	1 μ l, split, 250 °C		
Carrier gas:	1.1 ml/min He		
Detector:	FID 250 °C		
Temperature:	a	b	c
	95 °C	65 °C	85 °C
Resolution:	<0.5	2.33	0.66
			

partial TFA derivative MN Appl. No. 211330

Column a)	
Temperature:	75 °C
Resolution:	9.98
	

TFA derivative MN Appl. No. 211340

Column a)	
Temperature:	75 °C
Resolution:	8.79
	

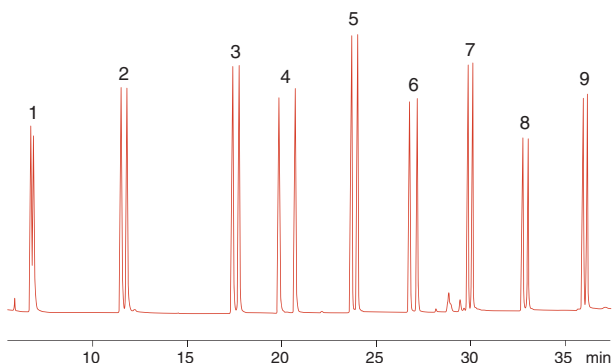
Courtesy of H. Casabianca, CNRS Service Central d'Analyse, Vernaison, France

Enantiomer analysis of 1,2-diols (TFA) MN Appl. No. 202851

Column: LIPODEX[®] A,
50 m x 0.25 mm ID,
REF 723360.50,
max. temperature 200/220 °C
Carrier gas: H₂
Temperature: 2 °C/min
48 °C (5 min) $\rightarrow$ 120 °C
Detector: FID

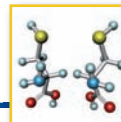
Peaks:

1. Butane-1,2-diol (C₄H₁₀O₂)
2. Pentane-1,2-diol (C₅H₁₂O₂)
3. Hexane-1,2-diol (C₆H₁₄O₂)
4. Cyclohexane-*trans*-1,2-diol (C₆H₁₂O₂)
5. Heptane-1,2-diol (C₇H₁₆O₂)
6. Cycloheptane-*trans*-1,2-diol (C₇H₁₄O₂)
7. Octane-1,2-diol (C₈H₁₈O₂)
8. Phenylglycol (C₈H₁₀O₂)
9. Nonane-1,2-diol (C₉H₂₀O₂)



Courtesy of Prof. W.A. König, Hamburg, Germany

Compounds C₄ and greater



Enantiomers separation of butane-1,2,4-triol (TFA)

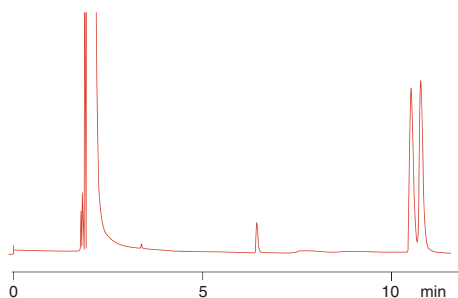
MN Appl. No. 202931

Column: LIPODEX® A,
50 m x 0.25 mm ID,
REF 723360.50,
max. temperature 200/220 °C

Carrier gas: H₂
Temperature: 70 °C

Detector: FID

C₄H₁₀O₃



Courtesy of Prof. W.A. König, Hamburg,
Germany

Enantiomer separation of 2-butylamine (TFA)

MN Appl. No. 202340

Column: FS-LIPODEX® D,
25 m x 0.25 mm ID,
REF 723366.25,
max. temperature 200/220 °C

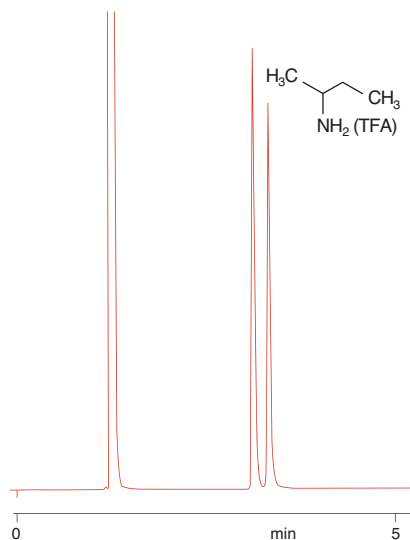
Injection: 0.1 µl (1% in CH₂Cl₂)
split 190 ml/min

Carrier gas: 50 kPa H₂ (1.5 ml/min)

Temperature: 105 °C

Detector: FID 250 °C

C₄H₁₁N

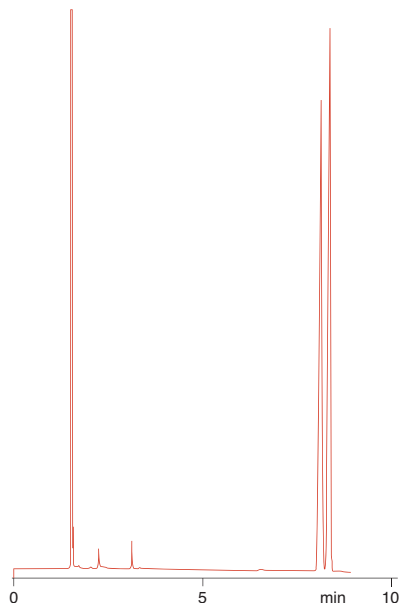


Chiral separations

Enantiomer separation of 2-aminobutanol-1 (N,O-TFA)

MN Appl. No. 210570

Column: FS-LIPODEX® A,
25 m x 0.25 mm ID,
REF 723360.25,
max. temperature 200/220 °C
Injection: 1 µl (1 % in CH₂Cl₂)
split 200 ml/min
Carrier gas: 11 psi He
Temperature: 90 °C
Detector: FID 250 °C
C₄H₁₁NO

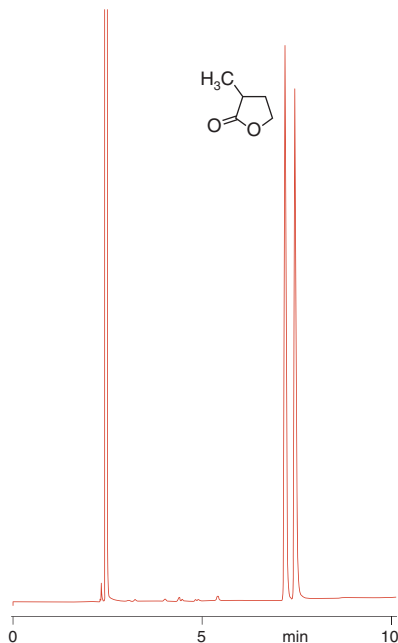


Courtesy of Mr. B. Sievers, Riedel-de Haen AG,
Seelze, Germany

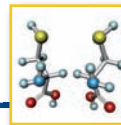
Enantiomer separation of α-methylbutyrolactone

MN Appl. No. 202842

Column: FS-LIPODEX® A,
50 m x 0.25 mm ID,
REF 723360.50,
max. temperature 200/220 °C
Injection: 0.1 µl (1 % in CH₂Cl₂)
split 245 ml/min
Carrier gas: 110 kPa H₂ (2.3 ml/min)
Temperature: 90 °C
Detector: FID 240 °C
C₅H₈O₂



Compounds C₄ and greater

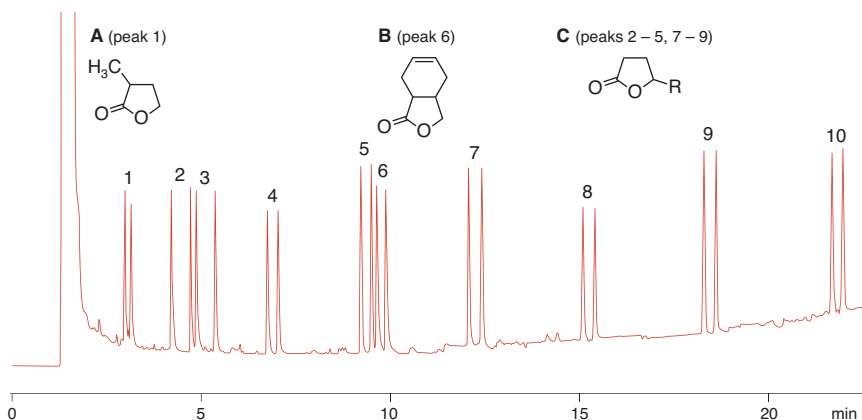


Enantiomer analysis of γ -lactones MN Appl. No. 202982

Column: LIPODEX® B, 50 m x 0.25 mm ID, REF 723362.50,
max. temperature 200/220 °C
Carrier gas: H₂ $\xrightarrow{3\text{ °C/min}}$ 200 °C
Temperature: 135 °C
Detector: FID

Peaks:

1. α -Methylbutyrolactone (structure A, C₅H₈O₂)
2. γ -Valerolactone (structure C, C₅H₈O₂, R = CH₃)
3. γ -Hexalactone (structure C, C₆H₁₀O₂, R = C₂H₅)
4. γ -Heptalactone (structure C, C₇H₁₂O₂, R = C₃H₇)
5. γ -Octalactone (structure C, C₈H₁₄O₂, R = C₄H₉)
6. Tetrahydroisobenzofuranone (structure B, C₈H₁₀O₂)
7. γ -Nonalactone (structure C, C₉H₁₆O₂, R = C₅H₁₁)
8. γ -Decalactone (structure C, C₁₀H₁₈O₂, R = C₆H₁₃)
9. γ -Undecalactone (structure C, C₁₁H₂₀O₂, R = C₇H₁₅)
10. γ -Dodecalactone (structure C, C₁₂H₂₂O₂, R = C₈H₁₇)

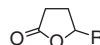


Courtesy of Prof. W.A. König, Hamburg, Germany

Chiral separations

Enantiomer analysis of γ -lactones

Columns: all 25 m x 0.25 mm ID:
 a) FS-HYDRODEX β -6TBDM (REF 723381.25),
 b) FS-LIPODEX[®] E (REF 723368.25),
 c) FS-HYDRODEX β -PM (REF 723370.25)
 Injection: 1 μ l, split, 250 °C
 Carrier gas: 1.1 ml/min He
 Detector: FID 250 °C



γ -Valerolactone MN Appl. No. 211350

	a	b	c
Temperature:	90 °C	105 °C	85 °C
Resolution:	4.29	2.88	1.15

γ -Hexalactone MN Appl. No. 211360

	a	b	c
Temperature:	100 °C	115 °C	95 °C
Resolution:	4.13	7.54	1.39

γ -Heptalactone MN Appl. No. 211390

	a	b	c
Temperature:	115 °C	115 °C	110 °C
Resolution:	5.74	10.27	1.63

γ -Octalactone MN Appl. No. 211440

	a	b	c
Temperature:	125 °C	115 °C	120 °C
Resolution:	5.99	11.83	1.39

γ -Nonalactone MN Appl. No. 211480

	a	b	c
Temperature:	135 °C	120 °C	130 °C
Resolution:	3.57	7.27	1.05

γ -Decalactone MN Appl. No. 211660

	a	b	c
Temperature:	140 °C	130 °C	140 °C
Resolution:	2.77	5.18	0.63

For separation of γ -decalactone also see appl. 212530 at www.mn-net.com

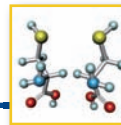
γ -Undecalactone MN Appl. No. 211720

	a	b	c
Temperature:	160 °C	140 °C	150 °C
Resolution:	<0.5	4.51	0.3

γ -Dodecalactone MN Appl. No. 211750

	a	b	c
Temperature:	175 °C	135 °C	150 °C
Resolution:	0.99	1.64	<0.5

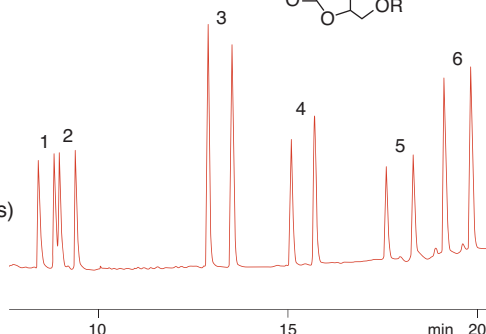
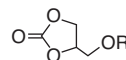
Courtesy of H. Casabianca, CNRS Service Central d'Analyse, Vernaison, France



Enantiomer separation of 1-O-alkylglycerols (cyclic carbonates)

MN Appl. No. 202951

Column: LIPODEX® B,
50 m x 0.25 mm ID,
REF 723362.50,
max. temperature 200/220 °C
Sample: O-alkylglycerols derivatised
with phosgene (very toxic)
Carrier gas: 1 bar H₂
Temperature: 160 °C $\xrightarrow{2\text{ °C/min}}$ 200 °C
Detector: FID



Peaks:

(R enantiomers are eluted before S enantiomers)

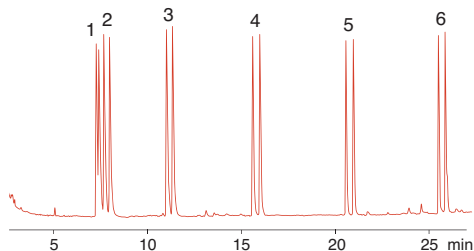
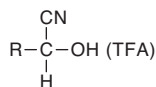
- 1-O-Methylglycerol
(C₅H₈O₄, R = -CH₃)
- 1-O-Ethylglycerol
(C₆H₁₀O₄, R = -C₂H₅)
- 1-O-Isobutylglycerol
(C₈H₁₄O₄, R = -CH₂-CH(CH₃)₂)
- 1-O-Butylglycerol
(C₈H₁₄O₄, R = -(CH₂)₃-CH₃)
- 1-O-Isopentylglycerol
(C₉H₁₆O₄, R = -(CH₂)₂-CH(CH₃)₂)
- 1-O-Pentylglycerol
(C₉H₁₆O₄, R = -(CH₂)₄-CH₃)

W.A. König et al., HRC 11 (1988) 621 – 625

Enantiomer separation of cyanohydrins (TFA)

MN Appl. No. 202280

Column: LIPODEX® C,
50 m x 0.25 mm ID,
REF 723364.50,
max. temperature 200/ 220 °C
Carrier gas: 1 bar H₂
Temperature: 50 °C $\xrightarrow{2\text{ °C/min}}$ 180 °C
Detector: FID



Peaks:

1. R = *n*-propyl (C₅H₉NO)
2. R = *i*-butyl (C₆H₁₁NO)
3. R = *n*-butyl (C₆H₁₁NO)
4. R = *n*-pentyl (C₇H₁₃NO)
5. R = *n*-hexyl (C₈H₁₅NO)
6. R = *n*-heptyl (C₉H₁₇NO)

R enantiomers are eluted before S enantiomers.

W.A. König et al., HRC 12 (1989) 35 – 39

Chiral separations

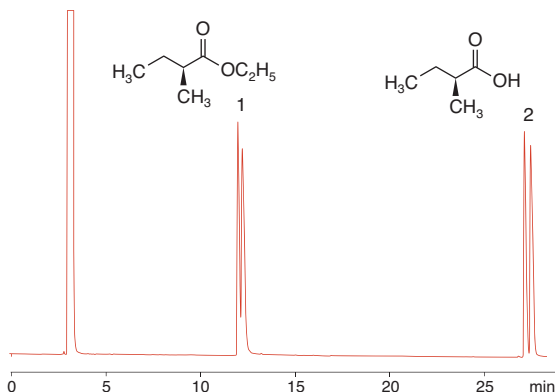
Enantiomer separation of 2-methylbutyric acid and ethyl ester

MN Appl. No. 202462

Column: FS-HYDRODEX β -PM,
50 m x 0.25 mm ID,
REF 723370.50,
max. temperature
230/250 °C
Injection: 1 μ l (0.5% in CH₂Cl₂)
split 1:100
Carrier gas: 110 kPa H₂
Temperature: 65 °C (12 min)
4 °C/min
→ 130 °C
Detector: FID 250 °C

Peaks:

1. 2-Methylbutyric acid ethyl ester
(C₇H₁₄O₂)
2. 2-Methylbutyric acid
(C₅H₁₀O₂)



Enantiomer separation of 2,3-epoxy-3-methylbutanol-1

MN Appl. No. 202681/202691

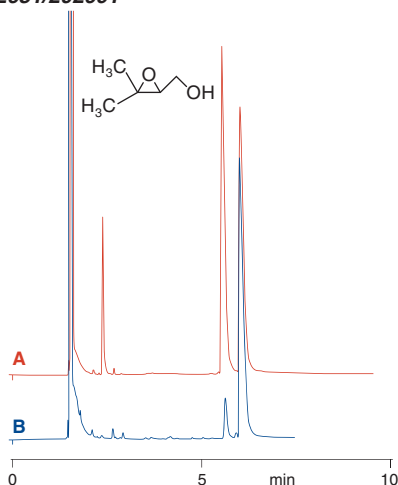
Column: FS-LIPODEX® E, 25 m x 0.25
mm ID, REF 723368.25,
max. temperature 200/220 °C
Injection: 0.5 μ l, split 135 ml/min
Carrier gas: 50 kPa H₂ (1.2 ml/min)
Temperature: 100 °C
Detector: FID 250 °C

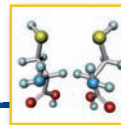
A) racemate (Appl. 202681)

B) ee = 83 % (Appl. 202691)

C₅H₁₀O₂

2,3-epoxy-3-methylbutanol-1 can also be separated on HYDRODEX β -PM (applications 202661/202671) or LIPODEX® A (applications 202701/202711). See our application database at www.mn-net.com.





Enantiomer separation of 1,4- and 1,5-anhydroarabinitol (TFA)

MN Appl. No. 203021

Column: LIPODEX® A,
50 m x 0.25 mm ID,
REF 723360.50,
max. temperature 200/220 °C

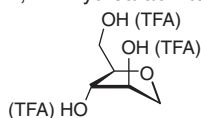
Carrier gas: H₂

Temperature: 90 °C

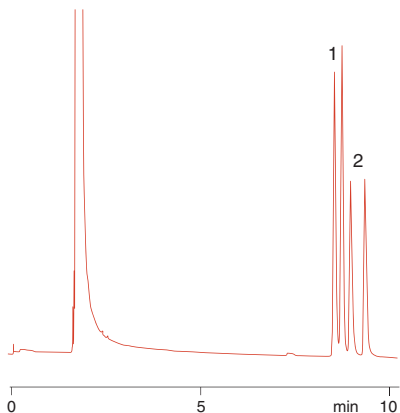
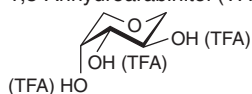
Detector: FID

Peaks:

- 1,4-Anhydroarabinitol (TFA), C₅H₁₀O₄



- 1,5-Anhydroarabinitol (TFA), C₅H₁₀O₄



Courtesy of Prof. W.A. König, Hamburg,
Germany

Enantiomer separation of 2-bromopentane

MN Appl. No. 202240

Column: FS-LIPODEX® C,
50 m x 0.25 mm ID,
REF 723364.50,
max. temperature 200/220 °C

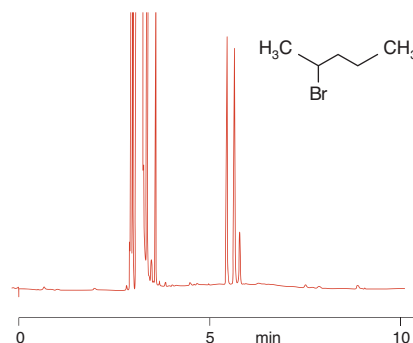
Injection: 1 µl, split 1:100

Carrier gas: 100 kPa H₂

Temperature: 60 °C

Detector: FID 250 °C

C₅H₁₁Br



Chiral separations

Enantiomer separation of chiral alkyl chlorides

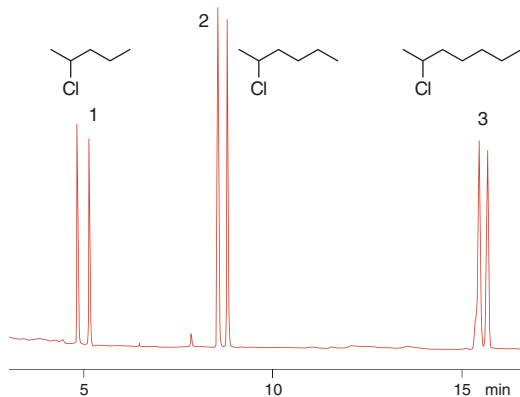
MN Appl. No. 202260

Column: LIPODEX® C,
50 m x 0.25 mm ID,
REF 723364.50,
max. temperature
200/ 220 °C

Carrier gas: 1 bar H₂
Temperature: 60 °C
Detector: FID

Peaks:

1. 2-Chloropentane (C₅H₁₁Cl)
2. 2-Chlorohexane (C₆H₁₃Cl)
3. 2-Chloroheptane (C₇H₁₅Cl)



W.A. König et al., HRC **12** (1989) 35 – 39

Enantiomer separation of 2-alkanols

MN Appl. No. 202451

Column: FS-LIPODEX® E,
50 m x 0.25 mm ID,
REF 723368.50,
max. temperature 200/ 220 °C

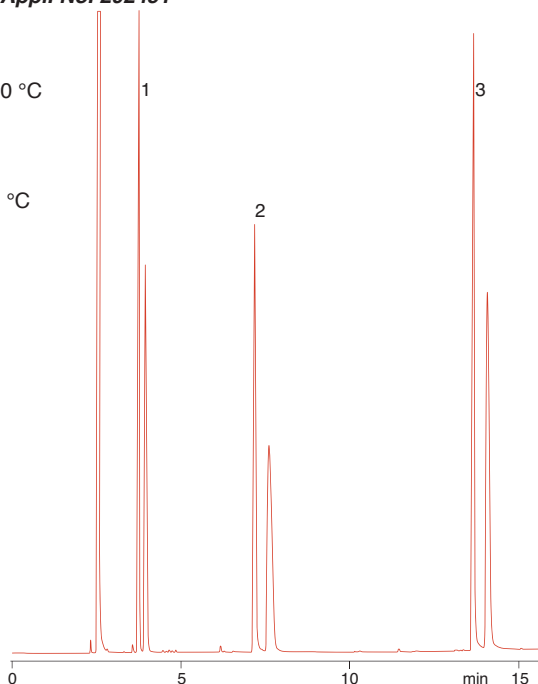
Injection: 0.1 µl (1 % in CH₂Cl₂)
split 200 ml/min

Carrier gas: 120 kPa H₂ (2.1 ml/min)

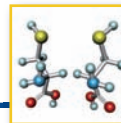
Temperature: 80 °C (4 min) $\xrightarrow{4\text{ °C/min}}$ 130 °C
Detector: FID 250 °C

Peaks:

1. R/S-Pentanol-2 (C₅H₁₂O)
2. R/S-Heptanol-2 (C₇H₁₆O)
3. R/S-Nonanol-2 (C₉H₂₀O)



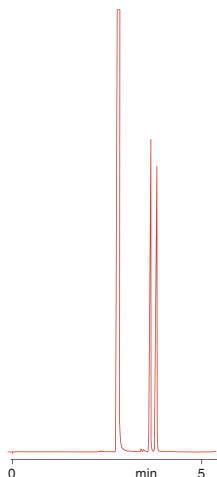
Compounds C₅ and greater



Enantiomer separation of pentanol-2 (TFA)

MN Appl. No. 202471

Column: FS-LIPODEX® E,
50 m x 0.25 mm ID,
REF 723368.50,
max. temperature 200/220 °C
Injection: 0.1 µl (1 % in CH₂Cl₂)
split 200 ml/min
Carrier gas: 120 kPa H₂ (2.1 ml/min)
Temperature: 60 °C
Detector: FID 250 °C
C₅H₁₂O



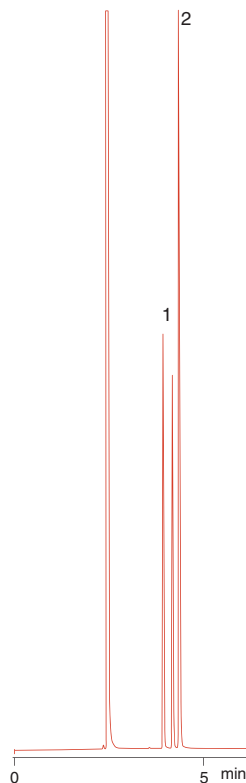
Enantiomer separation of pentane-2,4-diol (TFA)

MN Appl. No. 202911

Column: FS-LIPODEX® E,
50 m x 0.25 mm ID,
REF 723368.50,
max. temperature 200/220 °C
Injection: 0.1 µl (1 % in CH₂Cl₂)
split 200 ml/min
Carrier gas: 120 kPa H₂ (2.1 ml/min)
Temperature: 100 °C
Detector: FID 250 °C

Peaks: (C₅H₁₂O₂)

1. RR/SS
2. meso



Chiral separations

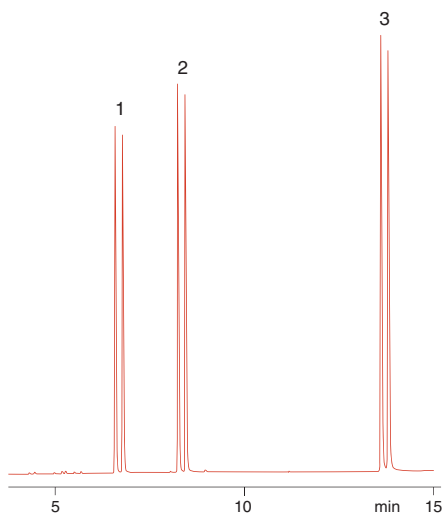
Enantiomer separation of 1,2-diols (TFA)

MN Appl. No. 202921

Column: FS-LIPODEX® E,
50 m x 0.25 mm ID,
REF 723368.50,
max. temperature 200/220 °C
Injection: 0.1 µl (1 % in CH₂Cl₂)
split 200 ml/min
Carrier gas: 120 kPa H₂ (2.1 ml/min)
Temperature: 80 °C $\xrightarrow{3\text{ °C/min}}$ 130 °C
Detector: FID 250 °C

Peaks:

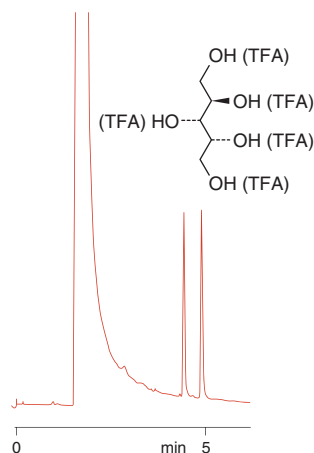
1. Pentane-1,2-diol (TFA), C₅H₁₂O₂
2. Hexane-1,2-diol (TFA), C₆H₁₄O₂
3. Octane-1,2-diol (TFA), C₈H₁₈O₂



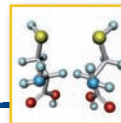
Enantiomer separation of arabinitol (TFA)

MN Appl. No. 203011

Column: LIPODEX® A,
50 m x 0.25 mm ID,
REF 723360.50,
max. temperature 200/220 °C
Carrier gas: H₂
Temperature: 110 °C
Detector: FID
C₅H₁₂O₅



Courtesy of Prof. W.A. König, Hamburg, Germany



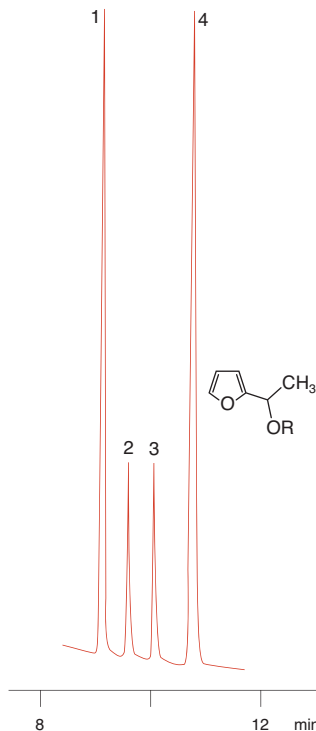
Enantiomer separation of R/S-1-(2-furyl)ethanol and R/S-1-(2-furyl)ethanol acetate

MN Appl. No. 213260

Column: equivalent to FS-HYDRODEX
β-6TBDM, 25 m x 0.25 mm
ID, REF 723381.25, max.
temperature 230/250 °C
Injector: 250 °C
Carrier gas: 60 kPa H₂
Temperature: 85 °C
Detector: FID 300 °C

Peaks:

1. (S)-1-(2-furyl)ethanol acetate (R = CO-CH₃)
2. (R)-1-(2-furyl)ethanol (R = H)
3. (S)-1-(2-furyl)ethanol (C₆H₈O₂)
4. (R)-1-(2-furyl)ethanol acetate (C₈H₁₀O₃)



Enantiomer separation of lactide

MN Appl. No. 202832

Column: FS-LIPODEX® A,
25 m x 0.25 mm ID,
REF 723360.25,
max. temperature 200/220 °C

Injection:

0.5 µl (1% in CH₂Cl₂)
split 180 ml/min

Carrier gas:

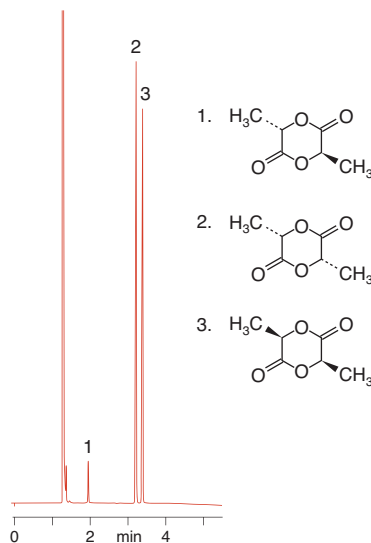
60 kPa H₂ (1.8 ml/min)

Temperature: 150 °C

Detector: FID 250 °C

Peaks: (C₆H₈O₄)

1. meso
2. SS
3. RR



Lactide can also be separated on HYDRODEX
β-PM
(see application 202822 at www.mn-net.com).

Courtesy of A. Ghanem, V. Schurig, Institute
of Organic Chemistry, University of Tübingen,
Germany

Chiral separations

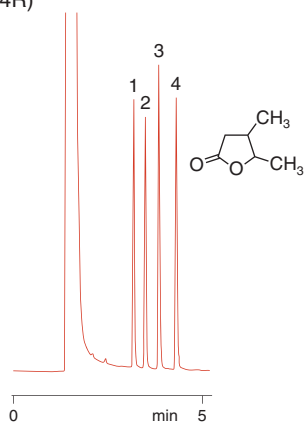
Analysis of 4 stereoisomers of 3,4-dimethylbutyrolactone

MN Appl. No. 202892

Column: LIPODEX® B,
50 m x 0.25 mm ID,
REF 723362.50,
max. temperature 200/220 °C
Carrier gas: 1 bar H₂
Temperature: 135 °C
Detector: FID

Peaks: (C₆H₁₀O₂)

1. (3S, 4R)
2. (3R, 4S)
3. (3S, 4S)
4. (3R, 4R)



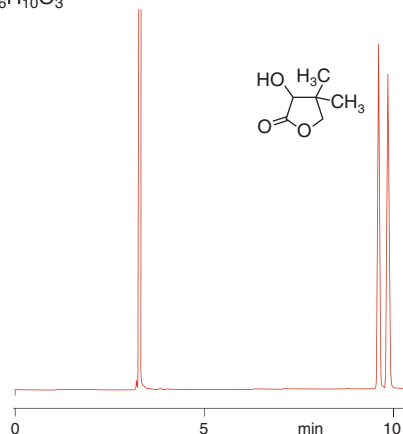
W.A. König et al., HRC 11 (1988) 621 – 625

Enantiomer analysis of pantolactone

MN Appl. No. 202912

Column: FS-HYDRODEX β-PM,
50 m x 0.25 mm ID,
REF 723370.50,
max. temperature 230/250 °C
Injection: 0.5 µl, split 1:100
Carrier gas: 100 kPa H₂
Temperature: 160 °C
Detector: FID 250 °C

C₆H₁₀O₃



Pantolactone can also be separated on LIPODEX® E (applications 202902). See our application database at www.mn-net.com.

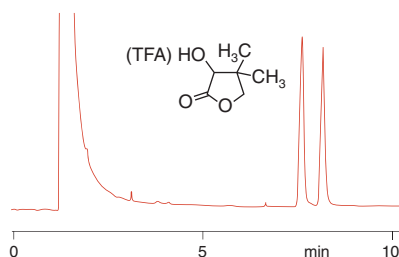
Enantiomer analysis of pantolactone (TFA)

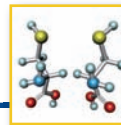
MN Appl. No. 202922

Column: LIPODEX® B,
50 m x 0.25 mm ID,
REF 723362.50,
max. temperature 200/220 °C
Carrier gas: H₂
Temperature: 110 °C
Detector: FID

C₆H₁₀O₃

Courtesy of Prof. W.A. König, Hamburg, Germany

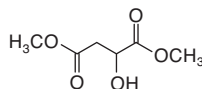




Enantiomer separation of dimethyl malate

MN Appl. No. 211370

Column: FS-HYDRODEX β-6TBDM, 25 m x 0.25 mm ID, REF 723381.25
 Injection: 1 μl, split, 250 °C
 Carrier gas: 1.1 ml/min He
 Temperature: 100 °C
 Detector: FID 250 °C



C₆H₁₀O₅

Resolution: 8.22



Courtesy of H. Casabianca, CNRS Service Central d'Analyse, Vernaison, France

Enantiomer separation of tartaric acid dimethyl esters (TFA)

MN Appl. No. 202712

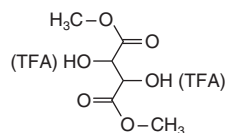
Column: FS-HYDRODEX β-PM,
 50 m x 0.25 mm ID,
 REF 723370.50,
 max. temperature
 230/250 °C
 Injection: 0.5 μl (1% in CH₂Cl₂)
 split 150 ml/min
 Carrier gas: 120 kPa H₂ (1.7 ml/
 min)
 Temperature: 135 °C
 Detector: FID 250 °C

Peaks: (C₆H₁₀O₆)

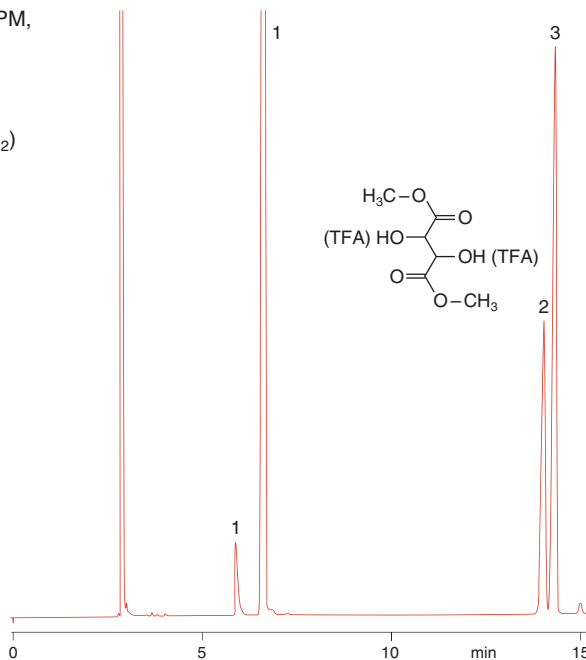
1. impurities

2. D-(-)

3. L-(+)



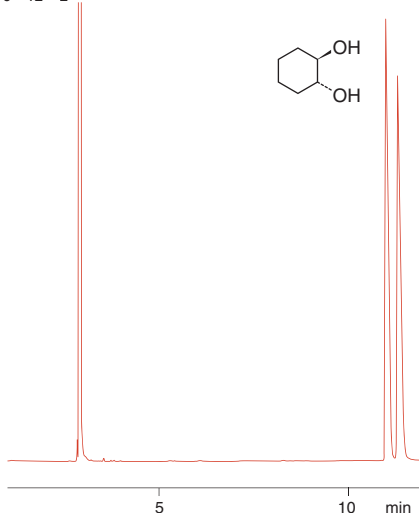
Tartaric acid dimethyl ester can also
 be separated on HYDRODEX β-PM
 (application 202722) or LIPODEX® A
 (application 202702).
 See our application database at
www.mn-net.com.



Chiral separations

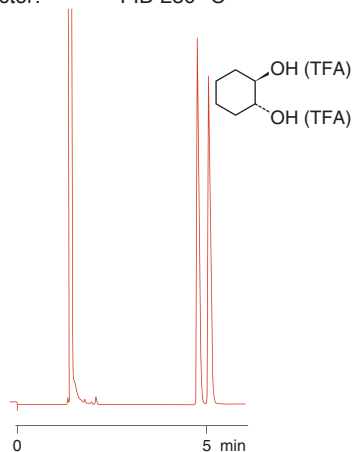
Enantiomer separation of *trans*-1,2-cyclohexanediol MN Appl. No. 202841

Column: FS-HYDRODEX β -PM,
50 m x 0.25 mm ID,
REF 723370.50,
max. temperature 230/250 °C
Injection: 0.5 μ l, split 1:100
Carrier gas: 100 kPa H₂
Temperature: 150 °C
Detector: FID 250 °C
C₆H₁₂O₂



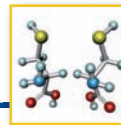
Enantiomer separation of *trans*-1,2-cyclohexanediol (TFA) MN Appl. No. 202831

Column: FS-LIPODEX® E,
25 m x 0.25 mm ID,
REF 723368.25,
max. temperature 200/ 220 °C
Injection: 0.5 μ l, split 1:100
Carrier gas: 60 kPa H₂
Temperature: 105 °C
Detector: FID 250 °C



trans-1,2-cyclohexanediol can also be separated on LIPODEX® A (see application 202821 at www.mn-net.com).

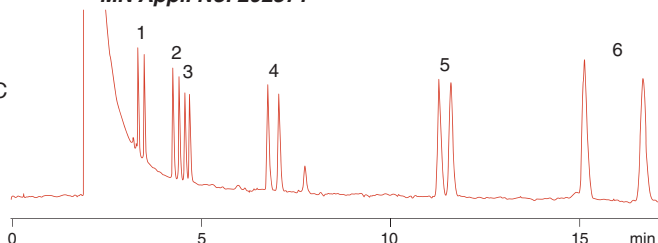
Compounds C₆ and greater



Enantiomer analysis of *trans*-1,2- and *trans*-1,3-cycloalkanediols (TFA)

MN Appl. No. 202871

Column: LIPODEX® D,
25 m x 0.25 mm ID,
REF 723366.25,
max. temperature 200/220 °C
Carrier gas: 0.7 bar H₂
Temperature: 150 °C
Detector: FID



Peaks:

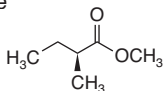
- | | |
|--|--|
| 1. Cyclohexane- <i>trans</i> -1,2-diol (C ₆ H ₁₂ O ₂) | 4. Cyclooctane- <i>trans</i> -1,2-diol (C ₈ H ₁₆ O ₂) |
| 2. Cycloheptane- <i>trans</i> -1,2-diol (C ₇ H ₁₄ O ₂) | 5. Cyclodecane- <i>trans</i> -1,2-diol (C ₁₀ H ₂₀ O ₂) |
| 3. Cycloheptane- <i>trans</i> -1,3-diol (C ₇ H ₁₄ O ₂) | 6. Cyclodecane- <i>trans</i> -1,3-diol (C ₁₀ H ₂₀ O ₂) |

W.A. König, S. Lutz, G. Wenz, E. von der Bey, HRC 11 (1988) 506 – 509

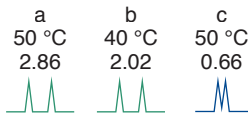
Enantiomer separation of 2-methylbutyric acid methyl ester

MN Appl. No. 211380

Columns: all 25 m x 0.25 mm ID:
a) FS-HYDRODEX β-6TBDM
(REF 723381.25),
b) FS-LIPODEX® E (REF
723368.25),
c) FS-HYDRODEX β-PM
(REF 723370.25)
Injection: 1 µl, split, 250 °C
Carrier gas: 1.1 ml/min He
Detector: FID 250 °C
C₆H₁₂O₂



Temperature:	50 °C	40 °C	50 °C
Resolution:	2.86	2.02	0.66



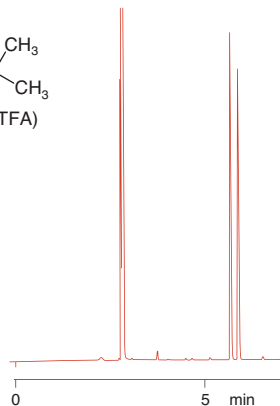
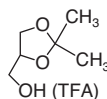
Courtesy of H. Casabianca, CNRS Service
Central d'Analyse, Vernaion, France

Enantiomer separation of solketal (TFA)

MN Appl. No. 202971

Column: FS-HYDRODEX β-PM,
50 m x 0.25 mm ID,
REF 723370.50,
max. temperature 230/250 °C
Injection: 0.1 µl (1 % in CH₂Cl₂)
split 150 ml/min
Carrier gas: 120 kPa H₂ (1.7 ml/min)
Temperature: 120 °C
Detector: FID 250 °C

C₆H₁₂O₃



Solketal can also be separated on LIPODEX® A
(see application 202981 at www.mn-net.com).

Chiral separations

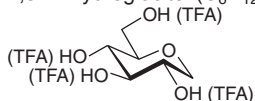
Enantiomer separation of 1,5-anhydroglucitol, -galactitol und -mannitol (TFA)

MN Appl. No. 203031

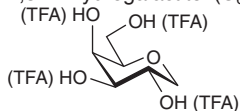
Column: LIPODEX® A, 50 m x 0.25 mm ID, REF 723360.50, max. temperature 200/220 °C
 Carrier gas: H₂
 Temperature: 120 °C
 Detector: FID

Peaks:

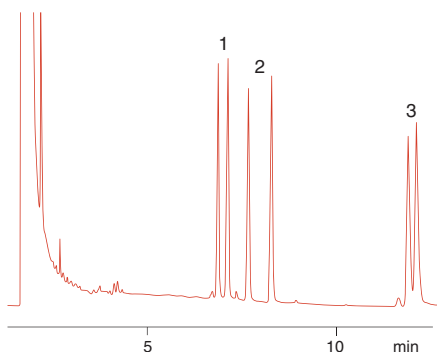
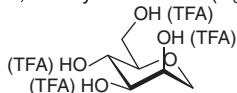
- 1,5-Anhydroglucitol (C₆H₁₂O₅)



- 1,5-Anhydrogalactitol (C₆H₁₂O₅)



- 1,5-Anhydromannitol (C₆H₁₂O₅)



Courtesy of Prof. W.A. König, Hamburg, Germany

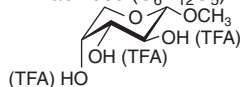
Enantiomer separation of arabinose (α- and β-anomers, methylglycosides, TFA)

MN Appl. No. 202372

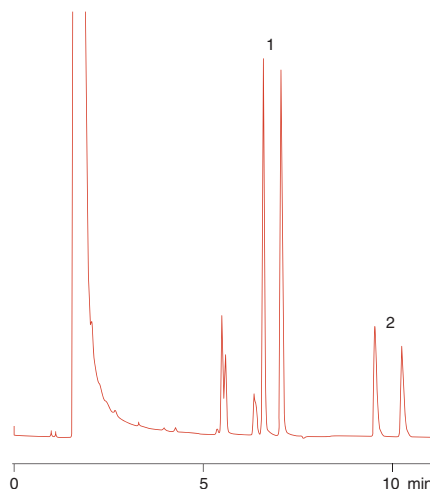
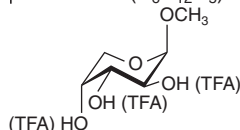
Column: LIPODEX® A, 50 m x 0.25 mm ID, REF 723360.50, max. temperature 200/220 °C
 Carrier gas: H₂
 Temperature: 100 °C
 Detector: FID

Peaks:

- α-Arabinose (C₆H₁₂O₅)



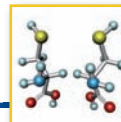
- β-Arabinose (C₆H₁₂O₅)



Courtesy of Prof. W.A. König, Hamburg, Germany

For separation of other sugar anomers on LIPODEX® A see applications 202432, 202392, 202382 at www.mn-net.com.

Compounds C₆ and greater



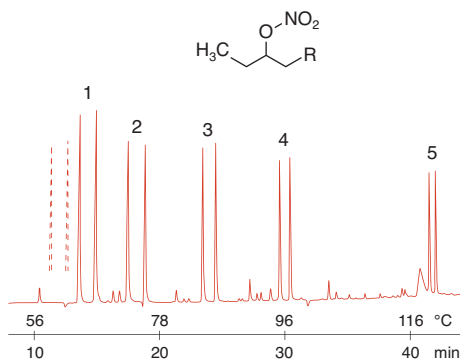
Separation of chiral alkylnitrates

MN Appl. No. 212050

Column: FS-LIPODEX® D, 25 m x 0.25 mm ID, REF 723366.25, max. temperature 200/220 °C
 Injection: on-column
 Carrier gas: 70 kPa H₂, 55 cm/s, 100 °C
 Temperature: 40 °C (2 min) $\xrightarrow{2\text{ °C/min}}$ 190 °C
 Detector: ECD 240 °C

Peaks:

1. *n*-Hexyl-3-nitrate (C₆H₁₃NO₃, R = C₂H₅)
2. *n*-Heptyl-3-nitrate (C₇H₁₅NO₃, R = C₃H₇)
3. *n*-Octyl-3-nitrate (C₈H₁₇NO₃, R = C₄H₉)
4. *n*-Nonyl-3-nitrate (C₉H₁₉NO₃, R = C₅H₁₁)
5. *n*-Undecyl-3-nitrate (C₁₁H₂₃NO₃, R = C₇H₁₅)



dashed lines indicate where peak 1 would be expected to elute

M. Schneider, K. Ballschmiter,
 J. Chromatography A **852** (1999) 525 – 534

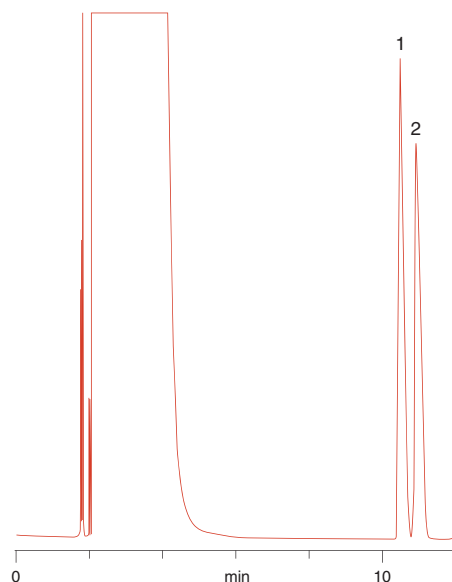
Enantiomer separation of 2-methylpentanol-1

MN Appl. No. 212450

Column: FS-HYDRODEX® β-TBDAC, 50 m x 0.25 mm ID, REF 723384.50, max. temperature 220/240 °C
 Sample: 0.1 % in CH₂Cl₂
 Carrier gas: 1.5 bar H₂
 Temperature: 80 °C
 Detector: FID

Peaks: (C₆H₁₄O)

1. (*R*)-2-Methylpentanol-1
2. (*S*)-2-Methylpentanol-1



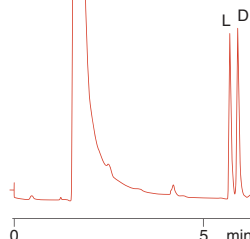
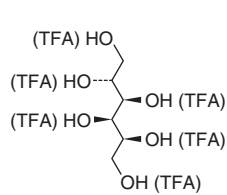
Chiral separations

Enantiomer separation of sorbitol (TFA)

MN Appl. No. 203001

Column: LIPODEX® A,
50 m x 0.25 mm ID,
REF 723360.50,
max. temperature 200/220 °C

Carrier gas: H₂
Temperature: 110 °C
Detector: FID
C₆H₁₄O₆



Courtesy of Prof. W.A. König, Hamburg, Germany

Enantiomer separation of thioacetic acid cyclopent-2-enyl ester

Carrier gas: 100 kPa H₂
Temperature: 70 °C (2 min) $\xrightarrow{10\text{ °C/min}}$ 100 °C (2 min) $\xrightarrow{10\text{ °C/min}}$ 130 °C (2 min)
Detector: FID

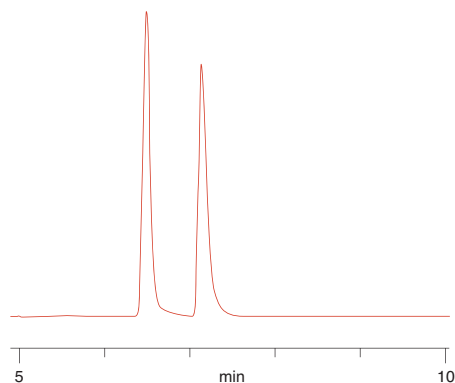
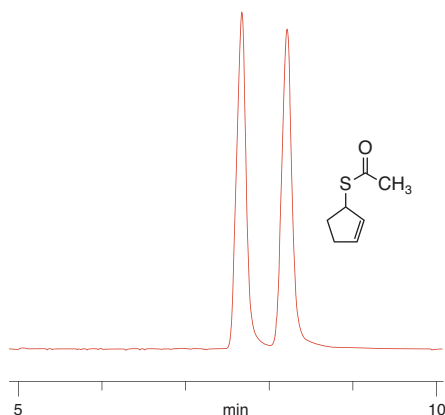
MN Appl. No. 212270

Column: FS-LIPODEX® E,
25 m x 0.25 mm ID,
REF 723368.25,
max. temperature 200/220 °C

MN Appl. No. 212280

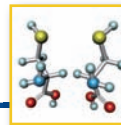
Column: FS-LIPODEX® G,
25 m x 0.25 mm ID,
REF 723379.25,
max. temperature 220/240 °C

C₇H₁₀OS



Courtesy of Mrs. Vermeeren, AK Prof. Gais, Inst. für Org. Chemie, RWTH Aachen, Germany

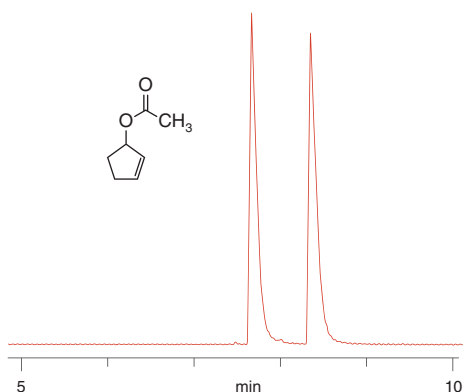
Compounds C₆ – C₇



Enantiomer separation of acetic acid cyclopent-2-enyl ester

MN Appl. No. 212300

Column: FS-LIPODEX® G,
25 m x 0.25 mm ID,
REF 723379.25,
max. temperature 220/240 °C
Carrier gas: 100 kPa H₂
Temperature: 50 °C (5 min) $\xrightarrow{10\text{ °C/min}}$ 80 °C
(5 min)
Detector: FID
C₇H₁₀O₂

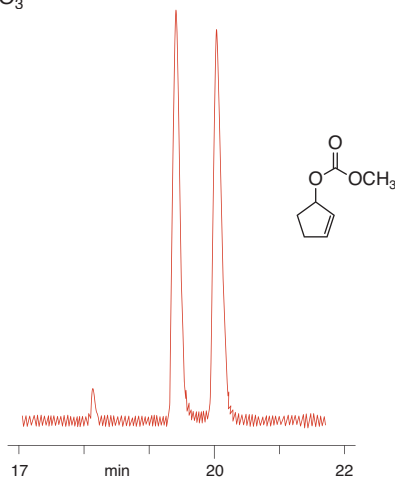


Courtesy of Mrs. Vermeeren, AK Prof. Gais,
Inst. für Org. Chemie, RWTH Aachen, Germany

Enantiomer separation of cyclopent-2-enyl methyl carbonate

MN Appl. No. 212330

Column: FS-LIPODEX® E, 25 m x 0.25 mm ID,
REF 723368.25, max. temperature 200/220 °C
Carrier gas: 100 kPa H₂
Temperature: 50 °C (15 min) $\xrightarrow{10\text{ °C/min}}$ 80 °C
(5 min) $\xrightarrow{10\text{ °C/min}}$ 120 °C (5 min)
Detector: FID
C₇H₁₀O₃



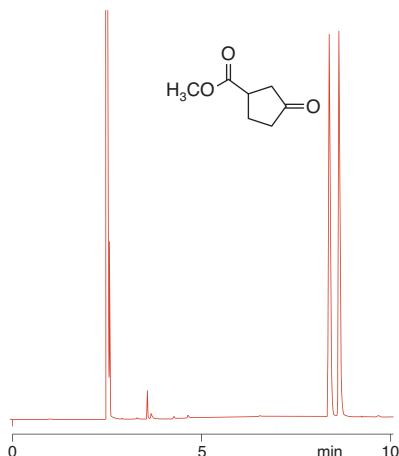
Courtesy of Mrs. Vermeeren, AK Prof. Gais,
Inst. für Org. Chemie, RWTH Aachen, Germany

Chiral separations

Enantiomer separation of cyclopentanonecarboxylic acid methyl ester

MN Appl. No. 202080

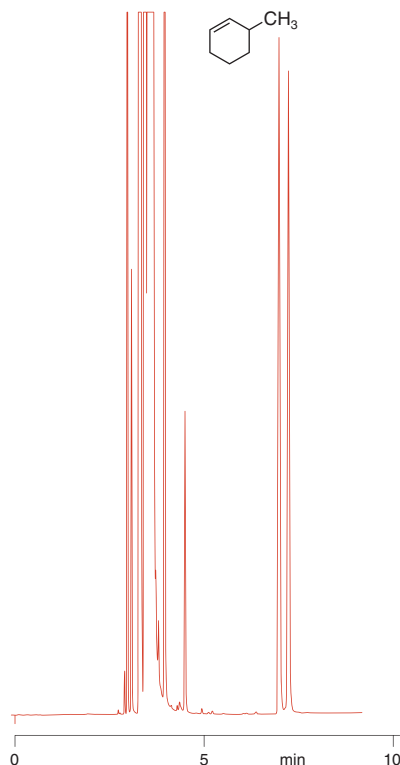
Column: FS-LIPODEX® A,
50 m x 0.25 mm ID,
REF 723360.50,
max. temperature 200/220 °C
Injection: 0.1 µl (1 % in CH₂Cl₂)
split 320 ml/min
Carrier gas: 120 kPa H₂ (2.2 ml/min)
Temperature: 120 °C
Detector: FID 250 °C
C₇H₁₀O₃



Enantiomer separation of 3-methylcyclohexene

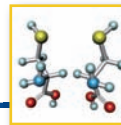
MN Appl. No. 201860

Column: FS-LIPODEX® C,
50 m x 0.25 mm ID,
REF 723364.50,
max. temperature 200/220 °C
Injection: 1.0 µl, split 1:100
Carrier gas: 100 kPa H₂
Temperature: 35 °C
Detector: FID 250 °C
C₇H₁₂



3-methylcyclohexene can also be separated on LIPODEX® A
(see application 201870 at www.mn-net.com).

Compounds C₇ – C₈



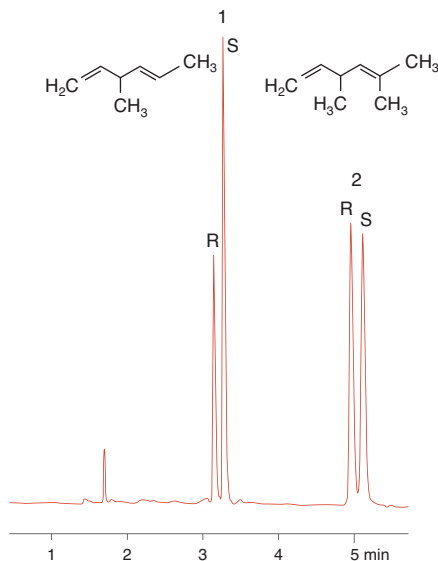
Enantiomer separation of 3-methyl-1,4(Z)-hexadiene and 3,5-dimethyl-1,4-hexadiene

MN Appl. No. 201890

Column: LIPODEX® C,
50 m x 0.25 mm ID,
REF 723364.50,
max. temperature 200/220 °C
Injection: headspace
Carrier gas: 1 bar H₂
Temperature: 30 °C
Detector: FID

Peaks:

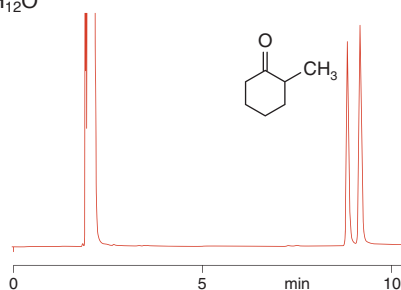
1. 3-Methyl-1,4-hexadiene (C₇H₁₂)
2. 3,5-Dimethyl-1,4-hexadiene (C₈H₁₄)



Enantiomer separation of 2-methylcyclohexanone

MN Appl. No. 202050

Column: FS-LIPODEX® A, 50 m x
0.25 mm ID, REF 723360.50,
max. temperature 200/220 °C
Injection: 1 µl (1 % in CH₂Cl₂)
split 245 ml/min
Carrier gas: 110 kPa H₂ (2.3 ml/min)
Temperature: 70 °C
Detector: FID 240 °C
C₇H₁₂O



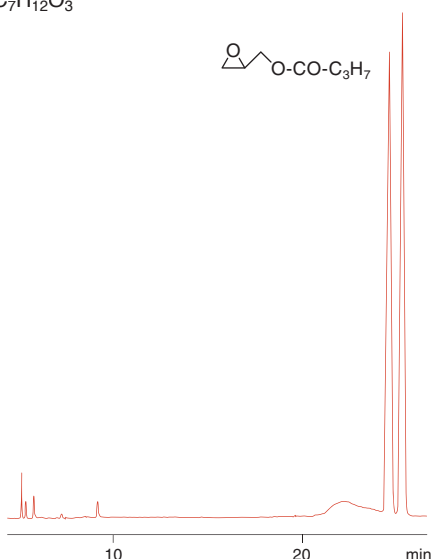
W.A. König et al., HRC 12 (1989) 35 – 39

Chiral separations

Enantiomer separation of glycidyl butyrate

MN Appl. No. 202731

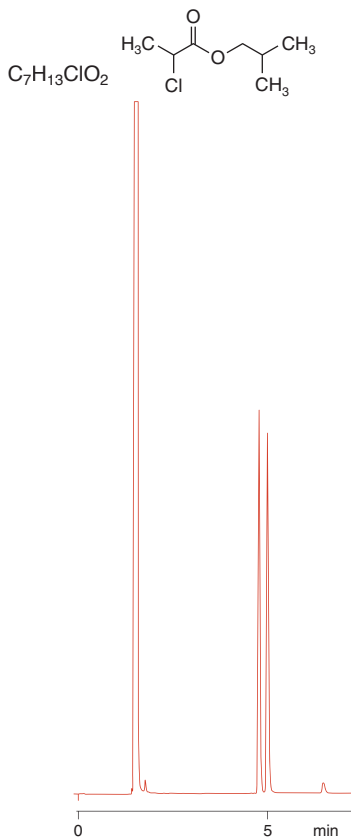
Column: FS-LIPODEX® A,
25 m x 0.25 mm ID,
REF 723360.25,
max. temperature 200/220 °C
Injection: 1 µl (1 % in CH₂Cl₂)
split 180 ml/min
Carrier gas: 60 kPa H₂ (1.8 ml/min)
Temperature: 60 °C
Detector: FID 250 °C
C₇H₁₂O₃

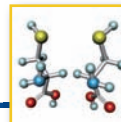


Enantiomer separation of 2-chloropropionic acid isobutyl ester

MN Appl. No. 202532

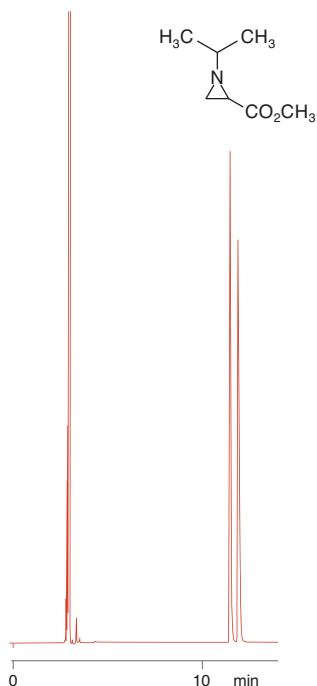
Column: FS-LIPODEX® E, 25 m x 0.25
mm ID, REF 723368.25,
max. temperature 200/220 °C
Injection: 1 µl, 1 % in CH₂Cl₂, split 125
ml/min
Carrier gas: 50 kPa H₂ (1.5 ml/min)
Temperature: 90 °C
Detector: FID 250 °C





Enantiomer separation of 1-*i*-propyl-2-carboxymethyl aziridine *MN Appl. No. 202361*

Column: FS-HYDRODEX β-PM,
50 m x 0.25 mm ID,
REF 723370.50,
max. temperature 230/250 °C
Injection: 0.1 μl (1 % in CH₂Cl₂)
split 100 ml/min
Carrier gas: 120 kPa H₂ (1.7 ml/min)
Temperature: 110 °C
Detector: FID 250 °C
C₇H₁₃NO₂



Enantiomers separation of 2-methylbutyric acid ethyl ester *MN Appl. No. 211400*

Columns: all 25 m x 0.25 mm ID:
a) FS-HYDRODEX β-6TBDM
(REF 723381.25),
b) FS-LIPODEX® E
(REF 723368.25),
c) FS-HYDRODEX β-PM
(REF 723370.25)
Injection: 1 μl, split, 250 °C
Carrier gas: 1.1 ml/min He
Detector: FID 250 °C
C₇H₁₄O₂

	a	b	c
Temperature:	55 °C	40 °C	50 °C
Resolution:	2.93	1.31	0.98



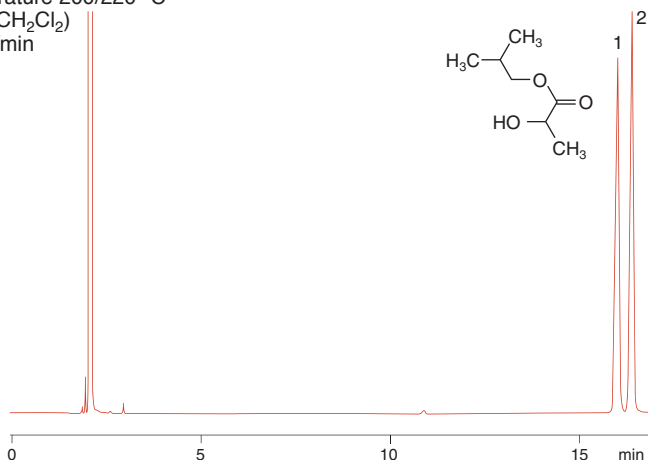
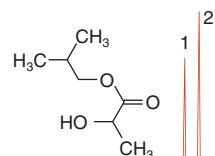
Courtesy of H. Casabianca, CNRS Service
Central d'Analyse, Vernaison, France

Chiral separations

Enantiomer separation of isobutyl lactate

MN Appl. No. 202802

Column: FS-LIPODEX® A, 50 m x 0.25 mm ID, REF 723360.50,
max. temperature 200/220 °C
Injection: 1 µl (1 % in CH₂Cl₂)
split 220 ml/min
Carrier gas: 120 kPa H₂
(2.5 ml/min)
Temperature: 60 °C
Detector: FID 250 °C
C₇H₁₄O₃



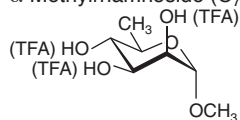
Enantiomer separation of α-methylrhannoside and α-methylquinovoside (TFA)

MN Appl. No. 203041

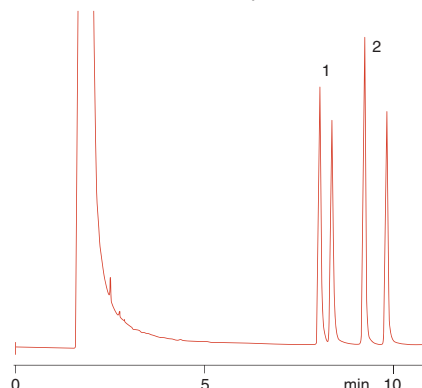
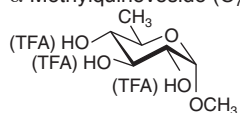
Column: LIPODEX® C, 50 m x 0.25 mm ID, REF 723364.50, max. temperature 200/220 °C
Carrier gas: 1 bar H₂
Temperature: 90 °C
Detector: FID

Peaks:

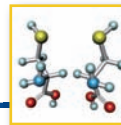
1. α-Methylrhannoside (C₇H₁₄O₅)



2. α-Methylquinovoside (C₇H₁₄O₅)



W.A. König et al., HRC **12** (1989) 35 – 39



Enantiomer separation of glucose and mannose (α - and β -anomers, methylglycosides, TFA)

MN Appl. No. 202412

Column: LIPODEX® B, 50 m x 0.25 mm ID, REF 723362.50, max. temperature 200/220 °C

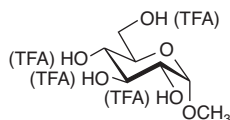
Carrier gas: H₂

Temperature: 155 °C

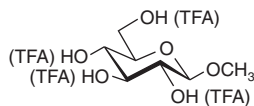
Detector: FID

Peaks: (C₇H₁₄O₆)

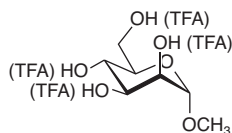
1. α -Glucose



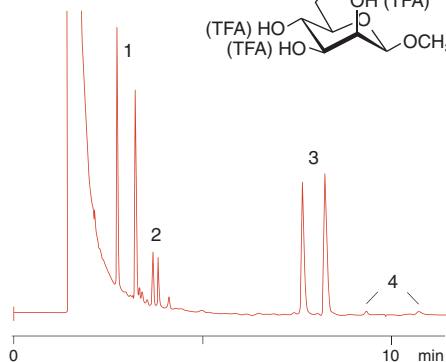
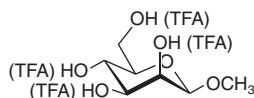
2. β -Glucose



3. α -Mannose



4. β -Mannose



Courtesy of Prof. W.A. König, Hamburg, Germany

See application 202402 at www.mn-net.com for separation of glucose anomer methylglycosides on LIPODEX® A

Enantiomer separation of 2-heptanol (TFA)

MN Appl. No. 202511

Column: FS-LIPODEX® A, 50 m x 0.25 mm ID, REF 723360.50, max. temperature 200/220 °C

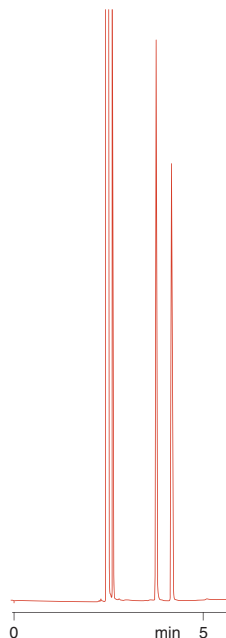
Injection: 0.1 μ l (1% in CH₂Cl₂) split 320 ml/min

Carrier gas: 120 kPa H₂ (2.2 ml/min)

Temperature: 80 °C

Detector: FID 250 °C

C₇H₁₆O



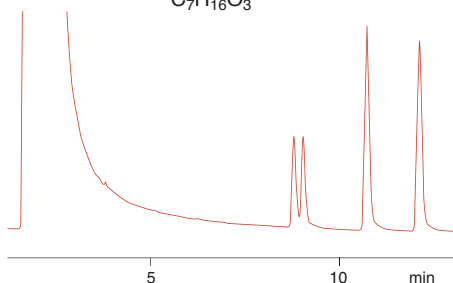
2-heptanol (TFA) can also be separated on LIPODEX® E (see application 202501 at www.mn-net.com).

Chiral separations

Analysis of 4 stereoisomers of heptane-1,2,3-triol (TFA)

MN Appl. No. 202941

Column: LIPODEX® A,
50 m x 0.25 mm ID,
REF 723360.50,
max. temperature 200/220 °C
Carrier gas: H₂
Temperature: 85 °C
Detector: FID
C₇H₁₆O₃



Courtesy of Prof. W.A. König, Hamburg, Germany

Enantiomer separation of styrene oxide

MN Appl. No. 211410

Columns: all 25 m x 0.25 mm ID:
a) FS-HYDRODEX β-6TBDM
(REF 723381.25),
b) FS-LIPODEX® E
(REF 723368.25),
c) FS-HYDRODEX β-PM
(REF 723370.25)
Injection: 1 μl, split, 250 °C
Carrier gas: 1.1 ml/min He
Detector: FID 250 °C
C₈H₈O

	a	b	c
Temperature:	90 °C	80 °C	85 °C
Resolution:	4.81	13.06	1.69



Courtesy of H. Casabianca, CNRS Service
Central d'Analyse, Vernaison, France

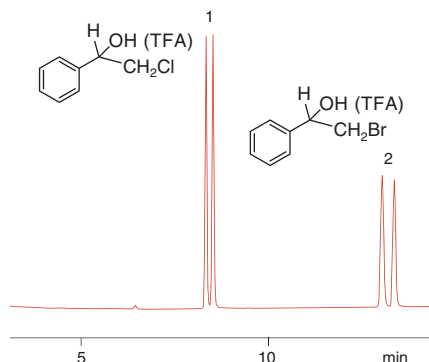
Enantiomer separation of 2-chloro- and 2-bromo-1-phenylethanol (TFA)

MN Appl. No. 202531

Column: LIPODEX® A,
50 m x 0.25 mm ID,
REF 723360.50,
max. temperature 200/220 °C
Carrier gas: H₂
Temperature: 110 °C
Detector: FID

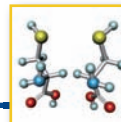
Peaks:

1. 2-Chloro-1-phenylethanol (TFA) (C₈H₉ClO)
2. 2-Bromo-1-phenylethanol (TFA) (C₈H₉BrO)



Courtesy of Prof. W.A. König, Hamburg, Germany

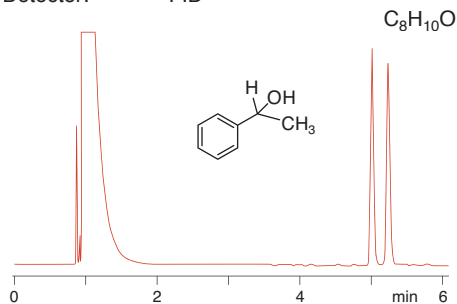
Compounds C₇ – C₈



Enantiomer separation of α -methylbenzyl alcohol = α -phenylethanol

MN Appl. No. 212400

Column: FS-HYDRODEX® β -TBDAC,
25 m x 0.25 mm ID,
REF 723384.25,
max. temperature 220/240 °C
Carrier gas: 0.6 bar H₂
Temperature: 130 °C
Detector: FID



α -Phenylethanol can also be separated on
HYDRODEX β -PM (application 202541),
HYDRODEX β -3P (application 202571),
LIPODEX® A (application 202561).
See our application database at www.mn-net.com.

Enantiomer separation of α -phenylethanol (TFA)

MN Appl. No. 211430

Columns: all 25 m x 0.25 mm ID:
a) FS-HYDRODEX β -6TBDM
(REF 723381.25),
b) FS-LIPODEX® E
(REF 723368.25),
c) FS-HYDRODEX β -PM
(REF 723370.25)
Injection: 1 μ l, split, 250 °C
Carrier gas: 1.1 ml/min He
Detector: FID 250 °C

	a	b	c
Temperature:	90 °C	80 °C	100 °C
Resolution:	2.4	3.52	3.47



Courtesy of H. Casabianca, CNRS Service
Central d'Analyse, Vernaison, France

α -Phenylethanol (TFA) can also be separated
on HYDRODEX β -3P (application 202601) or
LIPODEX® A (application 202591).
See our application database at www.mn-net.com.

MN Appl. No. 211420

Columns: all 25 m x 0.25 mm ID:
a) FS-HYDRODEX β -6TBDM
(REF 723381.25),
b) FS-LIPODEX® E
(REF 723368.25),
c) FS-HYDRODEX β -PM
(REF 723370.25)
Injection: 1 μ l, split, 250 °C
Carrier gas: 1.1 ml/min He
Detector: FID 250 °C

	a	b	c
Temperature:	115 °C	80 °C	100 °C
Resolution:	<0.5	3.52	3.47



Courtesy of H. Casabianca, CNRS Service
Central d'Analyse, Vernaison, France

Chiral separations

Enantiomer separation of 1-phenyl-ethanol and *o*-, *m*-, *p*-methyl derivatives (TFA)

MN Appl. No. 202521

Column: LIPODEX® A, 50 m x 0.25 mm ID, REF 723360.50, max. temperature 200/220 °C

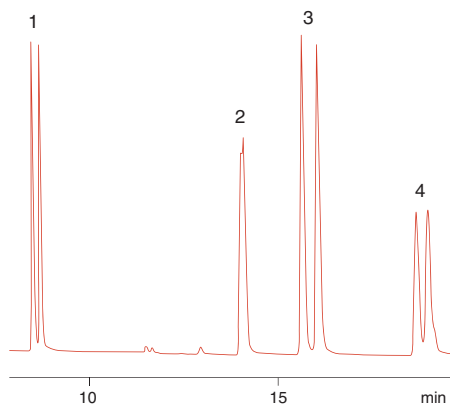
Carrier gas: H₂

Temperature: 75 °C

Detector: FID

Peaks:

1. 1-Phenylethanol (C₈H₁₀O)
2. 1-(2-Methylphenyl)ethanol (C₉H₁₂O)
3. 1-(3-Methylphenyl)ethanol
4. 1-(4-Methylphenyl)ethanol



Enantiomer separation of α -phenyl-ethylamine (TFA)

MN Appl. No. 202290

Column: FS-LIPODEX® A, 50 m x 0.25 mm ID, REF 723360.50, max. temperature 200/220 °C

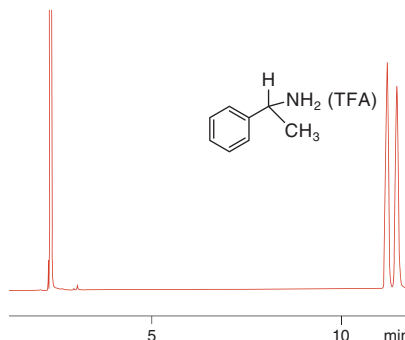
Injection: 1 μ l (1% in CH₂Cl₂) split 245 ml/min

Carrier gas: 110 kPa H₂ (2.3 ml/min)

Temperature: 120 °C

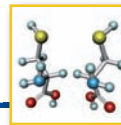
Detector: FID 240 °C

C₈H₁₁N



Courtesy of Prof. W.A. König, Hamburg, Germany

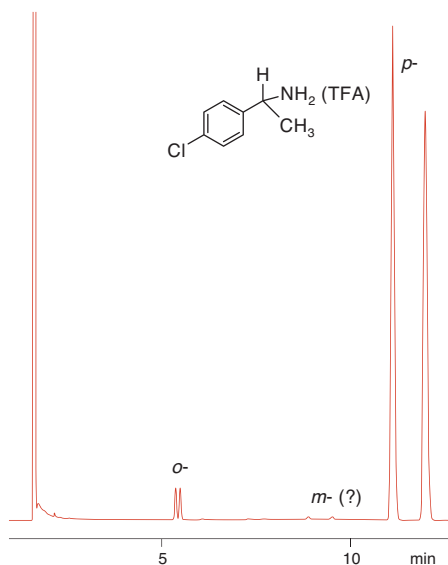
Compounds C₈ – C₉



Enantiomer separation of *p*-chloro-phenylethylamine (TFA)

MN Appl. No. 202300

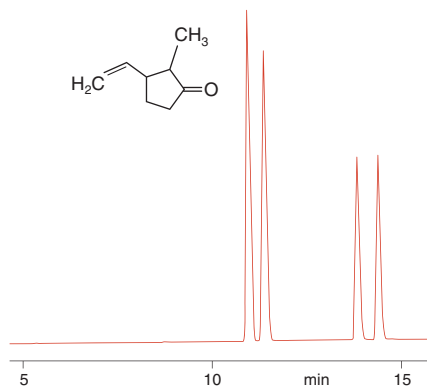
Column: FS-LIPODEX® D,
25 m x 0.25 mm ID,
REF 723366.25,
max. temperature 200/220 °C
Injection: 0.1 µl (5% in CH₂Cl₂)
split 130 ml/min
Carrier gas: 50 kPa H₂ (1.1 ml/min)
Temperature: 170 °C
Detector: FID 250 °C
R is eluted before S.
C₈H₁₁ClN



Enantiomer separation of 2-methyl-3-vinylcyclopentanone

MN Appl. No. 202060

Column: FS-LIPODEX® E,
25 m x 0.25 mm ID,
REF 723368.25,
max. temperature 200/220 °C
Injection: 0.2 µl, split 80 ml/min
Carrier gas: 50 kPa H₂ (1.2 ml/min)
Temperature: 70 °C (3 min) $\xrightarrow{2\text{ °C/min}}$ 110 °C
Detector: FID 250 °C
C₈H₁₂O



Chiral separations

Enantiomer composition of a bicyclic lactone (hexahydroisobenzofuranone)

MN Appl. No. 202962

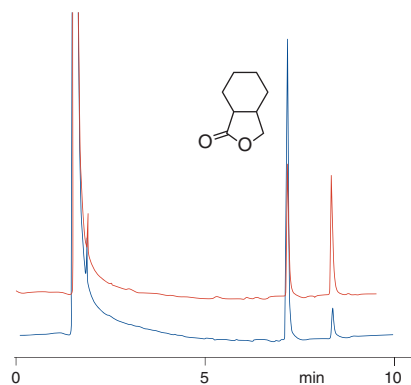
Column: LIPODEX® D,
50 m x 0.25 mm ID,
REF 723366.50,
max. temperature 200/220 °C

Carrier gas: H₂

Temperature: 170 °C

Detector: FID

C₈H₁₂O₂



Courtesy of Prof. W.A. König, Hamburg, Germany

Enantiomer separation of acetic acid cyclohex-2-enyl ester

MN Appl. No. 212290

Column: FS-LIPODEX® G,
25 m x 0.25 mm ID,
REF723379.25,
max. temperature 220/240 °C

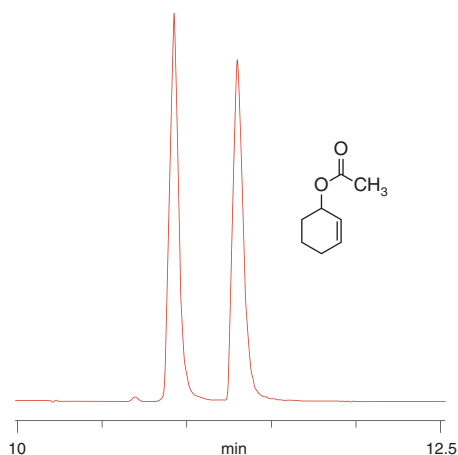
Carrier gas: 100 kPa H₂

Temperature: 50 °C (5 min) $\xrightarrow{10\text{ °C/min}}$ 80 °C

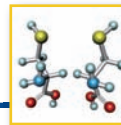
(5 min) 10 °C/min

Detector: FID

C₈H₁₂O₂



Courtesy of Mrs. Vermeeren, AK Prof. Gais, Inst. für Org. Chemie, RWTH Aachen, Germany



Enantiomer separation of (E)-2,3-epoxy-4-octin-1-ol (TFA)

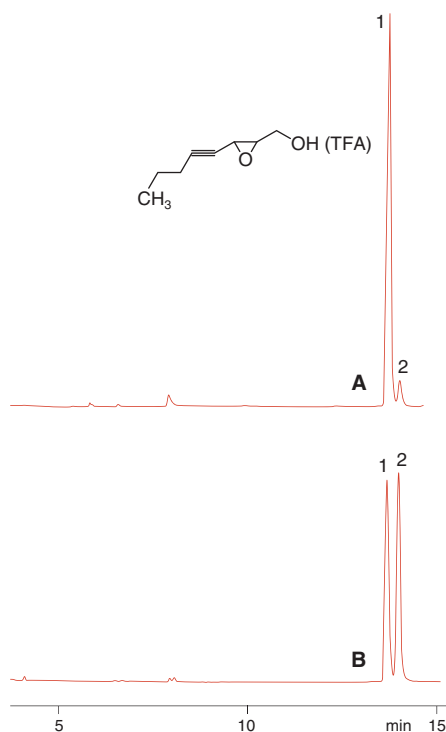
MN Appl. No. 202631

Column: LIPODEX® A,
50 m x 0.25 mm ID,
REF 723360.50,
max. temperature 200/220 °C
Carrier gas: 1 bar H₂
Temperature: 80 °C
Detector: FID

Peaks:

C₈H₁₂O₂, 1. (2S,3S), 2. (2R,3R)

Trace A: ee = 89.3 %, trace B = racemate

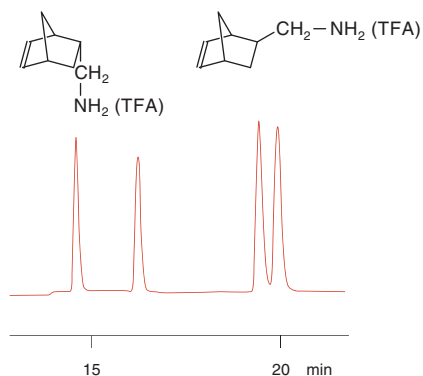


Separation of 4 stereoisomers of 5-aminomethylnorbornene (N-TFA)

MN Appl. No. 202350

Column: LIPODEX® D,
50 m x 0.25 mm ID,
REF 723366.50,
max. temperature 200/220 °C
Temperature: 150 °C
Carrier gas: 1 bar H₂
Detector: FID

C₈H₁₃N



W.A. König, S. Lutz, G. Wenz, E. von der Bey,
HRC 11 (1988) 506 – 509

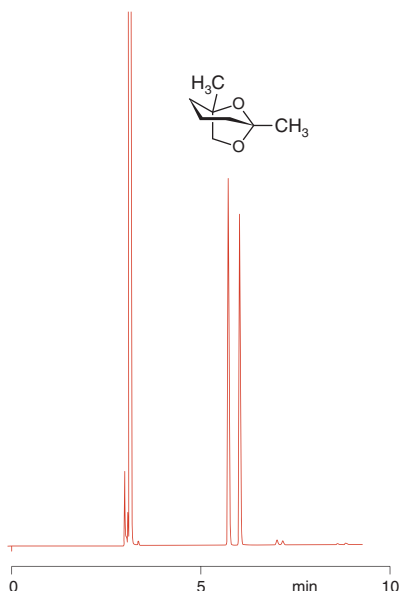
W.A. König et al., Angew. Chem. Int. Ed. Engl.
28 (1989) 178

Chiral separations

Enantiomer separation of frontalin

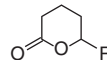
MN Appl. No. 202190

Column: FS-LIPODEX® E,
50 m x 0.25 mm ID,
REF 723368.50,
max. temperature 200/220 °C
Injection: 0.5 µl, split 180 ml/min
Carrier gas: 100 kPa H₂ (1.5 ml/min)
Temperature: 100 °C
Detector: FID 250 °C
C₈H₁₄O₂



Enantiomer analysis of δ-lactones

Columns: all 25 m x 0.25 mm ID:
a) FS-HYDRODEX β-6TBDM
(REF 723381.25),
b) FS-LIPODEX® E
(REF 723368.25),
c) FS-HYDRODEX β-PM
(REF 723370.25)
Injection: 1 µl, split, 250 °C
Carrier gas: 1.1 ml/min He
Detector: FID 250 °C



δ-Octalactone (C₈H₁₄O₂)

MN Appl. No. 211450

	a	b	c
Temperature:	120 °C	115 °C	120 °C
Resolution:	5.92	1.2	1.32



δ-Nonalactone (C₉H₁₆O₂)

MN Appl. No. 211490

	a	b	c
Temperature:	130 °C	120 °C	130 °C
Resolution:	6	7.65	1.07



δ-Decalactone (C₁₀H₁₈O₂)

MN Appl. No. 211670

	a	b	c
Temperature:	145 °C	125 °C	145 °C
Resolution:	1.04	2.01	<0.5



δ-Undecalactone (C₁₁H₂₀O₂)

MN Appl. No. 211730

	a	b	c
Temperature:	170 °C	130 °C	155 °C
Resolution:	0.88	0.67	<0.5



δ-Dodecalactone

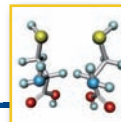
MN Appl. No. 211760

	a	b	c
Temperature:	155 °C	140 °C	155 °C
Resolution:	1.25	1.11	<0.5



Courtesy of H. Casabianca, CNRS Service
Central d'Analyse, Vernaison, France

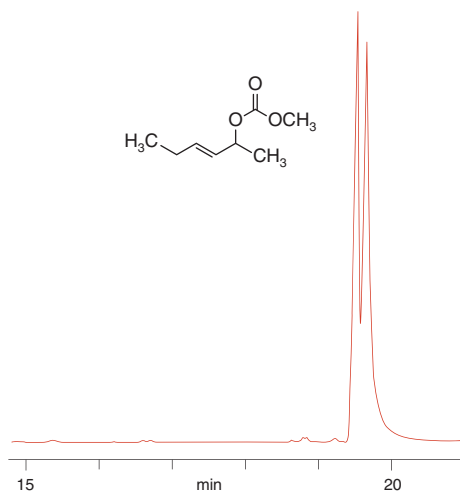
Compounds C₈ and greater



Enantiomer separation of 2-hex-3-enyl methyl carbonate

MN Appl. No. 212310

Column: FS-LIPODEX® G,
25 m x 0.25 mm ID,
REF723379.25,
max. temperature 220/240 °C
Carrier gas: 100 kPa H₂
Temperature: 50 °C (20 min) $\xrightarrow{10\text{ °C/min}}$ 80 °C
(5 min) $\xrightarrow{10\text{ °C/min}}$ 120 °C (5 min)
Detector: FID
C₈H₁₄O₃

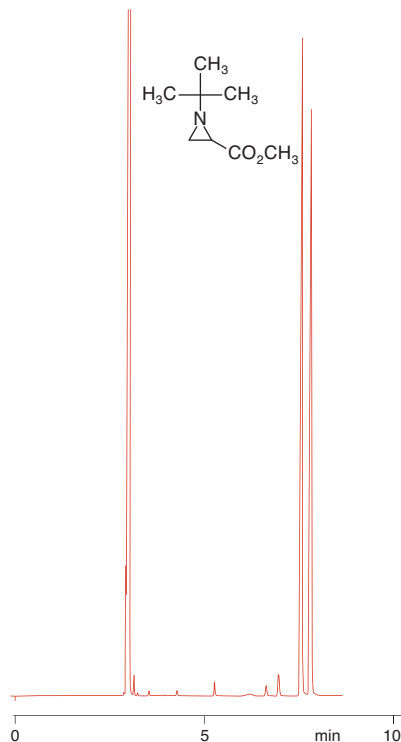


Courtesy of Mrs. Vermeeren, AK Prof. Gais,
Inst. für Org. Chemie, RWTH Aachen, Germany

Enantiomer separation of 1-*t*-butyl-2-carboxymethyl aziridine

MN Appl. No. 202371

Column: FS-HYDRODEX β-PM,
50 m x 0.25 mm ID,
REF 723370.50,
max. temperature 230/250 °C
Injection: 0.1 µl (1 % in CH₂Cl₂)
split 100 ml/min
Carrier gas: 120 kPa H₂ (1.7 ml/min)
Temperature: 130 °C
Detector: FID 250 °C
C₈H₁₅NO₂



Chiral separations

Enantiomer separation of 4-methyl-3-heptanone (ant pheromone)

MN Appl. No. 201900

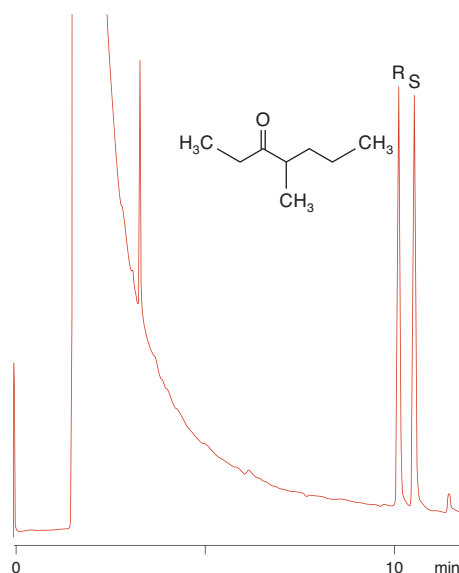
Column: LIPODEX® A,
50 m x 0.25 mm ID,
REF 723360.50,
max. temperature 200/220 °C

Carrier gas: H₂

Temperature: 85 °C

Detector: FID

C₈H₁₆O



Courtesy of Prof. W.A. König, Hamburg, Germany

Enantiomer separation of 1-octen-3-ol

MN Appl. No. 211460

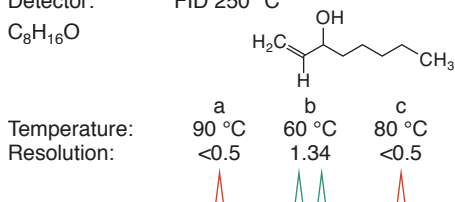
Columns: all 25 m x 0.25 mm ID:
a) FS-HYDRODEX β-6TBDM
(REF 723381.25),
b) FS-LIPODEX® E
(REF 723368.25),
c) FS-HYDRODEX β-PM
(REF 723370.25)

Injection: 1 µl, split, 250 °C

Carrier gas: 1.1 ml/min He

Detector: FID 250 °C

C₈H₁₆O



Courtesy of H. Casabianca, CNRS Service
Central d'Analyse, Vernaison, France

1-octen-3-ol can also be separated on
LIPODEX® A

(see application 202761 at www.mn-net.com).

Enantiomer separation of 1-octen-3-ol (TFA)

MN Appl. No. 211470

Columns: FS-HYDRODEX β-6TBDM,
25 m x 0.25 mm ID,
REF 723381.25

Injection: 1 µl, split, 250 °C

Carrier gas: 1.1 ml/min He

Temperature: 80 °C

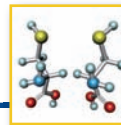
Detector: FID 250 °C

Resolution: 1.49



Courtesy of H. Casabianca, CNRS Service
Central d'Analyse, Vernaison, France

Compounds C₈ – C₉



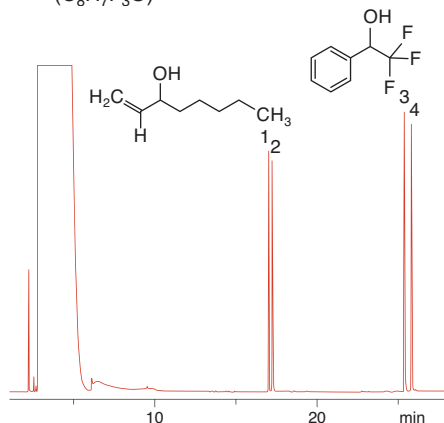
Enantiomer separation of 1-octen-3-ol and 2,2,2-trifluorophenylethanol

MN Appl. No. 212460

Column: FS-HYDRODEX® β-TBDAC,
50 m x 0.25 mm ID,
REF 723384.50,
max. temperature 220/240 °C
Sample: 0.1 % in CH₂Cl₂
Carrier gas: 1.2 bar H₂
Temperature: 80 °C (5 min) $\xrightarrow{3\text{ °C/min}}$ 140 °C
(5 min)
Detector: FID

Peaks:

1. (S)-1-Octen-3-ol (C₈H₁₆O)
2. (R)-1-Octen-3-ol
- 3./4. (R)/(S)-2,2,2-Trifluorophenylethanol
(C₈H₇F₃O)



Enantiomer separation of 1-indanol (TFA)

MN Appl. No. 202781

Column: FS-HYDRODEX β-PM,
50 m x 0.25 mm ID,
REF 723370.50,
max. temperature 230/250 °C
Injection: 1 µl (1 % in CH₂Cl₂)
split 150 ml/min
Carrier gas: 120 kPa H₂ (1.7 ml/min)
Temperature: 130 °C
Detector: FID 250 °C
C₉H₁₀O



1-indanol can also be separated on LIPODEX®
A (see application 202771 at www.mn-net.com).

Chiral separations

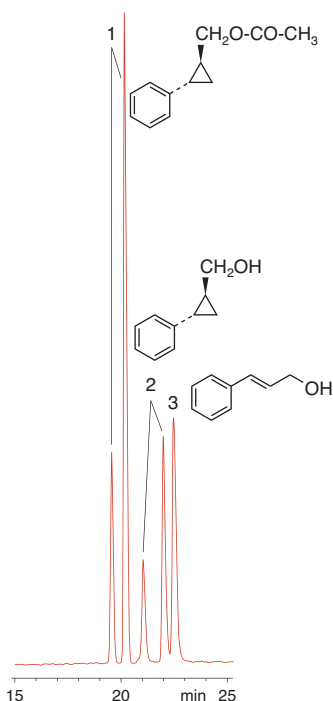
Enantiomer separation of phenylcyclopropylmethanol, phenylcyclopropylmethyl acetate and cinnamyl alcohol (TFA)

MN Appl. No. 212680

Column: FS-HYDRODEX β -TBDAC,
25 m x 0.25 mm ID,
REF 723384.25,
max. temperature 220/240 °C
Injection: 1 μ l (0.1 % in methyl *t*-butyl
ether), 220 °C, split 15 ml/min
Carrier gas: 0.6 bar H₂
Temperature: 130 °C
Detector: FID 220 °C

Peaks:

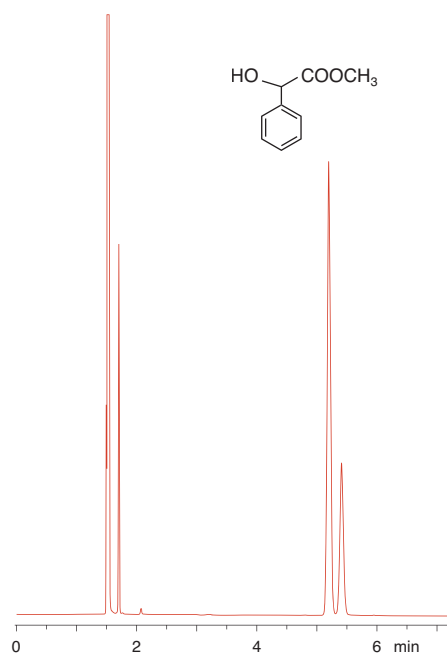
1. ($\pm$)-trans-2-Phenylcyclopropylmethyl acetate (C₁₂H₁₄O₂)
2. ($\pm$)-trans-2-Phenylcyclopropylmethanol (C₁₀H₁₂O)
3. Cinnamyl alcohol (C₉H₁₀O)



Enantiomer separation of mandelic acid methyl ester

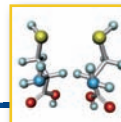
MN Appl. No. 213020

Column: FS-HYDRODEX γ -TBDAC,
25 m x 0.25 mm ID,
REF 723387.25,
max. temperature 220/240 °C
Injection: 1 μ l, split 50 ml/min
Carrier gas: 0.6 bar H₂
Temperature: 170 °C
Detector: FID
C₉H₁₀O₃



Mandelic acid methyl ester can also be separated on LIPODEX® A (application 202672) or LIPODEX® C (application 202682). See our application database at www.mn-net.com.

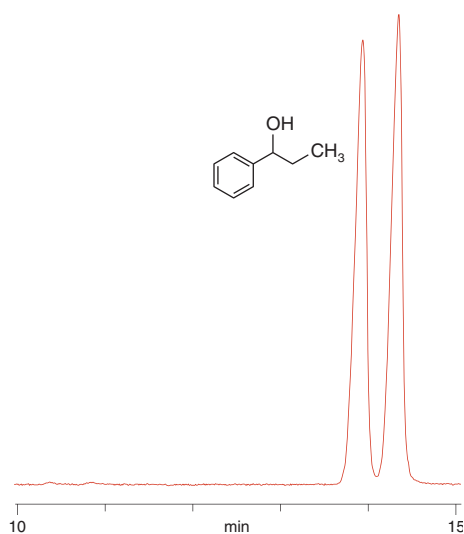
Compounds C₉ – C₁₂



Enantiomer separation of 1-phenylpropan-1-ol

MN Appl. No. 212320

Column: FS-LIPODEX® G,
25 m x 0.25 mm ID,
REF723379.25,
max. temperature 220/240 °C
Carrier gas: 100 kPa H₂
Temperature: 50 °C (5 min) $\xrightarrow{10\text{ °C/min}}$ 80 °C
(5 min)
Detector: FID
C₉H₁₂O



Courtesy of Mrs. Vermeeren, AK Prof. Gais,
Inst. für Org. Chemie, RWTH Aachen, Germany

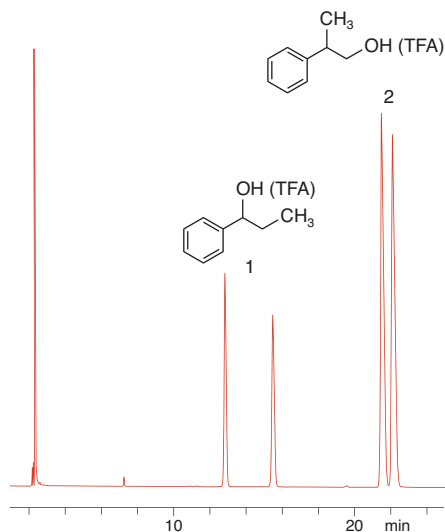
Enantiomer separation of phenylpropanols (TFA)

MN Appl. No. 212970

Column: FS-HYDRODEX γ-TBDAC,
50 m x 0.25 mm ID,
REF723378.50,
max. temperature 220/240 °C
Injection: 1 µl, split 50 ml/min
Carrier gas: 1.2 bar H₂
Temperature: 100 °C
Detector: FID

Peaks: (C₉H₁₂O)

1. 1-Phenylpropan-1-ol (TFA)
2. 2-Phenylpropan-1-ol (TFA)



Chiral separations

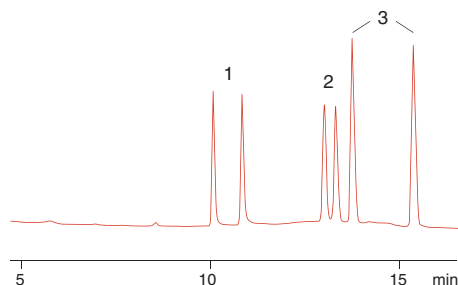
Enantiomer separation of amphetamine, ephedrine and norephedrine (N,O-TFA)

MN Appl. No. 202310

Column: LIPODEX® D,
50 m x 0.25 mm ID,
REF 723366.50,
max. temperature 200/220 °C
Carrier gas: H₂
Temperature: 150 °C
Detector: FID

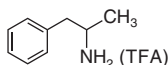
Peaks:

1. Amphetamine
2. Ephedrine
3. Norephedrine

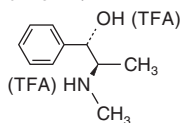


Courtesy of Prof. W.A. König, Hamburg, Germany

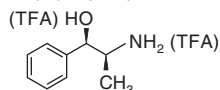
Amphetamine (C₉H₁₃N)



Ephedrine (C₁₀H₁₅NO)



Norephedrine (C₉H₁₃NO)



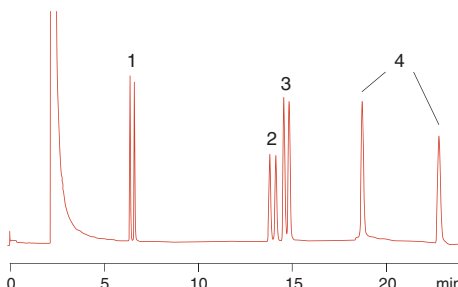
Enantiomer separation of sympathomimetic amines (N,O-TFA)

MN Appl. No. 202320

Column: LIPODEX® D,
50 m x 0.25 mm ID,
REF 723366.50,
max. temperature 200/220 °C
Carrier gas: 1 bar H₂
Temperature: 175 °C
Detector: FID

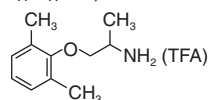
Peaks:

1. Amphetamine
2. Mexiletine
3. Pholedrine
4. Tranlylcypromine

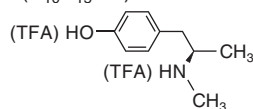


W.A. König, S. Lutz, G. Wenz, E. von der Bey,
HRC 11 (1988) 506 – 509

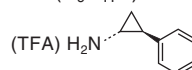
Mexiletine (C₁₁H₁₇NO)



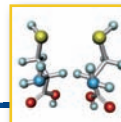
Pholedrine (C₁₀H₁₅NO)



Tranlylcypromine (C₉H₁₁N)



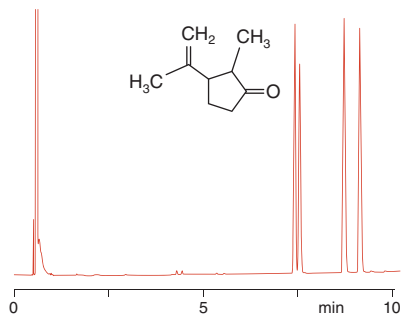
Compounds C₉ – C₁₁



Enantiomer separation of 3-isopropenyl-2-methylcyclopentanone

MN Appl. No. 202070

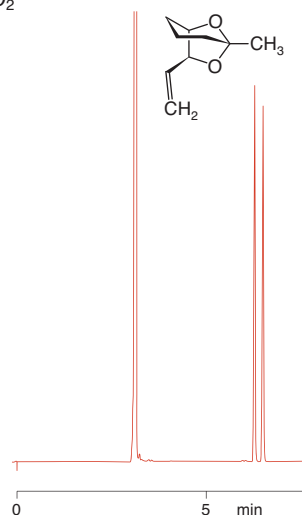
Column: FS-LIPODEX® E,
25 m x 0.25 mm ID,
REF 723368.25,
max. temperature 200/220
°C
Injection: 0.2 µl, split 80 ml/min
Carrier gas: 50 kPa H₂ (1.2 ml/min)
Temperature: 70 °C (3 min) $\xrightarrow{2\text{ °C/min}}$ 110
°C
Detector: FID 250 °C
C₉H₁₄O



Enantiomer separation of vinyl-endo-brevicomin

MN Appl. No. 202130

Column: FS-LIPODEX® E,
50 m x 0.25 mm ID,
REF 723368.50,
max. temperature 200/220 °C
Injection: 0.5 µl, split 180 ml/min
Carrier gas: 100 kPa H₂ (1.5 ml/min)
Temperature: 120 °C
Detector: FID 250 °C
C₉H₁₄O₂



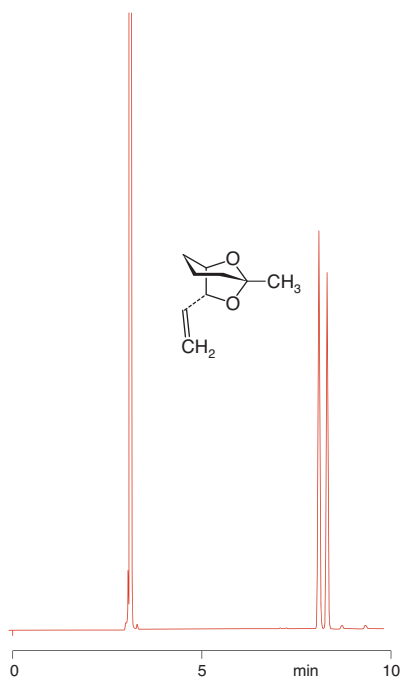
Vinyl-endo-brevicomin can also be separated on HYDRODEX β-PM (see application 202120 at www.mn-net.com).

Chiral separations

Enantiomer separation of vinyl-exo-brevicomine

MN Appl. No. 202150

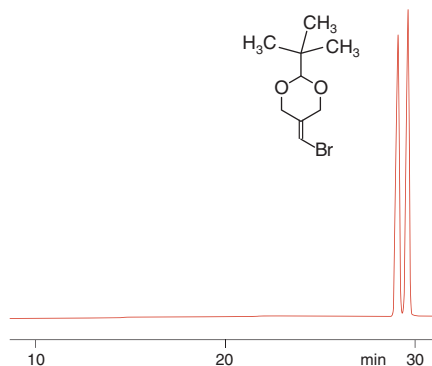
Column: FS-LIPODEX® E,
50 m x 0.25 mm ID,
REF 723368.50,
max. temperature 200/220 °C
Injection: 0.5 µl, split 180 ml/min
Carrier gas: 100 kPa H₂ (1.5 ml/min)
Temperature: 105 °C
Detector: FID 250 °C
C₉H₁₄O₂



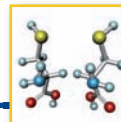
Enantiomer separation of 2-*t*-butyl-5-bromomethylene 1,3-dioxane

MN Appl. No. 202230

Column: FS-LIPODEX® A,
50 m x 0.25 mm ID,
REF 723360.50,
max. temperature 200/220 °C
Injection: 0.5 µl (2% in CH₂Cl₂)
split 92 ml/min
Carrier gas: 130 kPa H₂ (2.0 ml/min)
Temperature: 105 °C
Detector: FID 250 °C
C₉H₁₅BrO₂

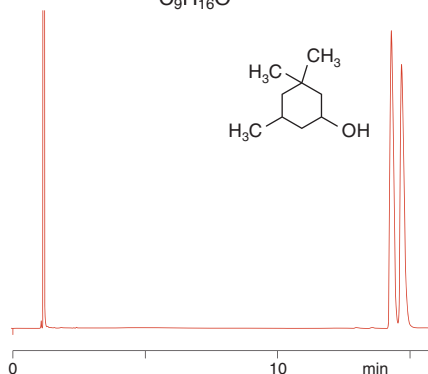


Vinyl-exo-brevicomine can also be separated on HYDRODEX β-PM (see application 202140 at www.mn-net.com).



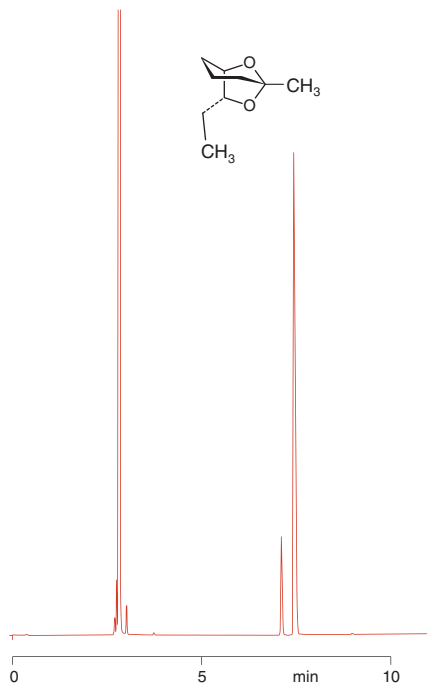
Enantiomer separation of *cis*-3,3,5-trimethylcyclohexanol *MN Appl. No. 201970*

Column: FS-LIPODEX® E,
25 m x 0.25 mm ID,
REF 723368.25,
max. temperature 200/220 °C
Injection: 0.1 µl (1 % in CH₂Cl₂)
split 125 ml/min
Carrier gas: 60 kPa H₂ (1.6 ml/min)
Temperature: 80 °C
Detector: FID 250 °C
C₉H₁₆O



Enantiomer separation of (-)-exo-brevicomin *MN Appl. No. 202160*

Column: FS-HYDRODEX β-PM,
50 m x 0.25 mm ID,
REF 723370.50,
max. temperature 230/250 °C
Injection: 0.1 µl (1 % in CH₂Cl₂)
split 150 ml/min
Carrier gas: 120 kPa H₂ (1.7 ml/min)
Temperature: 120 °C
Detector: FID 250 °C
C₉H₁₆O₂



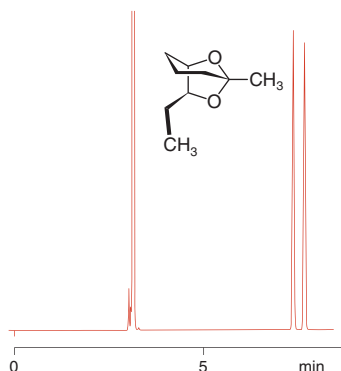
(-)-exo-brevicomin can also be separated on
LIPODEX® E
(see application 202170 at www.mn-net.com).

Chiral separations

Enantiomer separation of endo-brevi-comin

MN Appl. No. 202180

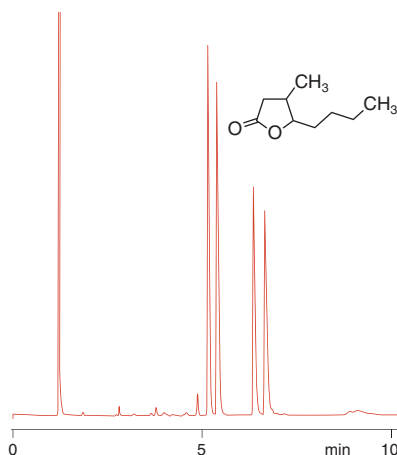
Column: FS-LIPODEX® E,
50 m x 0.25 mm ID,
REF 723368.50,
max. temperature 200/220 °C
Injection: 0.5 µl, split 180 ml/min
Carrier gas: 100 kPa H₂ (1.5 ml/min)
Temperature: 110 °C
Detector: FID 250 °C
C₉H₁₆O₂



Enantiomer analysis of 4-butyl-3-methylbutyrolactone (Whisky lactone)

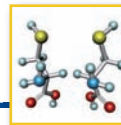
MN Appl. No. 202862

Column: FS-LIPODEX® E,
25 m x 0.25 mm ID,
REF 723368.25,
max. temperature 200/220 °C
Injection: 0.1 µl (1 % in CH₂Cl₂)
split 95 ml/min
Carrier gas: 60 kPa H₂ (1.7 ml/min)
Temperature: 145 °C
Detector: FID 250 °C
C₉H₁₆O₂



Whisky lactone can also be separated on HYDRODEX β-PM (application 202882) or LIPODEX® A (application 202852). See our application database at www.mn-net.com.

Compounds C₉ – C₁₁



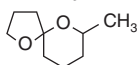
Enantiomer separation of spiroacetals

MN Appl. No. 202200

Column: LIPODEX® A, 50 m x 0.25 mm ID, REF 723360.50,
max. temperature 200/ 220 °C
Temperature: 80 °C
Carrier gas: H₂
Detector: FID

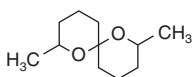
Peaks:

1. 7-Methyl-1,6-dioxaspiro[4,5]decane



C₉H₁₆O₂

2. 2,8-Dimethyl-1,7-dioxaspiro[5,5]undecane



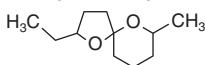
C₁₁H₂₀O₂

3. 1,7-Dioxaspiro[5,5]undecane

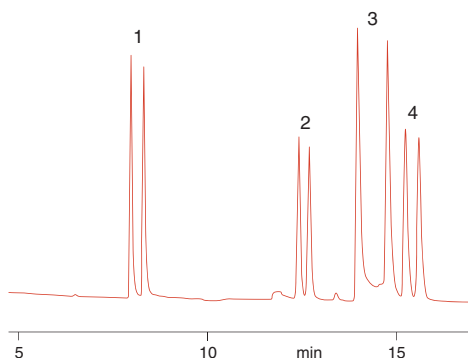


C₉H₁₆O₂

4. 2-Ethyl-7-methyl-1,6-dioxaspiro[4,5]decane



C₁₁H₂₀O₂

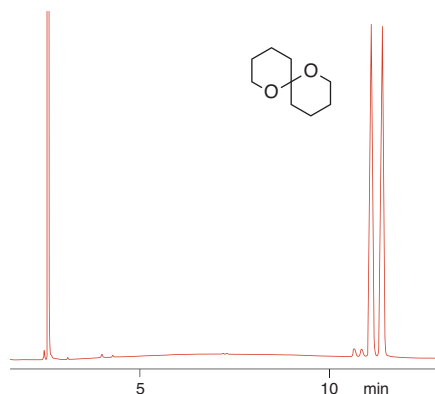


Courtesy of Prof. W.A. König, Hamburg, Germany

Enantiomer separation of 1,7-dioxaspiro[5,5]undecane

MN Appl. No. 202220

Column: FS-LIPODEX® A,
50 m x 0.25 mm ID,
REF 723360.50,
max. temperature 200/220 °C
Injection: 0.5 µl (1 % in CH₂Cl₂)
split 320 ml/min
Carrier gas: 120 kPa H₂ (2.2 ml/min)
Temperature: 95 °C
Detector: FID 240 °C
C₉H₁₆O₂

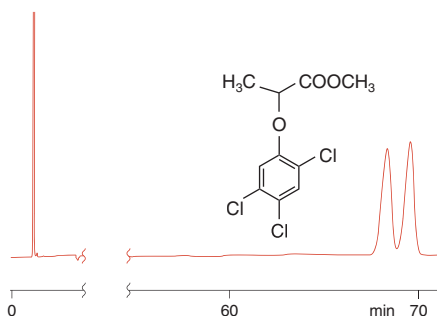


Chiral separations

Enantiomers separation of fenoprop methyl ester

MN Appl. No. 202552

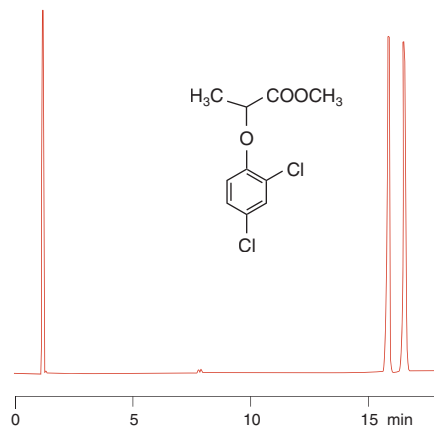
Column: FS-HYDRODEX β -3P,
25 m x 0.25 mm ID,
REF 723358.25, max. tem-
perature 230/250 °C
Injection: 1 μ l (1 % in CH₂Cl₂)
split 130 ml/min
Carrier gas: 60 kPa H₂ (1.9 ml/min)
Temperature: 140 °C
Detector: FID 250 °C
C₁₀H₉Cl₃O₃



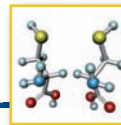
Enantiomer separation of dichlorprop methyl ester

MN Appl. No. 202542

Column: FS-HYDRODEX β -3P,
25 m x 0.25 mm ID,
REF 723358.25,
max. temperature 230/250 °C
Injection: 0.1 μ l (1 % in CH₂Cl₂)
split 130 ml/min
Carrier gas: 60 kPa H₂ (1.9 ml/min)
Temperature: 160 °C
Detector: FID 250 °C
C₁₀H₁₀Cl₂O₃



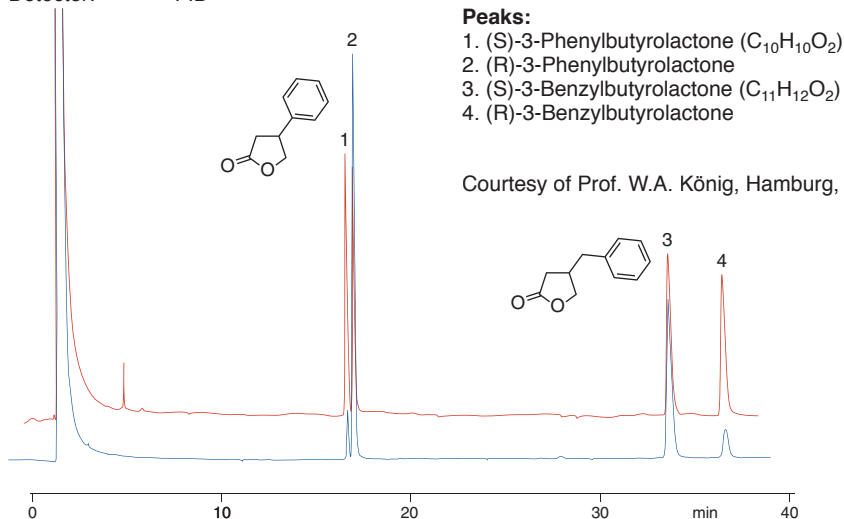
Compounds C₁₀ – C₁₁



Enantiomer composition of 3-phenyl- and 3-benzylbutyrolactone

MN Appl. No. 202932

Column: LIPODEX® B, 50 m x 0.25 mm ID, REF 723362.50,
max. temperature 200/220 °C
Carrier gas: H₂
Temperature: 175 °C
Detector: FID

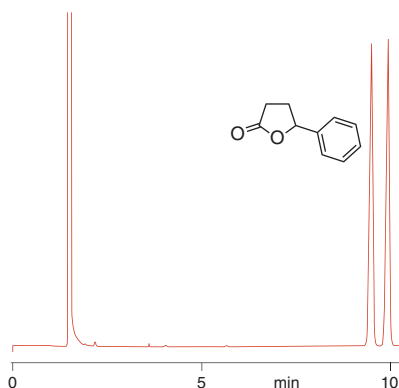


Enantiomer analysis of γ -phenylbutyrolactone

MN Appl. No. 202942

Column: FS-LIPODEX® E, 25 m x 0.25
mm ID, REF 723368.25,
max. temperature 200/220 °C
Injection: 0.5 μ l, split 1:100
Carrier gas: 60 kPa H₂
Temperature: 170 °C
Detector: FID 250 °C
C₁₀H₁₀O₂

γ -phenylbutyrolactone can also be separated on
HYDRODEX β -PM
(see application 202952 at www.mn-net.com).

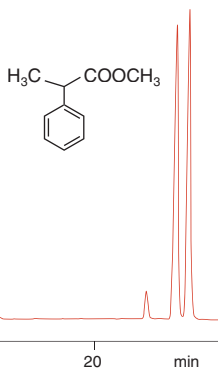


Chiral separations

Enantiomer separation of 2-phenylpropionic acid (hydratropic acid) methyl ester

MN Appl. No. 202572

Column: FS-LIPODEX® E,
25 m x 0.25 mm ID,
REF 723368.25,
max. temperature
200/220 °C
Injection: 0.1 µl (0.1 % in CH₂Cl₂)
split 100 ml/min
Carrier gas: 60 kPa H₂
Temperature: 80 °C
Detector: FID 250 °C
C₁₀H₁₂O₂



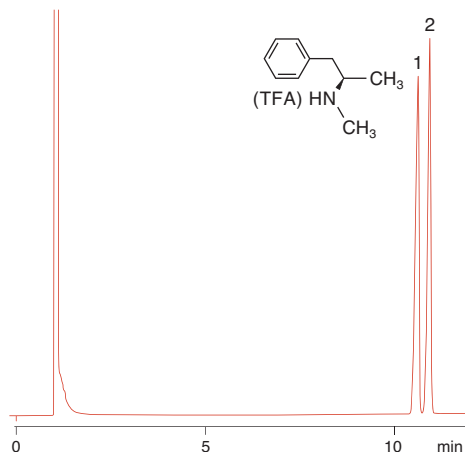
Enantiomer separation of desoxyephedrine (TFA)

MN Appl. No. 202330

Column: FS-LIPODEX® E,
25 m x 0.25 mm ID,
REF 723368.25,
max. temperature 200/220 °C
Injection: 0.5 µl (1 % in CH₂Cl₂)
split 70 ml/min
Carrier gas: 60 kPa H₂ (1.3 ml/min)
Temperature: 130 °C
Detector: FID 250 °C

Peaks:

1. D-Desoxyephedrine (TFA) C₁₀H₁₅N
2. L-Desoxyephedrine (TFA)



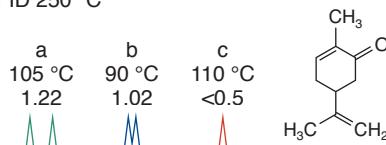
Enantiomer separation of carvone

MN Appl. No. 211500

Columns: all 25 m x 0.25 mm ID: a) FS-HYDRODEX β-6TBDM (REF 723381.25),
b) FS-LIPODEX® E (REF 723368.25),
c) FS-HYDRODEX β-PM (REF 723370.25)

Injection: 1 µl, split, 250 °C
Carrier gas: 1.1 ml/min He
Detector: FID 250 °C
C₁₀H₁₄O₂

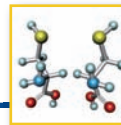
Temperature: 105 °C 90 °C 110 °C
Resolution: 1.22 1.02 <0.5



Courtesy of H. Casabianca, CNRS
Service Central d'Analyse, Vernaison,
France

Carvone can also be separated on
LIPODEX® A (see applications 202110
and 211940 at www.mn-net.com).

Compounds C₁₀ – C₁₂



Chiral constituents of peppermint oil

MN Appl. No. 250410

Column: FS-LIPODEX® G, 25 m x 0.25 mm ID, max. temp. 220/240 °C, REF 723379.25

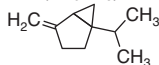
Carrier gas: 50 kPa H₂

Temperature: 75 °C, isothermal

Detector: FID

Peaks:

1. (+)-*trans*-Sabinene hydrate (C₁₀H₁₆)



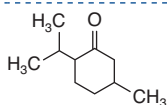
2. (+)-Menthone

3. (+)-Isomenthone

4. (–)-Menthone

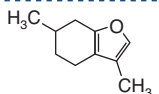
5. (–)-Isomenthone

(C₁₀H₁₈O)



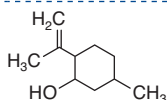
6. (+)-Menthofuran

(C₁₀H₁₄O)



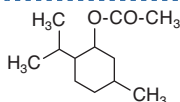
7. (–)-Isopulegol

(C₁₀H₁₈O)



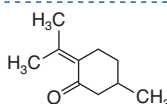
8. (–)-Menthyl acetate

(C₁₂H₂₂O₂)



9. (+)-Pulegone

(C₁₀H₁₆O)



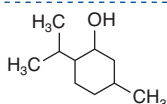
10. (+)-Neomenthol

11. (–)-Neomenthol

12. (+)-Neoisomenthol

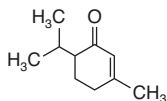
13. (+)-Menthol

14. (–)-Neoisomenthol



15. (+)-Piperitone

(C₁₀H₁₆O)



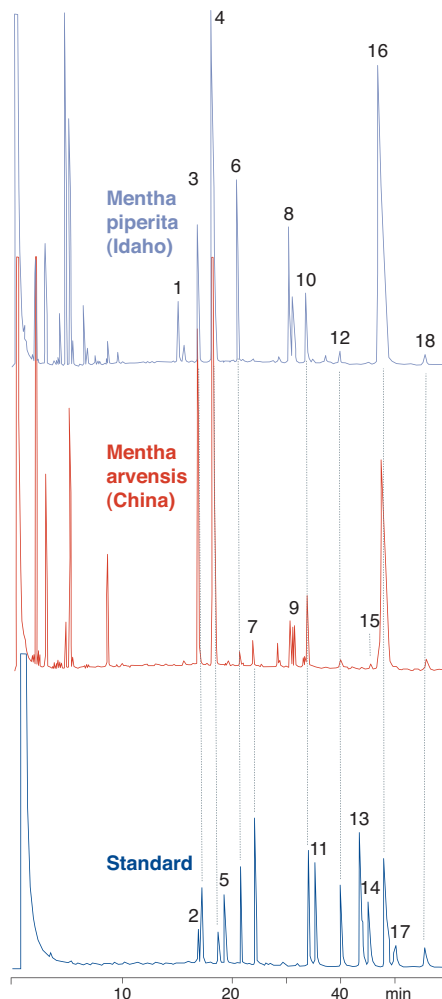
16. (–)-Menthol

17. (+)-Isomenthol

18. (–)-Isomenthol

(C₁₀H₂₀O)

structure see above



W. A. König et al., High Resol. Chromatogr. **20** (1997) 55 – 61

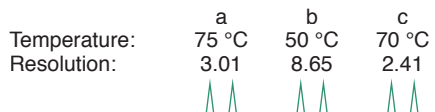
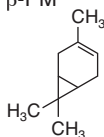
Chiral separations

Enantiomer separation of Δ^3 -carene MN Appl. No. 211530

Columns: all 25 m x 0.25 mm ID:
a) FS-HYDRODEX β -6TBDM
(REF 723381.25),
b) FS-LIPODEX® E
(REF 723368.25),
c) FS-HYDRODEX β -PM
(REF 723370.25)

Injection: 1 μ l, split, 250 °C
Carrier gas: 1.1 ml/min He
Detector: FID 250 °C

C₁₀H₁₆



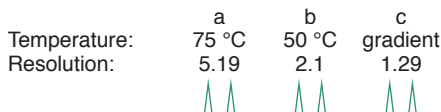
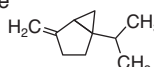
Courtesy of H. Casabianca, CNRS Service
Central d'Analyse, Vernaison, France

Enantiomer separation of sabinene MN Appl. No. 211540

Columns: all 25 m x 0.25 mm ID:
a) FS-HYDRODEX β -6TBDM
(REF 723381.25),
b) FS-LIPODEX® E
(REF 723368.25),
c) FS-HYDRODEX β -PM
(REF 723370.25)

Injection: 1 μ l, split, 250 °C
Carrier gas: 1.1 ml/min He
Detector: FID 250 °C

C₁₀H₁₆



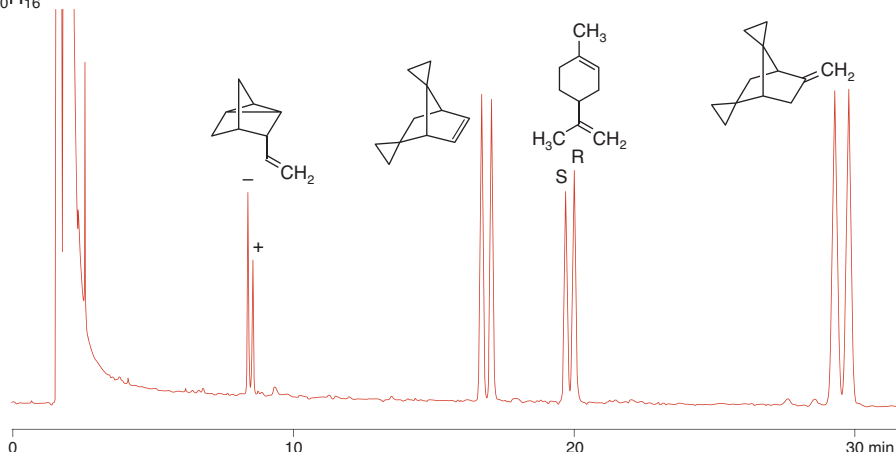
Courtesy of H. Casabianca, CNRS Service
Central d'Analyse, Vernaison, France

Enantiomer separation of cyclic olefins including limonene MN Appl. No. 201830

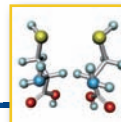
Column: LIPODEX® C, 50 m x 0.25 mm ID, REF 723364.50,
max. temperature 200/220 °C

Carrier gas: 1 bar H₂
Temperature: 70 °C
Detector: FID

C₁₀H₁₆



W.A. König et al., HRC 12 (1989) 35 – 39

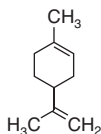


Enantiomer separation of limonene

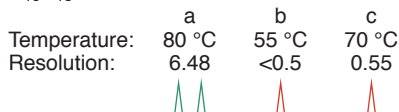
MN Appl. No. 211550

Columns: all 25 m x 0.25 mm ID:
a) FS-HYDRODEX β -6TBDM
(REF 723381.25),
b) FS-LIPODEX® E
(REF 723368.25),
c) FS-HYDRODEX β -PM
(REF 723370.25)

Injection: 1 μ l, split, 250 °C
Carrier gas: 1.1 ml/min He
Detector: FID 250 °C



C₁₀H₁₆



Courtesy of H. Casabianca, CNRS Service
Central d'Analyse, Vernaison, France

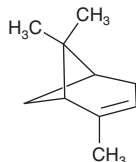
Limonene can also be separated on
HYDRODEX β -3P (see application 201850 at
www.mn-net.com).

Enantiomer separation of α -pinene

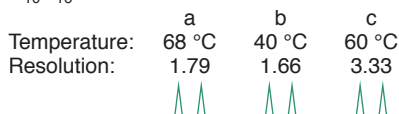
MN Appl. No. 211510

Columns: all 25 m x 0.25 mm ID:
a) FS-HYDRODEX β -6TBDM
(REF 723381.25),
b) FS-LIPODEX® E
(REF 723368.25),
c) FS-HYDRODEX β -PM
(REF 723370.25)

Injection: 1 μ l, split, 250 °C
Carrier gas: 1.1 ml/min He
Detector: FID 250 °C



C₁₀H₁₆



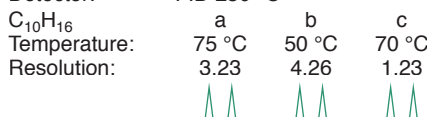
Courtesy of H. Casabianca, CNRS Service
Central d'Analyse, Vernaison, France

Enantiomer separation of β -pinene

MN Appl. No. 211520

Columns: all 25 m x 0.25 mm ID:
a) FS-HYDRODEX β -6TBDM
(REF 723381.25),
b) FS-LIPODEX® E
(REF 723368.25),
c) FS-HYDRODEX β -PM
(REF 723370.25)

Injection: 1 μ l, split, 250 °C
Carrier gas: 1.1 ml/min He
Detector: FID 250 °C



Courtesy of H. Casabianca, CNRS Service
Central d'Analyse, Vernaison, France

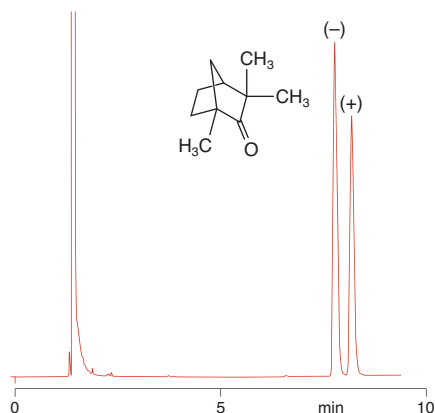
Enantiomer separation of fenchone

MN Appl. No. 202090

Column: FS-LIPODEX® E,
25 m x 0.25 mm ID,
REF 723368.25,
max. temperature 200/220 °C

Injection: 0.5 μ l, split 1:100
Carrier gas: 60 kPa H₂
Temperature: 90 °C
Detector: FID 250 °C

C₁₀H₁₆O

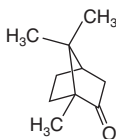


Chiral separations

Enantiomer separation of camphor

MN Appl. No. 211560

Columns: all 25 m x 0.25 mm ID:
 a) FS-HYDRODEX β -6TBDM
 (REF 723381.25),
 b) FS-LIPODEX® E
 (REF 723368.25),
 c) FS-HYDRODEX β -PM
 (REF 723370.25)
 Injection: 1 μ l, split, 250 °C
 Carrier gas: 1.1 ml/min He
 Detector: FID 250 °C
 $C_{10}H_{16}O$



	a	b	c
Temperature:	100 °C	90 °C	100 °C
Resolution:	2.11	6.66	<0.5



Courtesy of H. Casabianca, CNRS Service
 Central d'Analyse, Vernaison, France

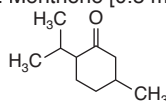
Separation of essential oils

MN Appl. No. 212980 / 212990 / 213000

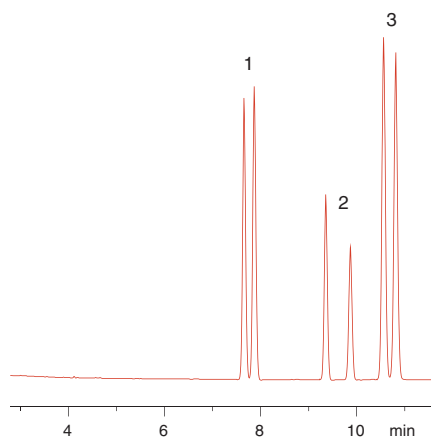
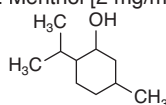
Column: FS-HYDRODEX γ -TBDAc, 50
 m x 0.25 mm ID, max. temp.
 220/240 °C, REF 723387.50
 Injector: 220 °C
 Carrier gas: 1.2 bar H_2
 Temperature: 125 °C
 Detector: FID 220 °C

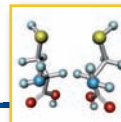
Peaks:

1. Fenchone [1.5 mg/ml] ($C_{10}H_{16}O$)
2. Menthone [0.5 mg/ml] ($C_{10}H_{18}O$)



3. Menthol [2 mg/ml] ($C_{10}H_{20}O$)

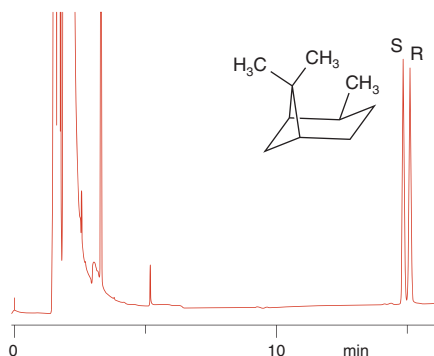




Enantiomer separation of *cis*-pinane

MN Appl. No. 201810

Column: LIPODEX® C,
50 m x 0.25 mm ID,
REF 723364.50,
max. temperature 200/220 °C
Carrier gas: H₂
Temperature: 70 °C
Detector: FID
C₁₀H₁₈



Courtesy of Prof. W.A. König, Hamburg,
Germany

Enantiomer separation of terpinen-4-ol

MN Appl. No. 211630

Columns: all 25 m x 0.25 mm ID:
a) FS-HYDRODEX β-6TBDM
(REF 723381.25),
b) FS-LIPODEX® E
(REF 723368.25),
c) FS-HYDRODEX β-PM
(REF 723370.25)
Injection: 1 µl, split, 250 °C
Carrier gas: 1.1 ml/min He
Detector: FID 250 °C
C₁₀H₁₈O

	a	b	c
Temperature:	110 °C	75 °C	70 °C
Resolution:	2.1	<0.5	0.65

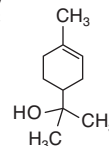


Courtesy of H. Casabianca, CNRS Service
Central d'Analyse, Vernaison, France

Enantiomer separation of α-terpineol

MN Appl. No. 211640

Columns: all 25 m x 0.25 mm ID:
a) FS-HYDRODEX β-6TBDM
(REF 723381.25),
b) FS-LIPODEX® E
(REF 723368.25),
c) FS-HYDRODEX β-PM
(REF 723370.25)
Injection: 1 µl, split, 250 °C
Carrier gas: 1.1 ml/min He
Detector: FID 250 °C
C₁₀H₁₈O



	a	b	c
Temperature:	115 °C	80 °C	110 °C
Resolution:	4.29	1.26	1.75

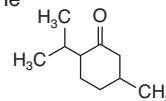


Courtesy of H. Casabianca, CNRS Service
Central d'Analyse, Vernaison, France

Enantiomer separation of menthone

MN Appl. No. 211600

Columns: all 25 m x 0.25 mm ID:
a) FS-HYDRODEX β-6TBDM
(REF 723381.25),
b) FS-LIPODEX® E
(REF 723368.25),
c) FS-HYDRODEX β-PM
(REF 723370.25)
Injection: 1 µl, split, 250 °C
Carrier gas: 1.1 ml/min He
Detector: FID 250 °C
C₁₀H₁₈O



	a	b	c
Temperature:	100 °C	80 °C	90 °C
Resolution:	1.88	11.45	<0.5



Courtesy of H. Casabianca, CNRS Service
Central d'Analyse, Vernaison, France

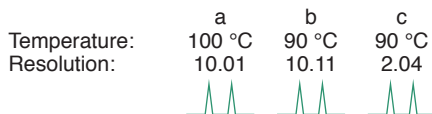
Chiral separations

Enantiomer separation of isomenthone

MN Appl. No. 211610

Columns: all 25 m x 0.25 mm ID:
 a) FS-HYDRODEX β -6TBDM
 (REF 723381.25),
 b) FS-LIPODEX[®] E
 (REF 723368.25),
 c) FS-HYDRODEX β -PM
 (REF 723370.25)

Injection: 1 μ l, split, 250 °C
 Carrier gas: 1.1 ml/min He
 Detector: FID 250 °C
 $C_{10}H_{18}O$

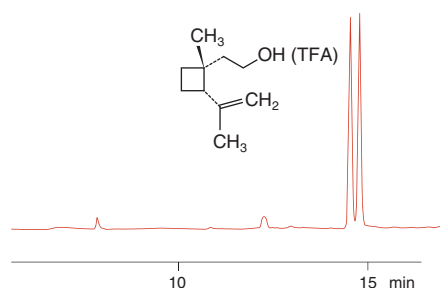


Courtesy of H. Casabianca, CNRS Service Central d'Analyse, Vernaison, France

Enantiomer separation of grandisol (TFA)

MN Appl. No. 202811

Column: LIPODEX[®] C,
 50 m x 0.25 mm ID,
 REF 723364.50,
 max. temperature 200/220 °C
 Carrier gas: 1 bar H_2
 Temperature: 90 °C
 Detector: FID
 $C_{10}H_{18}O$



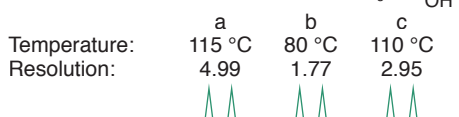
W.A. König et al., HRC 12 (1989) 35 – 39

Enantiomer separation of borneol

MN Appl. No. 211570

Columns: all 25 m x 0.25 mm ID:
 a) FS-HYDRODEX β -6TBDM
 (REF 723381.25),
 b) FS-LIPODEX[®] E
 (REF 723368.25),
 c) FS-HYDRODEX β -PM
 (REF 723370.25)

Injection: 1 μ l, split, 250 °C
 Carrier gas: 1.1 ml/min He
 Detector: FID 250 °C
 $C_{10}H_{18}O$



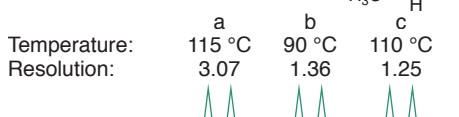
Courtesy of H. Casabianca, CNRS Service Central d'Analyse, Vernaison, France

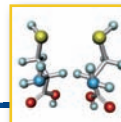
Enantiomer separation of isoborneol

MN Appl. No. 211580

Columns: all 25 m x 0.25 mm ID:
 a) FS-HYDRODEX β -6TBDM
 (REF 723381.25),
 b) FS-LIPODEX[®] E
 (REF 723368.25),
 c) FS-HYDRODEX β -PM
 (REF 723370.25)

Injection: 1 μ l, split, 250 °C
 Carrier gas: 1.1 ml/min He
 Detector: FID 250 °C
 $C_{10}H_{18}O$



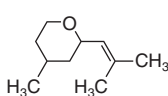
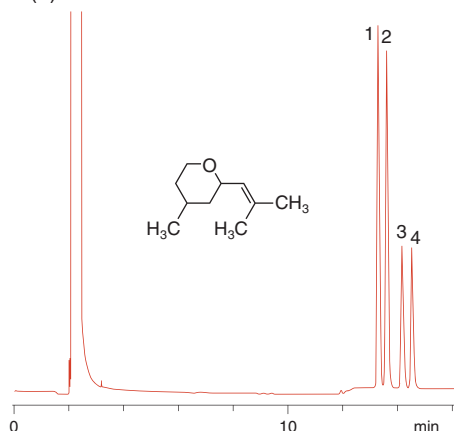


Enantiomer separation of rose oxide

MN Appl. No. 212430

Column: FS-HYDRODEX® β-TBDAC,
50 m x 0.25 mm ID,
REF 723384.50,
max. temperature 220/240 °C
Sample: 0.1 % in hexane
Carrier gas: 1.4 bar H₂
Temperature: 90 °C
Detector: FID

- Peaks:** (C₁₀H₁₈O)
1. (+)-*cis*-Rose oxide
2. (–)-*cis*-Rose oxide
3. (–)-*trans*-Rose oxide
4. (+)-*trans*-Rose oxide



Enantiomer separation of lavandulol

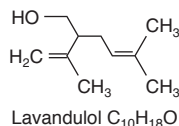
MN Appl. No. 211590

Columns: all 25 m x 0.25 mm ID:
a) FS-HYDRODEX β-6TBDM
(REF 723381.25),
b) FS-LIPODEX® E
(REF 723368.25),
c) FS-HYDRODEX β-PM
(REF 723370.25)
Injection: 1 µl, split, 250 °C
Carrier gas: 1.1 ml/min He
Detector: FID 250 °C

	a	b	c
Temperature:	110 °C	80 °C	100 °C
Resolution:	6	0.95	2.82



Courtesy of H. Casabianca, CNRS Service
Central d'Analyse, Vernaison, France

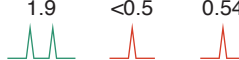


Enantiomer separation of lavandulyl acetate

MN Appl. No. 211740

Columns: all 25 m x 0.25 mm ID:
a) FS-HYDRODEX β-6TBDM
(REF 723381.25),
b) FS-LIPODEX® E
(REF 723368.25),
c) FS-HYDRODEX β-PM
(REF 723370.25)
Injection: 1 µl, split, 250 °C
Carrier gas: 1.1 ml/min He
Detector: FID 250 °C

	a	b	c
Temperature:	105 °C	85 °C	100 °C
Resolution:	1.9	<0.5	0.54



Courtesy of H. Casabianca, CNRS Service
Central d'Analyse, Vernaison, France

MN Appl. No. 211620

Columns: all 25 m x 0.25 mm ID:
a) FS-HYDRODEX β-6TBDM
(REF 723381.25),
b) FS-LIPODEX® E
(REF 723368.25)
Carrier gas: 1.1 ml/min He
Detector: FID 250 °C

	a	b
Temperature:	90 °C	60 °C
Resolution:	1.41	<0.5



Courtesy of H. Casabianca, CNRS Service
Central d'Analyse, Vernaison, France

Chiral separations

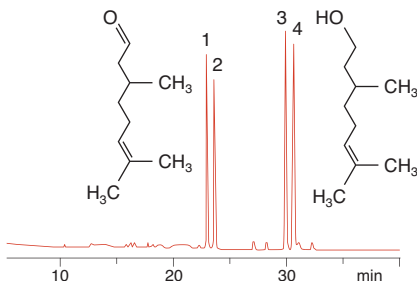
Enantiomer separation of citronellal and citronellol

MN Appl. No. 212440

Column: FS-HYDRODEX® β -TBDAC,
50 m x 0.25 mm ID,
REF 723384.50,
max. temperature 220/240 °C
Sample: 0.1 % in CH₂Cl₂
Carrier gas: H₂, 1.5 bar
Temperature: 100 °C
Detector: FID

Peaks:

1. (R)/(S)-Citronellal (C₁₀H₁₈O)
2. (S)/(R)-Citronellal
3. (S)-Citronellol (C₁₀H₂₀O)
4. (R)-Citronellol



Also see application 211680 at
www.mn-net.com

Enantiomer separation of citronellol (TFA)

MN Appl. No. 211690

Columns: all 25 m x 0.25 mm ID:
a) FS-HYDRODEX β -6TBDM
(REF 723381.25),
b) FS-LIPODEX® E
(REF 723368.25)
Carrier gas: 1.1 ml/min He
Detector: FID 250 °C

	a	b
Temperature:	95 °C	85 °C
Resolution:	0.67	1.75



Courtesy of H. Casabianca, CNRS Service
Central d'Analyse, Vernaion, France

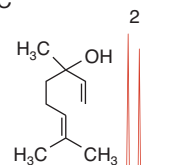
Enantiomer separation of linalool

MN Appl. No. 201920

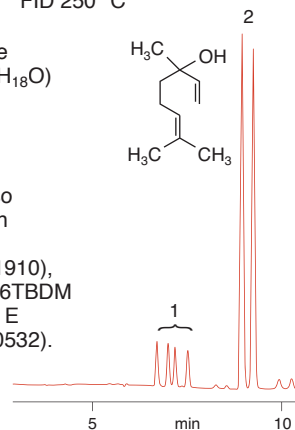
Column: FS-HYDRODEX β -3P,
25 m x 0.25 mm ID,
REF 723358.25,
max. temperature 230/250 °C
Injection: 0.1 μ l (1 % in CH₂Cl₂)
split 130 ml/min
Carrier gas: 60 kPa H₂ (1.9 ml/min)
Temperature: 120 °C
Detector: FID 250 °C

Peaks:

1. Linalool oxide
2. Linalool (C₁₀H₁₈O)



Linalool can also
be separated on
LIPODEX® C
(application 201910),
HXDRODEX β -6TBDM
and LIPODEX® E
(application 210532).
See our
application
database at
www.mn-net.com

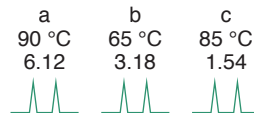


Enantiomer separation of linalool oxide

MN Appl. No. 210533

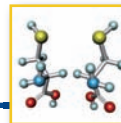
Columns: all 25 m x 0.25 mm ID:
a) FS-HYDRODEX β -6TBDM
(REF 723381.25),
b) FS-LIPODEX® E
(REF 723368.25),
c) FS-HYDRODEX β -PM
(REF 723370.25)
Injection: 1 μ l, split, 250 °C
Carrier gas: 1.1 ml/min He
Detector: FID 250 °C

C ₁₀ H ₁₈ O ₂	a	b	c
Temperature:	90 °C	65 °C	85 °C
Resolution:	6.12	3.18	1.54



Courtesy of H. Casabianca, CNRS Service
Central d'Analyse, Vernaion, France

Compounds C₁₀ – C₁₂



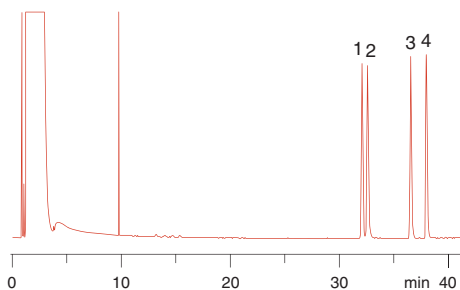
Enantiomer separation of linalool and linalyl acetate

MN Appl. No. 212410

Column: FS-HYDRODEX® β -TBDAC,
25 m x 0.25 mm ID,
REF 723384.25,
max. temperature 220/240 °C
Sample: 0.1 % in CH₂Cl₂
Carrier gas: 0.8 bar H₂
Temperature: 75 °C (20 min) $\xrightarrow{1\text{ °C/min}}$ 110 °C
(5 min)
Detector: FID

Peaks:

1. (*R*)-Linalyl acetate (C₁₂H₂₀O₂)
2. (*S*)-Linalyl acetate
3. (*R*)-Linalool (C₁₀H₁₈O)
4. (*S*)-Linalool



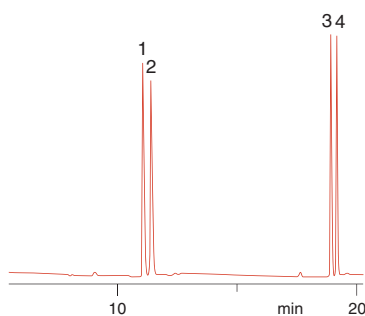
Enantiomer separation of δ -decalactone and δ -dodecalactone

MN Appl. No. 212420

Column: FS-HYDRODEX® β -TBDAC,
25 m x 0.25 mm ID, REF
723384.25,
max. temperature 220/240 °C
Sample: 0.1 % in hexane
Carrier gas: H₂, 0.8 bar
Temperature: 160 °C (12 min) $\xrightarrow{4\text{ °C/min}}$ 200 °C (10 min)
Detector: FID

Peaks:

1. (*S*)- δ -Decalactone (C₁₀H₁₈O₂)
2. (*R*)- δ -Decalactone
3. (*S*)- δ -Dodecalactone (C₁₂H₁₈O₂)
4. (*R*)- δ -Dodecalactone

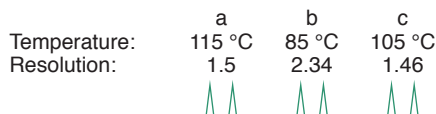


Chiral separations

Enantiomer separation of menthol

MN Appl. No. 211700

Columns: all 25 m x 0.25 mm ID:
 a) FS-HYDRODEX β -6TBDM
 (REF 723381.25),
 b) FS-LIPODEX® E
 (REF 723368.25),
 c) FS-HYDRODEX β -PM
 (REF 723370.25)
 Injection: 1 μ l, split, 250 °C
 Carrier gas: 1.1 ml/min He
 Detector: FID 250 °C
 C₁₀H₂₀O



Courtesy of H. Casabianca, CNRS Service
 Central d'Analyse, Vernaison, France

Menthol can also be separated on LIPODEX® A
 (see application 210090 at www.mn-net.com).

Enantiomer separation of menthol (TFA)

MN Appl. No. 211710

Columns: FS-HYDRODEX β -6TBDM,
 25 m x 0.25 mm ID,
 REF 723381.25
 Injection: 1 μ l, split, 250 °C
 Carrier gas: 1.1 ml/min He
 Temperature: 85 °C
 Detector: FID 250 °C
 Resolution: 2.52

Courtesy of H. Casabianca, CNRS Service
 Central d'Analyse, Vernaison, France

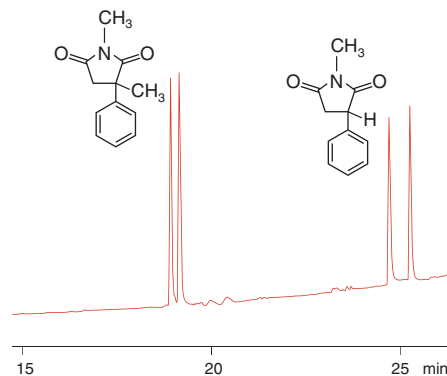
Enantiomer separation of antiepileptic drugs mesuximide and phensuximide

MN Appl. No. 202381

Column: LIPODEX® B,
 50 m x 0.25 mm ID,
 REF 723362.50,
 max. temperature 200/220 °C
 Carrier gas: 1 bar H₂
 Temperature: 135 °C $\xrightarrow{3\text{ °C/min}}$ 200 °C
 Detector: FID

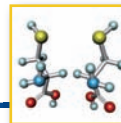
Peaks:

1. Mesuximide (C₁₂H₁₃NO₂)
2. Phensuximide (C₁₁H₁₁NO₂)



W.A. König et al., HRC 11 (1988) 621 – 625

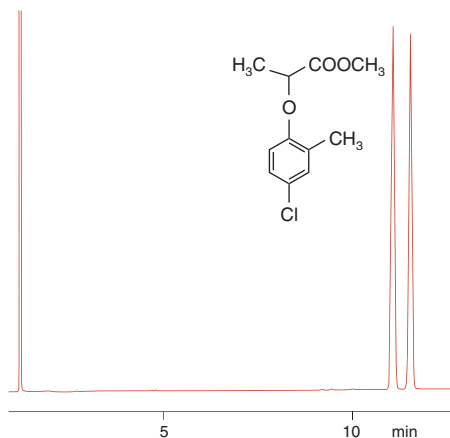
Compounds C₁₀ – C₁₂



Enantiomer separation of mecoprop methyl ester

MN Appl. No. 202562

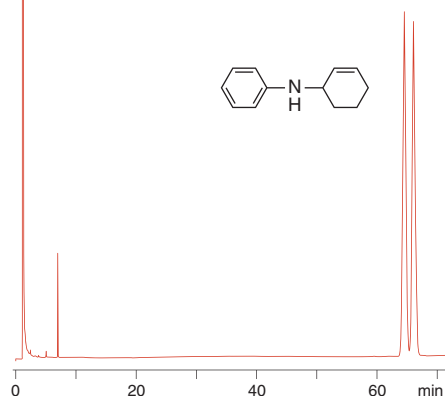
Column: FS-HYDRODEX β-3P,
25 m x 0.25 mm ID,
REF 723358.25,
max. temperature 230/250 °C
Injection: 0.1 µl (1 % in CH₂Cl₂)
split 130 ml/min
Carrier gas: 60 kPa H₂ (1.9 ml/min)
Temperature: 160 °C
Detector: FID 250 °C
C₁₁H₁₃ClO₃



Enantiomer separation of N-phenyl-cyclohex-2-enamine

MN Appl. No. 212250

Column: FS-LIPODEX® E,
25 m x 0.25 mm ID,
REF 723368.25,
max. temperature 200/220 °C
Sample: 10 µl in 3 ml CH₂Cl₂
Carrier gas: 0.6 bar H₂
Temperature: 110 °C
Detector: FID
C₁₂H₁₅N

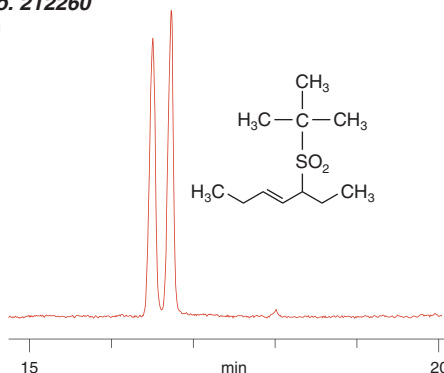


Courtesy of Mrs. Eschmann, Institut für Brennstoffchemie, RWTH Aachen, Germany

Enantiomer separation of 5-(2-methyl-propan-2-sulfonyl)-hept-3-ene

MN Appl. No. 212260

Column: FS-LIPODEX® G, 25 m x 0.25 mm
ID, REF 723379.25,
max. temperature 220/240 °C
Carrier gas: 100 kPa H₂
Temperature: 60 °C (15 min) $\xrightarrow{10\text{ °C/min}}$ 100 °C
(5 min)
Detector: FID
C₁₁H₂₂O₂S



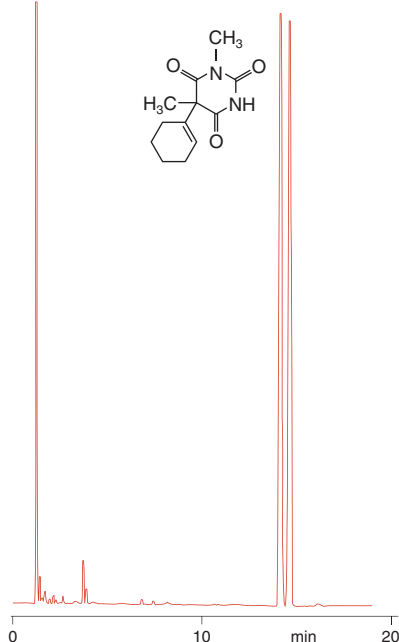
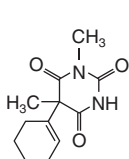
Courtesy of Mrs. Vermeeren, AK Prof. Gais,
Inst. für Org. Chemie, RWTH Aachen, Germany

Chiral separations

Enantiomer separation of hexobarbital (TFA)

MN Appl. No. 202401

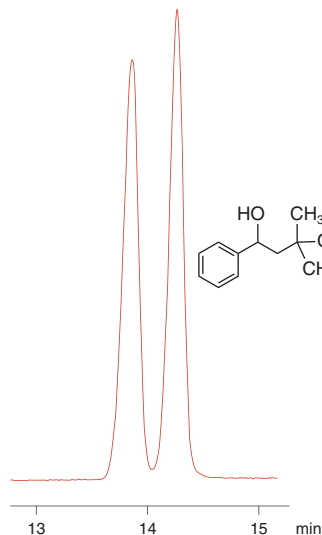
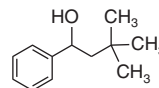
Column: FS-HYDRODEX β -3P,
25 m x 0.25 mm ID,
REF 723358.25,
max. temperature 230/250 °C
Injection: 0.1 μ l, split 130 ml/min
Carrier gas: 60 kPa H₂ (1.9 ml/min)
Temperature: 210 °C
Detector: FID 250 °C
C₁₂H₁₆N₂O₃



Enantiomer separation of 3,3-dimethyl-1-phenyl-butan-1-ol

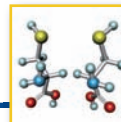
MN Appl. No. 212350

Column: FS-LIPODEX® G,
25 m x 0.25 mm ID,
REF 723379.25,
max. temperature 200/220 °C
Carrier gas: 100 kPa H₂
Temperature: 100 °C (5 min) $\xrightarrow{10\text{ °C/min}}$ 120 °C
(5 min) $\xrightarrow{10\text{ °C/min}}$ 140 °C (5 min)
 $\xrightarrow{10\text{ °C/min}}$ 170 °C (45 min)
Detector: FID
C₁₂H₁₈O



Courtesy of Mrs. Vermeeren, AK Prof. Gais,
Inst. für Org. Chemie, RWTH Aachen, Germany

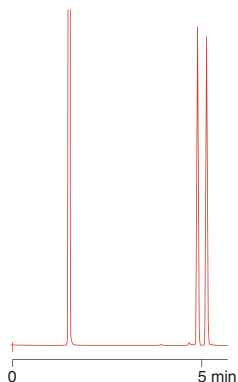
Compounds C₁₂ and greater



Enantiomer separation of menthyl acetate

MN Appl. No. 202010

Column: FS-HYDRODEX β-3P,
25 m x 0.25 mm ID,
REF 723358.25,
max. temperature 230/250 °C
Injection: 0.1 µl (1 % in CH₂Cl₂)
split 130 ml/min
Carrier gas: 60 kPa H₂ (1.9 ml/min)
Temperature: 150 °C
Detector: FID 250 °C
C₁₂H₂₂O₂



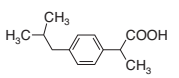
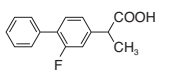
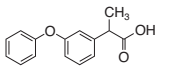
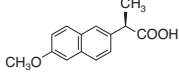
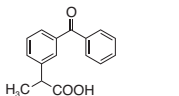
Menthyl acetate can also be separated on HYDRODEX β-PM (see application 202000 at www.mn-net.com).

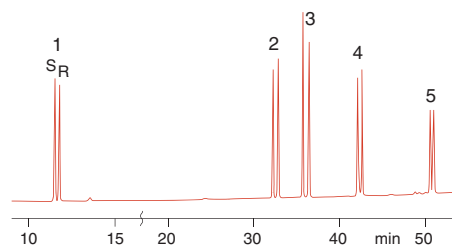
Separation of isomeric antiinflammatory drugs

MN Appl. No. 210150

Column: FS-HYDRODEX β-6TBDM,
25 m x 0.25 mm ID,
max. temperature 250 °C,
REF 723381.25
Carrier gas: H₂
Temperature: 135 °C $\xrightarrow{1\text{ °C/min}}$ 200 °C
Detector: FID

Peaks:

1. Ibuprofen C₁₃H₁₈O₂ 
2. Flurbiprofen C₁₅H₁₃FO₂ 
3. Fenopropfen C₁₅H₁₄O₃ 
4. Naproxen C₁₄H₁₄O₃ 
5. Ketoprofen C₁₆H₁₄O₃ 

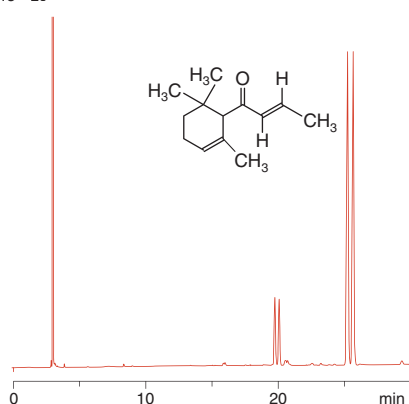


Courtesy of Prof. W.A. König, Hamburg, Germany

Chiral separations

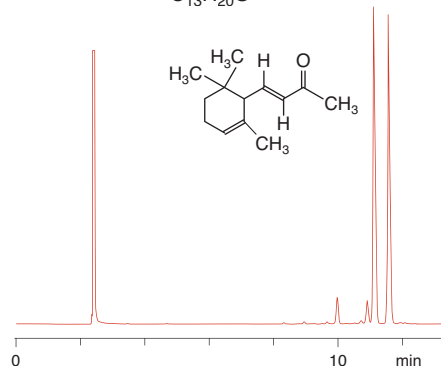
Enantiomer separation of α -damascone *MN Appl. No. 202030*

Column: FS-HYDRODEX β -PM,
50 m x 0.25 mm ID,
REF 723370.50,
max. temperature 230/250 °C
Injection: 0.1 μ l (1 % in CH₂Cl₂)
split 150 ml/min
Carrier gas: 120 kPa H₂ (1.7 ml/min)
Temperature: 130 °C
Detector: FID 250 °C
C₁₃H₂₀O



Enantiomer separation of α -ionone *MN Appl. No. 213010*

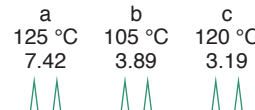
Column: FS-HYDRODEX γ -TBDAC,
50 m x 0.25 mm ID,
REF 723370.50,
max. temperature 230/250 °C
Injection: 0.5 μ l, split 50 ml/min
Carrier gas: 1.2 bar H₂
Temperature: 155 °C
Detector: FID
C₁₃H₂₀O



MN Appl. No. 211770

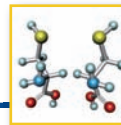
Columns: all 25 m x 0.25 mm ID:
a) FS-HYDRODEX β -6TBDM
(REF 723381.25),
b) FS-LIPODEX® E
(REF 723368.25),
c) FS-HYDRODEX β -PM
(REF 723370.25)
Injection: 1 μ l, split, 250 °C
Carrier gas: 1.1 ml/min He
Detector: FID 250 °C

	a	b	c
Temperature:	125 °C	105 °C	120 °C
Resolution:	7.42	3.89	3.19



Courtesy of H. Casabianca, CNRS Service
Central d'Analyse, Vernaion, France

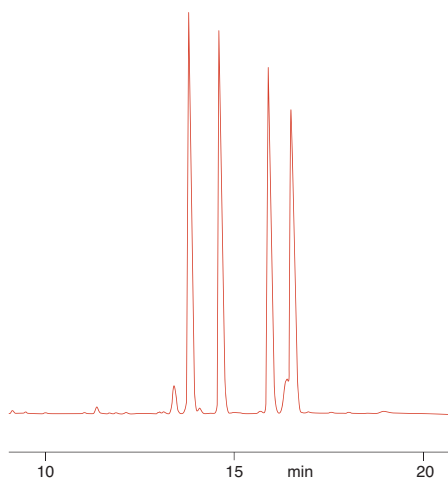
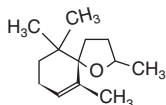
Compounds C₁₃ – C₁₄



Enantiomer separation of theaspiran

MN Appl. No. 202210

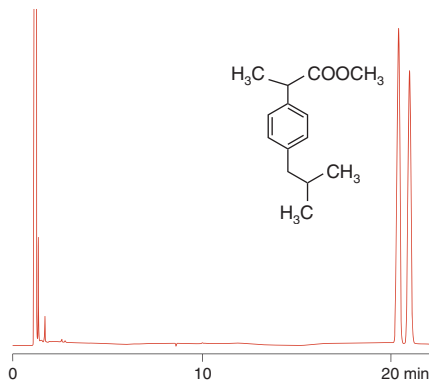
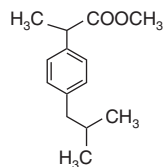
Column: FS-HYDRODEX β-PM,
50 m x 0.25 mm ID,
REF 723370.50,
max. temperature 230/250 °C
Injection: 0.5 µl (1 % in CH₂Cl₂)
split 1:100
Carrier gas: 200 kPa H₂
Temperature: 120 °C
Detector: FID 250 °C
C₁₃H₂₂O



Enantiomer separation of ibuprofen methyl ester

MN Appl. No. 202582

Column: FS-HYDRODEX β-3P,
25 m x 0.25 mm ID,
REF 723358.25,
max. temperature 230/250 °C
Injection: 0.5 µl (1 % in CH₂Cl₂)
split 130 ml/min
Carrier gas: 60 kPa H₂ (1.9 ml/min)
Temperature: 140 °C
Detector: FID 250 °C
C₁₄H₂₀O₂



Chiral separations

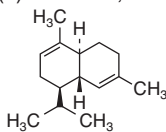
Separation of isomeric sesquiterpenes

MN Appl. No. 250170

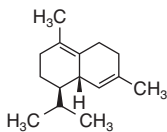
Column: FS-HYDRODEX β -6TBDM,
15 m x 0.25 mm ID,
max. temperature 230/250 °C
Carrier gas: H₂
Temperature: 100 °C $\xrightarrow{2\text{ °C/min}}$ 200 °C
Detector: FID

Peaks: (C₁₅H₂₄)

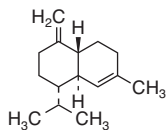
1. (–)- γ -Cadinene
2. (–)- δ -Cadinene
3. (–)- α -Cadinene
4. (+)- γ -Cadinene
5. (+)- α -Cadinene
6. (+)- δ -Cadinene
7. (+)-Cadina-1,4-diene
8. (–)-Cadina-1,4-diene



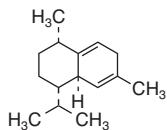
α -Cadinene



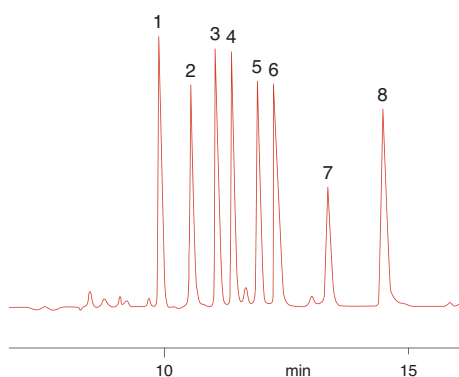
δ -Cadinene



γ -Cadinene



Cadina-1,4-diene



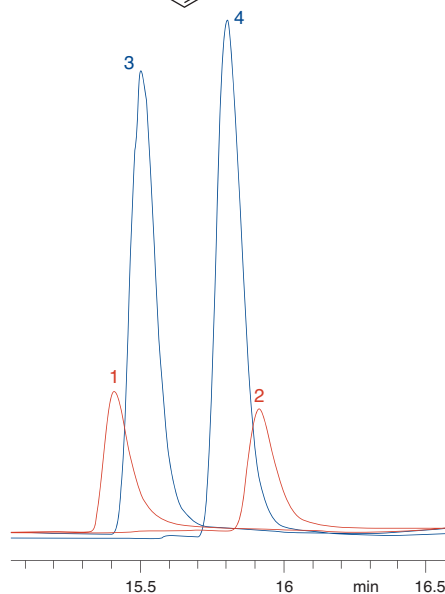
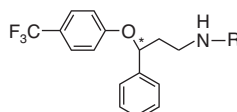
Enantiomer separation of fluoxetine and norfluoxetine

MN Appl. No. 212220

Column: FS-HYDRODEX β -6TBDM,
25 m x 0.25 mm ID,
REF 723358.25,
max. temperature 230/250 °C
Temperature: 170 °C $\xrightarrow{1\text{ °C/min}}$ 200 °C
Carrier gas: 1.1 ml/min H₂
Detector: PND 300 °C

Peaks:

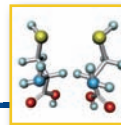
1. (S)-Norfluoxetine (R = H)
2. (R)-Norfluoxetine
3. (S)-Fluoxetine (R = CH₃)
4. (R)-Fluoxetine



S. Ulrich, J. Chromatography B, **783** (2003)
481 – 490

Courtesy of Prof. W.A. König, Hamburg,
Germany

Compounds C₁₅ and greater



Separation of isomeric paracyclophanes

MN Appl. No. 250190

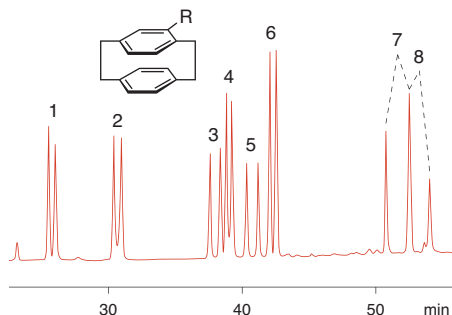
Column: FS-HYDRODEX β -6TBDM, 25 m x 0.25 mm ID, REF 723381.25
max. temperature 230/250 °C

Carrier gas: H₂ 1 °C/min
Temperature: 135 °C $\longrightarrow$ 200 °C

Detector: FID

Peaks:

- | | |
|---|--|
| 1. 4-Methyl-[2.2]-paracyclophane | (C ₁₇ H ₁₈ , R = -CH ₃) |
| 2. 4-Ethynyl-[2.2]-paracyclophane | (C ₁₈ H ₁₆ , R = -C $\equiv$ CH) |
| 3. 4-Allyloxy-[2.2]-paracyclophane | (C ₁₉ H ₂₀ O, R = -O-CH ₂ -CH=CH ₂) |
| 4. 4-Acetyl-[2.2]-paracyclophane | (C ₁₈ H ₁₈ O, R = -CO-CH ₃) |
| 5. 4-Formyl-[2.2]-paracyclophane | (C ₁₇ H ₁₆ O, R = -CHO) |
| 6. 4-Cyano-[2.2]-paracyclophane | (C ₁₇ H ₁₅ N, R = -CN) |
| 7. 4-Hydroxymethyl-[2.2]-paracyclophane | (C ₁₇ H ₁₈ O, R = -CH ₂ OH) |
| 8. 4-Hydroxy-[2.2]-paracyclophane | (C ₁₆ H ₁₆ O, R = -OH) |



Courtesy of Prof. W.A. König, Hamburg, Germany

Chromatogram index

A

Acenaphthene	38–39, 50–53, 65, 66
Acenaphthene- <i>d</i> ₁₀	38–39, 50–51, 112
Acenaphthylene	50–53, 65, 66
Acephate	110
Acetaldehyde	131, 159, 160
Acetaminophen	see Paracetamol
Acetic acid anhydride	117
Acetic acid (C2)	117, 132, 163
Acetic acid cyclohex-2-enyl ester (e)	242
Acetic acid cyclopent-2-enyl ester (e)	231
Acetochlor	112
Acetone	122–126, 128, 129, 131, 132, 159, 160
Acetonitrile	123
Acetophenone	50–51, 126
Acetylcodeine	185, 186
3-Acetyldeoxynivalenol	191
15- <i>o</i> -Acetyl-4-deoxynivalenol	191
Acetyethyltetramethyltetralin	155
Acetylmorphine	185
4-Acetyl-[2.2]-paracyclophane (e)	275
Acetylthebaol	186
Adipic acid	194
AETT	see Acetyethyltetramethyltetralin
Alachlor	38–39, 101, 110
Alanine propyl esters (e)	202
Alaninol (e)	204
Aldols (e)	207
Aldrin	38–39, 74–80, 82, 83, 87, 110, 113
Alkyl bromides (e)	210
Alkyl chlorides (e)	220
1- <i>O</i> -Alkylglycerols (e)	217
Alkyl nitrates	150
Alkyl nitrates (e)	229
4-Allyloxy-[2.2]-paracyclophane (e)	275
Allylphosphonic acid diethyl ester	see Diethyl allylphosphonate
Alodan	72–73, 81, 108
Alphamethrin	see α -Cypermethrin
Alprazolam	184
Ametryn	38–39, 104–106, 110
Amidithion	94–95
4-Aminoazobenzene	144–146
<i>o</i> -Aminoazotoluene	144–146
Aminobenzene	see Aniline
2-Aminobenzoic acid methyl ester	153
4-Aminobiphenyl	144–146
2-Aminobutanol-1 (e)	204, 214
Amino- <i>i</i> -butyric acid isopropyl ester (e)	202
2-Aminoheptane (e)	204
2-Aminohexane (e)	204
2-Amino-6-methylheptane (e)	204
2-Amino-5-methylhexane (e)	204
5-Aminomethylnorbornene (e)	243

2-Amino-4-nitrotoluene	see 5-Nitro- <i>o</i> -toluidine
2-Aminooctane (e)	204
2-Aminopentane (e)	204
1-Aminopropan-2-ol (e)	204
Aminotoluene	see Toluidine
Amitriptyline	190
Ammelide, Ammeline	179
Amphetamine	181, 183, 184
Amphetamine (e)	250
Androstanediol	192
Androstanolone	192
Androsterone	191
Anethole	125, 156
Anhydroarabinitol (e)	219
1,5-Anhydrogalactitol (e)	228
1,5-Anhydroglucitol (e)	228
1,5-Anhydromannitol (e)	228
Anilazine	74, 104
Aniline	50–51, 142, 143
Anisaldehyde	125
<i>o</i> -Anisidine	17, 144–146
Anteismargaric acid methyl ester (C17:0 iso)	171
Anthelmintics	178
Anthracene	38–39, 50–53, 65, 66
Arabinitol (e)	222
Arabinose anomers (e)	228
Arachidic acid methyl ester (C20:0)	126, 165–173
Arachidonic acid methyl ester (C20:4n6)	165–172
Aramite	50–51
Aspartic acid propyl esters (e)	202
Aspon	see Chlordane
Atraton	38–39, 102–104
Atrazine	38–39, 102, 104–106, 110
Azinphos-ethyl	92, 94–95, 98, 110, 113
Azinphos-methyl	92–96, 98, 113
Azobenzene	48, 50–51, 101–104, 112

B

Barbital	183
Behenic acid methyl ester (C22:0)	165–173
Benzaldehyde	153
Benz[a]anthracene	38–39, 50–53, 65, 66
Benzanthrone	134
Benzene	43–45, 59, 119–121, 124, 125, 127, 199
Benzidine	50–51, 144, 145, 146
Benzo[b]fluoranthene	38–39, 50–53, 65, 66
Benzo[k]fluoranthene	38–39, 50–53, 65, 66
Benzoic acid	50–51
Benzoic acid methyl ester	153
Benzo[ghi]perylene	38–39, 50–53, 65, 66
Benzophenone	134
Benzo[a]pyrene	38–39, 50–53, 65, 66
Benzo[h]quinoline	101
Benzothiazole	105

(e) = chiral (enantiomer) separation

Benzoylglycine	194	2-Bromopropionic acid methyl ester (e)	206
Benzyl acetate	155	Bromopropylate	74, 114
Benzyl alcohol	50–51, 134, 153, 154, 156, 158	Butachlor	38–39
Benzylamine	139, 142	<i>i</i> -Butane	196
Benzyl benzoate	153–155	<i>n</i> -Butane	121, 196
Benzyl butyl phthalate	see Butyl benzyl phthalate	Butane-1,2-diol (e)	205, 212
3-Benzylbutyrolactone (e)	257	Butane-1,3-diol (e)	204, 212
Benzyl chloride	45	Butane-1,4-diol	160
Benzyl salicylate	155	Butane-2,3-diol (e)	211
α -BHC	38–39, 72–83, 87, 110, 113, 114	Butanediol dinitrate isomers	150
β -BHC	38–39, 72–83, 87, 110, 113	Butane-1,2,4-triol (e)	213
δ -BHC	38–39, 72–79, 81–83, 87, 110	Butanoic acid	see Butyric and Isobutyric acid
ε -BHC	75, 87	Butanol-1	122, 126–128, 130, 131, 159
γ -BHC (lindane)	38–39, 72–83, 87, 110, 113, 114	Butanol-2	125
BHT	see Butylated hydroxytoluene	<i>t</i> -Butanol	127
Bicyclic lactone (e)	242	Butanone	see Methyl ethyl ketone
Bifenox	114	Butaperazine	183
Bioallethrin	99	1-Butin-3-ol (e)	205
α -Bisabolol	156	Butter: triglycerides	175, 176
Bis(2-chloroethoxy)methane	50–51, 52–53, 133	<i>i</i> -Butyl acetate	122, 127
Bis(2-chloroethyl) ether	50–51, 52–53	<i>n</i> -Butyl acetate	122, 127, 128
Bis(1-chloroisopropyl) ether	133	<i>i</i> -Butyl alcohol	see 2-Methyl-1-propanol
Bis(2-chloroisopropyl) ether	50–51, 52–53, 133	2-Butylamine (e)	213
Bis(diethyl-phosphonates)	136	Butylamine	139
Bis(2-ethylhexyl) adipate	38–39	Butylate	38–39, 102–103
Bis(2-ethylhexyl) phthalate		Butylated hydroxytoluene	154
see Di(2-ethylhexyl) phthalate		<i>n</i> -Butylbenzene	43–45, 59
Bisphenol A and B	193	<i>sec</i> -Butylbenzene	43–45
Bis(phosphonic acid diethyl esters)	136	<i>t</i> -Butylbenzene	43–45
Bitertanol	98, 101	Butyl benzyl phthalate	38–39, 50–53, 133, 135, 136
Bladex	see Cyanazine	2- <i>t</i> -Butyl-5-bromomethylene 1,3-dioxane (e)	252
Bolstar	see Sulprofos	γ -Butylbutyrolactone	see γ -Octalactone
Borneol	156	1- <i>t</i> -Butyl-2-carboxymethyl aziridine (e)	245
Borneol (e)	264	<i>n</i> -Butylcyanohydrin (e)	217
Bornyl acetate	153, 155–157	4- <i>t</i> -Butylcyclohexyl acetate	155
endo-Brevicomine (e)	254	Butyldiglycol	122
(–)-exo-Brevicomine (e)	253	Butyldiglycol acetate	122
Bromacil	38–39, 102–103, 107, 110	4- <i>t</i> -Butyl-2,6-dimethyl-3,5-dinitroacetophenone	
Bromoacetic acid methyl ester	116	see Musk ketone	
Bromobenzene	43–46, 59	1- <i>t</i> -Butyl-3,5-dimethyl-2,4,6-trinitrobenzene	
2-Bromobutane (e)	210	see Musk xylene	
Bromochloroacetic acid methyl ester	116	2- <i>sec</i> -Butyl-4,6-dinitrophenol	see Dinoseb
Bromochloromethane	41, 43–45	<i>t</i> -Butyl ethyl ether	123
4-Bromo-2-chlorophenol	64	1-O-Butylglycerol (e)	217
Bromocyclen	72–73, 75–77, 80	Butyl glycol	122, 127
Bromodichloroacetic acid methyl ester	116	Butyl glycol acetate	122
Bromodichloromethane	40, 43–45	<i>i</i> -Butyl lactate (e)	236
Bromoform	see Tribromomethane	4-Butyl-3-methylbutyrolactone (e)	254
2-Bromohexane (e)	210	<i>i</i> -Butyl methyl ketone	127, 128
Bromomethane	45	Butyl-triethyl-tin	147
2-Bromopentane (e)	210, 219	δ -Butylvalerolactone	see δ -Nonalactone
3-Bromophenol	64	Butyric acid (C4)	132, 158, 163
4-Bromophenoxybenzene	133	Butyric acid methyl ester	167–170
2-Bromo-1-phenylethanol (e)	238	Butyrophenone	126
4-Bromophenyl phenyl ether	50–53		
Bromophos-ethyl	94–95, 113, 114		
Bromophos-methyl	98, 113		

Chromatogram index

C

Cadina-1,4-diene (e)	274	2-Chloroheptane (e)	220
Cadinene (e)	274	2-Chlorohexane (e)	220
Caffeine	183, 185, 186	Chloromethane	45
Camphene	156	Chloromethylaniline isomers	143
Camphor	156, 157	5-Chloro-2-methylbenzylamine	137
Camphor (e)	262	2-Chloromethyloxirane see Epichlorohydrin	
Cannabichromene	186	4-Chloro-3-methylphenol	50–53, 60–64
Cannabinol	186	4-Chloro-2-methylphenoxyacetic acid see MCPA	
Capric acid methyl ester (C10:0)	18, 142, 151, 164–172	4-(4-Chloro-2-methylphenoxy)butanoic acid see MCPB	
Caproic acid (C6:0)	163	1-Chloro-2-methylpropane	42
Caproic acid methyl ester	158, 165–169	1-Chloronaphthalene	48
Caprylic acid (C8:0)	158	2-Chloronaphthalene	47, 50–53
Caprylic acid methyl ester	165–171	Chloroneb	38–39, 74, 76, 77
Captafol	74, 115	Chloronitroaniline isomers	143
Captan	74, 115	2-Chloropentane (e)	220
Carbaryl	87	2-Chlorophenol	50–53, 60–63
Carbazole	48	4-Chlorophenol	64
Carbofuran	107, 110	4-Chlorophenoxybenzene	133
Carbon tetrachloride see Tetrachloromethane		2-Chloro-1-phenylethanol (e)	238
Carbophenothion	92, 94–95, 98, 115	4-Chlorophenylethylamine (e)	241
Carboxin	38–39	4-Chlorophenyl phenyl ether	50–53
2-Carene	153	Chloroprene rubber, pyrolysis	200
Carene	156	1-Chloropropane	42
Δ3-Carene	157	1-Chloropropene	42
Δ3-Carene (e)	260	2-Chloro-1-propene	45
Carvone	125, 156, 157	3-Chloropropene	42
Carvone (e)	258	2-Chloropropionic acid isobutyl ester (e)	234
Caryophyllene	186	2-Chloropropionic acid methyl ester (e)	206
Ceramides	177	Chlorothalonil	38–39, 74, 110, 114, 115
Cervonic acid methyl ester (C22:6n3, DHA)	168, 169, 172	2-Chlorotoluene	43–45, 49, 59
Chinomethionat see Quinomethionate		3-Chlorotoluene	45, 49, 59
Chlorbenside sulphone	72–73	4-Chlorotoluene	43, 44, 49, 59
α-Chlordane (cis)	38–39, 75–77, 80, 82, 113	4-Chloro- <i>o</i> -toluidine	17, 144–146
γ-Chlordane (trans)	38–39, 75, 80, 82, 86, 113	Chlorotrimethylsilane	148
Chlordane	92	Chlorpropham	38–39, 98, 110
Chlordecone see Kepone		Chlorpyrifos-ethyl	38–39, 74, 92–95, 113, 114, 115
Chlorfenprop-methyl	108	Chlorpyrifos-methyl	92, 94–95, 113
Chlorfenoson	72–73, 100	Chlorthal-methyl	38–39
Chlorfenvinphos	92, 98, 110, 113	Chlorthalonil see Chlorothalonil	
Chloridazon	110	Chlorthion	98, 99, 113
Chlormephos	94–95	Cholesterol	191
Chloroacetic acid methyl ester	116	Chrysene	38–39, 50–53, 65, 66
5-Chloro-2-aminobenzophenone	183	Chrysene-d ₁₂	38–39, 50–51, 112
2-Chloroaniline	143	Cimaterol	188
3-Chloroaniline	143	Cineole	125, 153, 155–157
4-Chloroaniline	17, 50–51, 74, 76, 77, 143–146	Cinnamyl alcohol (e)	248
Chlorobenzene	43–46, 59, 134	Citric acid	186
Chlorobenzilate	38–39, 50–51	Citronellal (e)	266
2-Chlorobiphenyl see PCB-1		Citronellol (e)	266
Chlorobornane derivatives see Parlar		Clenbuterol	188
Chlorodibromomethane	46	Clofibric acid	189
Chloroethane	45	Clomethiazole	183
Chloroform see Trichloromethane		Cocaine	184
		Codeine	183, 184
		Conjugated linoleic acid methyl esters	164

(e) = chiral (enantiomer) separation

Coumaphos	92–96, 115	DEET see Diethyltoluamide	
<i>p</i> -Cresidine	17, 144–146	DEG see Diethylene glycol	
<i>m</i> -Cresol see 3-Methylphenol		Demephion	94–95
<i>o</i> -Cresol see 2-Methylphenol		Demeton-O	92, 93, 96
<i>p</i> -Cresol see 4-Methylphenol		Demeton-S	92, 93, 96
Crotoxyphos	92	Demeton-S-methyl	114
Crufomate	94–95	Demeton-S-methylsulfon	113
Cumene see <i>i</i> -Propylbenzene		Deoxynivalenol	191
Cyanazine	38–39, 104, 110, 112	Desethylatrazine	102–104, 106, 110
Cyanohydrins (e)	217	Desethyldeisopropylatrazine	102–103
5-Cyano-4-methyloxazole	142	Desethylsebutylazine	102–103
4-Cyano-[2.2]-paracyclophane (e)	275	Desethylterbumeton	102–103
(Cyanopropyl)methyldimethoxysilane	149	Desethylterbutylazine	102–103, 106, 110
Cyanuric acid	179	Desisopropylatrazine	102–106, 110
Cyclic olefins (e)	260	Desmetryn	102–103, 110
Cycloate	38–39, 101–103	Desoxyephedrine (e)	258
Cyclobarbitol	183, 187	Dextrorphan	187
Cyclodecane- <i>trans</i> -1,2-diol (e)	227	DHA see Cervonic acid	
Cyclodecane- <i>trans</i> -1,3-diol (e)	227	Diacetylmorphine see Heroin	
<i>trans</i> -1,2-Cycloheptanediol dinitrate	150	Dialifos	94–95
Cycloheptane- <i>trans</i> -1,2-diol (e)	212, 227	Diallyl phthalate	135
Cycloheptane- <i>trans</i> -1,3-diol (e)	227	2,4-Diaminoanisole	144–146
Cyclohexane	121, 123, 124, 126	4,4'-Diaminodiphenylmethane	144–146
Cyclohexanediol dinitrate isomers	150	2,4-Diaminotoluene	144–146
Cyclohexane- <i>trans</i> -1,2-diol (e)	212, 226, 227	<i>o</i> -Dianisidine see 3,3'-Dimethoxybenzidine	
Cyclohexanol	130	Diazinon	38–39, 92, 94–95, 98, 112, 113, 134
Cyclohexanone	122, 127, 128	Dibenz[ah]anthracene	38–39, 50–53, 65, 66
Cyclohexylbenzene	134	Dibenzofuran	50–51
2-Cyclohexylphenol	64	Dibenzothiophene	137
Cyclooctane- <i>trans</i> -1,2-diol (e)	227	Dibenzylamine	126, 140
Cyclopent-2-enyl methyl carbonate (e)	231	Dibrom see Naled	
Cyclopropylstearic acid see Methylene stearic acid		Dibromoacetic acid methyl ester	116
Cyfluthrin	99	4,4'-Dibromobiphenyl	52–53
Cyhalothrin	99	Dibromochloroacetic acid methyl ester	116
α -Cypermethrin	99	Dibromochloromethane	40, 43–45
Cypermethrin	114	1,2-Dibromo-3-chloropropane	43–45
Cyproconazole	110	1,2-Dibromoethane	43–45
		Dibromomethane	43–45
		4,4'-Dibromooctafluorobiphenyl	52–53
D		2,4-Dibromophenol	64
2,4-D esters	108, 109	Dibutylamine	139
α -Damascone (e)	272	Di- <i>i</i> -butylamine	140
2,4-DB	109	Di- <i>n</i> -butylamine	140
2,4'-DDD	72–81, 87, 113	Dibutyl-diethyl-tin	147
4,4'-DDD	38–39, 72–73, 75–83, 87, 113	Dibutyl-dipentyl-tin	147
2,4'-DDE	75–81, 85, 87, 113	3,5-Di- <i>t</i> -butylphenol	155
4,4'-DDE	38–39, 72–73, 75–83, 85, 87, 110, 113	Dibutyl phthalate	133–136
2,4'-DDT	72–75, 77–80, 87, 110, 113, 114	Di- <i>i</i> -butyl phthalate	135
4,4'-DDT	38–39, 72–80, 82, 83, 87, 110, 113	Di- <i>n</i> -butyl phthalate	38–39, 50–53
Decachlorobiphenyl see PCB-209		Dibutyl sebacate	135
δ -Decalactone (e)	244, 267	Dichlobenil	74, 110
γ -Decalactone	216	Dichlofenthion	92, 94–95
γ -Decalactone (e)	215	Dichlofluanid	74, 101, 115
Decane	126	Dichloran see Dicloran	
Decanolide see Decalactone		Dichloroacetic acid methyl ester	116
Decylamine	18, 139, 151	Dichloroaniline isomers	143
<i>n</i> -Decylamine	139	1,2-Dichlorobenzene	43–45, 47, 48, 50–53, 137

Chromatogram index

1,3-Dichlorobenzene	43–45, 47, 48, 50–53	Di- <i>i</i> -hexylamine	140
1,4-Dichlorobenzene	43–45, 47, 48, 50–53	Dihydrotestosterone see Androstanolone	
1,4-Dichlorobenzene-d ₄	50–51	Dimefox	94–95
3,3'-Dichlorobenzidine	50–51, 144–146	Dimethylametryn	102–103
2,3-Dichlorobiphenyl see PCB-5		Dimethoate	92, 98, 113, 114
Dichlorobromomethane	46	3,3'-Dimethoxybenzidine	144–146
Dichlorodimethylsilane	148	Dimethylamine	17, 138
1,1-Dichloroethane	43–45	2,4-Dimethylaniline	145
1,2-Dichloroethane	40, 43–46	2,6-Dimethylaniline	140, 142, 145
1,1-Dichloroethene	43–45	Dimethylaniline	18, 151
<i>cis</i> -1,2-Dichloroethene	40–46	Dimethylbenzene see Xylene	
<i>trans</i> -1,2-Dichloroethene	40, 42–46	3,3'-Dimethylbenzidine	144–146
<i>s</i> -Dichloroethyl ether	133	2,6-Dimethylbenzylamine	137
Dichlorofluoromethane	45	3,3-Dimethylbutane-1,2-diol (e)	205
Dichloromethane	40, 42–46, 123–125, 127, 130, 148	3,4-Dimethylbutyrolactone (e)	224
Dichloronitrobenzene isomers	54, 55	3,3'-Dimethyl-4,4'-diaminodiphenylmethane	144–146
2,4-Dichlorophenol	50–53, 60–64, 112	Dimethyldimethoxysilane	149
2,6-Dichlorophenol	50–51, 60–62	2,8-Dimethyl-1,7-dioxaspiro[5,5]undecane (e)	255
4-(2,4-Dichlorophenoxy)butyric acid see 2,4-DB		Dimethylformamide	126
2,3-Dichlorophenoxyacetic acid	189	N,N-Dimethylformamide	142
2,4-Dichlorophenoxyacetic acid see 2,4-D		3,5-Dimethyl-1,4-hexadiene (e)	233
1,1-Dichloropropane	42	2,2-Dimethyl-4-hydroxymethyl-1,3-dioxolane see Solketal	
1,2-Dichloropropane	43–45	Dimethyl malate (e)	225
1,3-Dichloropropane	43, 44	Dimethyl methylphosphonate	137
2,2-Dichloropropane	42–45	Dimethylnaphthalene isomers	199
1,1-Dichloropropene	43–45	1,2-Dimethyl-3-nitrobenzene	63
2,3-Dichloropropene	42	1,3-Dimethyl-2-nitrobenzene	112
<i>cis</i> -1,3-Dichloropropene (Z)	42–45	3,7-Dimethyl-1,6-octadien-3-ol	154
<i>trans</i> -1,3-Dichloropropene (E)	42–45	3,7-Dimethyl-6-octenal	153
Dichlorotoluene isomers	59	2,6-Dimethyl-7-octen-2-ol	153
Dichloroprop 2-ethyl 1-hexyl ester	108	3,7-Dimethyl-6-octen-1-ol	155
Dichloroprop methyl ester	108, 109	3,7-Dimethyl-octen-1-ol	154
Dichloroprop methyl ester (e)	256	2,4-Dimethylphenol	50–53, 60–63
Dichlorvos	38–39, 92–95, 98, 113, 115	2,6-Dimethylphenol	137
Diclobutrazol	101	3,3-Dimethyl-1-phenyl-butan-1-ol (e)	270
Diclofenac	189	Dimethyl phthalate	38–39, 50–53, 133, 135, 136
Dicloran	74, 101	Dimethylsilane	148
Dicrotophos	92	Dimethylsulfoxide	126
1,3-Dicyanopropyltetramethyldisiloxane	149	1,3-Dinitrobenzene	50–51, 55
Dicyclohexylamine	139, 140	4,6-Dinitro-2-methylphenol	52–53
Dicyclohexyl phthalate	135	2,4-Dinitrophenol	50–53, 60, 61–63
Dieldrin	38–39, 72–83, 87, 110, 113, 114	2,3-Dinitrotoluene	54, 55
Diethanolamine	140, 141	2,4-Dinitrotoluene	38–39, 50–55, 58
Diethyl acetal	159	2,5-Dinitrotoluene	54, 55
Diethyl allylphosphonate	136	2,6-Dinitrotoluene	38–39, 50–55, 58
Diethylamine	138, 140	3,4-Dinitrotoluene	54, 55
Diethylene glycol	129, 160	3,5-Dinitrotoluene	54, 55
Diethyl ethylphosphonate	136	2,3-Dinor-6-oxo-prostaglandin F _{1α}	192
Di(2-ethylhexyl) adipate	135	2,3-Dinor-thromboxane B ₂	192
Di(2-ethylhexyl) phthalate	38–39, 50–53, 133, 135, 136, 187	Dinoseb	50–51, 62
Diethyl methylphosphonate	136	Di- <i>n</i> -octyl phthalate	50–53
Diethyl phthalate	38–39, 50–53, 133, 135, 136, 153–155	Diocetyl phthalate	133, 135, 136
Diethylstilbestrol	193	1,4-Dioxane	125
Diethyltoluamide	107	Dioxane	123
		1,7-Dioxaspiro[5,5]undecane (e)	255

(e) = chiral (enantiomer) separation



Chromatogram index

Ethylbenzene	43–46, 59, 119–121, 127, 128, 199	Fluorene	38–39, 50–53, 65, 66
Ethyl butanoate	158	Fluorenone	134
γ -Ethylbutyrolactone	see γ -Hexalactone	2-Fluorobiphenyl	50–51
Ethylidiglycol	122	Fluorochloridone	74
S-Ethyl N,N-dipropylthiocarbamate	see EPTC	Fluorodifen	102–103
Ethylene glycol	128	2-Fluorophenol	50–53
1-O-Ethylglycerol (e)	217	4-Fluorophenol	137
Ethyl glycol	122, 127, 128	Fluotrimazole	102–103
Ethyl glycol acetate	122, 127, 128	Fluoxetine (e)	274
Ethyl hexanoate	158	Flurbiprofen (e)	271
3-Ethyl-2-hydroxy-2-cyclopenten-1-one	158	Fluridone	38–39
Ethyl methanesulfonate	50–51	Flusilazole	98, 110
2-Ethyl-7-methyl-1,6-dioxaspiro[4,5]decane (e)	255	Flutriafof	101
Ethyl methyl ketone	see Methyl ethyl ketone	Fluvalinate	74
Ethyl nonanoate	155	Folpet	74, 115
Ethyl octanoate	158	Fonofos	92, 112
N-Ethylloxazepam	190	4-Formyl-[2.2]-paracyclophane (e)	275
Ethylphosphonic acid diethyl ester		Frontalin (e)	244
see Diethyl ethylphosphonate		Fumaric acid	194
2-Ethyltoluene	120	Furaneol	158
3-Ethyltoluene	121	1-(2-Furyl)ethanol (e)	223
4-Ethyltoluene	120, 121	1-(2-Furyl)ethanol acetate (e)	223
Ethylvanillin	154		
Etridiazole	38–39, 72, 101	G	
Etrifos	98	Gemfibrozil	189
Eucalyptol	see Cineole	Geraniol	156
Eugenol	153, 156, 158	Glucose	185, 186
		Glucose anomers (e)	237
F		Glutamic acid propyl esters (e)	202
FAMES	164–174	Glutaric acid	194
Famphur	92	Glyceric acid methyl ester (e)	209
Fatty acid methyl esters	see FAMES	Glycerol	160
Fatty alcohols, ethoxylated	174	Glycidyl butyrate (e)	234
Fenamiphos	38–39, 94–95	Glycine propyl esters	202
Fenarimol	38–39, 101	Glycolic acid	127
Fenchlorphos	92–95, 98, 114	Glycolic acid <i>n</i> -butyl ester	127
Fenchone	156	Glycophene	see Iprodione
Fenchone (e)	261	Gondoic acid methyl ester (C20:1n9)	166
Fenfuram	101	Grandisol (e)	264
Fenitrothion	92, 113		
Fenoprofen	189	H	
Fenoprofen (e)	271	HCB	38–39, 47, 48, 50–53, 72–73, 75–81, 85, 87, 113
Fenoprop methyl ester	108, 109	HCH	see BHC
Fenoprop methyl ester (e)	256	Henicosane	18, 151
Fenpropathrin	99	Henicosanoic acid methyl ester	
Fenpropimorph	101	(C21:0)	165, 166, 168, 169, 171–173
Fensulfothion	92, 93, 96	Heptachlor	38–39, 74–80, 82, 83, 87, 110, 113
Fenthion	92, 93, 96, 98, 113	<i>cis</i> -Heptachlor epoxide (B)	
Fenvalerate	99		38–39, 72–73, 75, 78–81, 86, 87, 110, 113
Flamprop-isopropyl	102–103	<i>trans</i> -Heptachlor epoxide (A)	
Fluazifop-butyl	102–103		38–39, 75, 78–81, 86, 87, 110, 113
Fluchloralin	101–103	Heptachlor epoxide	72–73, 82, 83
Flucythrinate	74	Heptachlorobiphenyls	see PCB nomenclature
Fluometuron	107	Heptachlorodibenzodioxins	71
Fluoranthene	50–53, 65, 66	Heptachlorodibenzofurans	71

(e) = chiral (enantiomer) separation

<i>cis</i> -10-Heptadecenoic acid methyl ester (C17:1n7)	166, 168, 169	Hexylcinnamic aldehyde	155
Heptadecenoic acid methyl ester (C17:1)	165, 167, 170, 172	<i>n</i> -Hexylcyanohydrin (e)	217
γ -Heptalactone (e)	215, 216	<i>n</i> -Hexyl-3-nitrate (e)	229
Heptane	123	δ -Hexylvalerolactone	see δ -Undecalactone
<i>n</i> -Heptane	199	Hippuric acid	194
Heptane-1,2-diol (e)	205, 212	Hydratropic acid methyl ester (e)	258
Heptane-1,2,3-triol (e)	238	2-Hydroxybiphenyl	64
Heptanoic acid	163	3-Hydroxybutyraldehyde (e)	207
Heptanol	122	2-Hydroxybutyric acid methyl ester (e)	209
Heptanol-1	125	Hydroxycitronellal	155
Heptanol-2 (e)	220, 237	2-Hydroxycyclopentadecanone	154
4-Heptanolide	see γ -Heptalactone	7-Hydroxy-3,7-dimethyl-octanal	154
Heptanophenone	126	3-Hydroxyglutaric acid	194
Heptenophos	94–95, 98	Hydroxymethylcyclobutanol	188
γ -Heptylbutyrolactone	see γ -Undecalactone	3-Hydroxy-3-methylglutaric acid	194
<i>n</i> -Heptylcyanohydrin (e)	217	4-Hydroxymethyl-2.2-paracyclophane (e)	275
<i>n</i> -Heptyl-3-nitrate (e)	229	4-Hydroxy-4-methyl-2-pentanone	122
δ -Heptylvalerolactone	see δ -Dodecalactone	Hydroxymidazolam metabolites	190
Heroin	185, 186	1-Hydroxy-2-nitrooxycyclohexane	150
Hexachlorobenzene	see HCB	1-Hydroxy-2-nitrooxycyclopentane	150
Hexachlorobiphenyls	see PCB nomenclature	4-Hydroxy-[2.2]-paracyclophane (e)	275
Hexachlorobutadiene	43–45, 47, 48, 50–53, 78, 79	3-Hydroxy- <i>i</i> -valeric acid	194
Hexachlorocyclopentadiene	38–39, 47, 48, 50–53, 72–73	Hydroxyvaleronitrile (e)	217
Hexachlorodibenzodioxins	71		
Hexachlorodibenzofurans	71	Ibuprofen	189
Hexachloroethane	47, 48, 50–51, 52–53	Ibuprofen (e)	271
Hexachloropropene	50–51	Ibuprofen methyl ester (e)	273
Hexaconazole	101	Imazalil	98, 100
Hexadecenoic acid methyl ester (C16:1)	165, 169, 170, 171, 173, 174	Indane	121
also see Palmitoleic acid methyl ester		1-Indanol (e)	247
1,5-Hexadiene	42	Indeno[1,2,3-cd]pyrene	38–39, 50–53, 65, 66
Hexafluorobisphenol A	64	Indomethacin	189
Hexahydroisobenzofuranone (e)	242	Iodofenphos	94–95
γ -Hexalactone	216	α -Ionone (e)	272
γ -Hexalactone (e)	215	Iprodione	115
Hexamethyldisilazane	149	<i>i</i> -Propylbenzene	120
Hexamethyldisiloxane	148, 149	Isazofos	94–95
Hexamethylphosphoramide	92	Isoallyl chloride	42
Hexane	123	Isoamyl alcohol	159
<i>n</i> -Hexane	46, 121, 123, 125, 199	Isobenzan	78, 79
Hexane-1,2-diol (e)	205, 212, 222	Isoborneol (e)	264
1,2-Hexanediol dinitrate	150	Isobumeton	see Secbumeton
Hexanoic acid	158	Isobutanol	see 2-Methyl-1-propanol
also see Caproic and Isocaproic acid		Isobutyl	also see <i>i</i> -Butyl
Hexanoic acid 2-propenyl ester	153	Isobutylcyanohydrin (e)	217
Hexanol-1	130, 156	1-O-Isobutylglycerol (e)	217
Hexanolactone	see Hexalactone	Isobutyl methyl ketone	129
4-Hexanolide	see γ -Hexalactone	Isobutyric acid	see 2-Methylpropionic acid
Hexazinone	38–39, 102–103, 107	Isocaproic acid	163
(E)-2-Hexenoic acid	158	Isocarbamid	102–103
Hex-3-en-2-yl methyl carbonate (e)	245	Isodrin	50–51, 75–79, 110
Hexobarbital (e)	270	Isoheptane	122
γ -Hexylbutyrolactone	see γ -Decalactone	Isoleucine propyl esters (e)	202
		Isomargaric acid methyl ester (C17:0 iso)	171
		Isomenthol (e)	259

Chromatogram index

Isomenthone	125, 156	Linolelaidic acid methyl ester	
Isomenthone (e)	259, 264	(C18:2n6 <i>trans</i>)	166, 168, 169, 172
Isomethiozin	102–103	α -Linolenic acid methyl ester	
α -Isomethyl ionone	153–155	(C18:3n3)	164, 166, 168, 169, 171, 172
Isomyristic acid methyl ester (C14:0 iso)	171	γ -Linolenic acid methyl ester	
Isononyl acetate	153	(C18:3n6)	166, 168, 169, 171, 172
Isooctane	122	Linolenic acid methyl ester	
Isooctyl-dibutyl-ethyl-tin	147	(C18:3)	165, 167, 170, 173, 174
Isopalmitic acid methyl ester (C16:0 iso)	171	Lysine propyl esters (e)	202
Isopentyl acetate	134		
1-O-Isopentylglycerol (e)	217		
Isophorone	38–39, 48, 50–54, 58		
3-Isopropenyl-2-methylcyclopentanone (e)	251		
Isopropalin	102–103, 106		
Isopropyl see <i>i</i> -Propyl			
1,2-Isopropylidene glycerol see Solketal			
Isopulegol (e)	259		
Isosafrole	50–51		
Isovaleric acid	163		
Isovaline isopropyl ester (e)	202		
Isoxsuprine	188		
K			
Kepone	50–51		
Ketoendrin	80, 86		
Ketoprofen	189		
Ketoprofen (e)	271		
L			
Lactic acid	194		
Lactide (e)	223		
Lauric acid methyl ester (C12:0)	126, 165–173		
Lavandulol (e)	265		
Lavandulyl acetate (e)	265		
Lenacil	102–103		
Leptophos	92		
Leucine propyl esters (e)	202		
Leucinol (e)	204		
Levomethorphan	187		
Lidocaine	184		
Lidoflazine	183		
Lignoceric acid methyl ester			
(C24:0)	165–169, 171–173		
Lilial	153, 155		
Limonene	125, 153, 155–157		
Limonene (e)	260, 261		
Linalool	125, 155–157		
Linalool (e)	266, 267		
Linalool oxide (e)	266		
Linalyl acetate (e)	156, 267		
Linalyl anthranilate	155		
Linalyl butyrate	154		
Linalyl propanoate	153		
Lindane see γ -BHC			
Linoleic acid methyl ester (C18:2n6 <i>cis</i>)	164–174		

(e) = chiral (enantiomer) separation

M

Mabuterol	188
Malathion	87, 92, 94–95, 98, 113, 137
Maltol	158
Mandelic acid methyl ester (e)	248
Mannose anomers (e)	237
Margaric acid methyl ester (C17:0)	165–174
MCPA	109
MCPB	109
MCPP see Mecoprop	
Mecarbam	100
Meclofenamic acid	189
Meconin	186
Mecoprop methyl ester	109
Mecoprop methyl ester (e)	269
MEK see Methyl ethyl ketone	
Melamine	179
<i>p</i> -Menth-1(7)-en-9-ol	153
Menthofuran	125, 156
Menthofuran (e)	259
Menthol	153, 155–157
Menthol (e)	259, 268
L-Menthone	156
Menthone	125
Menthone (e)	259, 263
Menthyl acetate	125, 156
Menthyl acetate (e)	259, 271
Menthyl valerianate	156
Merphos	38–39, 92, 93
Merphos oxidation product	92, 93
Mestranol	193
Mesuximide (e)	268
Metalaxyl	101, 102–103, 110
Metamitron	102–103, 110
Metazachlor	102–103, 105, 106, 110, 114
Methacrifos	94–95, 98
Methadone	184
Methamphetamine	181, 184
Methane	196
Methanol	122–125, 127–131, 159, 160
Methaqualone	183, 184
Methfuroxam	101
Methidathion	94–95, 98, 113, 114
Methionine propyl esters (e)	202
Methomyl	107

Methoprotryn	102–104, 110	Methylene stearic acid methyl ester (C19:0 cyclo)	171
Methoxyaniline see Anisidine		1-Methylethylbenzene see <i>i</i> -Propylbenzene	
4-Methoxybenzaldehyde	153	Methyl-3-ethylbenzene	199
Methoxychlor		Methyl ethyl ketone	122, 123, 127–129
38–39, 74, 75, 77, 82, 87, 105, 110, 113		Methyl formate	123
<i>p</i> -Methoxycinnamic acid 2-ethylhexyl ester		3-Methylglutaric acid	194
<i>cis</i> - and <i>trans</i> - isomer	177	1-O-Methylglycerol (e)	217
Methoxy-5-indole	126	Methyl glycol	127
4-Methoxy- <i>m</i> -phenylene diamine		Methyl glycol acetate	127
see 2,4-Diaminoanisole		Methyl glycol phthalate	135
1-Methoxypropanol-2	122, 127, 128	Methyl heptanoate	158
1-Methoxypropyl acetate-2	122, 127, 128	4-Methyl-3-heptanone (e)	246
2-Methoxypropyl acetate-1	127, 128	3-Methyl-1,4(Z)-hexadiene (e)	233
Methyl acetate	122, 123, 127, 128, 160	Methyl hexanoate see Caproic acid methyl ester	
Methylamine	138	3-Methyl-1-hexene (e)	210
Methylaniline see Toluidine		Methylhydromethoxysilane	149
Methyl atrate	155	2-Methyl-3-hydroxyvaleraldehyde (e)	207
Methylbenzene see Toluene		β -Methylionone	154
α -Methylbenzyl acetate	153, 154	Methyl isobutyl ketone	122
3-(4'-Methylbenzylidene)camphor	177	Methyl lactate (e)	208, 209
Methylbutane	196	Methyl methanesulfonate	50–51
3-Methylbutane-1,2-diol (e)	205	Methylmyristic acid methyl ester	
2-Methyl-1-butanol	159, 160	(C15:0 iso/anteiso)	171
2-Methyl-2-butanol	126	1-Methylnaphthalene	50–51, 199
3-Methyl-1-butanol	159, 160	2-Methylnaphthalene	50–51, 199
3-Methylbutyl acetate	158	Methyl octadecanoate	
Methyl butyrate see Butyric acid methyl ester		see Stearic acid methyl ester (C18:0)	
2-Methylbutyric acid (e)	218	4-Methyl-5-oxazole-carboxylic acid amide	142
2-Methylbutyric acid ethyl ester (e)	218, 235	4-Methyl-[2.2]-paracyclophane (e)	275
2-Methylbutyric acid methyl ester (e)	227	Methyl pentachlorophenyl sulphide	72–73, 81
3-Methylbutyric acid	158	Methyl pentanoate see Valeric acid methyl ester	
α -Methylbutyrolactone (e)	214, 215	2-Methylpentanol-1 (e)	229
3-Methylcholanthrene	50–51	2-Methylphenol	50–51, 59–64
3-Methylcyclohexene (e)	210, 232	3-Methylphenol	50–51, 59, 61, 62
Methylcyclohexane	121	4-Methylphenol	50–51, 59–62
2-Methylcyclohexanone (e)	233	4-Methyl- <i>m</i> -phenylenediamine	
Methyl decanoate		see 2,4-Diaminotoluene	
see Capric acid methyl ester (C10:0)		1-(2-Methylphenyl)ethanol (e)	240
Methyl dihydrojasmonate	153–155	1-(3-Methylphenyl)ethanol (e)	240
Methyl 2,4-dihydroxy-3,6-dimethylbenzoate	155	1-(4-Methylphenyl)ethanol (e)	240
2-Methyl-4,6-dinitroanisole	56	5-(<i>p</i> -Methylphenyl)-5-phenylhydantoin	190
Methyldinitrobenzene isomers	48	Methylphosphonic acid diethyl ester	
2-Methyl-4,6-dinitrophenol	50–51, 60–63	see Diethyl methylphosphonate	
7-Methyl-1,6-dioxaspiro[4,5]decane (e)	255	Methylphosphonic acid dimethyl ester	
Methyl <i>cis</i> -13-docosenoate see EPTC: 1n9)		see Dimethyl methylphosphonate	
Methyl dodecanoate		2-Methyl-1-propanol	122, 127, 128, 159, 160
see Lauric acid methyl ester (C12:0)		5-(2-methyl-propan-2-sulphonyl)-hept-3-ene (e)	269
Methyl eicosanoate		2-Methylpropionic acid	158, 163
see Arachidic acid methyl ester (C20:0)		2-Methylpropyl acetate	158
4,4'-Methylene-bis-(2-chloroaniline)	144–146	1-Methylpropylbenzene see <i>sec</i> -Butylbenzene	
Methylene chloride see Dichloromethane		Methylpropyl-4,6-dinitrophenol see Dinoseb	
Methylenedianiline see Diaminodiphenylmethane		N-Methylpyrrolidone	142
3,4-Methylenedioxyamphetamine	181	α -Methylquinovoside (e)	236
3,4-Methylenedioxyethylamphetamine	181	α -Methylrhamnoside (e)	236
3,4-Methylenedioxyamphetamin	181	Methylsilane	148
4,4'-Methylenedi- <i>o</i> -toluidine		Methyl stearate see Stearic acid methyl ester (C18:0)	
see 3,3'-Dimethyl-4,4'-diaminodiphenylmethane		2-Methylthioethanol	158

Chromatogram index

3-Methylthiopropanol	158	4-Nitrophenol	50–53, 62, 63
Methyltrichlorosilane	148	4-Nitro-quinoline 1-oxide	50–51
Methyl undecanoate see Undecanoic acid methyl ester (C11:0)		N-Nitrosodimethylamine	50–53, 133
2-Methyl-3-vinylcyclopentanone (e)	241	N-Nitrosodiphenylamine	133
Metolachlor	38–39, 102–103, 105, 106, 110	N-Nitroso-di- <i>n</i> -propylamine	50–53, 133
Metribuzin	38–39, 104, 107, 110	Nitrotoluene isomers	55
Mevinphos	38–39, 92–95, 98	5-Nitro- <i>o</i> -toluidine	144–146
Mexiletine (e)	250	2-Nitro- <i>m</i> -xylene	38–39
MGK-264	38–39	Nivalenol	191
Midazolam	190	Nitromusk see Musk	
Mirex	72–73, 75–77, 81, 85, 87	<i>trans</i> -Nonachlor	38–39, 76
Molinate	38–39, 102–103	Nonachlorobornane see Parlar	
Monoacetylmorphine	185, 186	Nonadecanoic acid methyl ester (C19:0)	167, 168, 171
Monocrotophos	92	δ -Nonalactone (e)	244
Morphine	183, 184, 186	γ -Nonalactone	216
Musk ambrette	56–57	γ -Nonalactone (e)	215
Musk ketone	56–57, 80, 86	<i>n</i> -Nonane	121, 199
Musk moskene	56–57	Nonane-1,2-diol (e)	205, 212
Musk tibetene	56–57	Nonanoic acid ethyl ester	153
Musk xylene	56–57, 75, 80, 86	Nonanol-2 (e)	220
Methylphosphonic acid dimethyl ester see Dimethyl methylphosphonate		Nonylamine	139
Myclobutanil	98, 101, 114	<i>n</i> -Nonyl-3-nitrate (e)	229
Mycotoxines	191	4-Nonylphenol, technical	193
Myrcene	156	Norephedrine (e)	250
Myristic acid isopropyl ester	153	Norfluooxetine (e)	274
Myristic acid methyl ester (C14:0)	165–173	Norflurazon	38–39
Myristoleic acid methyl ester (C14:1)	165–170, 172	Nuarimol	74, 101, 114

N

Naled	92, 93, 115	Octachlorobiphenyls see PCB nomenclature	68
Nalorphine	184	Octachlorobornane see Parlar	
Naphthalene	43–45, 50–53, 65, 66, 121, 137, 199	Octachlorodibenzodioxin	71
Naphthalene- <i>d</i> ₈	50–51	Octachlorodibenzofuran	71
1,4-Naphthoquinone	50–51	Octachlorostyrene	72–73, 75, 81
1-Naphthylamine	143	<i>trans</i> -9,11-Octadecadienoic acid	164
2-Naphthylamine	144–146	Octadecatetraenoic acid methyl ester	171
2-Naphthyl methyl ketone	153	δ -Octalactone (e)	244
Napropamide	38–39	γ -Octalactone (e)	215, 216
Naproxen	189	Octamethylcyclotetrasiloxane	149
Naproxen (e)	271	<i>n</i> -Octane	199
Narcotine	185	Octane-1,2-diol (e)	205, 212, 222
Neoisomenthol (e)	259	Octanoic acid see Caprylic acid	
Neomenthol (e)	259	Octanol	17, 18, 151
Nervonic acid methyl ester (C22:1n9)	165–173	Octanol-1	126, 130, 137
Nicotine	183	4-Octanolide see γ -Octalactone	
Nitroaniline isomers	50–51	5-Octanolide see δ -Octalactone	
Nitrobenzene	48, 50–55, 58, 112	1-Octene-7,8-diol (e)	205
Nitrobenzene- <i>d</i> ₅	50–53, 64	1-Octen-3-ol (e)	246, 247
Nitrochlorobenzene isomers	54, 55	N-Octyl bicycloheptene dicarboximide see MGK-264	
Nitrooxybutanol isomers	150	γ -Octylbutyrolactone see γ -Dodecalactone	
2-Nitrooxyethanol	150	<i>n</i> -Octyl-3-nitrate (e)	229
Nitrooxypropanol isomers	150	4- <i>t</i> -Octylphenol	193
2-Nitrophenol	50–53, 60–63	Ofurace	101
		Oleic acid methyl ester (C18:1n9 <i>cis</i>)	164–174
		Ornithine propyl esters (e)	202

(e) = chiral (enantiomer) separation

Oxacyclohexadecan-2-one	155	PCB-201	38–39
Oxadiazon	102–103	PCB-209	70, 80
Oxamyl	107	PCDD see Polychlorinated dibenzodioxins	
2-Oxoglutaric acid	194	PCDF see Polychlorinated dibenzofurans	
Oxychlordane	75–77, 80	Pebulate	38–39, 102–103
4,4'-Oxydianiline	144–146	Penconazole	98, 101, 114
P			
PAH	38, 50–53, 65–67	Pendimethalin	110
Palmitic acid methyl ester (C16:0)	164–174	Pentabromodiphenyl ether isomers	85
Palmitoleic acid methyl ester		Pentachloroanisole	72–73
(C16:1n7 <i>cis</i>)	166–168, 172	Pentachlorobenzene	50–51, 75–79, 85
Pantolactone (e)	224	Pentachlorobiphenyls see PCB nomenclature	68
Papaverine	185, 186	Pentachlorodibenzodioxins	71
Paracetamol	185, 186	Pentachlorodibenzofurans	71
Parathion-ethyl	87, 92, 94–95, 110, 114	Pentachloronitrobenzene	50–51
Parathion-methyl	38–39, 87, 92, 93, 110, 113	Pentachlorophenol	38–39, 50–53, 60, 62, 63
Parlar 11, 12, 15, 21, 25	88, 89	Pentadecanoic acid methyl ester (C15:0)	165–172
Parlar 26	80, 85, 86, 88, 89	<i>cis</i> -10-Pentadecenoic acid methyl ester	
Parlar 31, 32, 38, 39, 40	88, 89	(C15:1n5)	166, 168, 169, 172
Parlar 41, 42a, 42b, 44	88, 89	Pentadecenoic acid methyl ester (C15:1)	165, 167, 170
Parlar 50	80, 85, 86, 88, 89	Pentaethylene glycol	129
Parlar 51, 56, 58, 59	88, 89	Pentafluorophenol	52–53
Parlar 62	80, 86, 88, 89	<i>n</i> -Pentane	46, 124, 196, 199
Parlar 63, 69	88, 89	Pentane-1,2-diol (e)	205, 212, 222
Parsols	177	Pentane-2,4-diol (e)	221
PCB-1, PCB-5	38–39	1,2-Pentanediol dinitrate	150
PCB-18, PCB-20	69	Pentanochlor	102–103
PCB-28	68–70, 72–81, 85	Pentanoic acid see Valeric acid and Isovaleric acid	
PCB-29	38–39	Pentanol-1	159
PCB-31	68–70, 80	Pentanol-2 (e)	220, 221
PCB-44	68, 69	4-Pentanolide see γ -Valerolactone	
PCB-47	38–39	Pentenol	125
PCB-49	68, 80	Pentobarbital	187
PCB-52	68–70, 72–81, 85	γ -Pentylbutyrolactone see γ -Nonalactone	
PCB-61	68	<i>n</i> -Pentylcyanohydrin (e)	217
PCB-66	68, 85	1-O-Pentylglycerol (e)	217
PCB-70	68	δ -Pentylvalerolactone see δ -Decalactone	
PCB-98	38–39	Permethrin	99
PCB-99	85	Permethrin <i>cis/trans</i>	38–39
PCB-101	69, 70, 72–77, 79–81, 85	Perylene- d_{12}	38–39, 50–51
PCB-103	85	Phenacetin	50–51, 183
PCB-105	69	Phenanthrene	38–39, 50–53, 65, 66
PCB-114	69	Phenanthrene- d_{10}	38–39, 50–51, 112
PCB-118	69, 70, 78, 79, 85	Phenethyl alcohol see 2-Phenylethanol	
PCB-128	114	<i>m</i> -Phenetidine	143
PCB-138	69, 70, 72–81, 85	Phenkapton	94–95, 98
PCB-149	69	Phenobarbital	187
PCB-153	69, 70, 72–81, 85, 114	Phenol	50–53, 60–64, 134
PCB-154	38–39	Phenol- d_6	50–51
PCB-156	69, 80	Phensuximide (e)	268
PCB-170	69, 75	Phenylacetic acid	158
PCB-171	38–39	Phenylacetic acid methyl ester	153, 154
PCB-180	69, 70, 72–81, 85	Phenylalanine propyl esters (e)	202
PCB-183, PCB-189	80	2-Phenylbutyric acid	194
PCB-194	69	3-Phenylbutyrolactone (e)	257
		4-Phenylbutyrolactone (e)	257
		N-Phenyl-cyclohex-2-enamine (e)	269
		($\pm$)- <i>trans</i> -2-Phenylcyclopropylmethanol (e)	248

Chromatogram index

(±)- <i>trans</i> -2-Phenylcyclopropylmethyl acetate (e)	248	Propanol-1	122, 127–132, 159, 160
1-Phenyl-1,2-ethanediol (e)	205, 212	Propanol-2	123–125, 127–131
2-Phenylethanol	155, 158	<i>m</i> -Propanolpyridine	141
α-Phenylethanol	153, 154	<i>o</i> -Propanolpyridine	140, 141
α-Phenylethanol (e)	239, 240	<i>p</i> -Propanolpyridine	141
1-Phenylethyl acetate	155	Propazine	38–39, 104–106, 110
2-Phenylethyl acetate	158	Propam	98, 102–103, 110
α-Phenylethylamine (e)	240	Propiconazole	102–103, 110, 114
Phenylethylamine (e)	204	Propionic acid	163
Phenylglycol see 1-Phenyl-1,2-ethanediol		<i>i</i> -Propyl acetate	127, 128
2-Phenylpropane see <i>i</i> -Propylbenzene		<i>n</i> -Propyl acetate	127
1-Phenylpropan-1-ol (e)	249	4- <i>i</i> -Propylaniline	143
2-Phenylpropan-1-ol (e)	249	<i>N</i> - <i>i</i> -Propylaniline	64
3-Phenyl-2-propen-1-ol	153	<i>i</i> -Propylbenzene	43–45, 59, 121
2-Phenylpropionic acid methyl ester (e)	258	<i>n</i> -Propylbenzene	43, 44, 59, 120, 121
(Phenylpropyl)methyldimethoxysilane	149	γ-Propylbutyrolactone see γ-Heptalactone	
Pholedrine (e)	250	1- <i>i</i> -Propyl-2-carboxymethyl aziridine (e)	235
Phorate	92, 93, 96	<i>n</i> -Propylcyanohydrin (e)	217
Phosalone	94–95, 110	2- <i>i</i> -Propyl-4,6-dinitrophenol	60, 61
Phosmet	92	Propylene glycol	125
Phosphamidon	92, 114	<i>i</i> -Propyl myristate	153, 155
Phosphonic acid diethyl esters	136	2- <i>i</i> -Propyltoluene	45
α-Pinene	156, 157	4- <i>i</i> -Propyltoluene	43, 44, 59
α-Pinene (e)	261	δ-Propylvalerolactone see δ-Octalactone	
β-Pinene	153, 156, 157	Propyzamide	38–39, 114
β-Pinene (e)	261	Prothiocarb	101
Piperidinone-2	132	Prothiophos	92–96, 114
Piperitone (e)	259	Pulegone	125
Piperonal	153, 154	Pulegone (e)	259
Pirimicarb	110	Pyrene	38–39, 50–53, 65, 66
Pirimiphos	98	Pyrene- <i>d</i> ₁₀	38–39
Pirimiphos-ethyl	94–95	Pyridaphenthion	94–95, 98
Pirimiphos-methyl	94–95, 113	Pyridine	50–51, 126, 141, 142
Polychlorinated biphenyls see PCB		Pyridine- <i>d</i> ₅	52–53
Polychlorinated bornanes see Parlar		Pyrolysis GC	200
Polychlorinated dibenzodioxins	71	Q	
Polychlorinated dibenzofurans	71	Quercus lactone see 4-Butyl-3-methylbutyrolactone	
Polycyclic aromatic hydrocarbons see PAH		Quinalphos	94–95, 98, 114
Polyethylene pyrolysis	200	Quinine	183
Polypropylene pyrolysis	200	Quinomethionate	101
Prochloraz	101, 110	Quintozene	72–74, 76, 77, 87, 110, 113
Procyimdone	74, 114, 115	R	
Profenofos	110, 114	Ractopamine	188
Profluralin	106	Rapeseed oil: FAMES	173
Progesterone	191–193	Ronnel see Fenchlorphos	
Proline propyl esters (e)	202	Rose oxide (e)	265
Prometon	38–39, 102–104, 112	S	
Prometryn	38–39, 104, 105, 106, 110	S 421	74
Pronamide see Propyzamide		Sabinene	156
Propachlor	38–39	Sabinene (e)	260
Propan-1,2-diol also see Propylene glycol		(+)- <i>trans</i> -Sabinene hydrate (e)	259
Propane	196		
Propane-1,2-diol (e)	204, 205		
Propanediol dinitrate isomers	150		
Propanil	102–103		

(e) = chiral (enantiomer) separation

Safrole	50–51, 156	Tetrachloromethane	40, 42–46
Salbutamol	188	1,2,3,4-Tetrachloronaphthalene	84, 85
Sebutylazine	102–103, 106, 110	1,2,4,5-Tetrachloro-3-nitrobenzene	see Tecnazene
Secbumeton	104	2,3,4,5-Tetrachlorophenol	61–63
Secobarbital	187	2,3,4,6-Tetrachlorophenol	50–51, 60–62
Silanes	148, 149	2,3,5,6-Tetrachlorophenol	60–62
Simazin	38–39, 104–106, 110	Tetrachlorosilane	148
Simetryn	38–39, 102–103, 104	2,4,5,6-Tetrachloroxylene	72–73
Solketal (e)	227	Tetrachlorvinphos	38–39, 92, 93
Sorbitol (e)	230	Tetracosane	105, 186
Spiroacetals (e)	255	Tetracosenic acid methyl ester (C24:1)	171
Stearic acid methyl ester (C18:0)	137, 164–174	Tetradecane	105
Stirofos see Tetrachlorvinphos		Tetradifon	72–74, 100, 114
Styralyl acetate	155	Tetraethylene glycol	129
Styrene	43–45, 59, 134	Tetrahydrocannabinol	186
Styrene oxide (e)	238	Tetrahydrofuran	123, 124, 126
Succinic acid	194	Tetrahydroisobenzofuranone (e)	215
Sulfotep	94–95, 98, 113	Tetralin	121
Sulprofos	92–96	Tetramethrin	99
Sympathomimetic amines (e)	250	1,1,3,3-Tetramethyldisiloxane	149
		Tetramethylsilane	148
		Tetrasul	74
		Tetatriacontane	179
		TFA see Trifluoroacetic acid	
		THC see Tetrahydrocannabinol	
		Theaspiran (e)	273
		THF see Tetrahydrofuran	
		Thiabenzazole	101
		Thioacetic acid cyclopent-2-enyl ester (e)	230
		Thiobencarb	101
		4,4'-Thiodianiline	144–146
		Thiometon	94–95
		Thionazin	92
		Thioquinox	101–103
		Thioridazine	183
		Thiothixene	183
		Thymol	156
		Tilidine	183
		Timnodonic acid methyl ester	
		(EPA, C20:5n3)	166, 168, 169, 172
		Tin compounds	147
		TNT see Trinitrotoluene	
		D,L- α -Tocopherol acetate	179
		Tokuthion see Prothiophos	
		Tolclofos-methyl	94–95, 98
		Tolfenamic acid	189
		Toluene	43–46, 59, 119, 120–128, 199
		Toluene-d ₈	64, 120
		2,4-Toluenediamine see 2,4-Diaminotoluene	
		m-Toluidine	142, 143
		o-Toluidine	143, 144, 145, 146
		p-Toluidine	143
		Tolyfluanid	74
		Toxaphenes	88, 89
		also see Parlar	
		2,4,5-TP see Fenoprop	
		Tranilcypromine (e)	250
2,4,5-T isooctyl ester	108		
2,4,5-T methyl ester	108, 109		
2,4,5-T octyl ester	108		
Tartaric acid dimethyl ester (e)	209, 225		
TCNB see Tecnazene			
Tebuconazole	98, 110		
Tebuthiuron	38–39		
Tecnazene	72–74, 80, 113, 114		
Telodrin	72–73, 81		
TEPP	92, 94–95		
Terbacil	38–39, 107		
Terbufos	38–39, 92, 112		
Terbumeton	104		
Terbutylazine	104–106, 110		
Terbutryn	38–39, 102–106, 110		
p-Terphenyl-d ₁₄	38–39, 50–51		
α -Terpinene	156		
γ -Terpinene	156		
Terpinen-4-ol	263		
α -Terpineol	156		
α -Terpineol (e)	263		
Terrazole see Etridiazole			
Testosterone	192, 193		
2,2',4,4'-Tetrabromodiphenyl ether	85		
Tetrabutyl-tin	147		
Tetrachloroaniline isomers	72–73		
1,2,3,4-Tetrachlorobenzene	75		
1,2,4,5-Tetrachlorobenzene	50–51, 75		
Tetrachlorobiphenyls see PCB nomenclature	68		
Tetrachlorodibenzodioxins	71		
2,3,7,8-Tetrachlorodibenzofuran	71		
1,1,1,2-Tetrachloroethane	43–45		
1,1,2,2-Tetrachloroethane	43–45		
Tetrachloroethene	40, 42–46, 127		

Appendices

(e) = chiral (enantiomer) separation

X

Structure index



Acenaphthene	64	Bornane	88	2-Chloropropionic acid	
Acenaphthylene	64	Borneol	264	isobutyl ester	234
Acephate	91	endo-Brevicomine	254	2-Chloropropionic acid	
Acetic acid cyclohex-2-enyl ester	242	exo-Brevicomine	253	methyl ester	206
Acetic acid cyclopent-2-enyl ester	231	Bromacil	107	Chlorothalonil	52
Acetochlor	100	2-Bromobutane	210	Chlorpropham	100
Adipic acid	194	Bromocyclen	75	Chlorpyrifos	91
Alachlor	100	2-Bromopentane	219	Chlorthal	52
Alanine	202	2-Bromo-1-phenylethanol	238	Chlorthion	94
Aldrin	75	Bromophos-ethyl	94	Cholesterol	191
Alkyl bromides	210	2-Bromopropionic acid methyl ester	206	Chrysene	64
Alkyl chlorides	220	Bromopropylate	72	Cimaterol	188
Alodan	75	Butachlor	100	1,8-Cineole	157
Alprazolam	181	Butaperazine	182	Cinnamyl alcohol	248
Ametryn	102	1-Butin-3-ol	205	Citronellal, Citronellol	266
Amidithion	90	2-Butylamine	213	Clenbuterol	188
<i>o</i> -Aminoazotoluene	146	Butylate	86	Clomethiazole	182
Aminoisobutyric acid	202	2- <i>t</i> -Butyl-5-bromomethylene		Cocaine	184
5-Aminomethylbornene	243	1,3-dioxane	252	Codeine	181
Amitriptyline	190	1- <i>t</i> -Butyl-2-carboxymethyl aziridine	245	Coumaphos	91
Ammelide, Ammeline	179	4-Butyl-3-methylbutyrolactone	254	<i>p</i> -Cresidine	144
Amphetamine	250	Cadina-1,4-diene	274	Crotoxyphos	90
Androstanediol	192	Cadinene	274	Cruformate	94
Androstanolone	192	Caffeine	182	Cyanazine	102
Androsterone	191	Camphor	262	Cyanohydrins	217
Anethole	154	Cannabichromene	186	Cyanuric acid	179
Anhydroarabinitol	219	Cannabinol	186	Cyclic carbonates of 1,2-diols	205
Anhydrogalactitol	228	Captafol	115	Cyclic carbonates of 1-O-alkylglycerols	217
Anhydroglucitol	228	Captan	115	Cycloate	86
Anhydromannitol	228	Carbamates	86, 107	Cyclobarbital	187
Anilanzine	102	Carbaryl	86	<i>trans</i> -1,2-Cyclohexanediol	226
Anisaldehyde	154	Carbazole	49	Cyclopentanonecarboxylic acid methyl ester	232
<i>o</i> -Anisidine	144	Carbofuran	86	Cyclopent-2-enyl methyl carbonate	231
Anthracene	64	Carbophenothion	90	Cyfluthrin	99
Arabinitol	222	Carboxin	116	Cyhalothrin	99
Arabinose	228	2-Carene	154	Cypermethrin	99
Aramite	49	Δ 3-carene	260	Cyproconazole	97
Aspartic acid	202	Carvone	258	2,4-D	109
Atratone	102	Caryophyllene	186	Damascone	272
Atrazine	102	Ceramides	177	2,4-DB	109
Azinphos-methyl/-ethyl	90	Chlorbenseide sulphone	72	<i>o,p'</i> -DDD; <i>p,p'</i> -DDD	72
Barbital	187	Chlordane	75	<i>o,p'</i> -DDE; <i>p,p'</i> -DDE	72
Benz[a]anthracene	64	Chlorfenprop	109	<i>o,p'</i> -DDT; <i>p,p'</i> -DDT	72
Benzanthrone	134	Chlorfenson	72	δ -Decalactone	244
Benzidine	144	Chlorfenvinphos	90	γ -Decalactone	215
Benzo[b]fluoranthene	64	Chloridazon	116	Demephion-O/-S	91
Benzo[k]fluoranthene	64	Chlormephos	90	Demeton-O/-S	91
Benzo[ghi]perylene	64	Chlorobenzilate	72	Demeton-O/-S-methyl	91
Benzophenone	134	2-Chloroheptane	220	Desmetryn	102
Benzo[a]pyrene	64	2-Chlorohexane	220	Desoxyephedrine	258
3-Benzylbutyrolactone	257	Chloroneb	52	Dialifos	90
Bifenox	72	2-Chloropentane	220	2,4-Diaminoanisole	144
Bioallethrin	99	2-Chloro-1-phenylethanol	238	Diazinon	91
Bisabolol	158	<i>p</i> -Chlorophenylethylamine	241	Dibenz[ah]anthracene	64
Bitertanol	97				

Structure index

Dibenzofuran	71	Fatty acids	161, 162	Hexazinone	96
Dichlobenil	52	Fenamiphos	94	2-Hex-3-enyl methyl carbonate	245
Dichlofenthion	94	Fenarimol	72	Hexobarbital	270
Dichlofluanid	74	Fenbendazole	178	Hexyl-3-nitrate	229
Dichlorphos	90	Fenchlorphos	94	Hippuric acid	194
Dichlorprop	109	Fenchone	261	Hydratropic acid	258
Dichlorprop methyl ester	256	Fenfuram	100	3-Hydroxybutyraldehyde	207
Diclobutrazole	97	Fenitrothion	94	2-Hydroxybutyric acid	
Dicloran	52	Fenopropfen	271	methyl esters	209
Dicrotophos	90	Fenoprop	109	Ibuprofen	271
Dieldrin	76	Fenoprop methyl ester	256	Imazalil	97
Diethyltoluamide	100	Fenpropathrin	99	Indane	121
Dimefox	91	Fenpropimorph	101	1-Indanol	247
Dimethametryn	102	Fensulfthion	94	Indeno[1,2,3-cd]pyrene	64
Dimethoate	90	Fenthion	94	Iodofenphos	94
3,4-Dimethylbutyrolactone	224	Fenvalerate	99	α -Ionone	272
2,8-Dimethyl-1,7-dioxaspiro[5,5]		Flamprop	100	Iprodione	112
undecane	255	Fluazifop-butyl	109	Isazofos	91
3,5-Dimethyl-1,4-hexadiene	233	Fluchloralin	96	Isobenzan	77
Dimethyl malate	225	Flucythrinate	99	Isoborneol	264
3,3-Dimethyl-1-phenyl-		Fluometuron	97	Isobutyl lactate	236
butan-1-ol	270	Fluoranthene	64	Isocarbamid	100
Dimethyl tartrate	225	Fluorene	64	Isodrin	77
2,3-Dinor-6-oxoprostaglandin	192	Fluorenone	134	Isoleucine	202
2,3-Dinorthromboxane	192	Fluorochloridone	112	Isomethiozin	96
Dinoseb	52	Fluorodifen	72	Isophorone	49
Dioxane	123	Fluotrimazole	97	Isopropalin	96
1,7-Dioxaspiro[5,5]undecane	255	Fluoxetine	274	3-Isopropenyl-2-methylcyclo-	
Dioxathion	91	Flurbiprofen	271	pentanone	251
Diphenamid	100	Fluridone	116	Isopulegol	259
Diphenhydramine	182	Flusilazole	97	Isovaline	202
Disulfoton	90	Flutriafol	97	Isoxsuprine	188
δ -Dodecalactone	244	Fluvalinate	99	Kepone	77
γ -Dodecalactone	215	Folpet	115	Ketoprofen	271
2,4-DP	109	Fonofos	90	Lactic acid	194
Edifenphos	91	Frontalin	244	Lactide	223
Endosulfan	76	Fumaric acid	194	Lavandulol	265
Endrin	76	Furaneol	158	Lenacil	107
Ephedrine	250	1-(2-Furyl)ethanol	223	Leptophos	94
Epibromohydrin	203	Geraniol	156	Leucine	202
Epichlorohydrin	203	Glucose	237	Lidocain	184
EPN	94	Glutamic acid	202	Lidoflazine	182
2,3-Epoxy-3-methylbutanol-1	218	Glutaric acid	194	Lilial	156
2,3-Epoxy-4-octin-1-ol	243	Glyceric acid methyl ester	209	Limonene	261
EPTC	86	Glycidyl butyrate	234	Linalool	266
Erythrose	209	Glycine	202	Lysine	202
Estrone	191	Grandisol	264	Mabuterol	188
Ethalfuralin	96	HCB	52	Malathion	90
Ethion	91	Heptachlor	76	Malic acid dimethyl ester	225
Ethoprophos	91	γ -Heptalactone	215	Maltol	158
Ethoxyquin	116	Heptenophos	91	Mandelic acid methyl ester	248
2-Ethyl-7-methyl-1,6-dioxaspiro[4,5]decane	255	Heptyl-3-nitrate	229	Mannose	237
Etridiazole	116	Heroin	181	MCCP	109
Etrimfos	91	Hexachlorobenzene	52	MCPA	109
Eugenol	154	Hexaconazole	97	MCPB	109
Famphur	94	Hexahydroisobenzofuranone	242	Mecarbam	90
		γ -Hexalactone	215	Meconin	181

Structure index



Mecoprop	109	Myrcene	156	Phosalone	90
Mecoprop methyl ester	269	Naled	91	Phosmet	90
Melamine	179	Nalorphine	181	Phosphamidon	90
Menthofuran	259	Naphthalene	64	<i>cis</i> -Pinane	263
Menthol	259	Napropamide	100	α -Pinene	261
Menthone	259	Naproxen	271	Piperitone	259
Menthyl acetate	259	Neomenthol	259	Piperonal	154
Mesuximide	268	Nicotine	182	Pirimicarb	86
Metalaxyl	100	Nonachlor	75	Pirimiphos-ethyl/-methyl	91
Metamitron	96	δ -Nonalactone	244	Prochloraz	97
Metazachlor	100	γ -Nonalactone	215	Procymidone	112
Methadone	181	Nonyl-3-nitrate	229	Profenofos	94
Methaqualone	182	Norephedrine	250	Profluralin	96
Methfuroxam	100	Norfluoxetine	274	Progesterone	191
Methidathion	90	Norflurazon	116	Proline	202
Methionine	202	Noscapine	181	Prometon	102
Methomyl	107	Nuarimol	72	Prometryn	102
Methoprotryn	102	Octachlorodibenzodioxin	71	Propachlor	100
<i>p,p'</i> -Methoxychlor	72	δ -Octalactone	244	Propanil	100
2-Methylbutyric acid methyl ester	227	γ -Octalactone	215	Propazine	102
2-Methylbutyric acid	218	1-Octen-3-ol	246	Propham	100
α -Methylbutyrolactone	214	Octyl-3-nitrate	229	Propiconazole	97
2-Methylcyclohexanone	233	Ofurace	100	1- <i>i</i> -Propyl-2-carboxymethyl aziridine	235
3-Methylcyclohexene	210, 232	Ornithine	202	Propyzamide	100
Methyl dihydrojasmonate	154	Oxadiazon	112	Prothiocarb	86
2-Methyl-4,6-dinitroanisole	56	Oxamyl	107	Prothiofos	94
7-Methyl-1,6-dioxaspiro[4,5]-decane	255	Oxazole	142	Pulegone	259
4-Methyl-3-heptanone	246	Oxfendazole	178	Pyrene	64
3-Methyl-1,4-hexadiene	233	4,4'-Oxydianiline	146	Pyridaphenthion	91
3-Methyl-1-hexene	210	PAH	64	Quinalphos	91
2-Methyl-3-hydroxy-valeraldehyde	207	Pantolactone	224	Quinine	182
β -Methylionone	154	Papaverine	181	Quinomethionat	116
Methyl lactate	208	Paracyclophanes	275	Quintozene	52
5-(2-Methyl-propan-2-sulphonyl)-hept-3-ene	269	Parathion	94	Ractopamine	188
N-Methylpyrrolidone	142	PCB	68	Rose oxide	265
α -Methylquinovoside	236	Pebulate	86	Sabinene	260
α -Methylrhamnoside	236	Penconazole	97	Safrrole	154
2-Methyl-3-vinylcyclopentanone	241	Pendimethalin	96	Salbutamol	188
Metolachlor	100	Pentanochlor	100	Saturated fatty acids	161
Metribuzin	96	Pentobarbital	187	Sebuthylazine	102
Mevinphos	90	Permethrin	99	Secbumeton	102
Mexiletine	250	Phenacetin	49	Secobarbital	187
Midazolam	181	Phenanthrene	64	Simazin	102
Mirex	77	Phenakapton	90	Simetryn	102
Molinate	86	Phenobarbital	187	Solketal	227
Monocrotophos	90	Phenoxycarboxylic acids	109	Sorbitol	230
Morphine	181	Phensuximide	268	Spiroacetals	255
Musk ambrette	56	Phenylalanine	202	Succinic acid	194
Musk ketone	56	3-Phenylbutyrolactone	257	Sulfotep	91
Musk moskene	56	N-Phenyl-cyclohex-2-enamine	269	Sulprofos	94
Musk tibetene	56	Phenylcyclopropylmethanol	248	2,4,5-T	109
Musk xylene	56	α -Phenylethanol	239	Tartaric acid dimethyl ester	209, 225
Myclobutanil	97	α -Phenylethylamine	240	Tebuconazole	97
		Phenylpropanol isomers	249	Tebuthiuron	97
		2-Phenylpropionic acid	258	Tecnazene	52
		Pholedrine	250		
		Phorate	90		

Structure index

TEPP	91	Thiobencarb	86	Tricyclazole	116
Terbacil	107	4,4'-Thiodianiline	146	Trietazine	102
Terbufos	90	Thiometon	90	Triflumizole	97
Terbumeton	102	Thionazin	91	2,2,2-Trifluorophenylethanol	247
Terbutylazine	102	Thioquinox	116	Trifluralin	96
Terbutryn	102	Thiordiazine	182	3,3,5-Trimethylcyclohexanol	253
α -Terpinene	156	Thiothixene	182	2,4,6-Trinitrotoluene	56
α -Terpineol	263	Thymol	156	δ -Undecalactone	244
Testosterone	192	Tilidine	182	γ -Undecalactone	215
Tetrachlorvinphos	90	TNT	56	Undecyl-3-nitrate	229
Tetradifon	72	Tolclofos-methyl	94	Unsaturated fatty acids	162
Tetrahydrocannabinol	186	<i>o</i> -Toluidine	144	γ -Valerolactone	215
Tetrahydrofuran	123	Tolylfluanid	74	Valine	202
Tetrahydroisobenzofuranone	215	Toxaphenes	88	Vamidothion	91
Tetralin	121	2,4,5-TP	109	Vanillin	154
Tetramethrin	99	Tranlylcypromine	250	Vernolate	86
Tetrasul	72	Triadimefon	97	Vinclozolin	112
Theaspiran	273	Triadimenol	97	Vinyl-endo-brevicomine	251
THF	123	Triazines	96, 102	Vinyl-exo-brevicomine	252
Thiabendazole	116	Triazophos	91	Vinylguaiaicol	154
Thioacetic acid cyclopent-2-enyl ester	230	Trichlorfon	91	Whisky lactone	254
		Trichloronat	94		

Product use restriction

MACHEREY-NAGEL chromatography products are intended, developed, designed and sold FOR RESEARCH AND DEVELOPMENT PURPOSES AND ANALYTICAL QUALITY CONTROL / ROUTINE MEASUREMENTS ONLY, except, however, any other function of the product being EXPRESSLY set forth in original MACHEREY-NAGEL product leaflets.

MACHEREY-NAGEL products are intended for GENERAL LABORATORY USE ONLY!

MACHEREY-NAGEL products are suited for QUALIFIED PERSONNEL ONLY !

MACHEREY-NAGEL products shall in any event be used wearing adequate PROTECTIVE CLOTHING. For detailed information please refer to the respective Material Safety Data Sheet of the product !

MACHEREY-NAGEL products shall exclusively be used in an ADEQUATE TEST ENVIRONMENT.

MACHEREY-NAGEL does not assume any responsibility for damages due to improper application of our products in other fields of application.

APPLICATION ON THE HUMAN BODY IS STRICTLY FORBIDDEN. The respective user is liable for any and all damages resulting from such application.

The user has to ensure that the products used are suitable for the intended application.

MACHEREY-NAGEL does not warrant the reproducibility of published applications.

Index of application numbers



200030	196	201620	170	202531	238	210080	42
200041	196	201660	171	202532	234	210110	64
200050	197	201700	164	202542	256	210130	163
200060	199	201760	174	202552	256	210150	271
200071	199	201780	176	202562	269	210170	146
200080	148	201790	175	202572	258	210180	71
200090	148	201800	176	202582	273	210210	100
200100	149	201810	263	202611	207	210230	76
200150	46	201830	260	202621	203	210260	124
200210	119	201860	232	202631	243	210270	87
200230	119	201880	210	202681	218	210280	140
200260	120	201890	233	202691	218	210290	150
200290	55	201900	246	202712	225	210300	123
200300	54	201920	266	202731	234	210330	42
200320	48	201970	253	202732	209	210340	137
200340	49	202010	271	202751	205	210350	55
200350	49	202030	272	202752	209	210360	141
200370	59	202050	233	202762	208	210370	190
200410	60	202060	241	202772	208	210380	142
200530	72	202070	251	202781	247	210410	105
200610	81	202080	232	202802	236	210430	113
200700	75	202090	261	202811	264	210440	179
200740	106	202130	251	202831	226	210460	181
200770	102	202150	252	202832	223	210470	186
200820	104	202160	253	202841	226	210490	138
200860	107	202180	254	202842	214	210500	80
200910	101	202190	244	202851	212	210510	68
200920	115	202200	255	202861	205	210520	142
200930	115	202210	273	202862	254	210533	266
200940	94	202220	255	202871	227	210550	174
201060	108	202230	252	202881	204	210560	40
201170	133	202240	219	202891	211	210570	214
201180	133	202250	210	202892	224	210580	206
201200	135	202260	220	202901	211	210590	120
201240	177	202280	217	202911	221	210600	198
201250	134	202290	240	202912	224	210630	114
201270	187	202300	241	202921	222	210640	114
201290	187	202310	250	202922	224	210660	191
201300	183	202320	250	202931	213	210670	188
201310	191	202330	258	202932	257	210680	110
201330	157	202340	213	202941	238	210690	74
201340	156	202350	243	202942	257	210700	77
201350	153	202361	235	202951	217	210710	86
201360	154	202362	209	202961	207	210720	99
201370	157	202371	245	202962	242	210730	147
201390	122	202372	228	202971	227	210740	98
201410	127	202381	268	202982	215	210770	69
201420	128	202401	270	203001	230	210820	145
201430	130	202412	237	203011	222	210830	110
201460	130	202421	204	203021	219	210840	110
201480	159	202431	204	203031	228	210850	110
201490	160	202451	220	203041	236	210880	129
201500	160	202462	218	210010	185	210930	159
201520	139	202471	221	210020	184	210950	129
201530	139	202511	237	210030	200	210990	168
201560	143	202512	206	210040	200	211000	168
201570	163	202521	240	210060	167	211180	69

Index of application numbers

211260	151	211920	88	213010	272
211280	44	211950	192	213020	248
211290	47	211980	177	213030	92
211320	212	212000	189	213150	123
211330	212	212010	173	213160	136
211340	212	212040	178	213170	63
211350	216	212050	229	213180	96
211360	216	212071	70	213190	66
211370	225	212072	186	213200	58
211380	227	212140	147	213210	116
211390	216	212150	173	213220	83
211400	235	212250	269	213230	144
211410	238	212260	269	213240	62
211420	239	212270	230	213250	117
211430	239	212280	230	213260	223
211440	216	212290	242	213270	84
211450	244	212300	231	213280	109
211460	246	212310	245	213300	179
211470	246	212320	249	250020	140
211480	216	212330	231	250040	141
211490	244	212350	270	250050	138
211500	258	212400	239	250060	61
211510	261	212410	267	250070	48
211520	261	212420	267	250110	107
211530	260	212430	265	250120	165
211540	260	212440	266	250130	89
211550	261	212450	229	250170	274
211560	262	212460	247	250190	275
211570	264	212480	40	250200	45
211580	264	212520	126	250210	106
211590	265	212540	65	250220	78
211600	263	212570	41	250300	192
211610	264	212580	190	250350	79
211620	265	212590	194	250360	185
211630	263	212620	193	250410	259
211640	263	212630	131	250420	93
211660	216	212640	164	250430	82
211670	244	212680	248	250440	47
211690	266	212750	172	250450	58
211700	268	212761	202	250461	43
211710	268	212790	50	250470	166
211720	216	212800	65	250500	125
211730	244	212810	38	250510	85
211740	265	212820	112	250520	71
211750	216	212830	52	250530	155
211760	244	212840	63	250620	117
211770	272	212870	134		
211820	169	212880	203		
211851	56	212920	67		
211852	56	212930	67		
211853	56	212940	131		
211860	136	212950	132		
211871	121	212960	124		
211872	121	212970	249		
211873	121	212980	262		
211880	158	212990	262		
211900	184	213000	262		

MACHERY-NAGEL • Chromatography and More

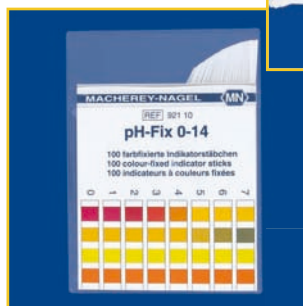
HPLC • SPE • TLC



Filtration



Rapid Tests



Bioanalysis



Water Analysis

